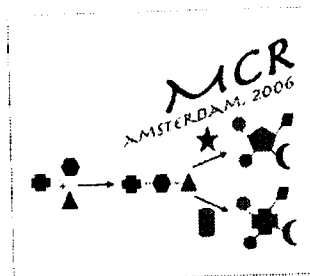
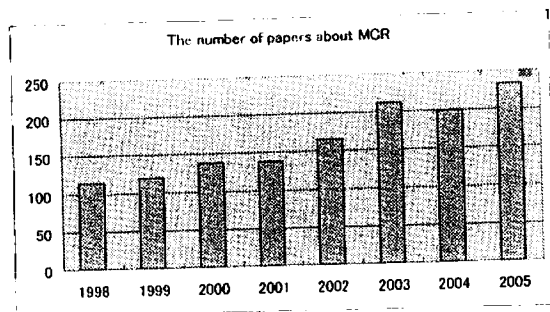


Asymmetric Multicomponent Reaction (AMCR) using Enantioselective Approach

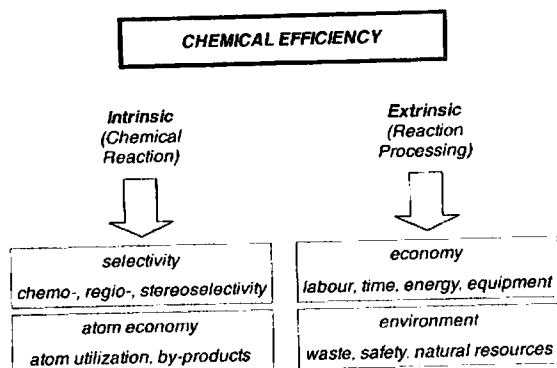
Chen ZhiHua

1. Introduction:



3rd International Conference
on Multi-Component
Reactions will be held on July
9 2006 in Amsterdam The
Netherlands.

Scheme 1 Key criteria for chemical efficiency



What is Asymmetric Multicomponent Reaction (AMCR)

It should be defined as the reaction between three or more either achiral reagent in a single vessel which have been added together (or nearly) to form stereoselectively a new chiral compound that contains portions of all the components, forming at least one new stereogenic element.

Approach of AMCR

1. Using diastereoselective Approach
2. Using Enantioselective Approach

Reviews: *Angew. Chem. Int. Ed.* **2005**, *44*, 1602.
Tetrahedron **2006**, *62*, 2143.
Chem. Eur. J. **2003**, *9*, 4286.
Chem. Rev. **1996**, *96*, 115.

Moreover, developing of MCR is fit for the needs of Combinatorial Library Synthesis.

Contents:

2. AMCR based on Isocyanide

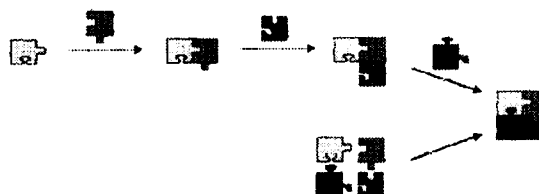
- 2.1 Passerini Reaction
- 2.2 Ugi Reaction

3. AMCR based on Tandem Conjugate (based on Enon and Enal)

- 3.1 Tandem Enantioselective Conjugate Addition
- 3.2 Knoevenagel/Diels-Alder Reaction and Wittig/Knoevenagel/Diels-Alder Reaction

4. AMCR based on Nucleophilic addition to Imine

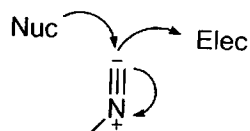
- 4.1 Strecker Reaction
- 4.2 Mannich Reaction
- 4.3 Biginelli Reaction
- 4.4 Organometallic 1,2-Addition Process



Multi-step (top) vs multicomponent (bottom) assembly of the same compound.

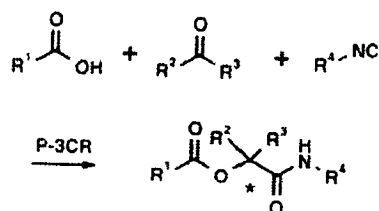
2. AMCR based on Isocyanide

The most important property of isocyanides is the reaction with nucleophiles and electrophiles at the isocyanide carbon atom.

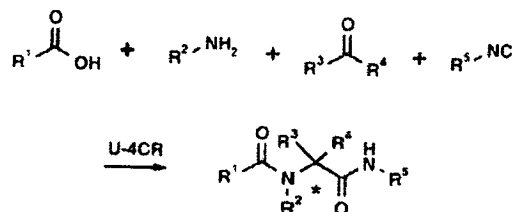


Reviews: *Chem. Rev.* **2006**, *106*, 17.
Angew. Chem. Int. Ed. **2000**, *39*, 3168.

Passerini Reaction

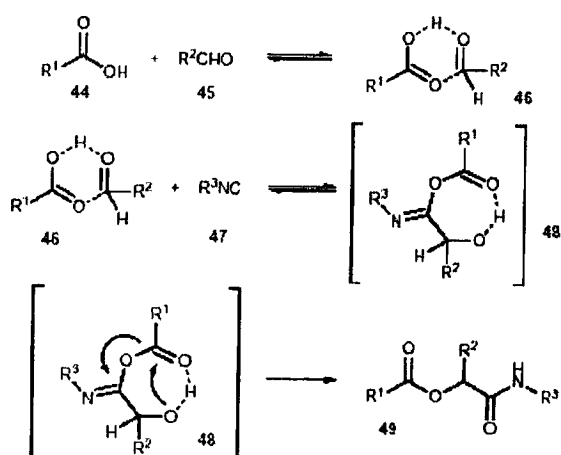


Ugi Reaction



2.1 Passerini Reaction: involving an oxo component, an isocyanide, and a nucleophile.

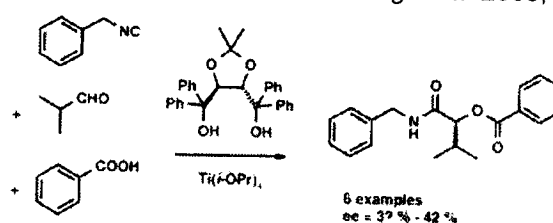
Scheme 2 Suggested mechanism of the P-3CR



Two example of catalytic asymmetric Passerini Reaction

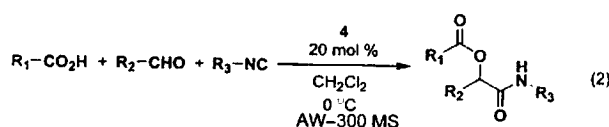
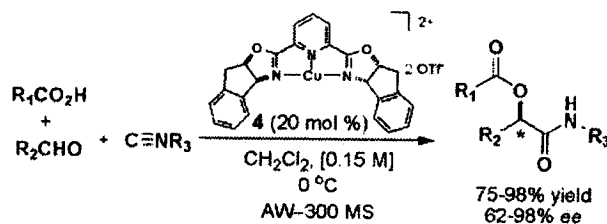
a. Domling Group: Lewis acid/Chiral ligand

Org. Lett. **2003**, *5*, 4021



b. Schreiber Group: Cu (II)-pybox complex

Org. Lett. **2004**, *6*, 4231



entry ^a	R ₁	R ₂	R ₃	product	% yield ^b	% ee ^c
1	PhCH ₂ (9)	2-furyl (10)	PhCH ₂ (11)	16	83	62 (R) ^d
2	PhCH ₂ (9)	BnOCH ₂ (6)	PhCH ₂ (11)	17	87	72
3	Ph (5)	2-thiophenecarboxyl (12)	<i>t</i> -butyl (13)	18	95	82 (R)
4	Ph (5)	BnOCH ₂ (6)	<i>n</i> -butyl (14)	19	87	88
5	Ph (5)	BnOCH ₂ (6)	<i>n</i> -pentyl (15)	20	83	89
6	Ph (5)	BnOCH ₂ (6)	PhCH ₂ (11)	21	89	93
7	Ph (5)	BnOCH ₂ (6)	<i>t</i> -butyl (13)	22	95	98
8	Ph (5)	2-furyl (10)	<i>p</i> -MeOPh (7)	23	98	91 (R)
9	Ph (5)	2-furyl (10)	PhCH ₂ (11)	24	90	75 (R)
10	Ph (5)	2-furyl (10)	<i>t</i> -butyl (13)	25	97	89 (R)
11	Ph (5)	2-furyl (10)	<i>n</i> -butyl (14)	26	82	78 (R)
12	Ph (5)	2-furyl (10)	<i>n</i> -pentyl (15)	27	82	78 (R)
13	Ph (5)	2-thiophenecarboxyl (12)	<i>p</i> -MeOPh (7)	28	95	89 (R)
14	Ph (5)	2-thiophenecarboxyl (12)	PhCH ₂ (11)	29	87	75 (R)
15	Ph (5)	2-thiophenecarboxyl (12)	<i>n</i> -butyl (14)	30	76	64 (R)
16	Ph (5)	2-thiophenecarboxyl (12)	<i>n</i> -pentyl (15)	31	75	60 (R)

