

Metal complex catalysis in Biological conditions

Literature Seminar
2013.12.2 (Mon.)
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Contents

1. Introduction

2. Redox catalyst

3. Pd catalyst

4. Others

5. Summary

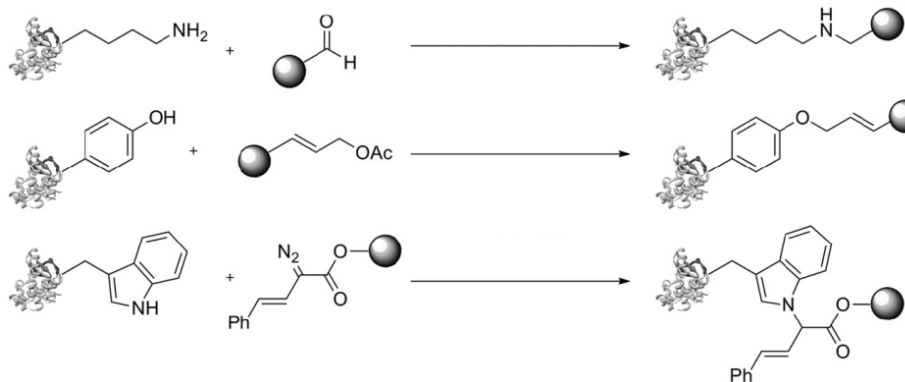
Biological condition

- buffer (pH7.4)
- protein (many functional groups)
- cell (surface or inside)

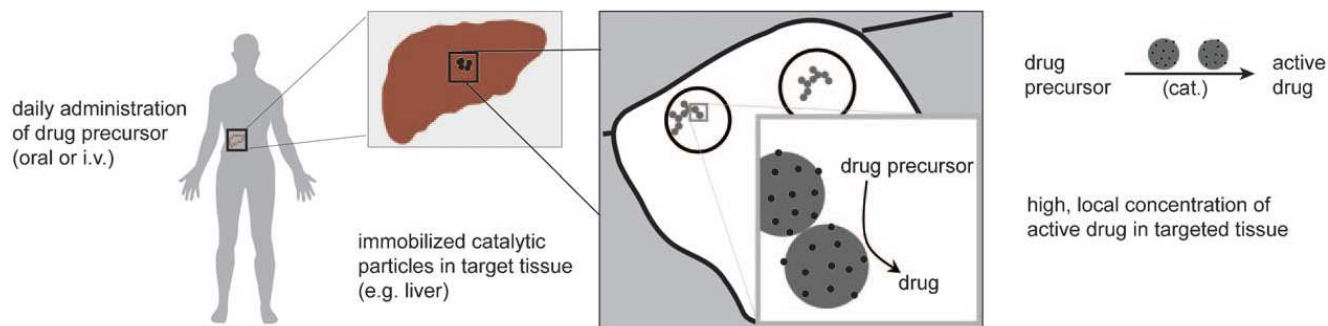
Good points

1. Introduction

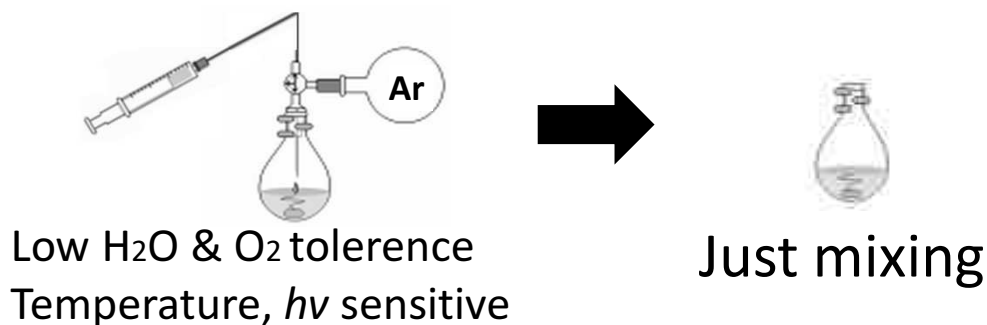
- Modification in Living cell



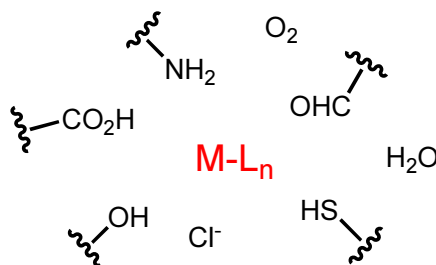
- Site specific prodrug activation



- (Eco-friendly & safe)

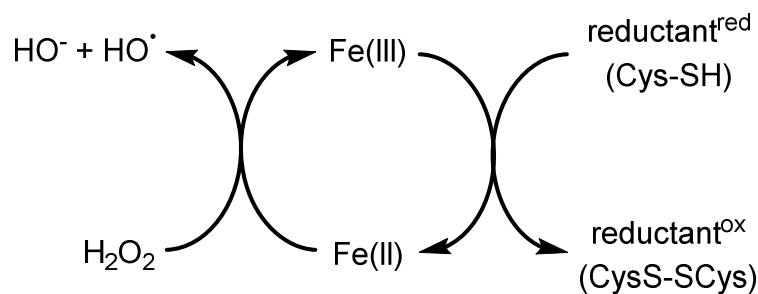


- Low functional group tolerance



- Toxicity

e.g.) Fenton reaction



- Low water solubility

$Pd(PPh_3)_4$, Hoveyda-Grubbs 2nd cat. etc...

