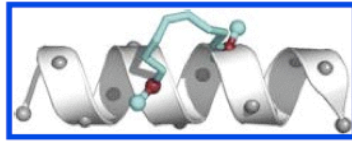


Cross-Linking System to Stabilize Alpha Helics Conformation in Short Peptides

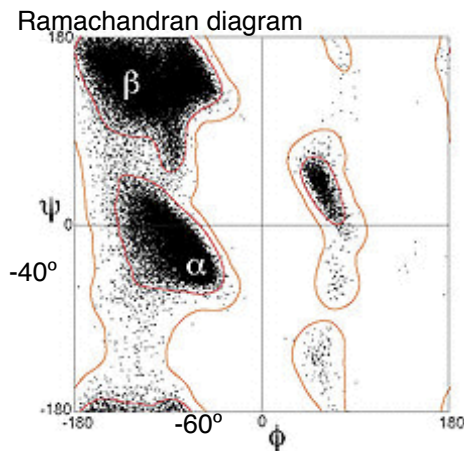
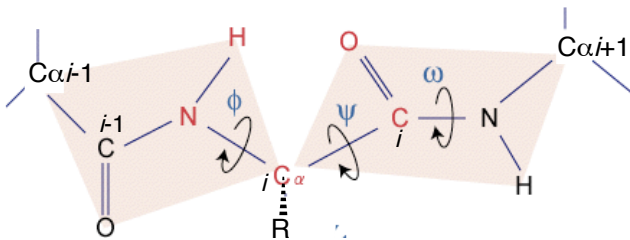
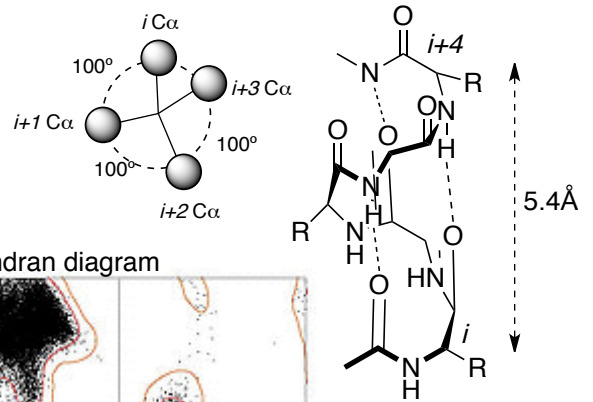


- Contents**
- 1 alpha helics conformation
 - 2 some strategies to stabilize alpha helics conformation
 - 3 cross-linking with lactam bridge
 - 4 cross-linking with all hydrocarbon staple
 - 5 cross-linking with acetylenic cross-linking agent

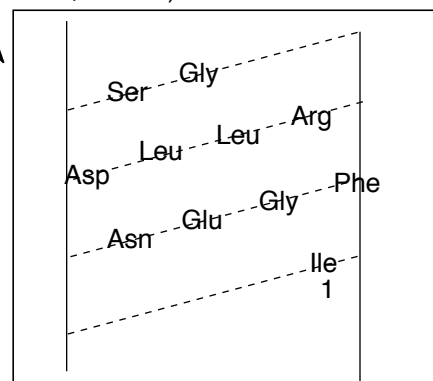
1 alpha helics conformation

Alpha helics, a common motif in the secondary structure of peptides, is right-handed coiled conformation.

- Every backbone N-H group donates a hydrogen bond to the backbone C=O group. (*i*-*i*+4 hydrogen bonding)
- Each amino acid residues corresponds to a 100° turn in the helics.
- The pitch of the alpha-helics (the vertical distance between two consecutive turns of the helics) is 5.4Å.
- Residues in alpha helics typically adopt backbone (ϕ, ψ) dihedral angles around (-60°, -40°).

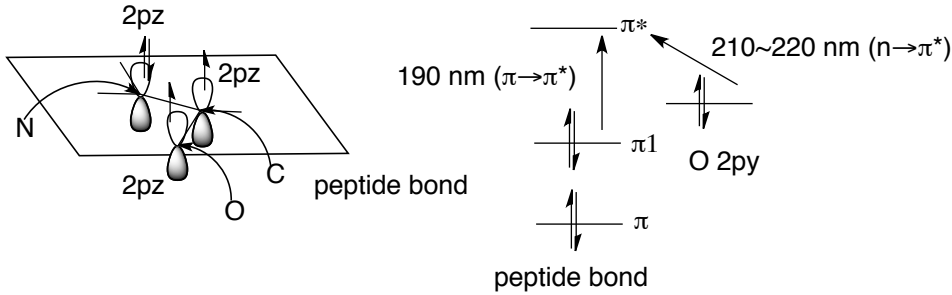


- Alpha helics is composed of 4 ~ 40 amino acid residues. (average 10 residues, 3 turns)
- Many of alpha helics are amphipathic. (e.g. alcohol dehydrogenase)
- Alpha helical sequences of ≤ 15 amino acids are estimated to account for ~30% of protein structure and frequently mediate biological processes through interactions with proteins, DNA or RNA.
- The folded states of most proteins are only ≤ 10 kcal/mol lower in energy than their unfolded states. (PNAS, 2006, 103, 16623)
But short peptide sequences are usually unfolded in aqueous solution, away from the helix stabilizing environments of proteins.



If short peptide alpha helics could be stabilized or mimicked by small molecules, such compounds might be valuable chemical or biological probes and lead to development of novel pharmaceuticals, vaccines, diagnostics, biopolymers and industrial agents.

• Circular Dichroism (CD) used to provide a quantitative measure of helical content.



$$[\theta(\lambda)] = \chi_{\alpha}[\theta_{\alpha}(\lambda)] + \chi_{\beta}[\theta_{\beta}(\lambda)] + \chi_{rc}[\theta_{rc}(\lambda)]$$

$[\theta_{\alpha}(\lambda)], [\theta_{\beta}(\lambda)], [\theta_{rc}(\lambda)];$ 2次構造が既知のポリペプチド鎖
の α, β, rc 構造に対する $[\theta]$
 $\chi_{\alpha}, \chi_{\beta}, \chi_{rc};$ それぞれの存在割合

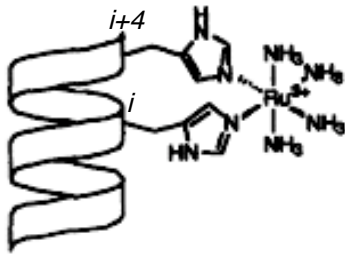
$$\text{alpha helix(\%)} = \frac{[\theta]_{208\text{nm}} + 4000}{-33000 + 4000} \times 100$$

Alanine	Ala	A	Tyrosine	Tyr	Y
Lysine	Lys	K	Tryptophan	Trp	W
Aspartic acid	Asp	D	Threonine	Thr	T
Glutamic acid	Glu	E	Valine	Val	V
Asparagine	Asn	N	Histidine	His	H
Arginine	Arg	R	Phenylalanine	Phe	F
Isoleucine	Ile	I	Proline	Pro	P
glycine	Gly	G	Methionine	Met	M
Glutamine	Gln	Q	Leucine	Leu	L
Cysteine	Cys	C			
Serine	Ser	S			