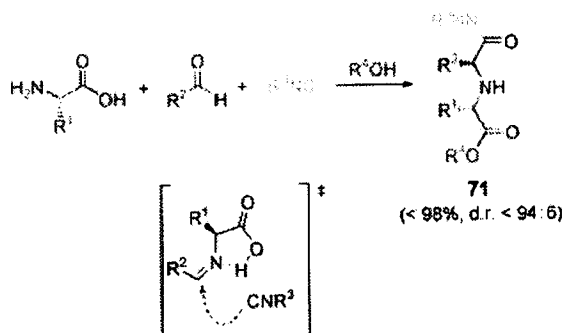
^a [0.15 M] final. ^b Isolated yield. ^c See the Supporting Information for HPLC conditions. ^d Inferred absolute stereochemistry based on X-ray crystal analysis with 5-bromo-2-furaldehyde as the substrate.

2.2 Ugi Reaction: involved a schiff base or an enamine with a nucleophile and an isocyanide

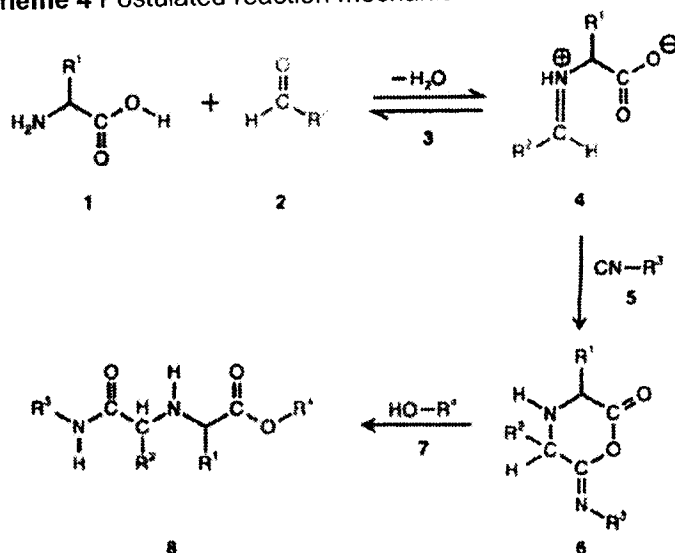
Angew. Chem. Int. Ed. **1996**, 35, 173.

- * high degree of diversity
- * two possible amide bonds

Scheme 3 Diastereoselective Ugi four-component reaction with a chiral α -amino acid



Scheme 4 Postulated reaction mechanism for the U-4CR

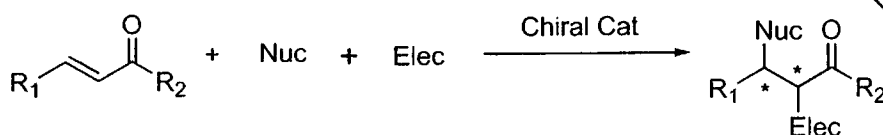


Why enantioselective approach is difficult:

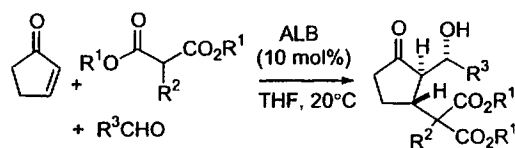
1. high reactivity
2. the amide bonds of product

3. AMCR based on Tandem Conjugate

3.1 Tandem Enantioselective Conjugate Addition

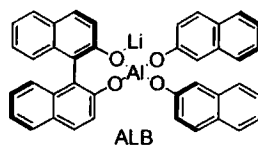


Scheme 5 The first catalytic asymmetric Michael aldol reaction



$R^1 = \text{Et}, R^2 = \text{Me}, R^3 = \text{Ph}(\text{CH}_2)_2$: 64% ee = 91%

$R^1 = \text{Et}, R^2 = \text{Me}, R^3 = \text{Ph}$: 82% ee = 89%

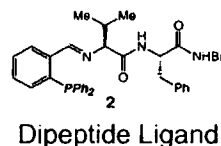
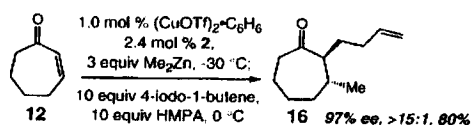


Angew. Chem. Int. Ed. **1996**, 35, 104
Angew. Chem. Int. Ed. **1997**, 36, 1236
J. Org. Chem. **1998**, 63, 3666

a. Hayashi Tamio Group: Rh catalyst

Hoveyda Group: Cu-dipeptide catalyst

Scheme 6 Cu-catalyzed conjugate addition to seven-membered cyclic enones



Dipeptide Ligand

J. Am. Chem. Soc. **2001**, 123, 755.

This type of catalyst works for the first step (1,4 addition) but has nothing to do with the second step.

For others this type examples

J. Am. Chem. Soc. **2001**, 123, 4358.

J. Am. Chem. Soc. **2001**, 123, 5841. 3/13

Both 1,4 addition and the aldol reactions are catalyzed by a rhodium complex.

Table 1 Tandem 1,4-Addition-Aldol Reaction

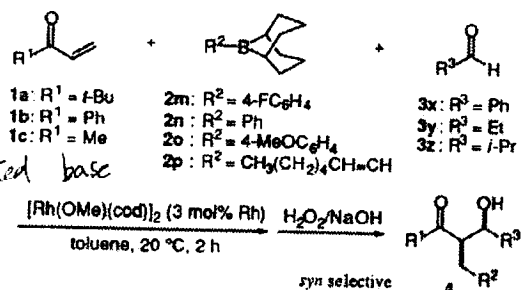
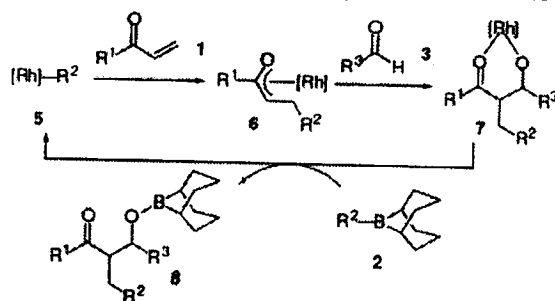


Table 1. Rhodium-Catalyzed Tandem Conjugate Addition-Aldol Reaction of Enone 1, *B-R*-9BBN 2, and Aldehyde 3^a

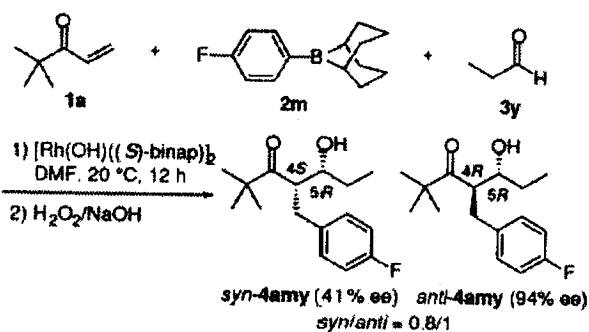
entry	enone 1	<i>B-R</i> -9BBN 2	aldehyde 3	product 4	yield (%) ^b	syn/anti ^c
1	1a	2m	3x	4amx	96	9.6:1
2	1a	2n	3x	4anx	97	10.7:1
3	1a	2o	3x	4aox	99	8.9:1
4 ^d	1a	2p	3x	4apx	85	21.4:1
5	1a	2m	3y	4amy	72	12.4:1
6	1b	2m	3x	4bm _x	88	9.0:1
7	1b	2m	3z	4bn _z	93	9.0:1
8	1c	2m	3x	4cm _x	99	5.7:1