J. A. Imlay *et al.*, *J. Bacteriol.* **2003**, 185, 1942-1950



Ligand must be

- protect active site
- water soluble

Contents

1. Introduction

2. Redox catalyst ■ ■ ■

R-SH oxidation

Reduction by HCO_2^-

Reduction by NADH

3. Pd catalyst

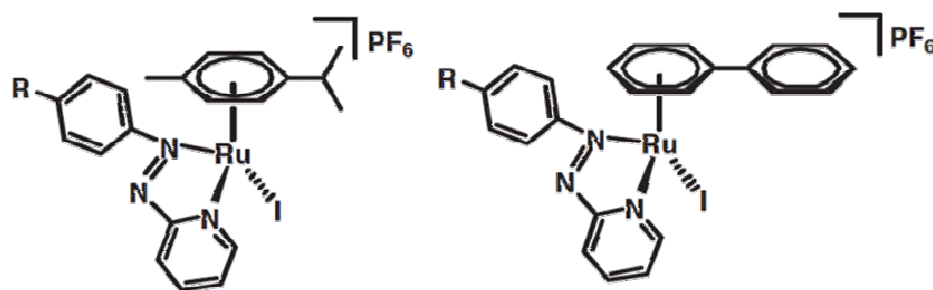
4. Others

5. Summary

2-1. R-SH oxidation (1)-1

2. Redox

Cytotoxic [Ru] complex oxidize GSH to GSSG

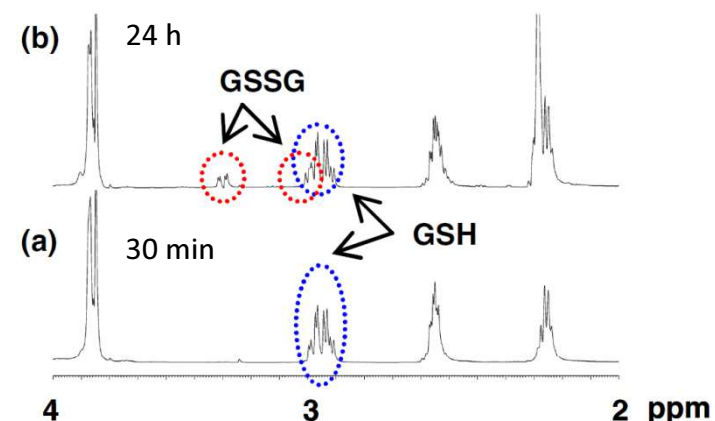


1: R = NMe₂
2: R = OH
3: R = H

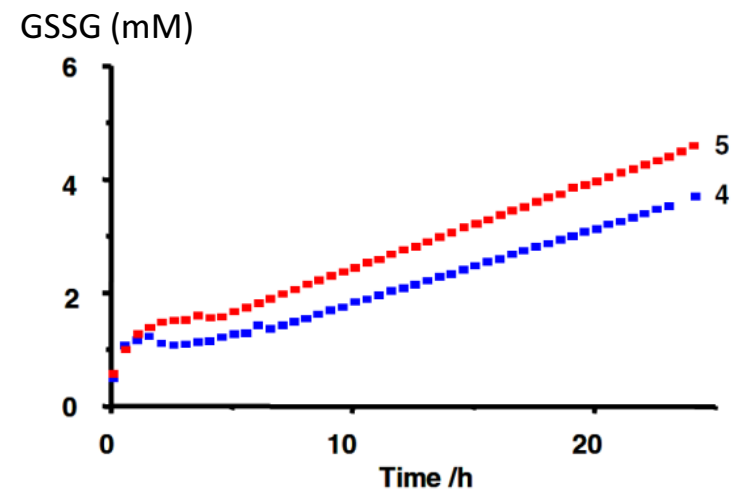
4: R = NMe₂
5: R = OH
6: R = H

Complex*	IC ₅₀ , μM		E _{red} , V
	A2780	A549	
1 (1-Cl)	4 (>100)	3 (>100)	-0.40, -1.00
4 (4-Cl)	3 (44)	2 (49)	-0.36
Azpy-NMe ₂	>100	14	-1.28
2 (2-Cl)	4 (58)	4 (>100)	-0.33, -0.77
5 (5-Cl)	5 (18)	6 (56)	-0.26, -0.72
Azpy-OH	>100	>100	nd [†]
3 (3-Cl)	>100 (>100)	>100 (>100)	-0.22, -0.74
6 (6-Cl)	39 (>100)	51 (>100)	-0.18, -0.67
Azpy	>100	>100	-1.31 (vs.SCE) [‡]

(A2780 ovarian and A549 lung cancer cell lines)

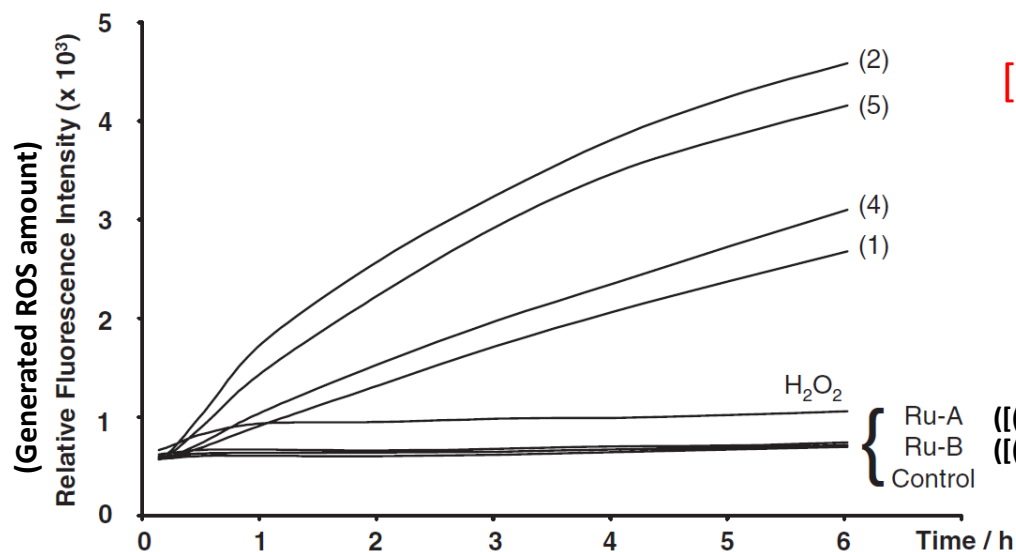


¹H NMR spectra of GSH after incubation (37°C) with 4 for (a) 30 min and (b) 24 h



Catalytic oxidation of GSH (10 mM) to GSSG by 4 (100 μM; blue) or 5 (100 μM; red) for 24 h, 37°C

2-1. R-SH oxidation (1)-2

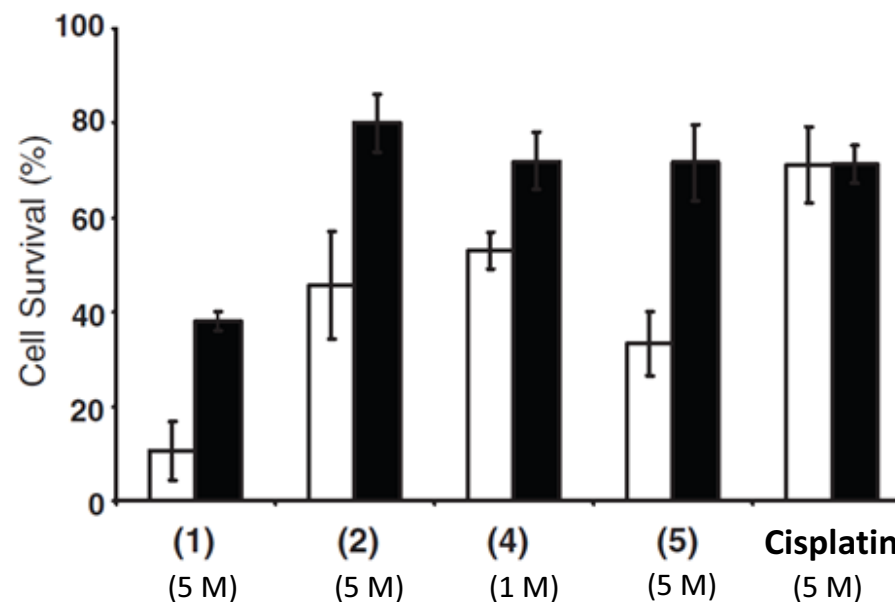


[Ru] increases cell ROS level

Incubation of A549 cancer cells with Ru complex (25 μ M)