Ala; CH₃ Glu; (CH₂)₂CO₂H
Lys; (CH₂)₄NH₂
Asp; CH₂CO₂H

2. some strategies to stabilize alpha helics conformation

2-1 Metal chelation

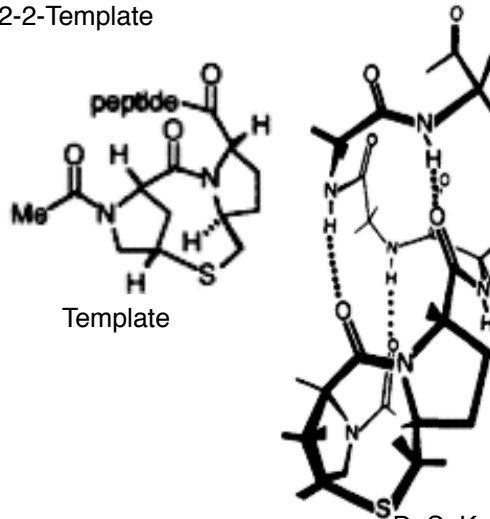


Peptide: Ac-AEAAAKEAAA KHAAAHA-NH₂

Figure 20 Binding of Ru(III) by His¹⁴²

M. Ghandiri JACS. 1990, 112, 9633
Rev) Tetrahedron 1999, 55, 11711

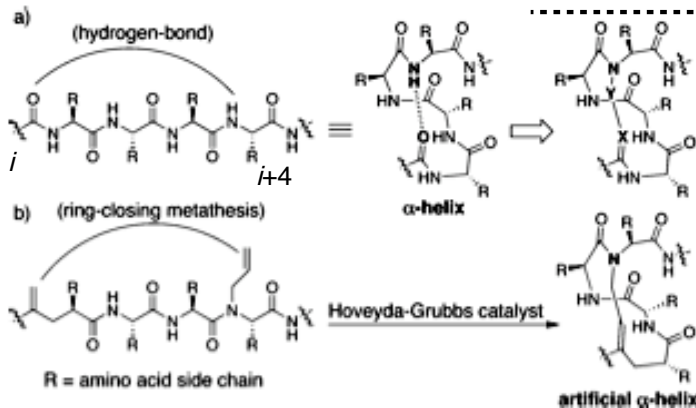
2-2-Template



Template

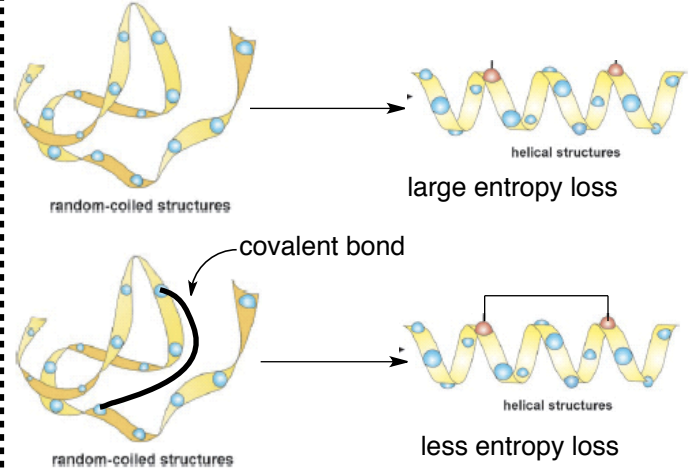
D. S. Kemp et al. JOC, 1991, 56, 6672

2-3 Hydrogen-Bond Surrogate



Paramijit S. Arora et al. JACS. 2004, 126, 12252

cross-linking

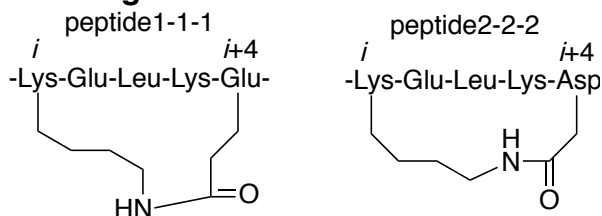


3 cross-linking with lactam bridge

Taylor, J. W. JACS, 1990, 112, 6046

Taylor, J. W. JACS, 1992, 114, 6966

3-1-1 design



3-1-2 synthesis of peptide 1-1-1

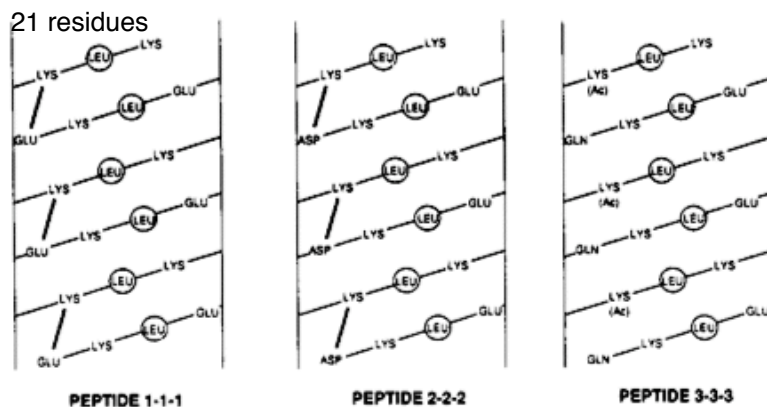
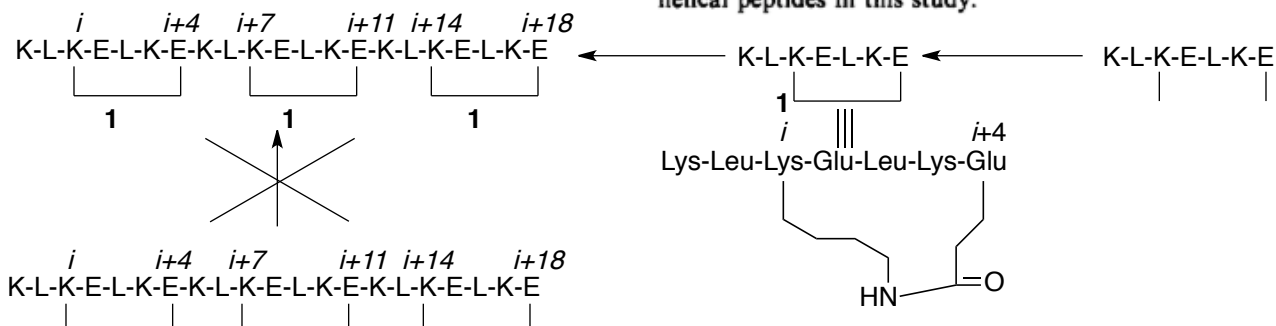
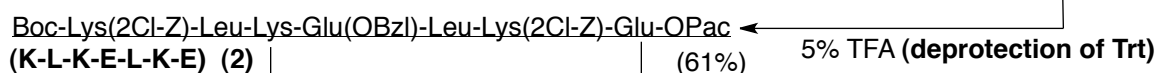
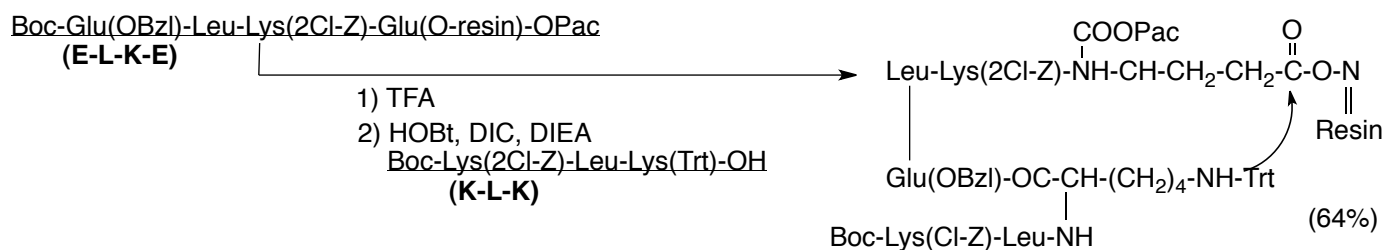
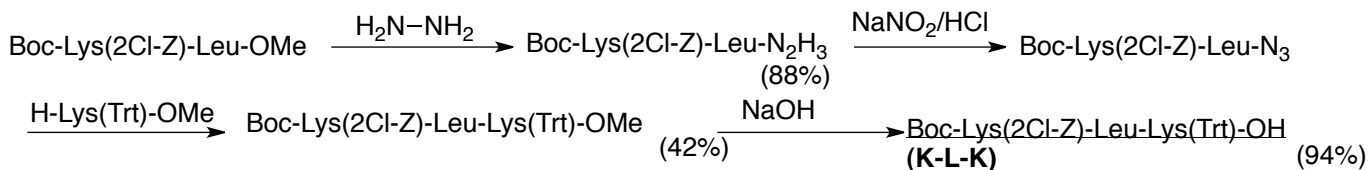
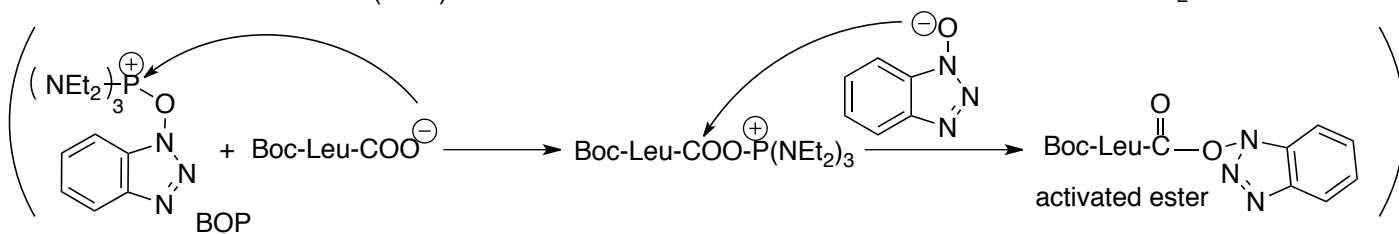
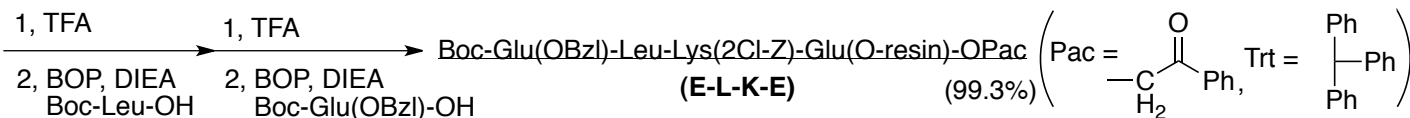
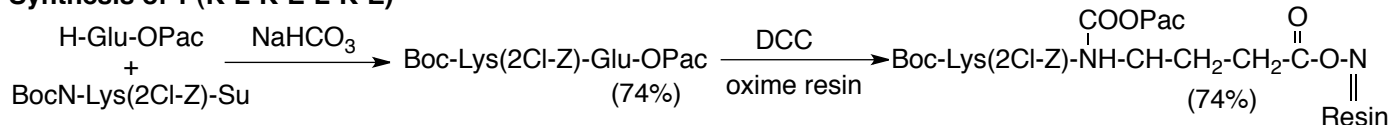
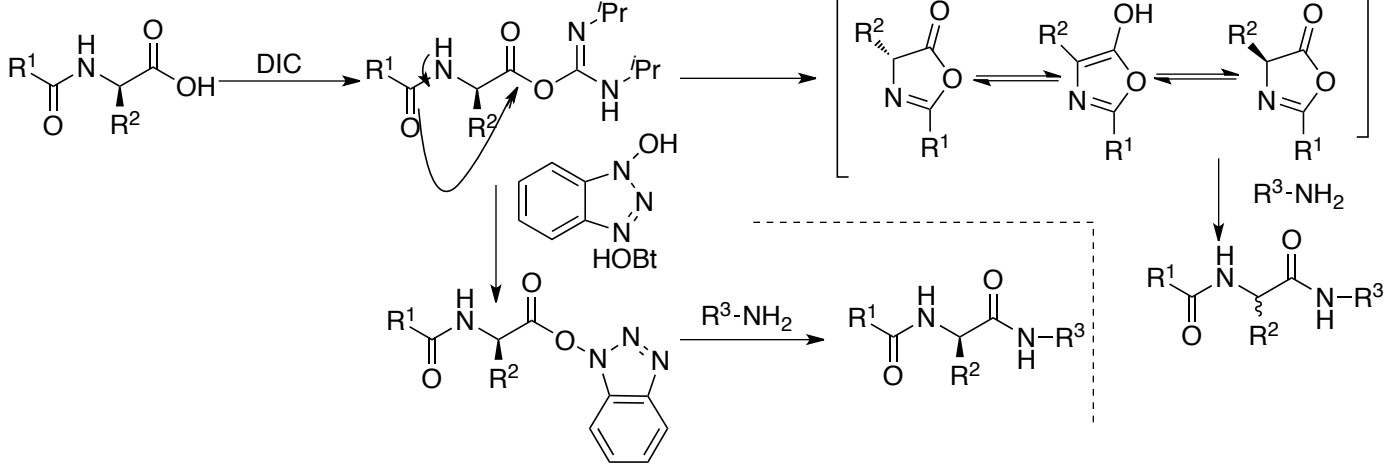


Figure 1. Helical net diagrams of the synthetic model amphiphilic α -helical peptides in this study.

Synthesis of 1 (K-L-K-E-L-K-E)



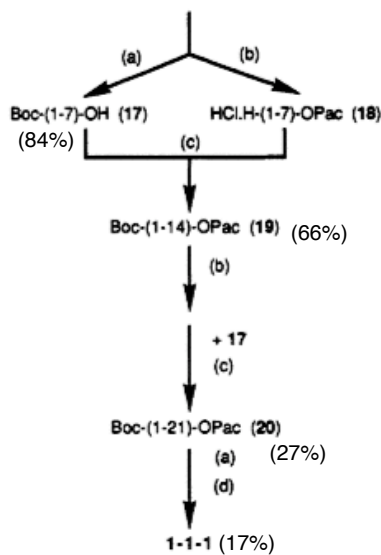
peptide-peptide coupling (DIC/HOBt)



synthesis of 1-1-1

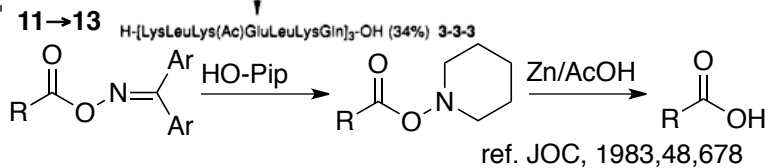
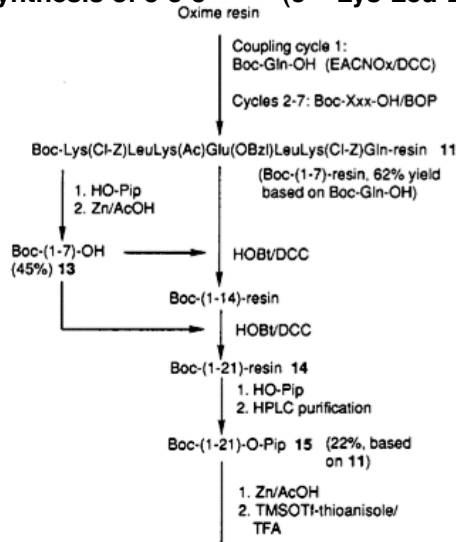
Scheme II. Synthesis of 1-1-1 by Segment Condensation*

Boc-(1-7)-OPac (2) Boc-Lys(2Cl-Z)-Leu-Lys-Glu(OBzl)-Leu-Lys(2Cl-Z)-Glu-OPac



* (a) Zn/AcOH; (b) TFA followed by HCl/dioxane; (c) HOBt/DCC; (d) TMSOTf + thioanisole in TFA.

synthesis of 3-3-3 (3 = Lys-Leu-Lys-Glu-Leu-Lys-Gln)



3-1-3 CD spectra

CD spectra of 1-1-1, 2-2-2, 3-3-3

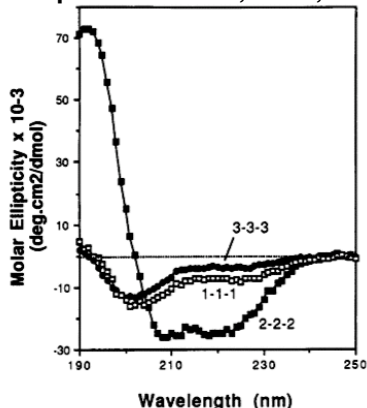


Figure 2. CD spectra of model peptides in aqueous 10 mM phosphate buffer, pH 7.0, at 25 °C: 1-1-1, 40 μM (□); 2-2-2, 30 μM (■); and 3-3-3, 54 μM (●).