Scheme 7

Mechanism: through (Oxa-π-allyl) rhodium intermediate



Scheme 8 Enantioselective 1,4-Addition-Aldol Reaction

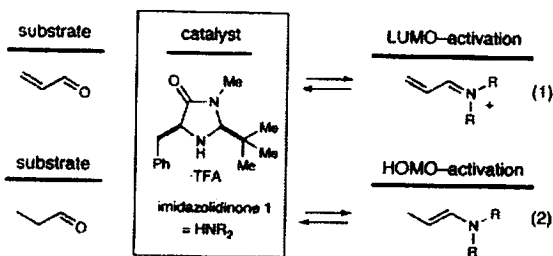


b. MacMillan Group: Cascade catalyst

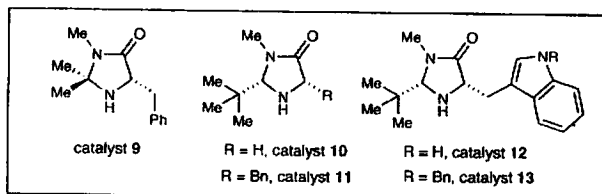
Scheme 9

J. Am. Chem. Soc. 2005, 127, 15051.

Imidazolidinones: Organocatalysts for HOMO or LUMO Activation



Scheme 11 Catalysts



Scheme 10 Concept

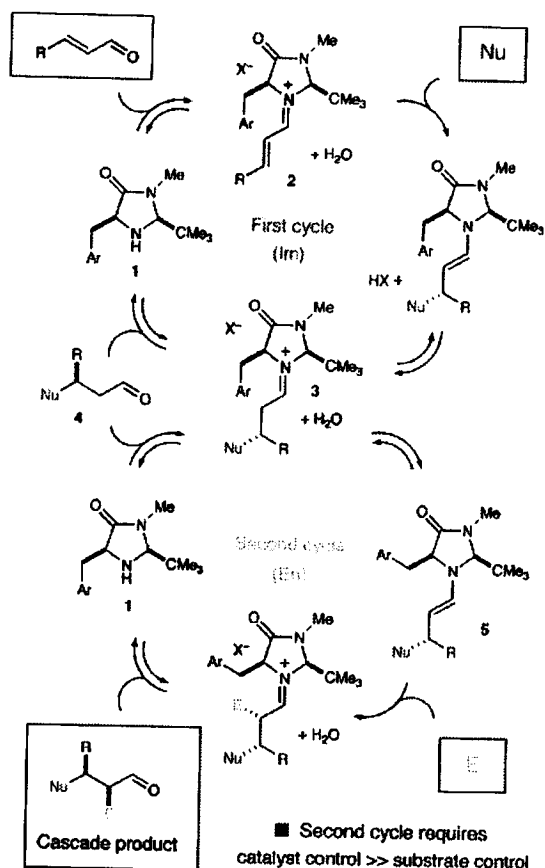
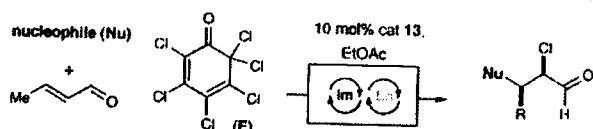
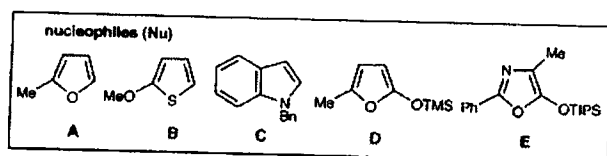


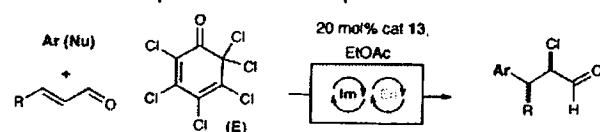
Table 2 Scope of Representative Nucleophiles



entry	nucleophile	product	temp (°C)	% yield	dr ^a	% ee
1	A		-50	86	14:1	99
2	B		-50	77	11:1	99
3 ^b	D		-55	71	>25:1	>99
4	C		-60	75	12:1	>99
5	E		-40	97	9:1	>99

^a Absolute and relative configuration assigned by chemical correlation.
^b Superior yields were obtained when the electrophile was added after consumption of the silyloxy furan.

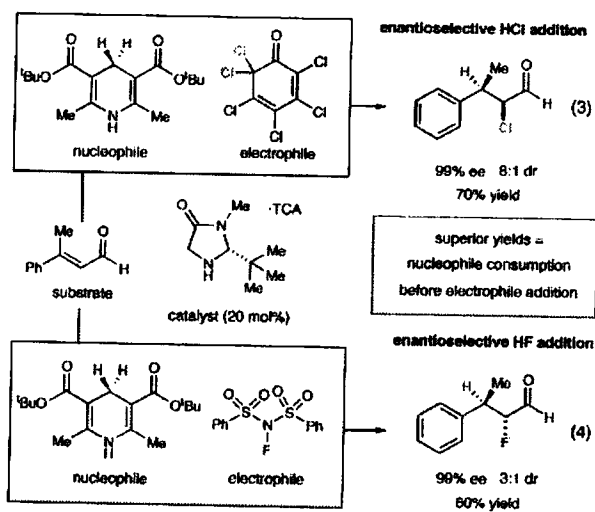
Table 3 Scope of Enal Component



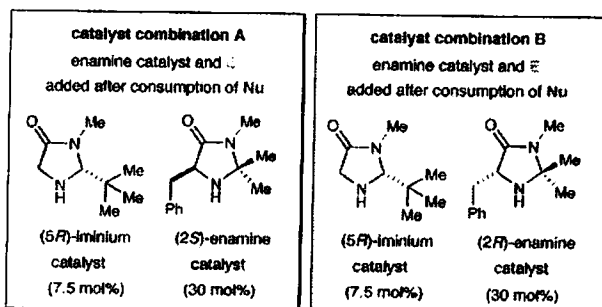
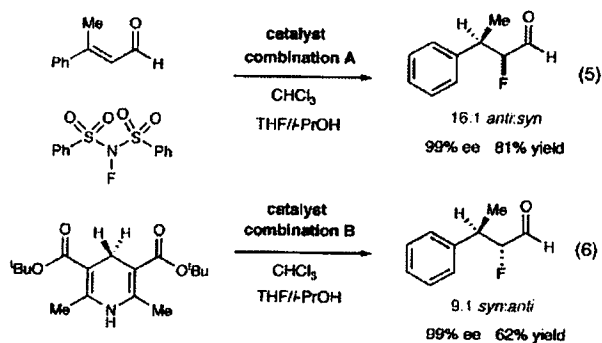
entry	R =	product	temp (°C)	% yield	dr ^a	% ee ^b
1	Me		-50	86	14:1	99
2	Pr		-50	74	13:1	99
3	CO ₂ Et		-60	80	22:1	99
4	CH ₂ OAc		-40	82	11:1	>99
5	Ph		-40	83	9:1	99
6	<i>i</i> -Pr		-40	67	12:1	>99

^a Absolute and relative configuration assigned by chemical correlation.
^b Enantiomeric excess determined by chiral GLC analysis.

Scheme 12 Scope of Electrophiles



Scheme 13 Diastereoselectivity control



Similar studies:

List Group: *J. Am. Chem. Soc.* **2005**, *127*, 15036.
 Jorgensen: *J. Am. Chem. Soc.* **2005**, *127*, 15710.