Thiol cancels [Ru] cytotoxicity

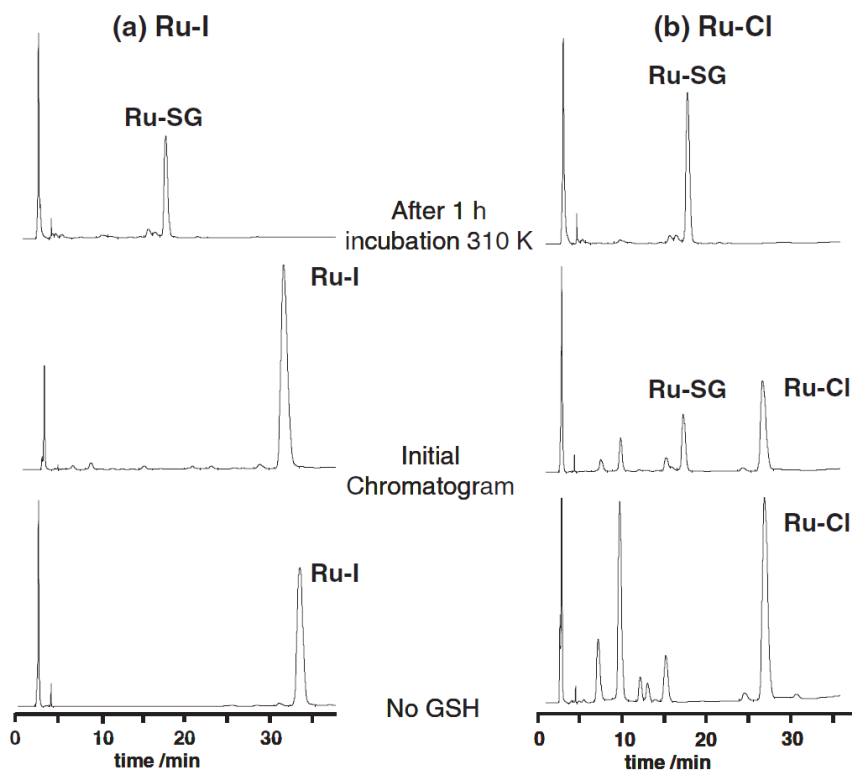
After 24 h exposure to ruthenium compounds,
 Lighter : 96 h recovery for A549 lung cancer cells
 Darker : same but pretreated to **increase intracellular thiol level**
 (retreated 5 mM *N*-Ac-Cys for 2 h)



2-1. R-SH oxidation (1)-3

2. Redox

Ru-X is replaced to Ru-SG

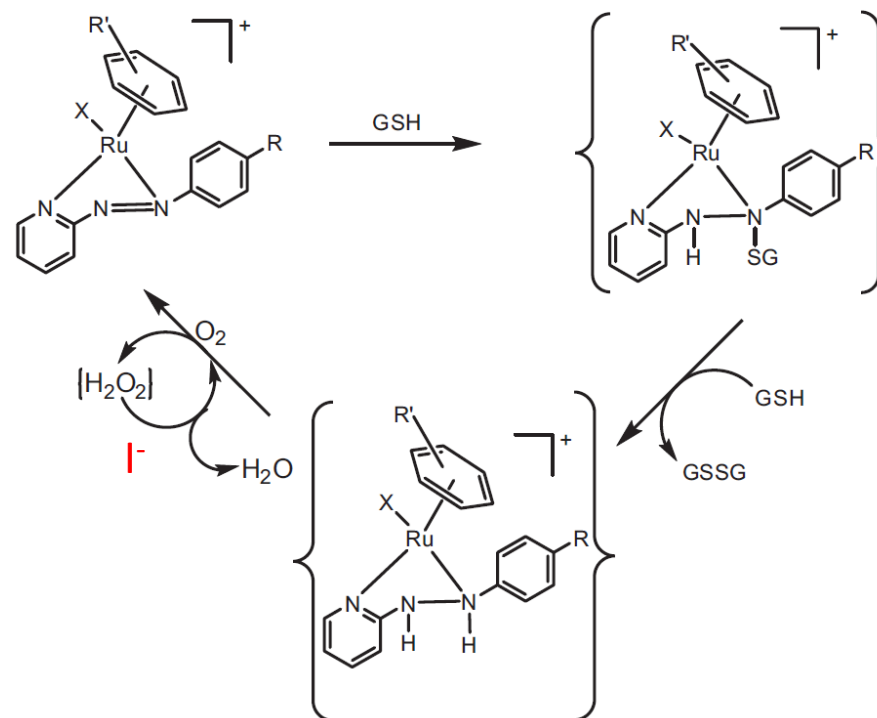


HPLC-analysis

Reaction of GSH (5 mM) with **4** (Ru-I) (a) and corresponding chlorido complex (50 M) (Ru-Cl) (b) (10 mM phosphate buffer (pH 7.9), 95% H₂O, 5% MeOH)

After 1 h incubation at 37°C, GS⁻ adduct is major peak and peaks of starting material have disappeared.

Proposed catalytic cycle



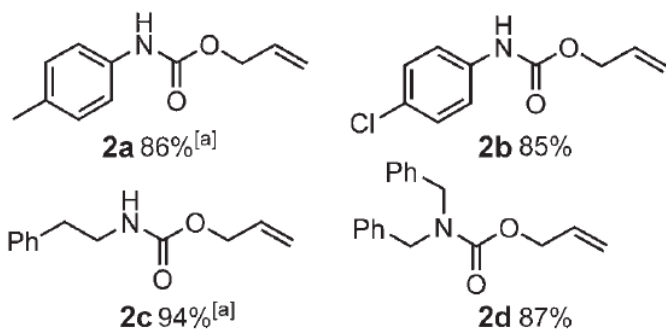
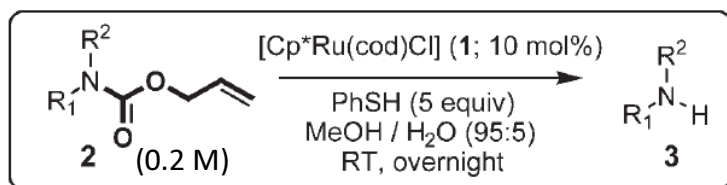
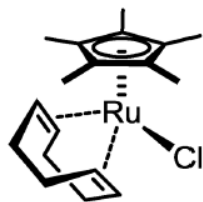
*X is initially I⁻, which is displaced GS⁻ during the early stage
 **I⁻-Catalyzed Decomposition of H₂O₂

J. C. Hansen, *J. Chem. Educ.*, 1996, 73, 728-732

2-1. R-SH oxidation (2)-1

2. Redox

Allylcarbamate cleavage by Ru(cod) complex



yield determined by GC-MS

Table 1: Catalytic cleavage of allylcarbamate **2a** to *p*-methylaniline with $[\text{Cp}^*\text{Ru}(\text{cod})\text{Cl}]$.^[a]

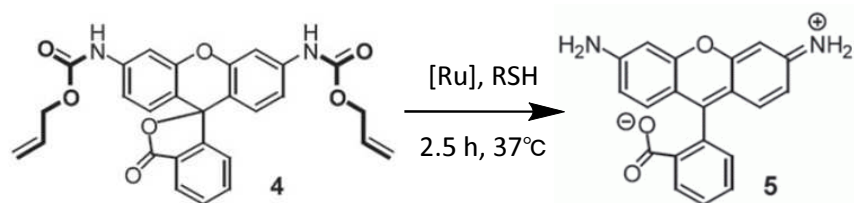
Entry	Thiol	Solvent	Atm.	<i>T</i>	Yield
1	PhSH	MeOH/H ₂ O (95:5)	air	RT	89%
2	PhSH	MeOH	air	RT	93%
3	PhSH	MeOH/H ₂ O (95:5)	argon	RT	96%
4	no thiols	MeOH/H ₂ O (95:5)	air	RT	0%
5	PhSH and PhCH ₂ CH ₂ SH	MeOH/H ₂ O (95:5)	air	RT	93%
6	PhCH ₂ CH ₂ SH	MeOH/H ₂ O (95:5)	air	RT	34%
7	PhCH ₂ CH ₂ SH	MeOH/H ₂ O (95:5)	air	37°C	67%

*Presence of excess thiol, no negative influence observed
yield determined by GC-MS
Obtain as byproduct PhS-SPh