Cross-linking through Lys-Asp could stabilize alpha-helics in 21 residues peptide.

Cross-linking length is important?

Effect of TFE on helix formation

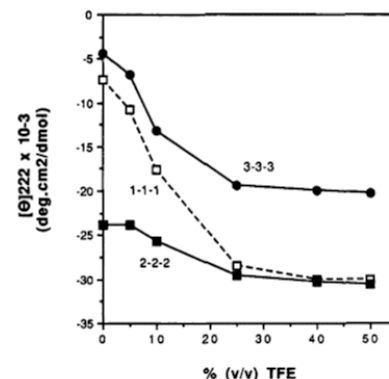
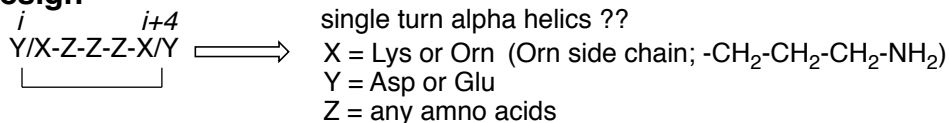


Figure 4. Effect of TFE on helix formation of synthetic peptides. Increasing proportions of TFE (% v/v) were added to peptide solutions in aqueous 10 mM phosphate buffer, pH 7.0, at 25 °C and [θ]₂₂₂ was monitored: 1-1-1, 40 μM (□); 2-2-2, 30 μM (■); 3-3-3, 54 μM (●).

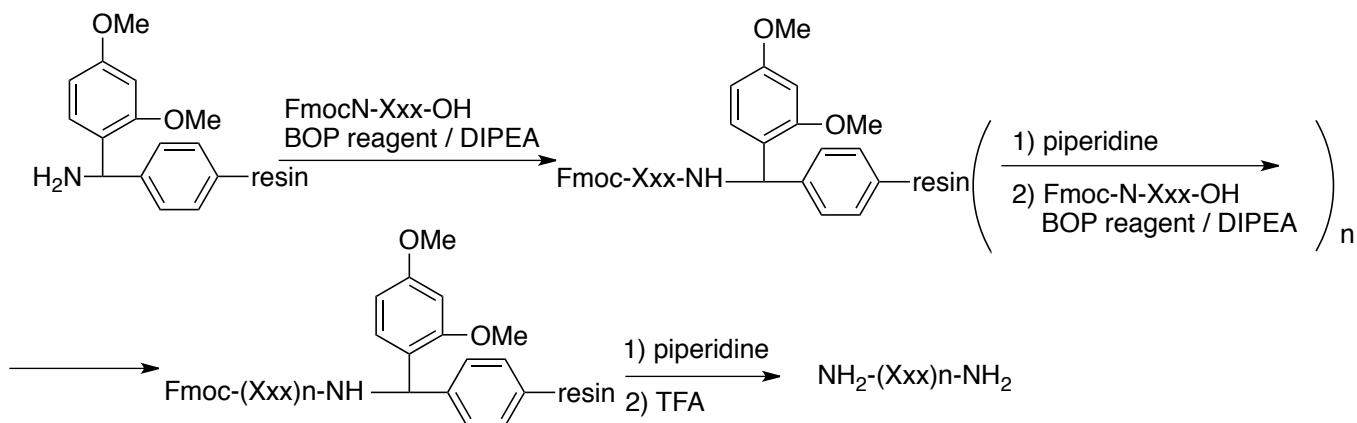
3-2 single turn peptide alpha helices

Nicholas E. Shepherd and David P. Fairlie JACS, 2005, 127, 2974

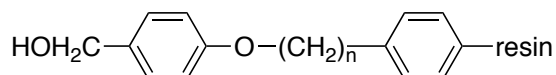
3-2-1 design



3-2-2 solid phase peptide synthesis (Fmoc protocol)



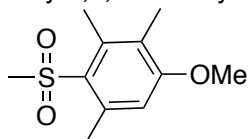
• Some Resins are used. Such as



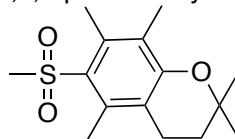
• Acids are not used except final step (cleavage from resin).

Side chains are protected with O^tBu (for Asp, Glu), Boc (for Lys, Ser and His), Mtr, Pmc (for Arg) and so on which can be deprotected with TFA.

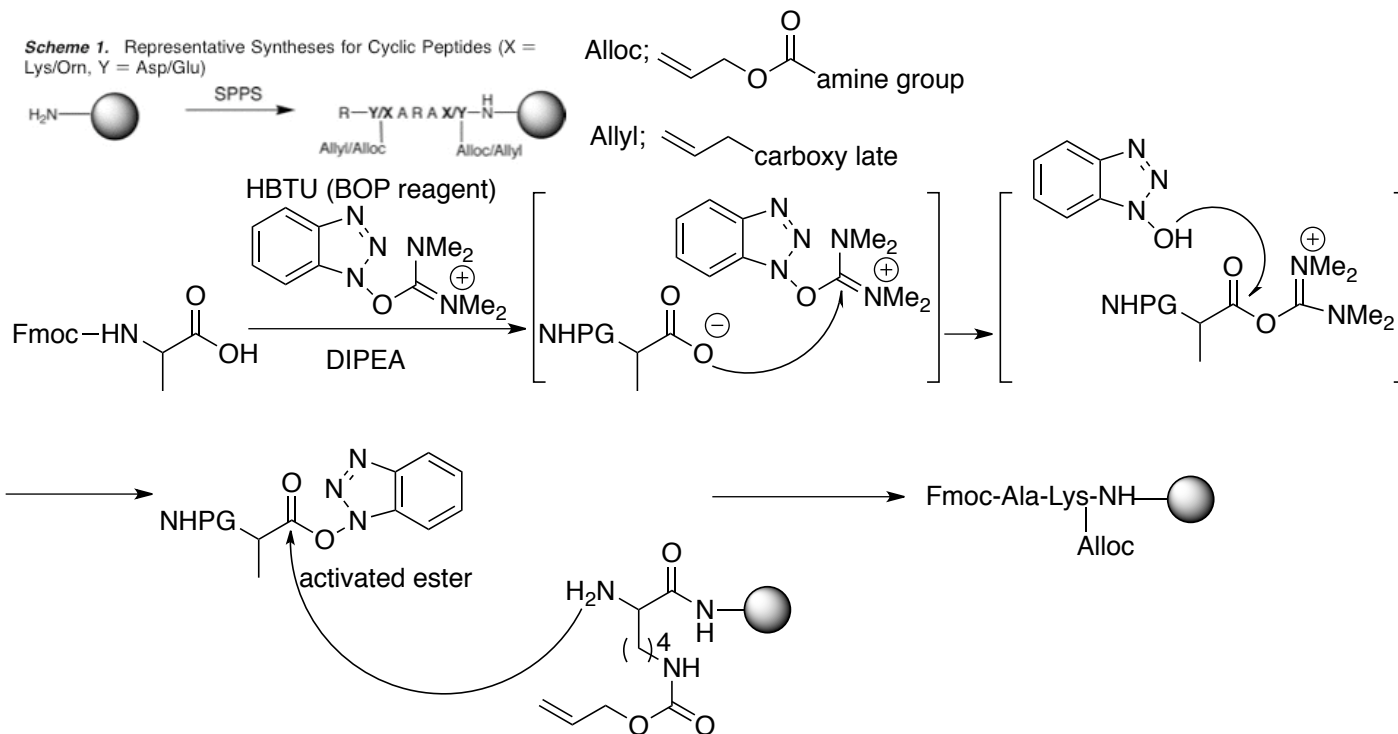
Mtr; 4-methoxy-2,3,6-trimethylbenzenesulfonyl



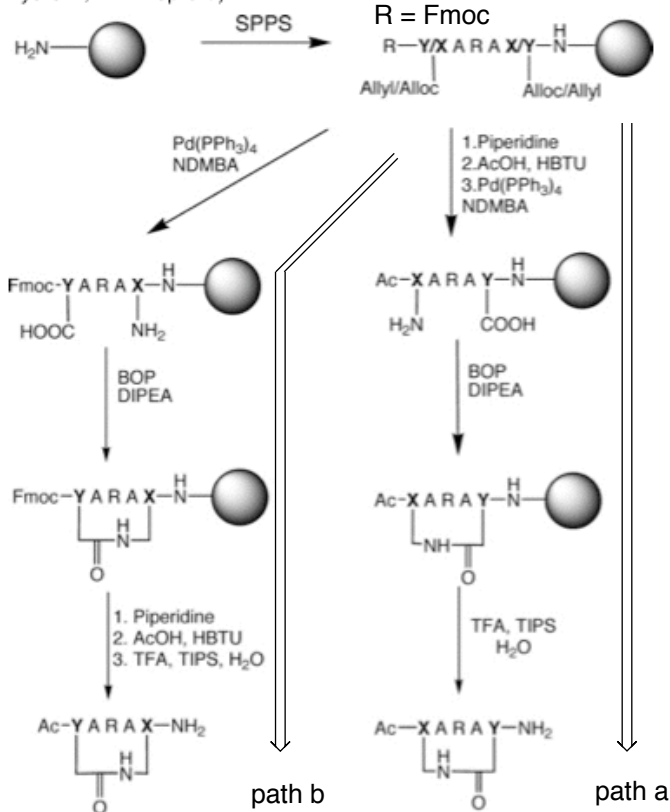
Pmc; 2,2,5,7,8-pentamethylchroman-6-sulfonyl



Scheme 1. Representative Syntheses for Cyclic Peptides (X = Lys/Orn, Y = Asp/Glu)