2.2 Knoevenagel/Diels-Alder Reaction and Wittig/Knoevenagel/Diels-Alder Reaction

Barbas Group: Amino acid Catalyst

Angew. Chem. Int. Ed. 2003, 42, 4233

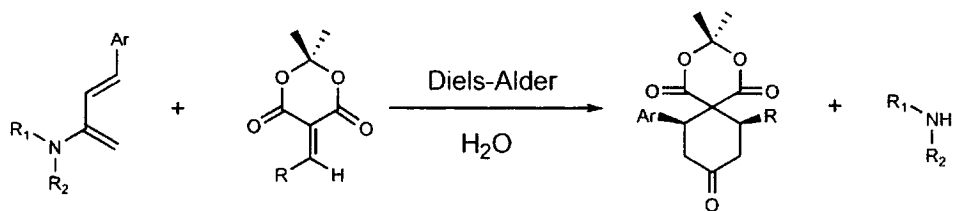
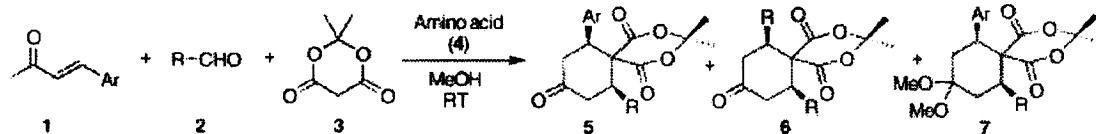


Table 4 Amino acid catalyzed asymmetric three component knoevenagel/Diels- Alder Reactions

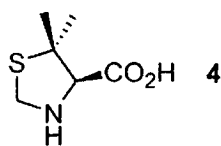


1a: Ar = Phenyl
1b: Ar = Piperonyl
1c: Ar = 1-Naphthalenyl
1d: Ar = 2-Furanyl
1e: Ar = 2-Thiophenyl

2a: R = 4-NO₂C₆H₄
2b: R = 4-CNC₆H₄
2c: R = C₆H₅
2d: R = 3,4-(OCH₂O)₂C₆H₃

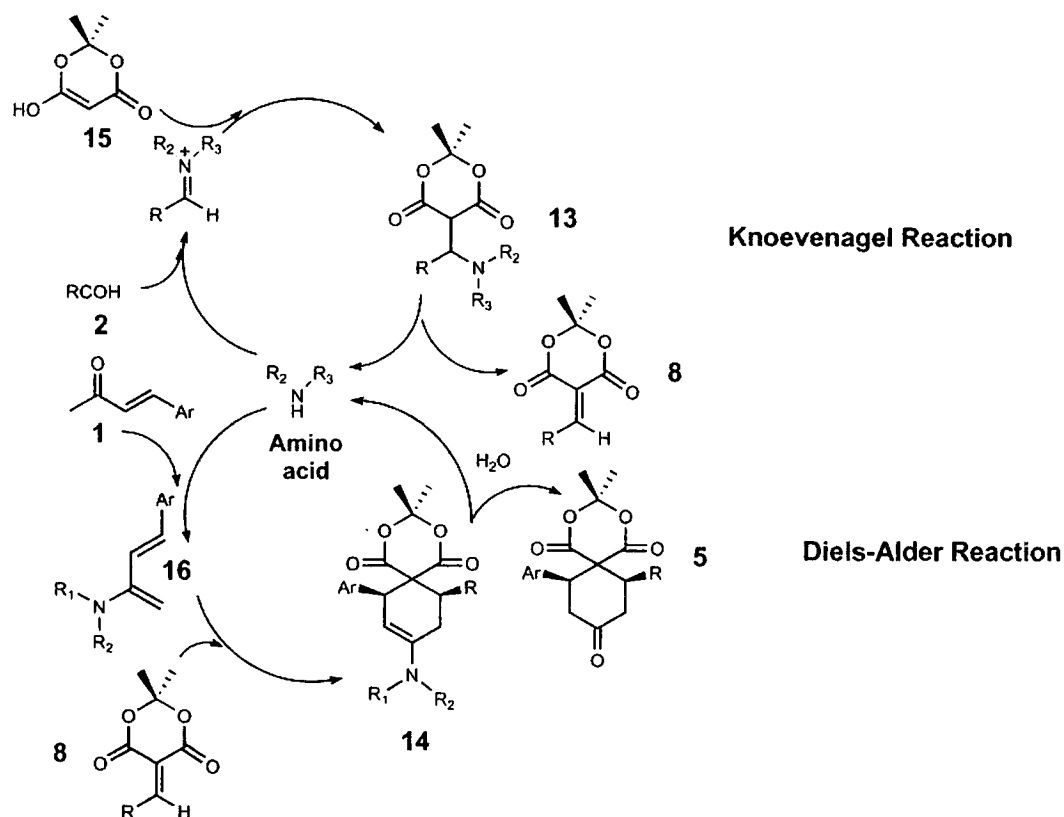
Entry	Enone	Aldehyde	t [h]	Products	Yield [%] ^[a]	Ratio 5:7	ee for 5 [%] ^[c]
1 ^[d]	1a	2c	96	5ac, 7ac	85	16:1	prochiral
2 ^[d]	1b	2d	96	5bd	99	>100:1	prochiral
3	1a	2a	72	5aa, 7aa	95	13:1	86
4	1a	2b	96	5ab, 7ab	85	16:1	84
5	1c	2a	72	5ca	93	>100:1	99
6	1d	2a	72	5da, 7da	92	12:1	88
7	1e	2a	72	5ea, 7ea	80	15:1	99

[a] See Supporting Information. [b] Yield of the combined isolated products. [c] Enantiomeric excesses were determined using chiral-phase HPLC. [d] Reaction catalyzed by L-proline (0.1 mmol).



5,5-dimethyl thiazolidinium-4-carboxylate (DMTC)

Scheme 14 Proposed reaction mechanism for the formation of the product.



Scheme 15 Proposed reaction mechanism for the formation of the unexpected product.

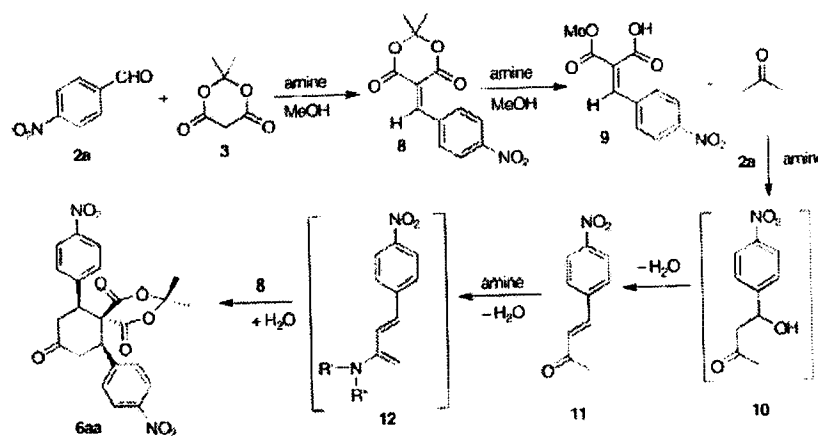
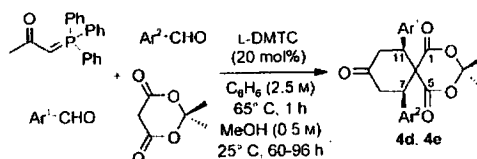


Table 5 Amino acid catalyzed asymmetric four component Wittig/Knoevenagel/Diels- Alder Reactions



Chem. Eur. J. **2004**, *10*, 5323.