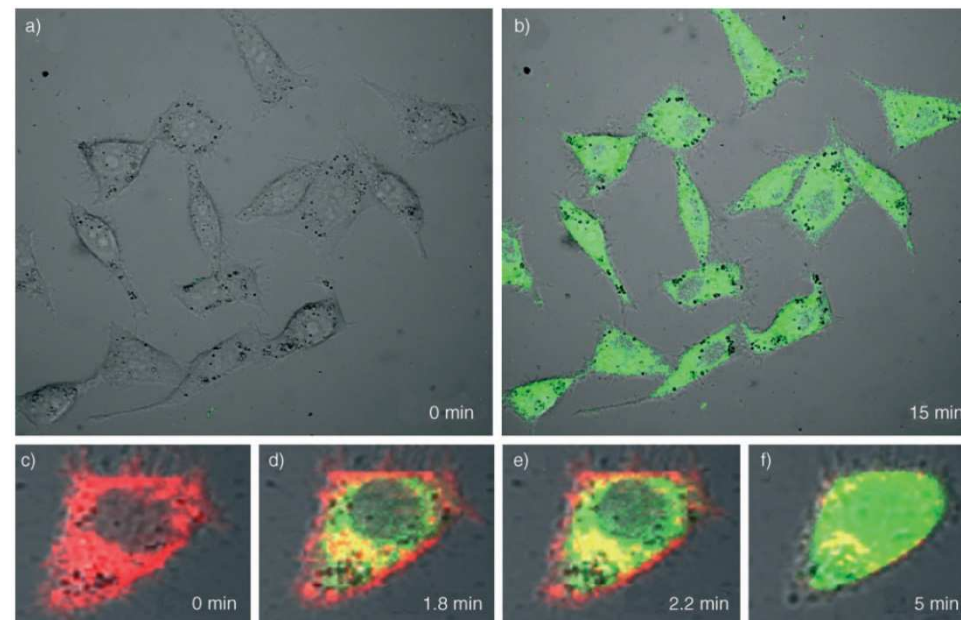
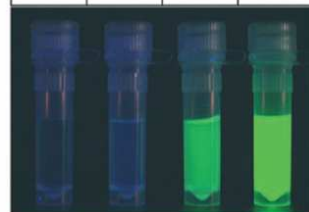
2-1. R-SH oxidation (2)-2

2. Redox

Reaction occurred in cell extract, HeLa cell and *E. coli*



Entry	1	2	3	4
Carbamate 4	+	+	+	+
Cell Extract + GSH	-	+	+	+
[Cp* <i>Ru</i> (cod)Cl]	-	-	+	+
PhSH	-	-	-	+
Yields of 5	0%	0%	6%	80%



- Entry 1: **4** (0.5 mM) in DMSO/H₂O (1:1)
 Entry 2: **4** (0.5 mM) in DMSO/cell extract (1:1), and GSH (3.5 mM), pH 7.0
 Entry 3: Same as entry 2, but with [Cp**Ru*(cod)Cl] (100 μM)
 Entry 4: Same as entry 2, but with [Cp**Ru*(cod)Cl] (100 μM) and PhSH (3.5 mM)
 *Yields determined by fluorescence intensity

Fluorescence imaging of HeLa cells

a) and b): preincubated with **4** (100 μM) for 30 min, washed with PBS buffer, then treated with [Cp**Ru*(cod)Cl] (20 μM) and PhSH (500 μM)

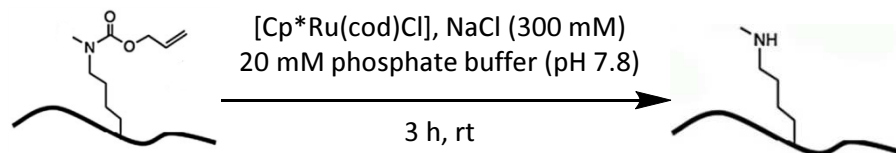
- a) right after this addition
 b) after 15 min

c)–f): preincubated with **4** (100 μM) for 30 min and at same time with membrane carbocyanine dye DiIC18(5). After washing with PBS buffer, cells were treated with [Cp**Ru*(cod)Cl] (40 μM) and PhSH (100 μM)

- c) right after this addition
 d)–f) after the indicated times.

No cytotoxicity observed (no figure)

E. Meggers et al. Angew. Chem., Int. Ed. **2006**, *45*, 5645-5648



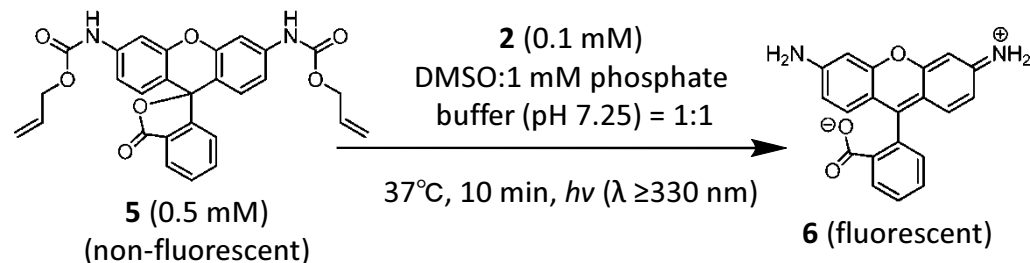
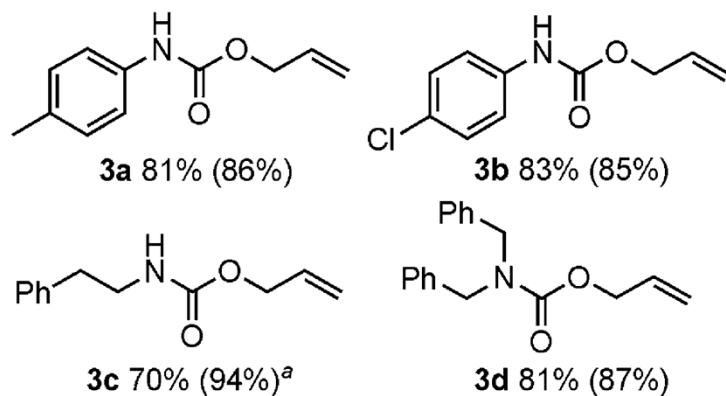
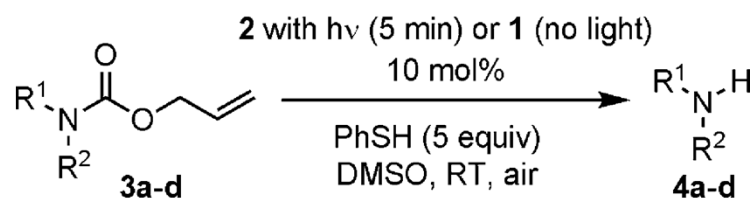
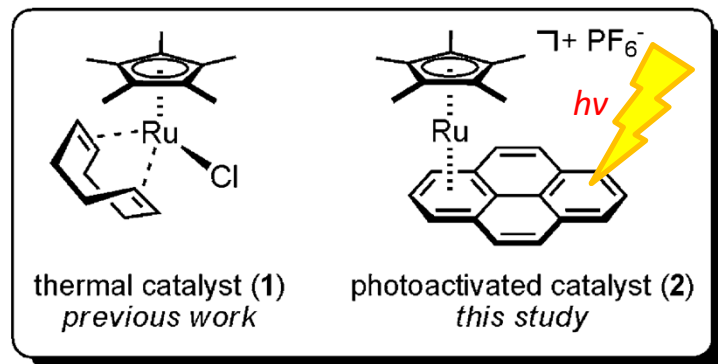
Histone H2B-Lys27(N-alloc)Lys in *E. coli*