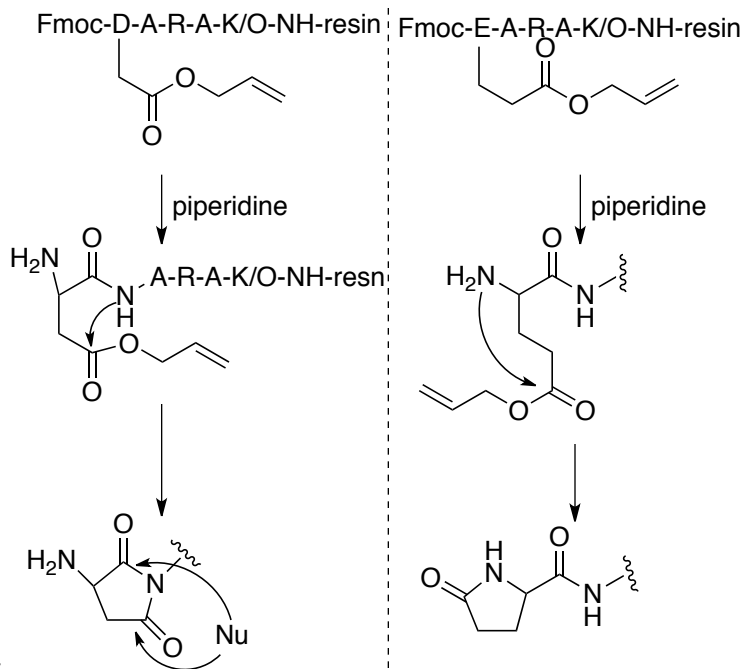


Scheme 1. Representative Syntheses for Cyclic Peptides (X = Lys/Orn, Y = Asp/Glu)



path a; Fmoc-K/O-A-R-A-D/E-NH-resin
path b; Fmoc-D/E-A-R-A-K/O-NH-resin

When Fmoc-D/E-A-R-A-K/O-NH- was cyclized through path a, some side reactions occurred.



Thanks Dr. Nich.

3-2-3 result

CD-spectra-1 Ac-(cyclo-1,5)-[X/Y-ARA-Y/X]

Table 1. Molar Ellipticities ($[\theta]$ deg cm² dmol⁻¹ residue⁻¹) at $\lambda = 215, 207,$ and 190 nm, Ratios of Ellipticities at $215/207$ nm, and Percentage Helicity for Peptides 1–12 in 10 mM Phosphate Buffer (pH 7.4, 25 °C)

peptide	$[\theta]_{215}$	$[\theta]_{207}$	$[\theta]_{190}$	$\theta_{215}/\theta_{207}$	relative helicity ^a
Ac-(cyclo-1,5)-[KARAE]-NH ₂ (1)	-1068	-3393	-10611	0.31	0.08
Ac-(cyclo-1,5)-[EARAK]-NH ₂ (2)	-7430	-12803	20735	0.58	0.58
Ac-(cyclo-1,5)-[KARAD]-NH ₂ (3)	-12757	-12211	38300	1.04	1.00
Ac-(cyclo-1,5)-[DARAK]-NH ₂ (4)	-7723	-10705	15600	0.72	0.60
Ac-(cyclo-1,5)-[OARAD]-NH ₂ (5)	92	-2077	-4613	-0.04	0
Ac-(cyclo-1,5)-[DARAO]-NH ₂ (6)	-4671	-9748	-6954	0.48	0.37
Ac-(cyclo-1,5)-[OARAE]-NH ₂ (7)	741	-3368	-16228	-0.22	0
Ac-(cyclo-1,5)-[EARAO]-NH ₂ (8)	2442	-1917	-11256	-1.27	0
Ac-KARAD-NH ₂ (9)	-524	-5555	-7372	0.09	0.04
Ac-(cyclo-1,5)-[KARAD]-OH (10)	207	-5643	-13953	-0.04	0
<i>a</i>	-625	-3195	-2659	0.20	0.06
H-(cyclo-1,5)-[KARAD]-NH ₂ (11)	-812	-2228	1355	0.36	0.06
<i>b</i>	-2590	-3327	8452	0.78	0.2
H-(cyclo-1,5)-[KARAD]-OH (12)	-1033	-5737	-4966	0.18	0.08

^a In 0.01 M HCl pH 2. ^b In 0.001 M NaOH pH 10. ^c $[\theta]_{215} (3)/[\theta]_{215} (x)$ refer to "Quantitation of Helicity" section.

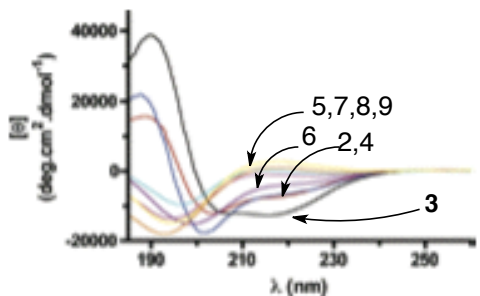
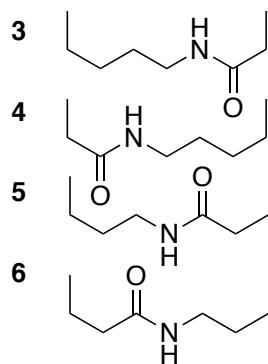


Figure 2. CD spectra of cyclic pentapeptides 1 (pink), 2 (blue), 3 (black), 4 (red), 5 (cyan), 6 (purple), 7 (orange), and 8 (yellow) in 10 mM phosphate buffer (pH 7.4, 25 °C).



CD-spectra-2 Ac-(cyclo-2,6)-R[K-ZZZ-D] (helics dependence on sequences)

Table 2. Molar Ellipticities ($[\theta]$ deg cm² dmol⁻¹ residue⁻¹) at $\lambda = 215, 207,$ and 190 nm, Ratios of Ellipticities at 215/207 nm, and Percent Helicity for Cyclic Peptides **13–21** in 10 mM Phosphate Buffer pH 7.4 at 25 °C

peptide	$[\theta]_{215}$	$[\theta]_{207}$	$[\theta]_{190}$	$\theta_{215}/\theta_{207}$	relative helicity ^a
Ac-(cyclo-2,6)-R[KAAAD]-NH ₂ (13)	-13 537	-13 684	39 352	0.99	0.91
Ac-(cyclo-2,6)-R[KALAD]-NH ₂ (14)	-14 798	-15 165	46 621	0.98	1.00
Ac-(cyclo-2,6)-R[KAMAD]-NH ₂ (15)	-11 853	-12 296	38 464	0.96	0.80
Ac-(cyclo-2,6)-R[KAQAD]-NH ₂ (16)	-11 394	-12 279	36 865	0.93	0.84
Ac-(cyclo-2,6)-R[KAFAD]-NH ₂ (17)	-8644	-9087	27 718	0.95	0.76
Ac-(cyclo-2,6)-R[KAGAD]-NH ₂ (18)	-4874	-7678	10 036	0.63	0.32
Ac-(cyclo-2,6)-R[KGSAD]-NH ₂ (19)	-4810	-6975	12 831	0.69	0.32
Ac-(cyclo-2,6)-R[KSSSD]-NH ₂ (20)	-4432	-8017	6827	0.55	0.30
Ac-(cyclo-2,6)-R[KGGGD]-NH ₂ (21)	-2131	-4868	-1593	0.44	0.14

Rank order of helical propencities (rev. Tetrahedron, 1999,55,11711)

AQ/AK	$\Delta\Delta G$ kcal mol ⁻¹	AXA	$\Delta\Delta G$ kcal mol ⁻¹	T4	$\Delta\Delta G$ kcal mol ⁻¹
				lysosyme	
Ala	-1.88	Ala	-1.05	Ala	-0.96
Arg	-1.67	Glu	-0.84	Leu	-0.92
Leu	-1.60	Leu	-0.69	Met	-0.86
Lys	-1.52	Ile	-0.67	Ile	-0.84
Met	-1.37	Arg	-0.64	Gln	-0.80
Gln	-1.31	Met	-0.53	Arg	-0.77
Glu	-1.20	His	-0.51	Lys	-0.73
Ile	-1.18	Lys	-0.47	Tyr	-0.72
Tyr	-1.3 to -1.1	Val	-0.32	Val	-0.63
Ser	-1.10	Ser	-0.29	Trp	-0.58
His	-1.07	Asp	-0.27	Phe	-0.59
Cys	-1.06	Gln	-0.25	His	-0.57
Asp	-1.00	Asn	-0.20	Thr	-0.54
Asn	-0.99	Thr	-0.17	Glu	-0.53
Trp	-1.1to-0.97	Gly	0.00	Ser	-0.53
Phe	-0.95	Phe	0.06	Asp	-0.42
Val	-0.83	Trp	0.06	Cys	-0.42
Thr	-0.56	Cys	0.12	Asn	-0.39
Gly	0.00	Tyr	0.90	Gly	0.00
Pro	+3.00	Pro	1.51	Pro	2.50

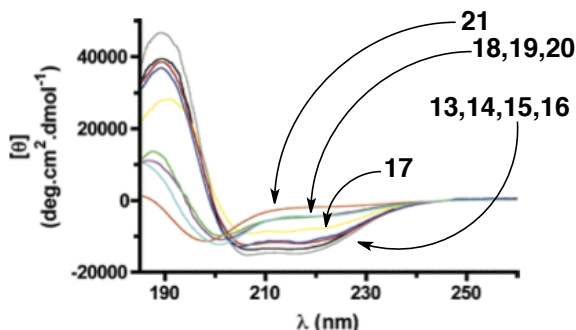


Figure 3. CD spectra of compounds **13** (black), **14** (grey), **15** (red), **16** (blue), **17** (yellow), **18** (purple), **19** (green), **20** (aqua), and **21** (orange) in 10 mM phosphate buffer (pH 7.4, 25 °C).

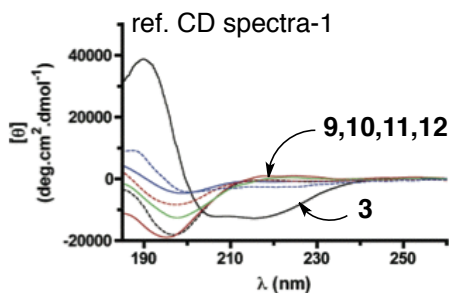


Figure 7. CD spectra of compounds **3** (black, solid), **9** (black, broken), **10** (red, solid), **11** (blue, solid), and **12** (green, solid) in 10 mM phosphate buffer (pH 7.4, 25 °C); **10** (red, broken) in 0.01 M HCl (pH 2) and **11** (blue, broken) in 0.001 M NaOH (pH 11).

3-2-4 NMR analysis of short peptides

A low temperature dependence for amide NH chemical shifts (values ≤ 4 ppb/K) are indicative of hydrogen bonds.