Entry	Ar ¹	Ar ²	Yield[%]	d.r.	er ^[b]	Absolute ^[c] configuration
1 ^[d]	C ₆ H ₅	4-NO ₂ C ₆ H ₄	83	> 100:1	69	(7 <i>R</i> ,11 <i>S</i>)- 4d
2 ^[d]	4-NO ₂ C ₆ H ₄	C ₆ H ₅	80	> 100:1	42	(7 <i>S</i> ,11 <i>R</i>)- 4d
3 ^[d]	C ₆ H ₅	4-CNC ₆ H ₄	85	> 100:1	70	(7 <i>R</i> ,11 <i>S</i>)- 4e
4 ^[d]	4-CNC ₆ H ₄	C ₆ H ₅	80	> 100:1	42	(7 <i>S</i> ,11 <i>R</i>)- 4e

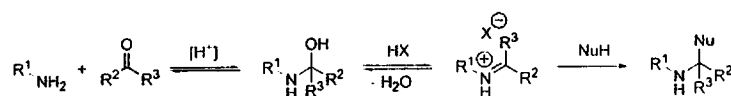
[a] Experimental conditions: aldehyde, Ar¹-CHO (0.5 mmol), and 1-(triphenylphosphanyl)propan-2-one (0.5 mmol) in benzene (0.2 mL) were stirred at 65°C for 1 h, then L-DMTC (0.1 mmol), aldehyde, Ar²-CHO (0.5 mmol), Meldrum's acid (0.5 mmol), and methanol (1.0 mL) were added and stirred at 25°C (see Experimental Section). [b] Enantiomeric excesses were determined using chiral-phase HPLC. [c] Absolute configuration determined based on HPLC analysis and comparison to earlier reports. [d] Spirotriones **4a** and **4b** are formed in 2:1 ratio with 10% yield. [e] Spirotrione **4f** is formed in 10% yield.

Our Group's paper about preparation α,β -Unsaturated N-Acylpyrroles in situ. *J. Am. Chem. Soc.* **2004**, *126*, 7559

4. AMCR based on Nucleophilic addition to Imine.

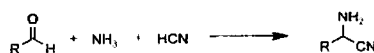
Introduction:

Scheme 16 Iminium intermediate

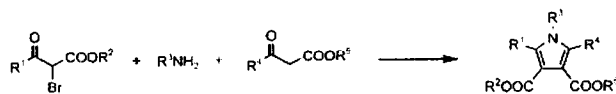


Scheme 17 Historical significant multicomponent reaction based on the α aminoalkylation of carbonyl compounds

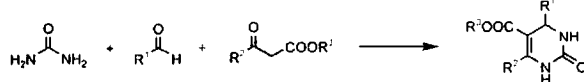
Strecker (1850)



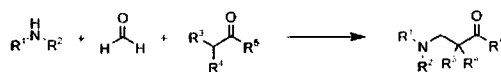
Hantzsch (1890)



Biginelli (1891)

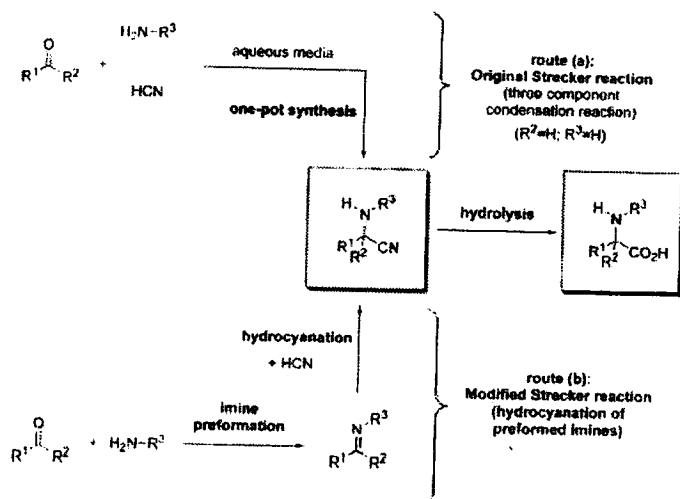


Mannich (1912)



4.1 Strecker Reaction

Scheme 18 Principles of Different Type of Strecker Reactions



Developments in Catalytic Asymmetric Strecker-Type Reaction:

Reviews: *Chem. Rev.* **2003**, *103*, 2795.
Angew. Chem. Int. Ed. **2001**, *40*, 85.

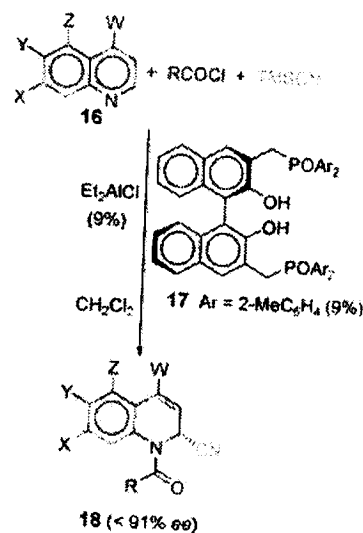
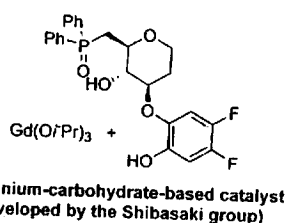
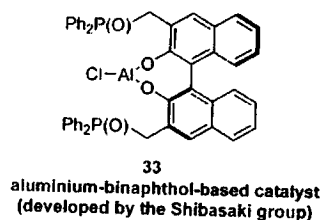
Lipton Group: diketopiperzine-based organocatalyst
 Jacobsen Group: Al(III) - salen based catalyst
 Hoveyda Group: Titanium-peptide-based catalyst
 Jacobsen Group: Thiourea / Urea -type organocatalyst
 and so on...

Scheme 20 Enantioselective Reissert-Henze MCR (a variant of the Strecker reaction)

J. Am. Chem. Soc. **2000**, *122*, 6327
J. Am. Chem. Soc. **2001**, *123*, 6801
J. Am. Chem. Soc. **2004**, *126*, 11808

Our lab's work: Bifunctional Lewis acid - Lewis base catalyst

Scheme 19



J. Am. Chem. Soc. **1999**, *121*, 2641
Angew. Chem. Int. Ed. **2000**, *39*, 1650.
Chem. Pharm. Bull. **2000**, *48*, 1586

J. Am. Chem. Soc. **2003**, *125*, 5634

Angew. Chem. Int. Ed. **1998**, *37*, 3186.
J. Am. Chem. Soc. **2000**, *122*, 762.

Catalytic Asymmetric Strecker Reaction: (Kobayashi Group)

Scheme 21 Preparation of Chiral Zirconium Catalyst

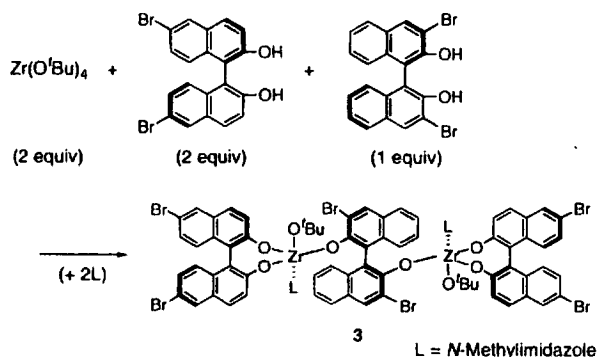
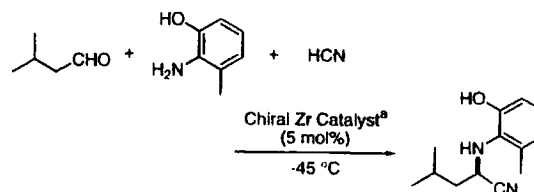


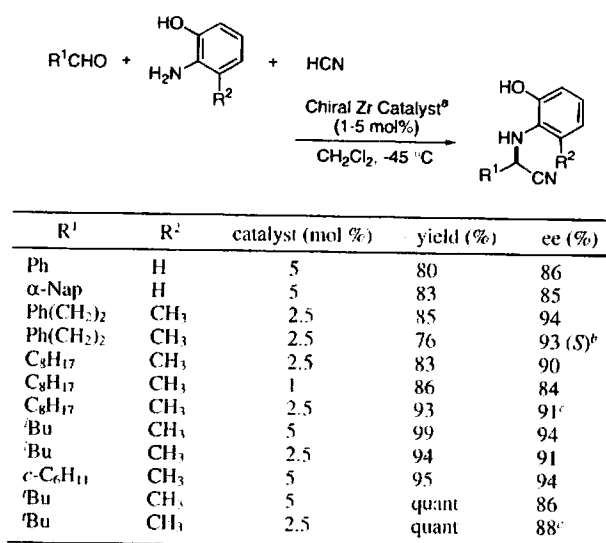
Table 6 Effect of Solvents and Concentration



entry	solvent	additive	concn (M)	yield (%)	ee (%)
1 ^b	toluene	MS 4A	0.04	63	65
2 ^c	toluene	MS 4A	0.04	49	79
3 ^c	CH ₂ Cl ₂	MS 4A	0.04	63	85
4 ^c	CH ₂ Cl ₂	none	0.01	99	94

^a See text. ^b Aldehyde, amine, and HCN were added to catalyst 3. ^c HCN was added to catalyst 3 first and this catalyst solution was added to the mixture of aldehyde and amine.