P. G. Schultz *et al.*, *Chem. Commun.*, **2010**, *46*, 5506-5508

2-1. R-SH oxidation (3)-1

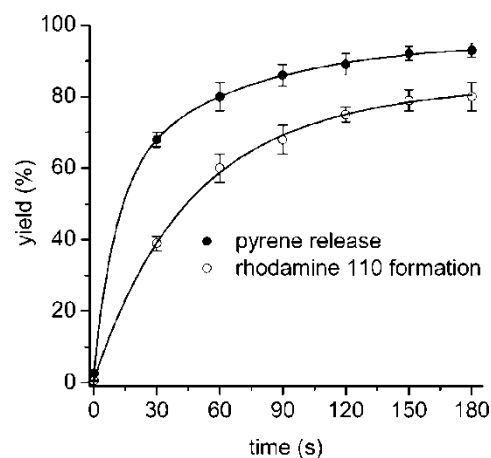
2. Redox

Photoactivatable catalyst



entry	amt of cat. (mol %)	thiol	yield (%) ^b
1	none	ME	0.5 ± 0.1
2	20	none	1.2 ± 0.3
3	20	ME	13 ± 2
4	20	Cys	14 ± 2
5	20	PhSH	93 ± 4
6	20	ME + PhSH	90 ± 3

β -mercaptoethanol (ME), PhSH, and cysteine (Cys) were all 5 mM
Yields determined by fluorescence intensity
(rhodamine 110: λ_{ex} 488 nm, λ_{em} 520 nm)



Reaction dynamics

2 (0.1 mM), 5 (0.5 mM), β -ME (5 mM), PhSH (5 mM), DMSO/H₂O (1/1), $h\nu$ ($\lambda \geq 330$ nm)
Yields determined by fluorescence intensity
(pyrene, λ_{ex} 319 nm, λ_{em} 390 nm;
rhodamine 110, λ_{ex} 488 nm, λ_{em} 520 nm)

2-1. R-SH oxidation (3)-2

2. Redox

Inside HeLa cells

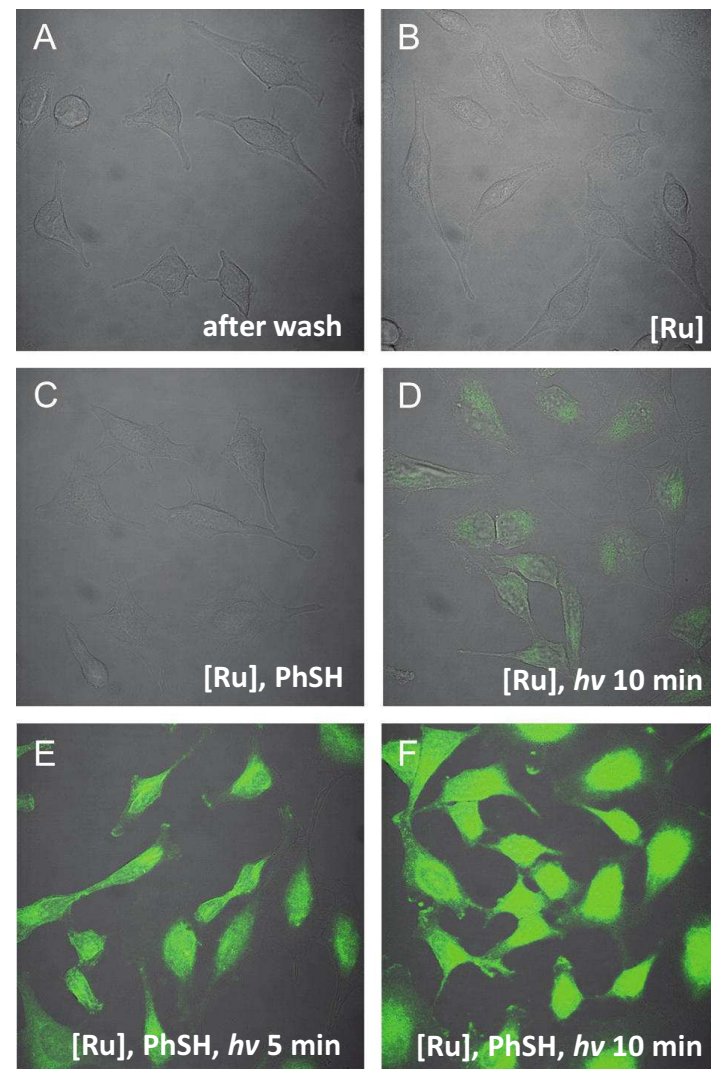
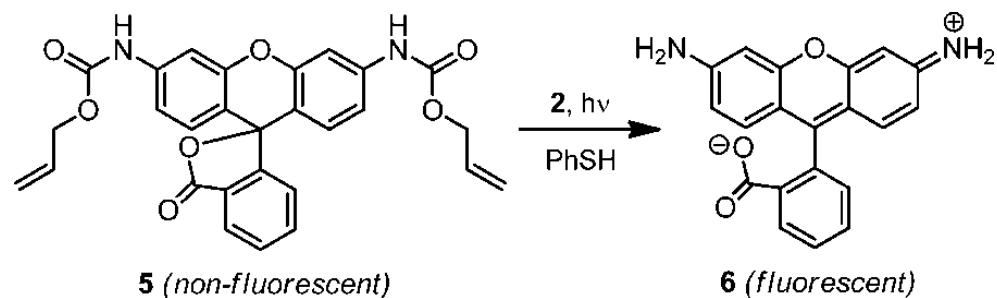
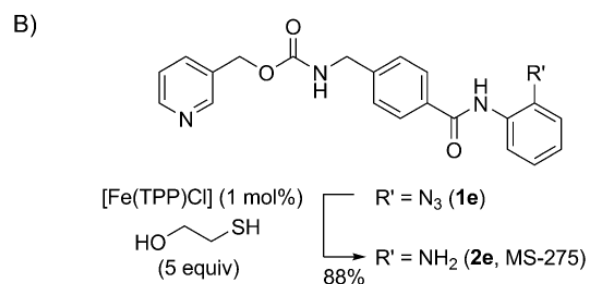
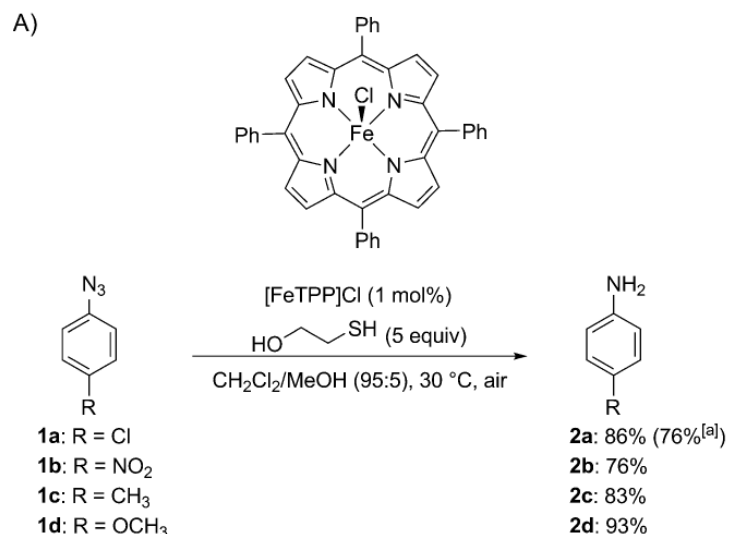


Figure 2. Confocal fluorescence imaging of $[\text{Cp}^*\text{Ru}(\text{pyrene})]\text{PF}_6^-$ -induced uncaging of the bisallylcarbamate-protected rhodamine 110 (**5**) inside HeLa cells. HeLa cells were incubated with caged rhodamine **5** ($100 \mu\text{M}$) for 25 min and then washed with PBS buffer. (A) after the washing step; (B) after the addition of ruthenium complex **2** ($20 \mu\text{M}$) and a 10 min incubation time; (C) after the addition of ruthenium complex **2** ($20 \mu\text{M}$) and thiophenol (1 mM) and a 10 min incubation time without photolysis; (D) ruthenium complex **2** ($20 \mu\text{M}$) added and photolyzed for 10 min with $\lambda \geq 330 \text{ nm}$; (E) ruthenium complex **2** ($20 \mu\text{M}$) and thiophenol (1 mM) added and photolyzed for 5 min with $\lambda \geq 330 \text{ nm}$; (F) ruthenium complex **2** ($20 \mu\text{M}$) and thiophenol (1 mM) added and photolyzed for 10 min with $\lambda \geq 330 \text{ nm}$.