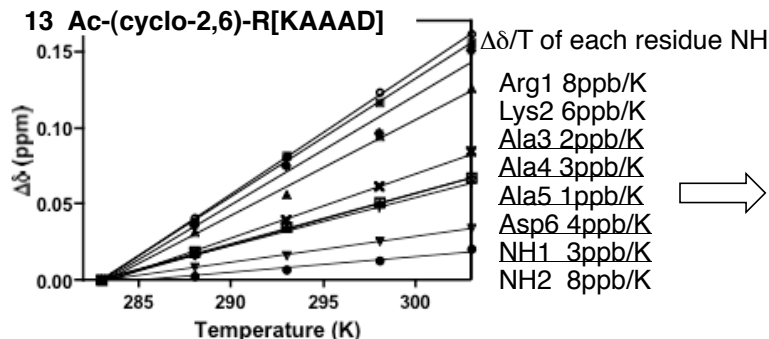


Figure S5. Temperature dependence of the amide NH chemical shifts for **13** in H₂O/D₂O (9:1). Line slopes indicating temperature coefficients ($\Delta\delta/T$) for each residue are shown (in brackets) ■ Arg1 (8 ppb/K), ▲ Lys2 (6 ppb/K), ▼ Ala3 (2 ppb/K), ◆ Ala4 (3 ppb/K), ● Ala5 (1 ppb/K), ★ Asp6 (4 ppb/K), □ Terminal NH1 (3 ppb/K), ○ Terminal NH2 (8 ppb/K), † Arg1 side chain NH (3 ppb/K), * Lys2 side chain NH (7 ppb/K), values ≤ 4 are indicative of hydrogen bonds.

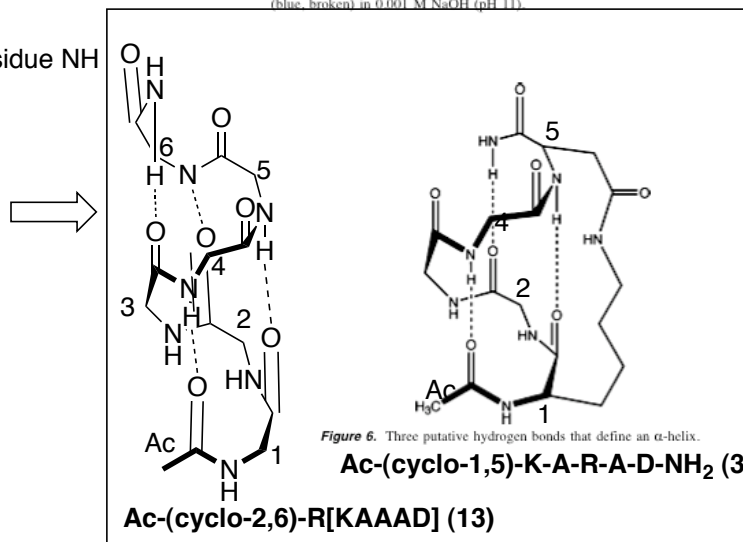


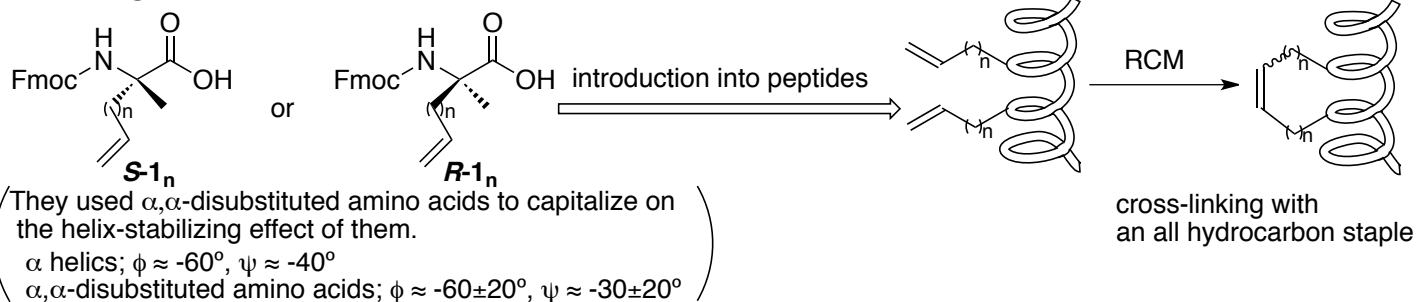
Figure 6. Three putative hydrogen bonds that define an α -helix.

The NMR analysis indicated the presence of three hydrogen bonds within the pentapeptide.

4 cross-linking with an all hydrocarbon staple

Gregory L. Verdine et al. JACS, 2000, 122, 5891

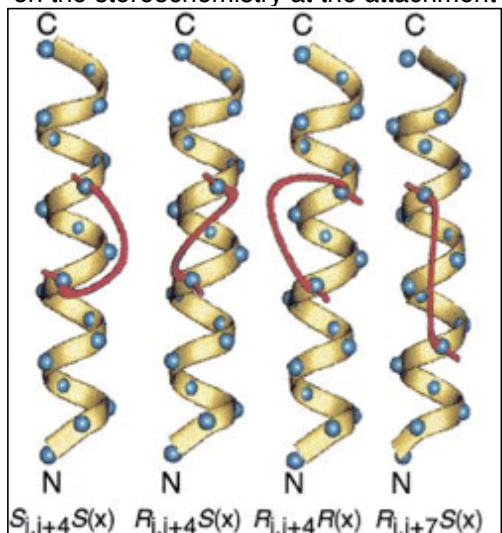
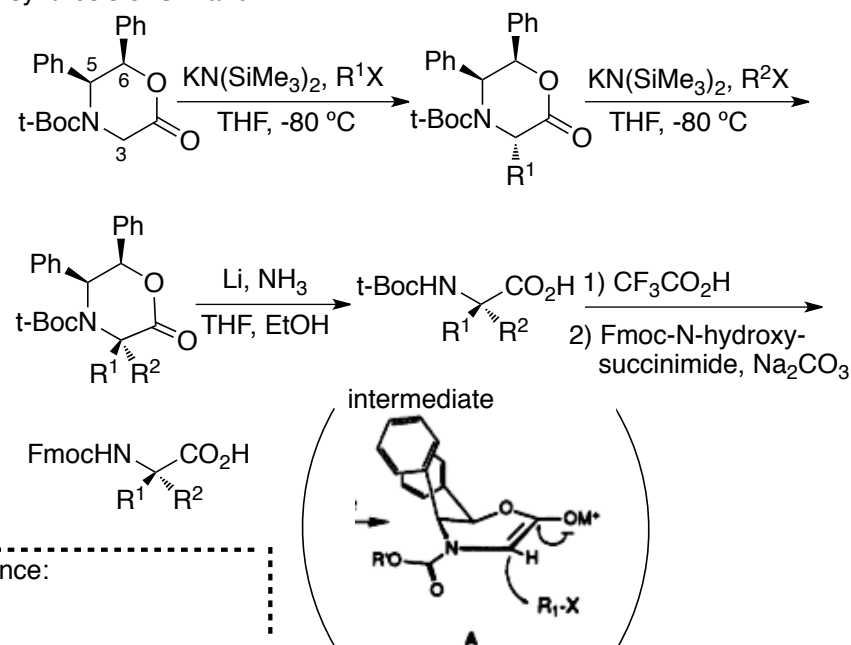
4-1 design



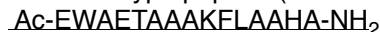
4-2 synthesis

The actual structure of cross-links is dependent on the stereochemistry at the attachment points.

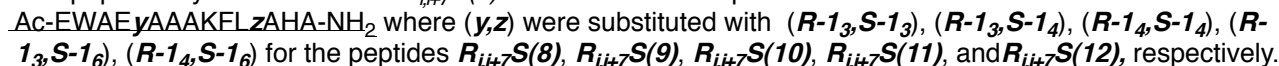
synthesis of **S-1** and **R-1**



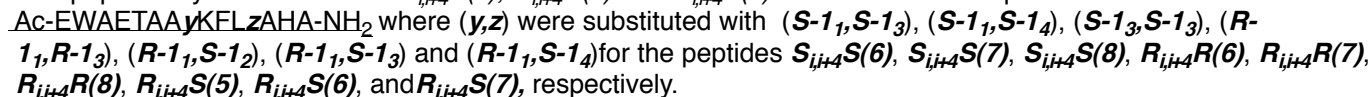
The wild-type peptide (RNase A) has the sequence:



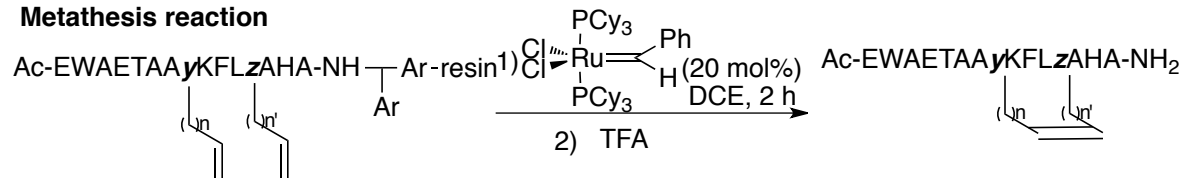
The peptides synthesized in the $R_{i,i+7}S(x)$ series have the sequences:



The peptides synthesized in the $S_{i,i+4}S(x)$, $R_{i,i+4}R(x)$ and $R_{i,i+4}S(x)$ series have the sequences:

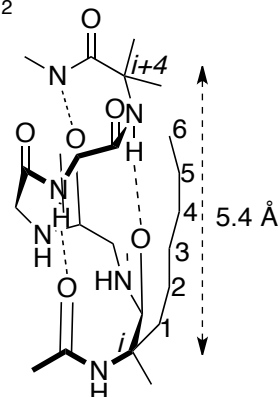


Metathesis reaction



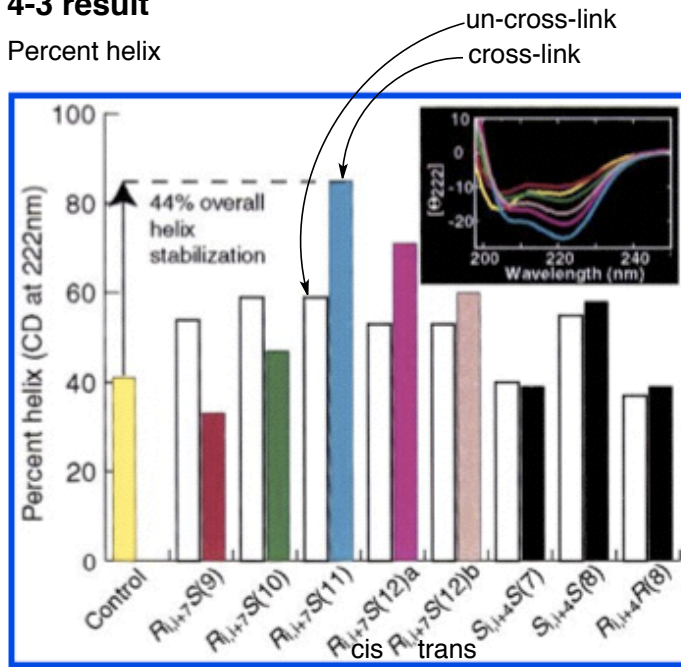
percent conversion product / product+starting material (determined by HPLC)

cross-link	% conversion ^a	cross-link	% conversion ^a	cross-link	% conversion ^a
$R_{i,i+7}S(8)$	0	$S_{i,i+4}S(6)$	0	$R_{i,i+4}R(6)$	0
$R_{i,i+7}S(9)$	51	$S_{i,i+4}S(7)$	68	$R_{i,i+4}R(7)$	17
$R_{i,i+7}S(10)$	77	$S_{i,i+4}S(8)$	>98	$R_{i,i+4}R(8)$	>98
$R_{i,i+7}S(11)$	>98				
$R_{i,i+7}S(12)$	>98				