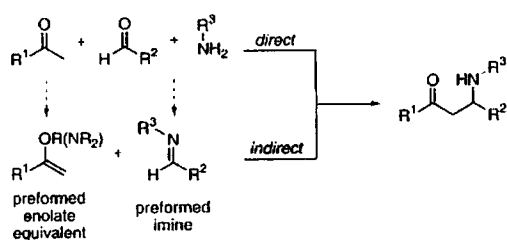
Table 7 Catalytic Asymmetric Strecker Reactions Using HCN



^a See text. ^b (*S*)-3-Br-BINOL and (*S*)-6-Br-BINOL were used. ^c Zr(OⁱPr)₄ was used instead of Zr(O^tBu)₄.

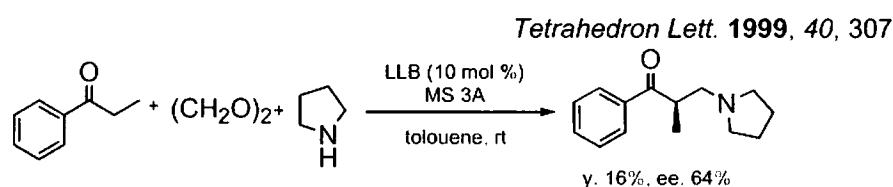
4.2 Mannich Reaction (a classic method for the preparation of β-amino carbonyl compounds)

Scheme 22 Direct and Indirect



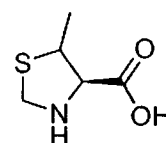
Review: *Acc. Chem. Res.* **2004**, *37*, 102

Our group: The first report of a direct catalytic asymmetric Mannich

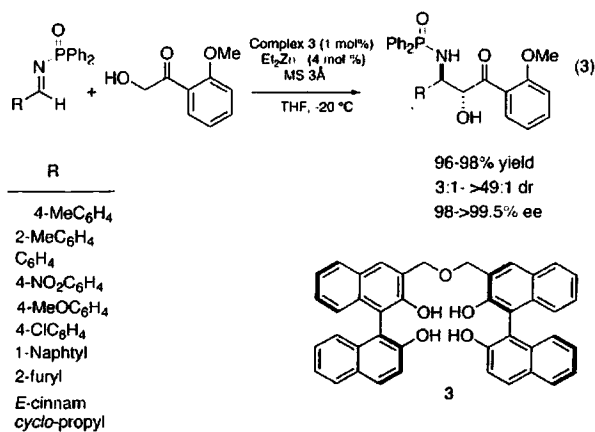


Developments in Direct Catalytic Asymmetric Mannich-Type Reaction:

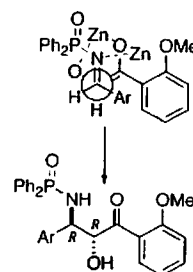
- Trost group: Dinuclear zinc catalyst
- Barbas group: 5,5-dimethylthiazolidine-4-carboxylic acid
- Jorgensen group: Chiral copper (II) bisoxazoline (BOX) catalyst
- Our group: Et₂Zn/linked-BINOL complex



Scheme 23 Mannich-type reaction using Et₂Zn/linked-BINOL complex



J. Am. Chem. Soc. **2003**, *125*, 2582

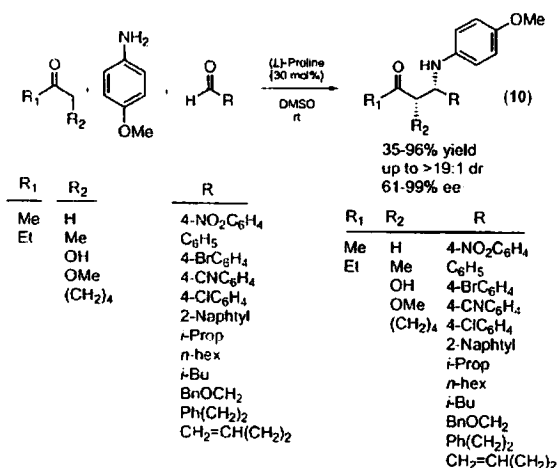


The Direct Catalytic Asymmetric Three-Component Mannich Reaction

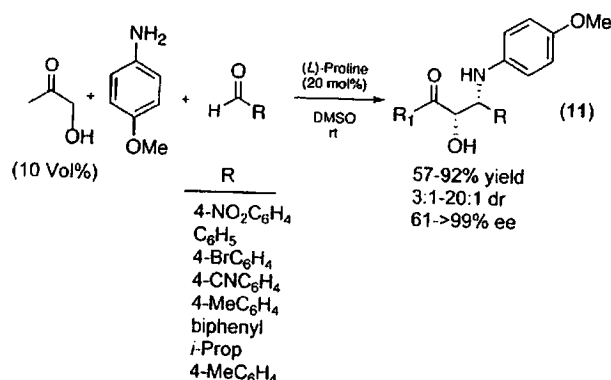
a. List Group

Acc. Chem. Res. **2004**, *37*, 548
J. Am. Chem. Soc. **2000**, *122*, 9336.
J. Am. Chem. Soc. **2002**, *124*, 827.

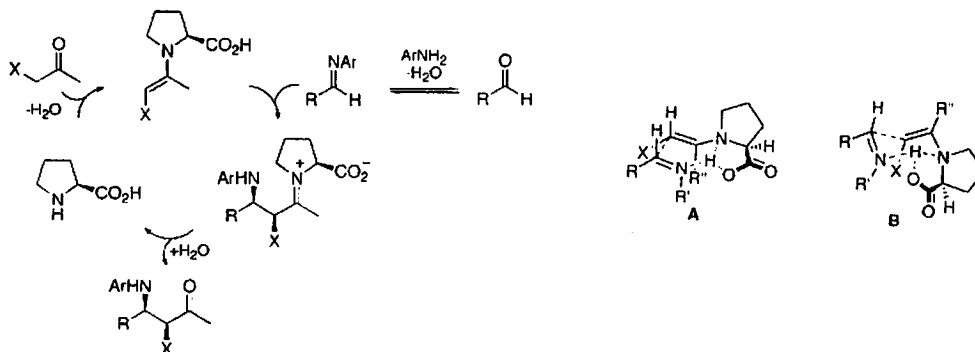
Scheme 24 The proline-catalyzed Mannich one-pot three-component reaction.



Scheme 25 Catalytic enantioselective synthesis of syn-1,2-amino alcohols



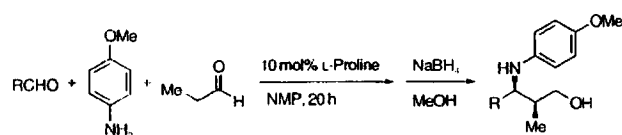
Scheme 26 Proposed Mechanism



b. Hayashi Yujiro Group:

Angew. Chem. Int. Ed. **2003**, *42*, 3677.