2-1. R-SH oxidation (4)

Ar-N₃ reduction by FeTPP and R-SH



Scheme 1. [Fe(TPP)Cl]-catalyzed reduction of aromatic azides to amines with thiols. A) Reduction of simple aromatic azides. B) [Fe(TPP)Cl]-catalyzed formation of the anticancer drug candidate MS-275. [a] Only 0.05 mol% [Fe(TPP)Cl] used in the reaction.

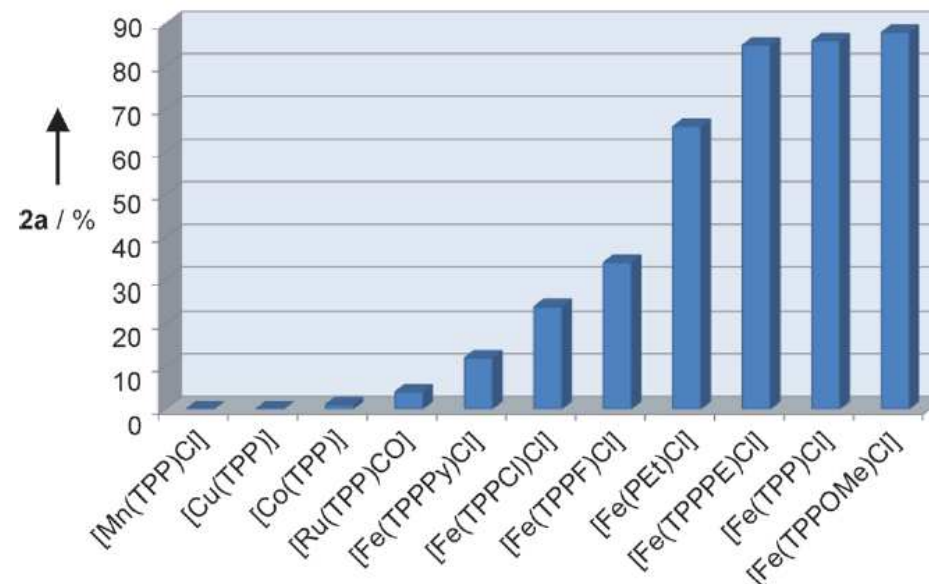


Figure 1. Comparison of different metalloporphyrins for the catalytic reduction of aromatic azides with thiols. Shown are isolated yields of **2a**. TPPPy: 5,10,15,20-tetra(*N*-methyl-4-pyridyl)porphine, TPPCl: 5,10,15,20-tetra(*o*-dichlorophenyl)-21*H*,23*H*-porphine, TPPE: 5,10,15,20-tetraphenyl-21*H*,23*H*-porphine modified with ethylene glycol chains, TPPF: 5,10,15,20-tetra(pentafluorophenyl)-21*H*,23*H*-porphine, PEt: 2,3,7,8,12,13,17,18-octaethyl-21*H*,23*H*-porphine, TPPOMe: 5,10,15,20-tetra(4-methoxyphenyl)-21*H*,23*H*-porphine. Reaction conditions: azide **1a** (1 μ M), β -mercaptoethanol (5 equiv), and metalloporphyrin (1 mol%) were reacted in CH₂Cl₂/MeOH (95:5) for 30 min at 30 °C. See the Supporting Information for more details.

2-1. R-SH oxidation (4)

2. Redox

Reductant scope

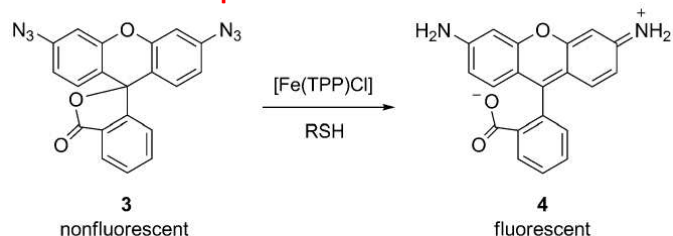


Table 1. Influence of biorelevant reaction conditions on the catalytic reduction of rhodamine bisazide **3** by [Fe(TPP)Cl].^[a]

	[Fe(TPP)Cl] [mol%]	Thiol	Ascorbate [mM]	Yield [%] ^[b]
1	5	–	–	0.2 ± 0.1
2	5	Cys (5 mM)	–	13 ± 2
3	5	Cys (5 mM)	10	40 ± 3
4	5	Cys (15 mM)	10	77 ± 5
5	5	PhSH (5 mM)	10	85 ± 5 ^[d]
6	5	Cys (5 mM)	10 ^[c]	41 ± 2
7	–	Cys (5 mM)	10	2.3 ± 0.5

[a] Bisazide **3** (0.5 mM) was treated with the indicated amounts of [Fe(TPP)Cl] in DMSO/phosphate buffer (1:1, 20 mM, pH 7.25) at 37 °C for 1 h.

[b] Standard deviations from three independent experiments. [c] Reaction performed in DMEM instead of phosphate buffer. [d] In the absence of [Fe(TPP)Cl] the yield is only 1%.

Ar-N₃ were reduced in cell

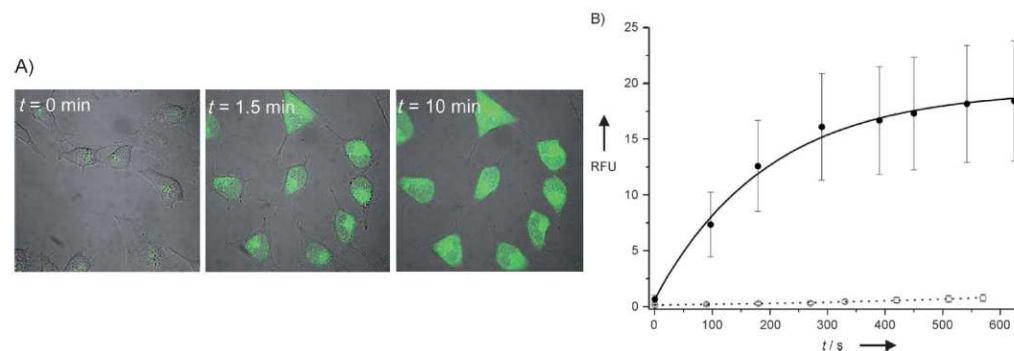


Figure 2. [Fe(TPP)Cl]-induced reduction of bisazide **3** to fluorescent rhodamine 110 inside HeLa cells. A) Superimposed phase contrast and confocal fluorescence images. HeLa cells were pre-incubated with bisazide **3** (100 μM) for 25 min, washed with PBS, and subsequently treated with [Fe(TPP)Cl] (10 μM). See the Supporting Information for more details. B) Quantified time-dependent increase of fluorescence emission within the cellular cytosol of HeLa cells upon [Fe(TPP)Cl] addition (\bullet) and the corresponding time-dependent fluorescence in the absence of catalyst (\circ). The image analysis was performed on regions of interest (ROIs) for eight individual HeLa cells with the software ImageJ. The fluorescence evolution was analyzed for eight different HeLa cells. The data for the catalytic reaction were fitted to an exponential equation with the exponent being the rate constant [s^{-1}] of the intensity increase.