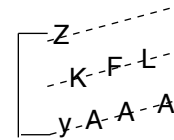
4-3 result

Percent helix



cleavage by trypsin (protease resistance *in vitro*)

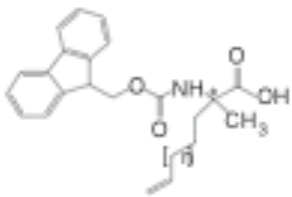
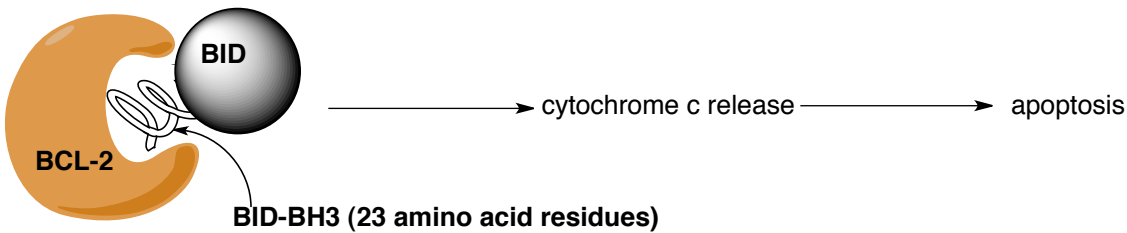
cross-link	cleavage rate constant ($M^{-1} s^{-1}$)	
	unmetathesized	metathesized and hydrogenated
control	2.39	
$R_{i,j+7}S(9)$		0.37
$R_{i,j+7}S(10)$		0.34
$R_{i,j+7}S(11)$	0.50	0.058
$R_{i,j+7}S(12)$		0.12



4-4 biological activity of α helix peptide stabilized by a hydrocarbon staple

Gregory L. Verdine et al. Science, 2004, 305, 1466

Bcl-2 (B-cell lymphoma 2) proteins constitute a critical control point for the regulation of apoptosis. BID is a pro-apoptotic protein that, in response to death receptor signaling, interconnects the extrinsic and core intrinsic apoptotic pathways. Activated BID can bind to BCL-2 and trigger activation of pro-apoptotic proteins, resulting in cytochrome c release.

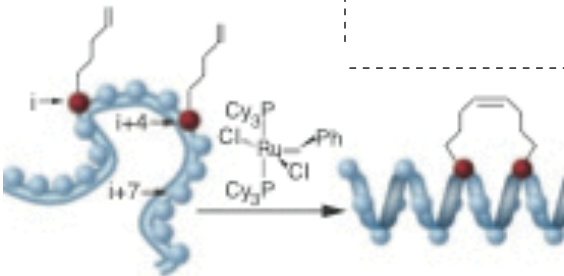


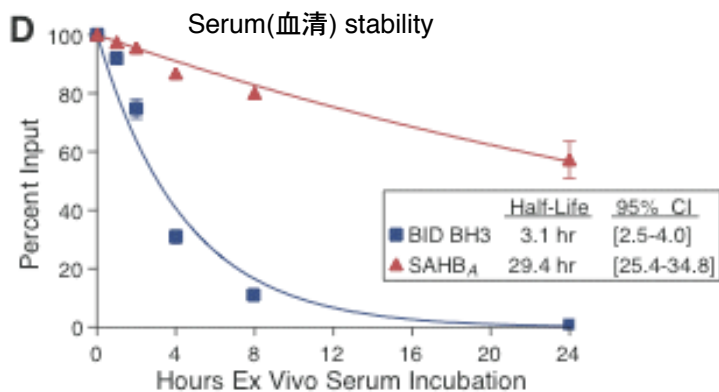
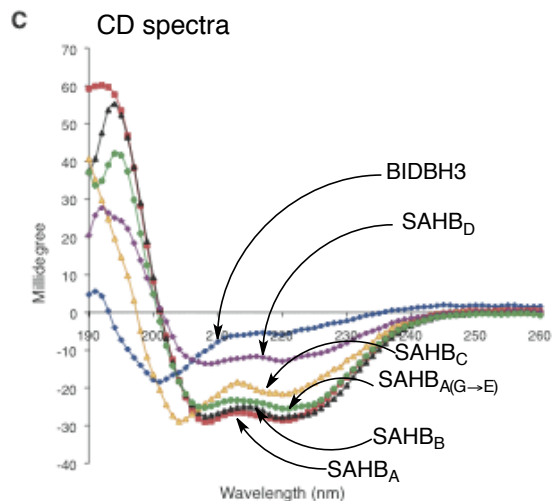
n=1: S5, R5
n=4: S8

B

Compound	1	5	10	15	20	
BID BH3	EDI	IRNI	ARHLA	QVGDSN _L	DRSIW	
SAHB _A	EDI	IRNI	ARHLA*	VGD*N _L	DRSIW	$S_{i,i+4}S(8)$
SAHB _A (G→E)	EDI	IRNI	ARHLA*	VED*N _L	DRSIW	$S_{i,i+4}S(8)$
SAHB _B	EDI	IRNI*	RHL*	QVGDSN _L	DRSIW	$S_{i,i+4}S(8)$
SAHB _C	EDI	IRNIA*	HLA*	VGDSN _L	DRSIW	$S_{i,i+4}S(8)$
SAHB _D	EDI	IRNIAR*	LAQVGD*	N _L	DRSIW	$R_{i,i+7}S(11)$

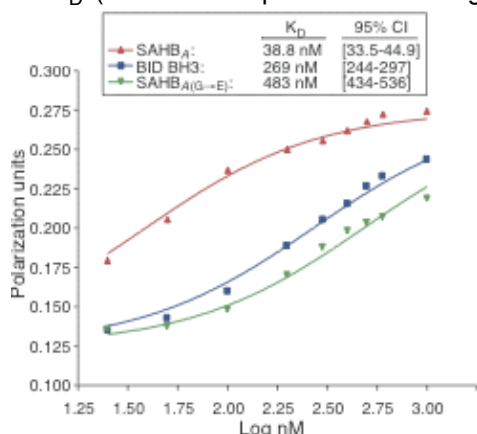
*=S5, *=R5, *=S8, N_L=norleu



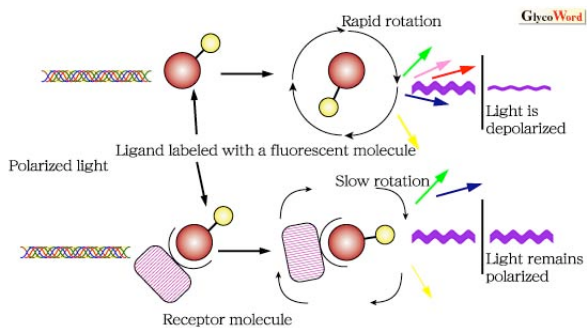
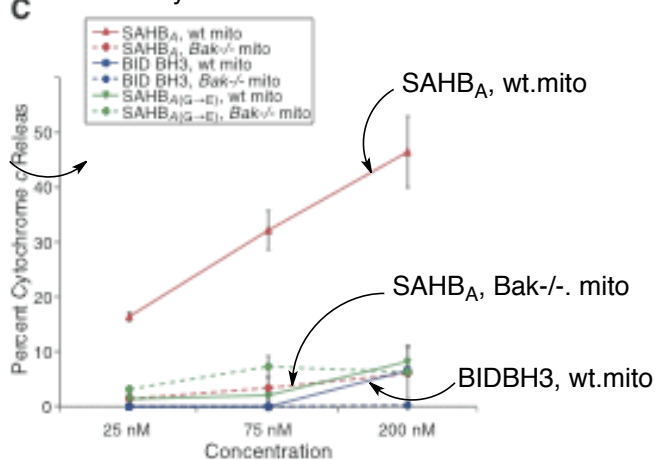


Serum includes all proteins not used in blood clotting and all the electrolytes, antibodies, antigens, hormones, and any exogenous substances (e.g., drugs and microorganisms).

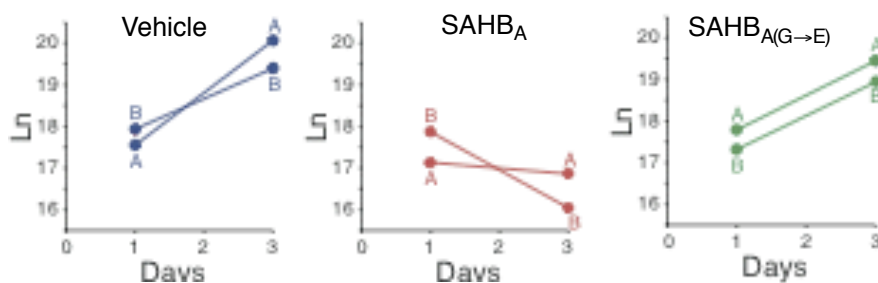
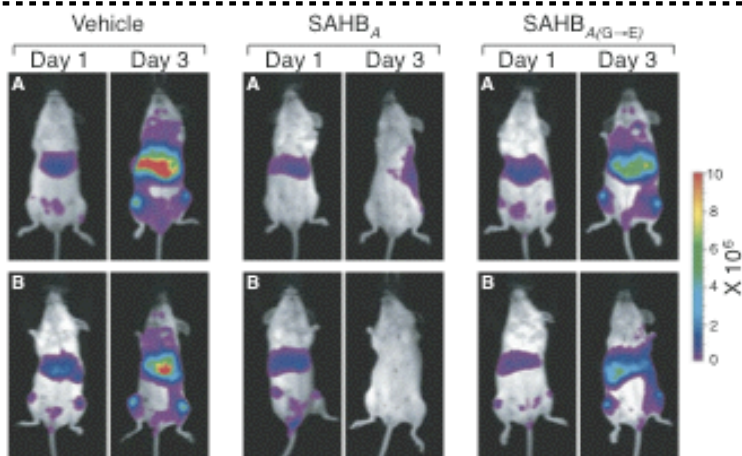
K_D (fluorescence polarization binding assay)



C Percent cytochrome *c* release



intravenous injection (静注) leukemia cells that stably expressed luciferase



4-5 the physical forces that underlie stapled peptide stabilization

Gregory L. Verdine and Eugene I. Shakhnovich et al. JACS, 2009, 131, 4622

Why some cases (RNase A) an $i, i+7$ staple can be more stabilizing than an $i, i+4$ staple, while in other cases (BIDBH3) the opposite is true?