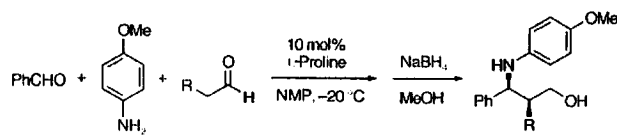
Table 8 Three-component Mannich reaction with various acceptor aldehydes



Entry	Aldehyde	T [°C]	Yield [%] ^[b]	syn:anti	ee [%] ^[c]
1	benzaldehyde	23	< 10	n.d. ^[d]	n.d.
2	benzaldehyde	0	15	> 95:5	n.d.
3	benzaldehyde	-10	50	> 95:5	90
4	benzaldehyde	-20	90	> 95:5	98
5	benzaldehyde ^[e]	-20	71	> 95:5	96
6	<i>p</i> -nitrobenzaldehyde	23	< 10	n.d.	n.d.
7	<i>p</i> -nitrobenzaldehyde	0	89	> 95:5	95
8	<i>p</i> -nitrobenzaldehyde	-10	93	> 95:5	99
9	<i>p</i> -bromobenzaldehyde	-10	85	> 95:5	95
10	<i>p</i> -bromobenzaldehyde	-20	33	> 95:5	98
11	<i>p</i> -chlorobenzaldehyde	-20	91	> 95:5	98
12	2-naphthaldehyde	-20	59	> 95:5	96
13	<i>p</i> -tolualdehyde ^[f]	-20	95	> 95:5	86
14	furfural	-20	87	> 95:5	84
15	<i>p</i> -pyridinecarbaldehyde	-20	84 ^[f]	> 95:5	> 99

[a] Reaction conditions: aldehyde:4-methoxyaniline:propanal:proline = 1.0:1.1:3.0:0.1. NMP was used as the solvent except in entries 5 and 13. Reaction time: 20 h. [b] Yield of the isolated amino alcohol. [c] The ee values were determined by chiral HPLC analysis (CHIRAPAK AD-H or AS-H). [d] Not determined. [e] DMF was used as the solvent. [f] Yield of the corresponding amino *tert*-butyldimethylsilyl ether, see text.

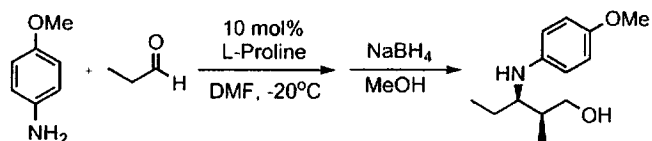
Table 9 Three-component Mannich reaction with various donor aldehydes



Entry	R	Yield [%] ^[b]	syn:anti	ee [%] ^[c]
1	Me	90	> 95:5	98
2	Et	85	> 95:5	97
3	<i>n</i> Pr	55	> 95:5	71

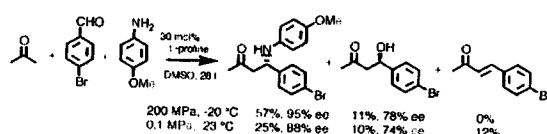
[a] Reaction conditions: benzaldehyde:4-methoxyaniline:aldehyde:proline = 1.0:1.1:3.0:0.1 [b] Yield of the isolated amino alcohol. [c] ee values were determined by chiral HPLC analysis (CHIRAPAK AD-H).

Scheme 27 The self-Mannich reaction of propanal



70%, 95:5, 96% ee 10/13

Scheme 28 Application of High pressure Induced by water-freezing to the Three-component Mannich Reaction



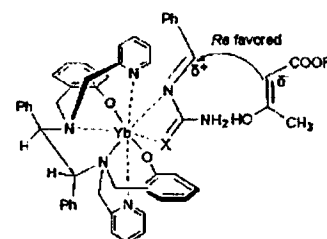
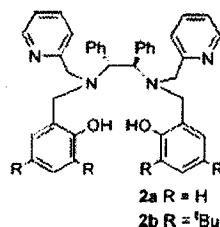
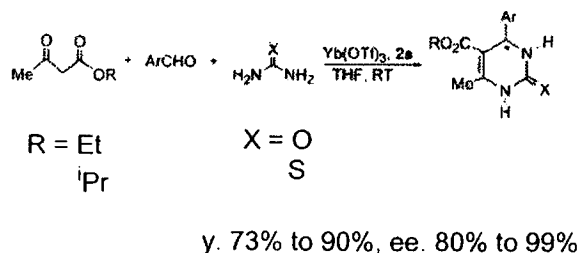
J. Am. Chem. Soc. **2003**, *125*, 11208.

4.3 Biginelli Reaction

(Asymmetric synthesis of Dihydropyrimidines)

Scheme 29 Enantioselective Three-component Biginelli Reaction catalyzed by Yb-hexadentate complex

J. Am. Chem. Soc. **2005**, *127*, 16386.



For further information, ref to Mr Noda's report

4.4 Organometallic 1,2-Addition Processes (Preparation of Progargylamines)

Aliphatic imine is difficult to be isolated, so one of solution is to generate it in situ.
Very recently, the preparation of the Dpp imine and Boc imine was reported

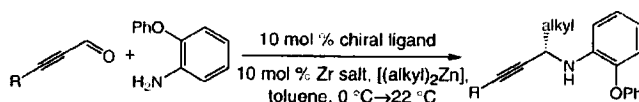
Tetrahedron Lett. **2006**, *47*, 3985
J. Am. Chem. Soc. **2006**, *128*, 6048.

Three-component enantioselective synthesis of Progargylamines

a. Hoveyda Group

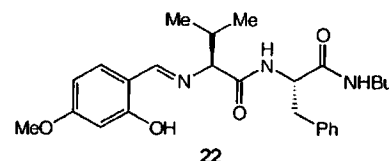
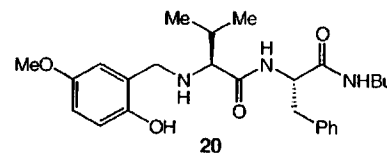
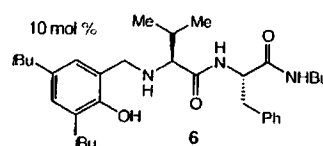
Angew. Chem. Int. Ed. **2003**, *42*, 4244

Table 10 Zr-dipeptide complex catalyzed enantioselective alkylation of alkynylamines.



Entry	R	[(alkyl) ₂ Zn]	Zr salt	Product	Chiral ligand	Yield [%]	ee [%]
1	nPent	3 [Me ₂ Zn]	[Zr(O ^t Bu) ₄]	15	6	87	82
2	nPent	3 [(Me ₂ CH(CH ₂) ₃) ₂ Zn]	[Zr(O ^t Bu) ₄]	16	6	60	98
3		11 [Me ₂ Zn]	[Zr(O ^t Bu) ₄]	17	6	75	92
4	TBSO	11 [Et ₂ Zn]	[Zr(O ⁱ Pr) ₄] HO ⁱ Pr	16	6	70	> 98
5	TMSO	12 [Et ₂ Zn]	[Zr(O ⁱ Pr) ₄] HO ⁱ Pr	19	20	86	84
6		13 [Et ₂ Zn]	[Zr(O ⁱ Pr) ₄] HO ⁱ Pr	21	22	89	86
7	Ph	2 [Me ₂ Zn]	[Zr(O ^t Bu) ₄]	23	6	84	80
8	Ph	2 [Et ₂ Zn]	[Zr(O ⁱ Pr) ₄] HO ⁱ Pr	24	6	85	> 98
9	Ph	2 [(Me ₂ CH(CH ₂) ₃) ₂ Zn]	[Zr(O ^t Bu) ₄]	25	6	81	91
10		14 [Me ₂ Zn]	[Zr(O ^t Bu) ₄]	26	6	75	93

Dipeptide Ligand 6, 20, 22



[a] Conditions: alkyl zinc reagent (6 equiv), 4 h for reactions with [Et₂Zn], 24 h for other alkyl zinc reagents. [b] Yields of product isolated after silica-gel chromatography; > 98% conversion in all cases. [c] Enantioselectivities determined by HPLC (chiralcel OD).