Fluorescence were observed *in vivo*, due to reductive metabolism of Ar-N₃

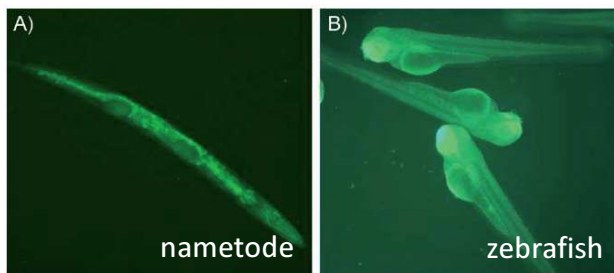


Figure 3. Activation of bisazide **3** *in vivo*. A) Adult *C. elegans* nematode incubated with bisazide **3** (25 μM) in M9 buffer with 1% DMSO for 20 min, before being pelleted, washed with M9 buffer (1 mL), concentrated to 100 μL by decanting, placed on a slide, and treated with one drop of 10% azide, followed by fluorescence imaging. B) 2.5-day-old transparent zebrafish embryos incubated with bisazide **3** (100 μM) for 30 min followed by fluorescence imaging.

2-2. Reduction by HCO₂H (1)

2. Redox

Reductive alkylation by [Ir-H] and HCO₂⁻

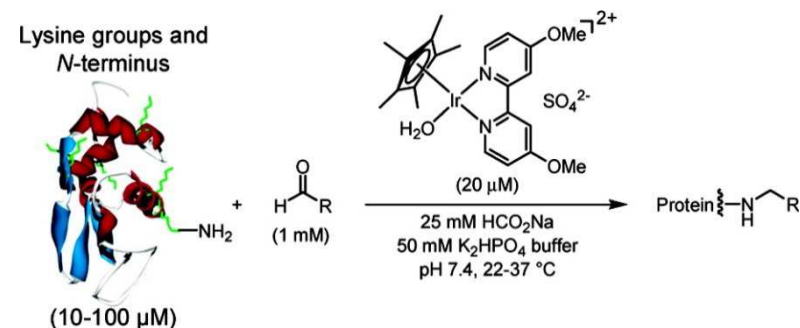
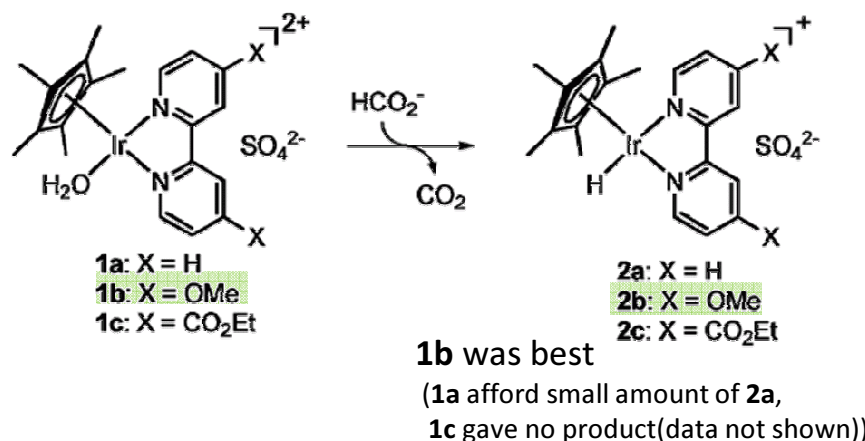


Table 1. Reductive Alkylation of Proteins Using 4a and 1b^a

entry	protein	[1b] (μM)	T (°C)	Unmod (%)	+1 (%)	+2 (%)	+3 (%)	+4 (%)
1	3a (100 μM)	20	22	6	21	37	25	11
2	3a (100 μM)	10	37	20	33	29	16	2
3	3a (100 μM)	5	37	48	38	31	1	0
4	3a (100 μM)	2	37	84	16	0	0	0
5	3a (10 μM)	20	22	15	32	34	19	trace
6	3b (100 μM)	20	22	11	19	27	22	12 ^b
7	3c (100 μM)	20	37	24	47	27	2	0
8	3d (100 μM)	20	37	85	15	0	0	0
9	3e (100 μM)	20	37	59	41	0	0	0
10	3f (100 μM)	20	37	70	30	trace	0	0

^a Additional conditions: 1 mM 4a, 25 mM HCO₂Na, 50 mM K₂HPO₄, pH 7.4, 18–22 h. Protein substrates: lysozyme (3a), cytochrome *c* (3b), α-chymotrypsinogen A (3c), h. h. myoglobin (3d), ribonuclease A (3e), and bacteriophage MS2 (3f). Product distributions were determined from ESI-MS analyses. ^b 3b also gave small amounts of +5 (5%) and +6 (3%).

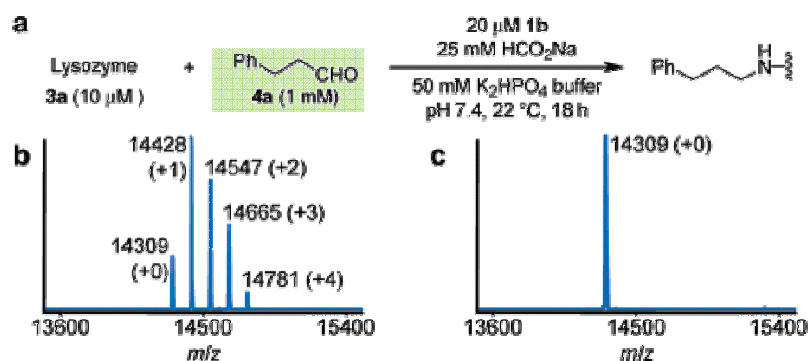

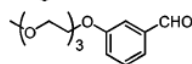
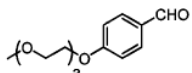
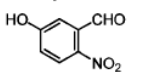

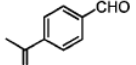
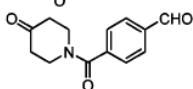


Figure 1. Modification of lysozyme using reductive alkylation. Under the reaction conditions summarized in (a), a distribution of alkylated products results (b). (c) Control experiments lacking catalyst yielded no reaction products. Spectra shown are reconstructed from charge ladders obtained using ESI-MS analysis.

2-2. Reduction by HCO₂H (1)

Aldehyde substrate scope

Table 2. Modification of Lysozyme (3a) By Reductive Alkylation^a

Aldehyde/Ketone	Unmod (%)	+1 (%)	+2 (%)	+3 (%)	+4 (%)
4b CH ₃ CH ₂ CHO	72	28	0	0	0
4c 	41	43	16	0	0
4d 	6	26	40	23	5
4e 	60	34	6	0	0
4f 	14	41	36	9	0
4g 	100	0	0	0	0
4h 	18	46	29	7	0
4i 	33	52	15	0	0

^a Conditions: 100 μM lysozyme, 1 mM **4b–i**, 20 μM **1b**, 25 mM sodium formate, 50 mM K₂HPO₄, pH 7.4, 22 °C, 22 h. Product distributions were determined using ESI-MS analysis.