What factors are most important?

$$H = E_{\text{con}} + W_{\text{hb}}E_{\text{hb}} + W_{\text{bbtor}}E_{\text{bbtor}} + W_{\text{sct}}E_{\text{sct}} + W_{\text{linktor}}E_{\text{linktor}}$$

E_{con} ; intermolecular potential energy

E_{hb} ; hydrogen-bonding potential

E_{bbtor} ; sequence-dependent

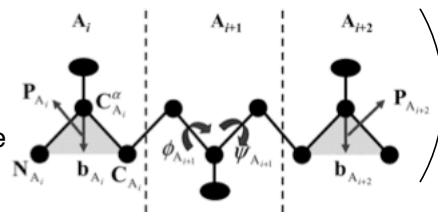
backbone torsional potential

E_{sct} ; side-chain torsional angle potential

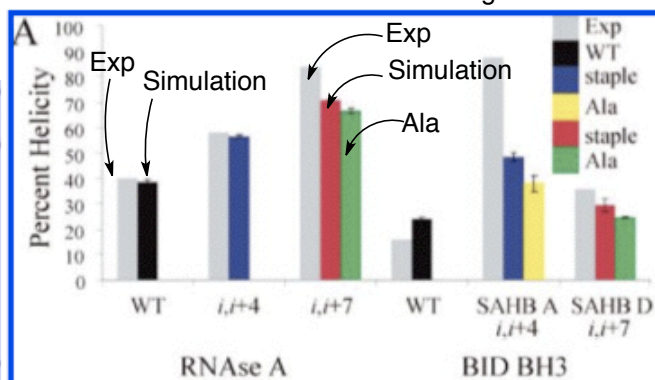
E_{linktor} ; torsional potential for hydrocarbon linker

S = configurational entropy of the peptide + that of the staple

Four variables in E_{bbtor} function are ϕ_{A+1}, ψ_{A+1} , the angle between b_{A_i} and $b_{A_{i+2}}$ and the angle between P_{A_i} and $P_{A_{i+2}}$.



Conformations that were obtained during MC simulations



Their simulation was able to predict which staple would be more stabilizing for each peptide.

In all cases, the stapled peptides are more helical than the corresponding alanine-substituted versions, suggesting that the physical constraint conferred by macrocyclic ring formation contributes to increased helical propensity.

	Compound	Sequence
RNase A	WT	EWAETAAAKFLAAHA
	$i, i+4$ staple	EWAETAA*KFL*AHA
	$i, i+7$ staple	EWAE*AAAKFL*AHA
	$i, i+7$ Ala	EWAEAAAKFLAAHA
BID BH3	WT	EDIIRNIARHLAQVGDSDRSIW
	SAHB A $i, i+4$ staple	EDIIRNIARHLA*VGD*MDRSIW
	$i, i+4$ Ala	EDIIRNIARHLAAVGDAMDRSIW
	SAHB D $i, i+7$ staple	EDIIRNIAR*LAQVGD*MDRSIW
$i, i+7$ Ala	EDIIRNIARALAQVGDAMDRSIW	

Table 1. Thermodynamic Analysis of RNase A and BID BH3 Simulations Using the Number of Helical Residues as the Order Parameter

	ΔH	$-\Delta S$	ΔG	$\Delta\Delta H$	$-\Delta\Delta S$	$\Delta\Delta G$
RNase A						
WT	-29.2 ± 0.6	32.5 ± 0.6	3.3 ± 0.1			
$i, i+4$ staple	-30.0 ± 0.5	31.9 ± 0.4	1.9 ± 0.1	-0.8 ± 0.5	-0.6 ± 0.5	-1.37 ± 0.07
$i, i+7$ staple	-33.25 ± 0.06	33.4 ± 0.1	0.12 ± 0.09	-4.0 ± 0.6	0.9 ± 0.5	-3.14 ± 0.07
$i, i+7$ Ala	-32.59 ± 0.09	34.16 ± 0.09	1.57 ± 0.04	-3.4 ± 0.5	1.7 ± 0.5	-1.7 ± 0.2
BID BH3						
WT	-47.0 ± 0.4	48.2 ± 0.4	1.2 ± 0.1			
$i, i+4$ staple	-48.3 ± 0.3	47.7 ± 0.3	-0.63 ± 0.03	-1.4 ± 0.5	-0.5 ± 0.4	-1.8 ± 0.1
$i, i+4$ Ala	-49.2 ± 0.4	49.0 ± 0.5	-0.16 ± 0.04	-2.2 ± 0.3	0.8 ± 0.4	-1.4 ± 0.1
$i, i+7$ staple	-47.5 ± 0.2	48.0 ± 0.1	0.56 ± 0.05	-0.5 ± 0.5	-0.1 ± 0.5	-0.7 ± 0.1
$i, i+7$ Ala	-46.5 ± 0.4	47.9 ± 0.3	1.38 ± 0.04	0.4 ± 0.5	-0.2 ± 0.4	0.2 ± 0.1

^a The temperatures used for RNase A and BID BH3 were 0.78 and 0.70, respectively. Errors are reported as standard deviations of averages of three groups of 50 runs (RNase A) or 200 runs (BID BH3). Thermodynamic values for folding to the helical state (ΔH , ΔS , ΔG) are given, as well as these values relative to the WT peptide ($\Delta\Delta H$, $\Delta\Delta S$, $\Delta\Delta G$).

For all stapled peptides $\Delta\Delta H$ makes a larger contribution to $\Delta\Delta G$ than $-\Delta\Delta S$.

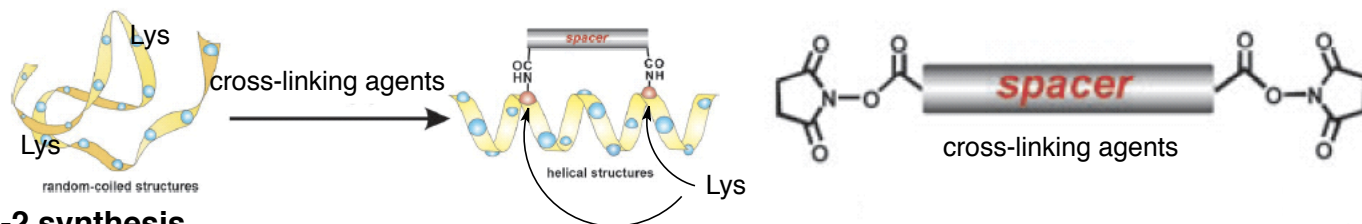
Table S2. Contributions of each potential term to overall enthalpic change ΔH for folding to the helical state, as well as the enthalpic change relative to the WT peptide $\Delta\Delta H$.

	WT	$i,j+4$ staple	$i,j+4$ Ala	$i,j+7$ staple	$i,j+7$ Ala		
RNase A	ΔH^p	-29.2 ± 0.6	-30.0 ± 0.5		-33.25 ± 0.06	-32.59 ± 0.09	
	$\Delta H(\text{con})$	-1.05 ± 0.05	-0.75 ± 0.08		-0.37 ± 0.02	-0.57 ± 0.07	
	$\Delta H(\text{hb})$	-11.4 ± 0.5	-11.9 ± 0.1		-14.03 ± 0.07	-13.9 ± 0.1	
	$\Delta H(\text{bbtor})$	-16.76 ± 0.09	-17.3 ± 0.3		-18.50 ± 0.09	-17.83 ± 0.09	
	$\Delta H(\text{sct})$	0.0 ± 0.3	-0.14 ± 0.09		-0.28 ± 0.04	-0.2 ± 0.1	
	$\Delta H(\text{linktor})$		0.1 ± 0.1		-0.07 ± 0.09		
	$\Delta\Delta H^p$		-0.8 ± 0.5		-4.0 ± 0.6	-3.4 ± 0.5	
	$\Delta\Delta H(\text{con})$		0.3 ± 0.1		0.69 ± 0.05	0.49 ± 0.06	
	$\Delta\Delta H(\text{hb})$		-0.5 ± 0.4		-2.7 ± 0.5	-2.6 ± 0.6	
	$\Delta\Delta H(\text{bbtor})$		-0.6 ± 0.4		-1.7 ± 0.2	-1.1 ± 0.2	
	$\Delta\Delta H(\text{sct})$		-0.1 ± 0.4		-0.3 ± 0.3	-0.2 ± 0.2	
	$\Delta\Delta H(\text{linktor})$		0.1 ± 0.1		-0.07 ± 0.09		
	BID BH3	ΔH^p	-47.0 ± 0.4	-48.3 ± 0.3	-49.2 ± 0.4	-47.5 ± 0.2	-46.5 ± 0.4
		$\Delta H(\text{con})$	-0.3 ± 0.1	-0.43 ± 0.07	-0.5 ± 0.1	-0.4 ± 0.1	-0.6 ± 0.1
$\Delta H(\text{hb})$		-24.3 ± 0.2	-24.85 ± 0.09	-24.9 ± 0.1	-23.20 ± 0.1	-22.7 ± 0.1	
$\Delta H(\text{bbtor})$		-22.6 ± 0.2	-23.3 ± 0.3	-23.6 ± 0.1	-24.0 ± 0.2	-23.9 ± 0.2	
$\Delta H(\text{sct})$		0.2 ± 0.2	0.2 ± 0.2	-0.2 ± 0.3	0.41 ± 0.07	0.7 ± 0.4	
$\Delta H(\text{linktor})$			-0.01 ± 0.02		-0.25 ± 0.05		
$\Delta\Delta H^p$			-1.4 ± 0.5	-2.2 ± 0.3	-0.5 ± 0.5	0.4 ± 0.5	
$\Delta\Delta H(\text{con})$			-0.2 ± 0.1	-0.19 ± 0.03	-0.16 ± 0.09	-0.36 ± 0.04	
$\Delta\Delta H(\text{hb})$			-0.6 ± 0.2	-0.6 ± 0.4	1.1 ± 0.1	1.6 ± 0.1	
$\Delta\Delta H(\text{bbtor})$			-0.6 ± 0.4	-1.0 ± 0.3	-1.4 ± 0.3	-1.3 ± 0.3	
$\Delta\Delta H(\text{sct})$			0.0 ± 0.2	-0.4 ± 0.3	0.2 ± 0.2	0.5 ± 0.5	
$\Delta\Delta H(\text{linktor})$			-0.01 ± 0.02		-0.25 ± 0.05		

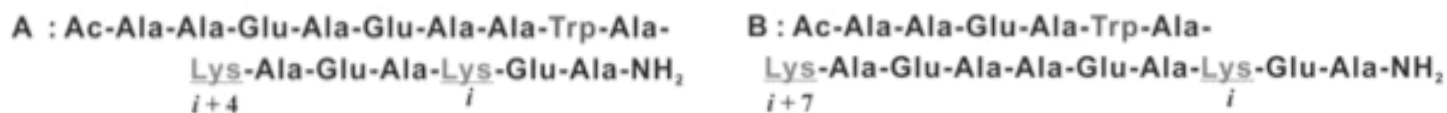
5 cross-linking with acetylenic cross-linking agent

M. Inouye et al. Chem Comm, 2004, 1280
 M. Inouye et al. Chem. Eur. J. 2008, 14, 857
 G. A Wooley et al. JACS. 2007, 129, 14154