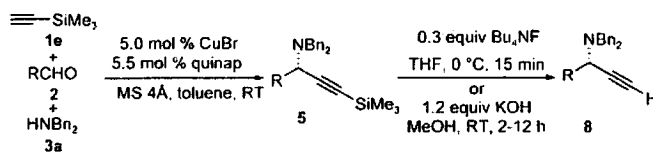
b. Knochel Group: Copper(I)-Quinap complex

Table 11 Enantioselective three-component one-pot synthesis of silylated propargylamine 5 and disilylation to terminal propargylamines 8

Angew. Chem. Int. Ed. 2002, 41, 2535

Angew. Chem. Int. Ed. 2003, 42, 5763

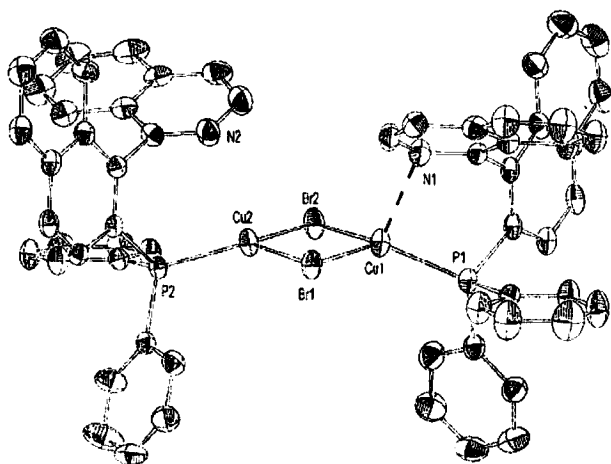
Chem. Eur. J. 2006, 12, 4380



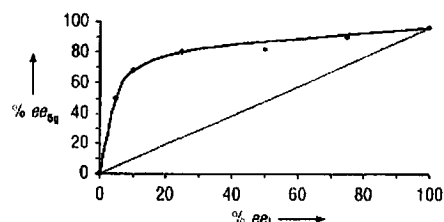
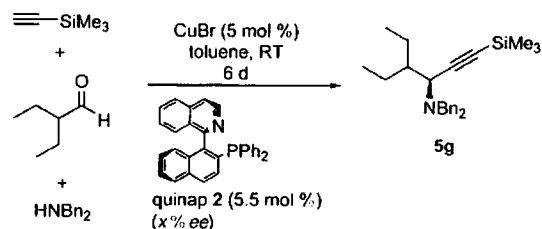
Entry	Aldehyde 2	Propargylamine 5	Yield [%] ^[a]	ee [%] ^[b]	Propargylamine 8	Yield [%] ^[a]
1	2m: <i>n</i> PrCHO	5c: R = <i>n</i> Pr	90	90 ^[c]	8a: R = <i>n</i> Pr	98 ^[d]
2	2a: <i>n</i> BuCHO	5d: R = <i>n</i> Bu	82	90 ^[c]	8b: R = <i>n</i> Bu	92 ^[d]
3	2n: <i>n</i> PentCHO	5e: R = <i>n</i> Pent	99	88 ^[d]	8c: R = <i>n</i> Pent	96 ^[d]
4	2b: <i>i</i> BuCHO	5f: R = <i>i</i> Bu	85	94 ^[d]	8d: R = <i>i</i> Bu	99 ^[d]
5	2o: <i>neo</i> -PentCHO	5g: R = <i>neo</i> -Pent	94	94 ^[d]	8e: R = <i>neo</i> -Pent	99 ^[d]
6	2c: <i>i</i> PrCHO	5h: R = <i>i</i> Pr	87	96 ^[c]	8f: R = <i>i</i> Pr	96 ^[d]
7	2p: <i>s</i> -PentCHO	5a: R = <i>s</i> -Pent	95	98 ^[c]	8g: R = <i>s</i> -Pent	98 ^[d]
8	2q: <i>c</i> -PrCHO	5i: R = <i>c</i> -Pr	98	92 ^[c]	8h: R = <i>c</i> -Pr	99 ^[d]
9	2r: <i>c</i> -PentCHO	5j: R = <i>c</i> -Pent	98	96	8i: R = <i>c</i> -Pent	99 ^[d]
10	2s: <i>c</i> -HexCHO	5b: R = <i>c</i> -Hex	86	97 ^[c]	8j: R = <i>c</i> -Hex	93 ^[d]
11	2t: PhCH=CH-CHO	5k: R = H	96	82 ^[c]	8k: R = H	97 ^[d]
12	2u: Ph2C=CH-CHO	5l: R = Ph	82	84 ^[c]	8l: R = Ph	97 ^[d]
13	2v: Ph(CH2)2CHO	5m: R = H	78	88 ^[c]	8m: R = H	98 ^[d]
14	2w: 4-Br-C6H4-(CH2)2CHO	5n: R = Br	73	88 ^[c]	8n: R = Br	98 ^[d]
15	2x: 4-CO2Et-C6H4-(CH2)2CHO	5o: R = CO2Et	96	87	8o: R = CO2Et	90 ^[d]
16	2d: PhCHO	5p: Ar = Ph	98	68		
17	2y: 2-naphthyl-CHO	5q: Ar = 2-naphthyl	69	54		
18	2z: 2-thiophen-CHO	5r: Ar = 2-thienyl	81	80 ^[c]	8p: Ar = 2-thienyl	99 ^[d]
19	2aa: 3-thiophen-CHO	5s: Ar = 3-thienyl	85	74 ^[c]	8q: Ar = 3-thienyl	99 ^[d]
20	2ab: 2-benzothiophenyl-CHO	5t: Ar = 2-benzothieryl	42	89		
21	2i: 3-benzothiophenyl-CHO	5u: Ar = 3-benzothieryl	92	82	8r: Ar = 3-benzothieryl	93 ^[d]

[a] Isolated yield of analytically pure product. [b] Enantiomeric excess determined by HPLC analysis using Chiracel OD-H column (*n*-heptane/*i*-PrOH). [c] The *ee* was determined after desilylation. [d] The *ee* was determined after acylation with PhCOCl. [e] Desilylation was carried out with Bu₄NF. [f] Desilylation was carried out with KOH.

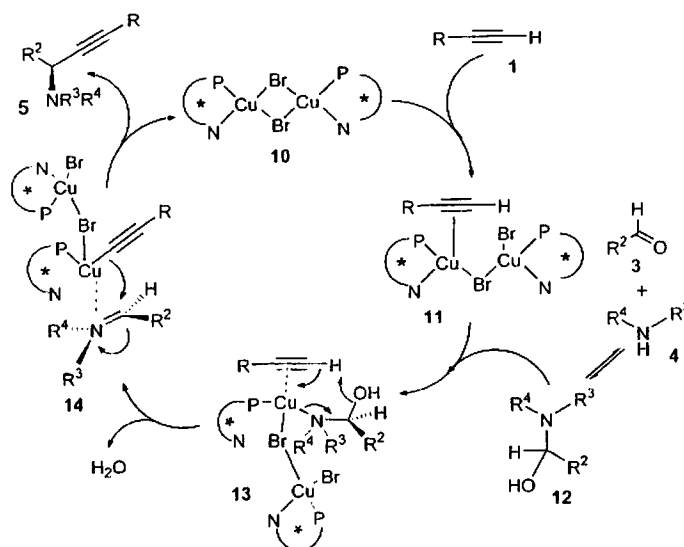
Scheme 30 Structure of the complex [BrCu(Quinap)]₂



Scheme 31 Nonlinear effects



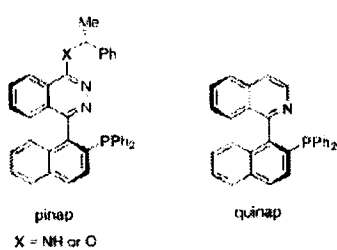
Scheme 32 Tentative mechanism of the three-component reaction



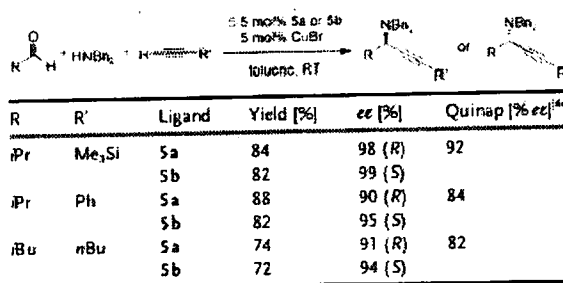
c. Carreira Group: Copper(I)-Pinap complex

Angew. Chem. Int. Ed. 2004, 43, 5971

Scheme 33 Structure of Pinap



Scheme 34 Cu-catalyzed addition of alkynes to imines



Advantage of Pinap:

1. Convenient preparation
2. Easily modify

Other Application of Pinap ligand:

Catalytic, Enantioselective, Conjugate Alkyne Addition *J. Am. Chem. Soc.* 2005, 127, 9682.