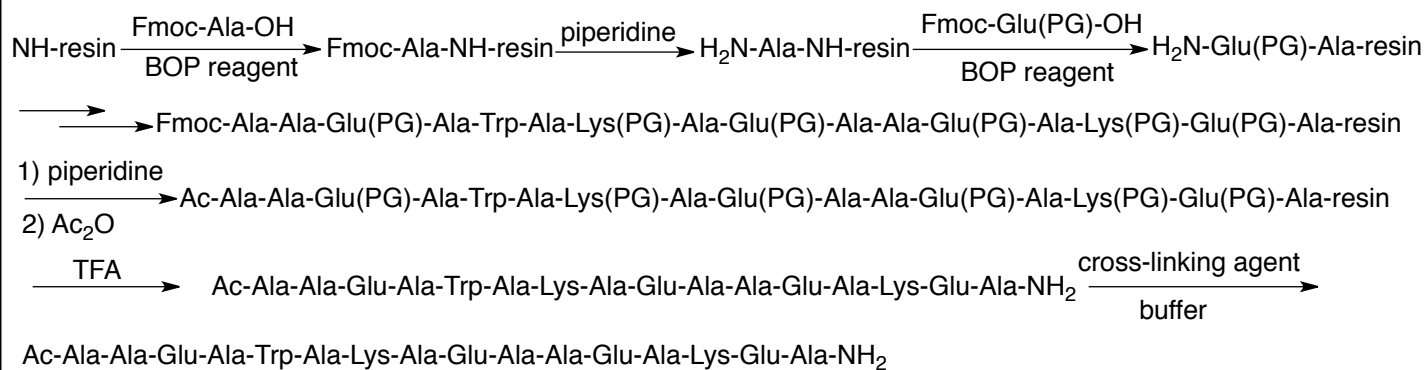
5-1 design



5-2 synthesis



solid phase peptide synthesis using the standard Fmoc chemistry



cross-linking agent

cross-linking yield

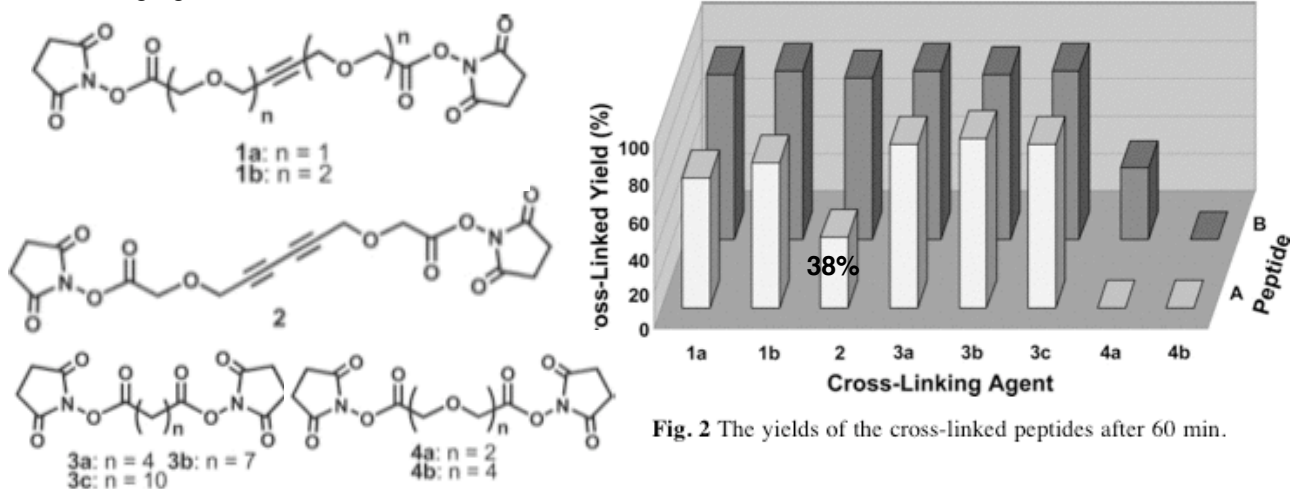
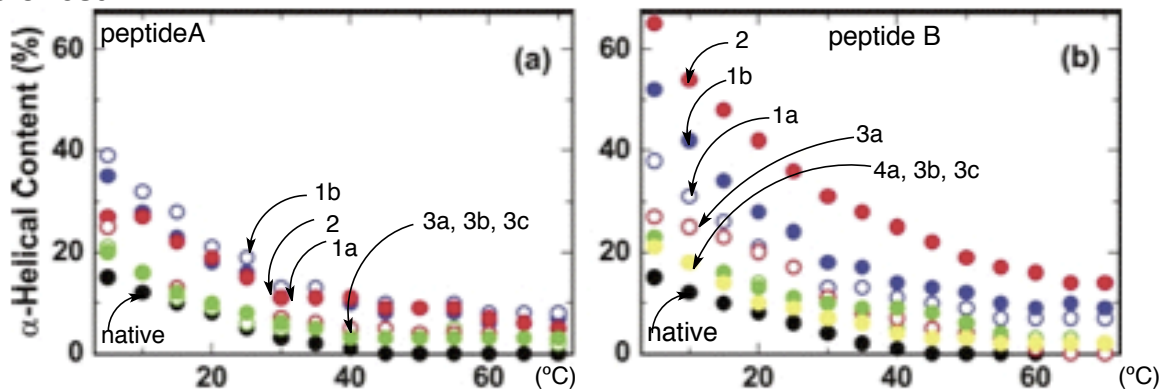


Fig. 2 The yields of the cross-linked peptides after 60 min.

5-3 result



peptide B (Lys→X); Ac-Ala-Ala-Glu-Ala-Trp-Ala-X-Ala-Glu-Ala-Ala-Glu-Ala-X-Glu-Ala-NH₂

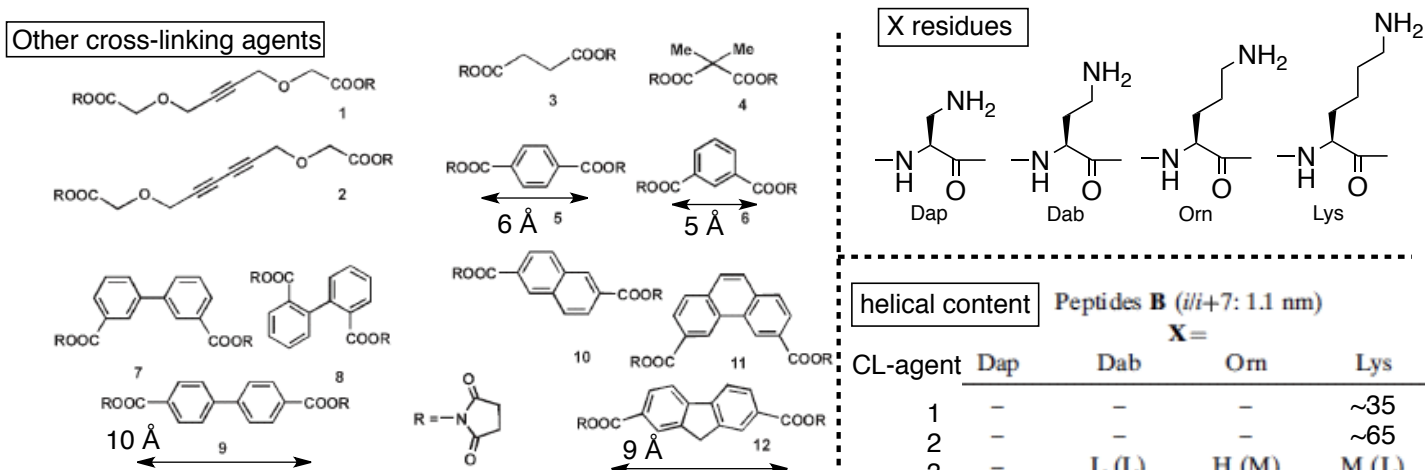
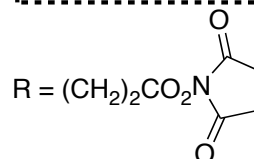
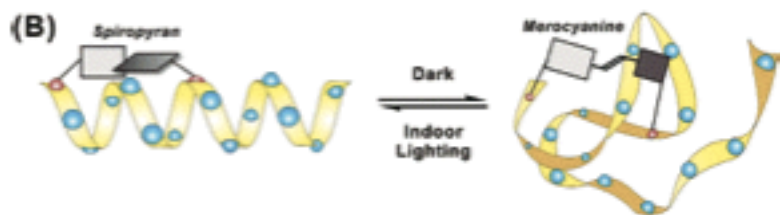
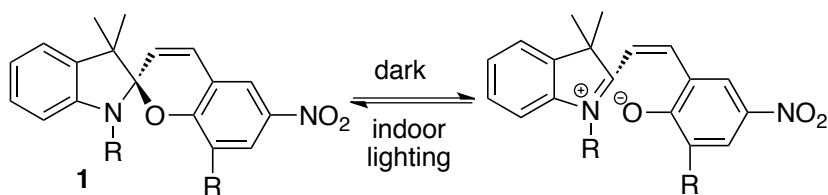


Figure 3. Cross-linking agents: acetylenic **1** and **2**, short alkylene-based **3** and **4**, benzene-based **5** and **6**, biphenylene-based **7-9**, and highly rigid **10-12**.

Reversible photoregulation of helical structure in short peptides (M. Inouye et al. Org. Lett. 2006, 8, 285)

Isomerization of the spiroopyran to the merocyanine



photochromic features of spiroopyran skeletons

- 1) only merocyanine forms strongly absorb visible light
- 2) merocyanines are more stable than the spiroopyrans in aqueous media under dark conditions
- 3) [merocyanine]/[spiroopyran] = 1.13 (dark) = 0.01 (light)

Scheme 1. Synthetic Route for **1**

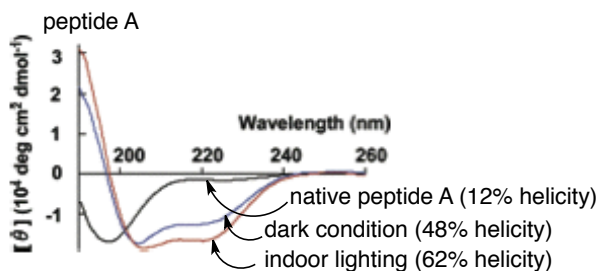
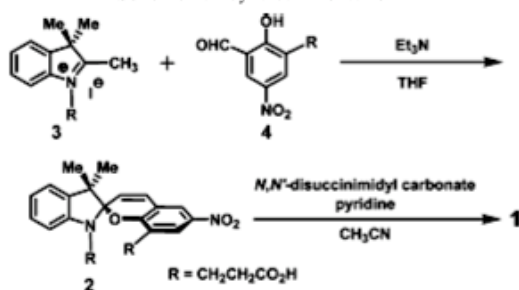


Figure 2. CD spectra of the native peptide A (black) and the cross-linked peptide A·1 under indoor lighting (red) and dark (blue) conditions in 100 mM phosphate buffer (pH 6.6) at 25 °C using a 1 mm cell.

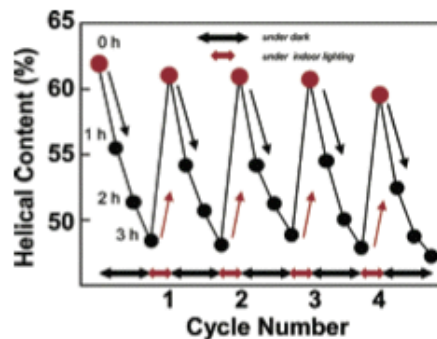


Figure 3. Switching behavior for the helical contents of the dark-adapted A·1 under the same conditions described in text. Red circles correspond to the helical contents after exposure of A·1 to indoor lighting for 20 min.