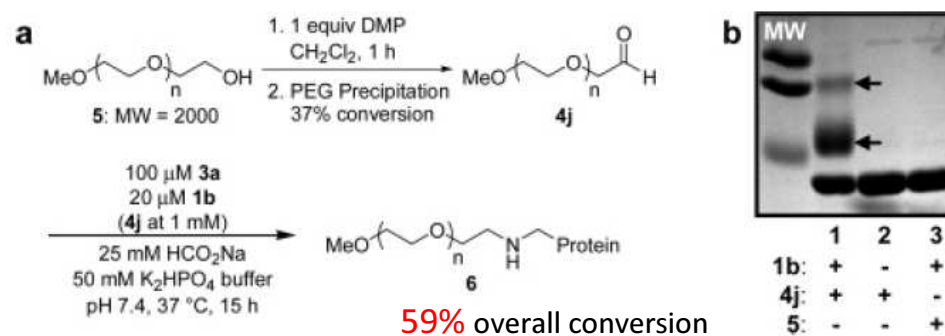
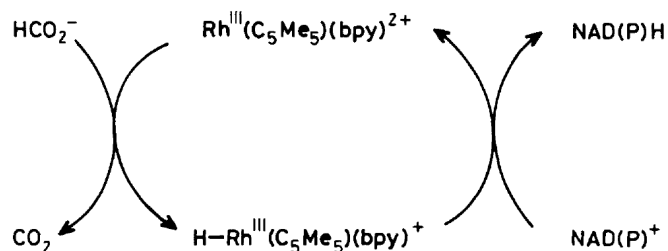


Figure 2. Conjugation of PEG to lysozyme using a simple two-step procedure. (a) Commercially available PEG alcohols can be oxidized to afford aldehydes before conjugation to proteins via reductive alkylation. (b) SDS-PAGE analysis of **6** shows the formation of singly and doubly alkylated conjugates. Control reactions run in the absence of catalyst (lane 2) or aldehyde (lane 3) afforded no conjugates. MW ladder: 25, 20, and 15 kD (from top). Lysozyme MW = 14.3 kD.

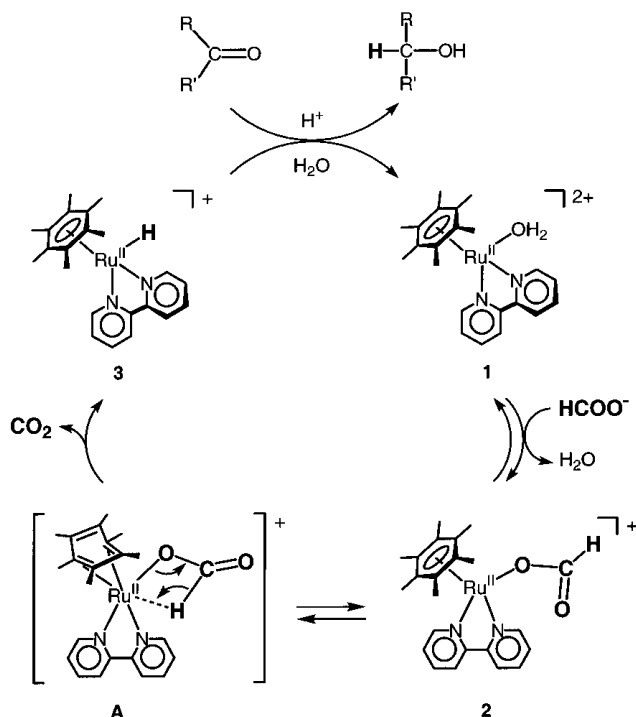
2-2. Reduction by HCO₂H (2)-0

2. Redox

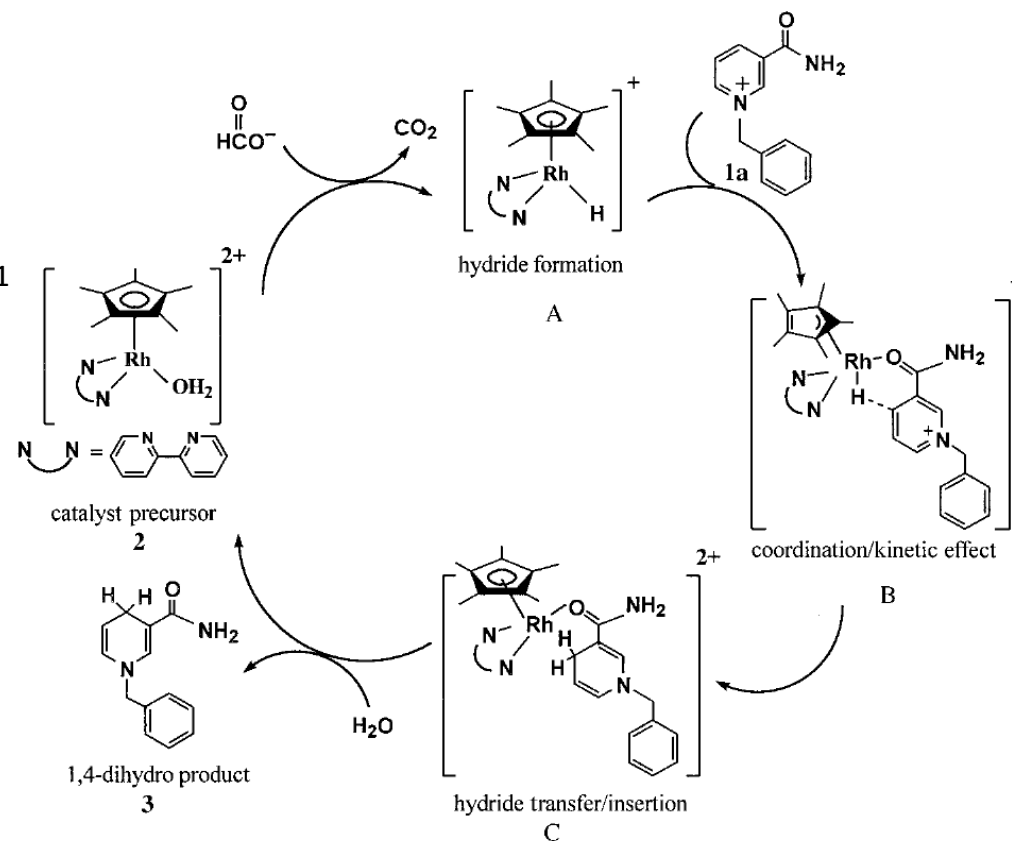
Precedents



E. Steckhan *et al.* *J. Chem. Soc., Chem. Commun.* **1988**, 1150-1151



Y. Watanabe *et al.* *Organometallics*, **2002**, 21, 2964-2969



R. H. Fish *et al.* *Angew. Chem. Int. Ed.* **1990**, 38, 1429-1432

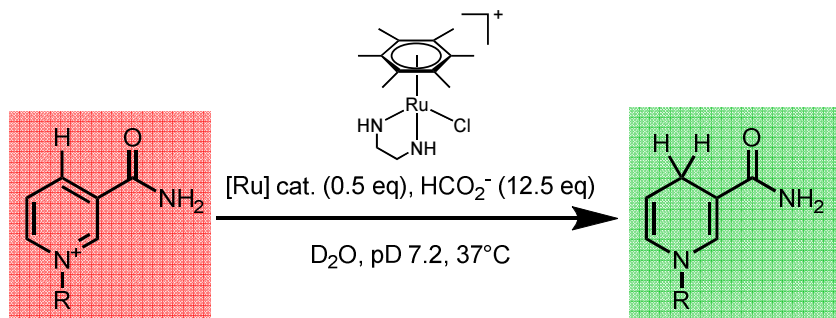
R. H. Fish *et al.* *Inorg. Chem.* **2001**, 40, 6705-6716

*Ir, Ru complex also reduce NAD⁺ to NAD catalytically

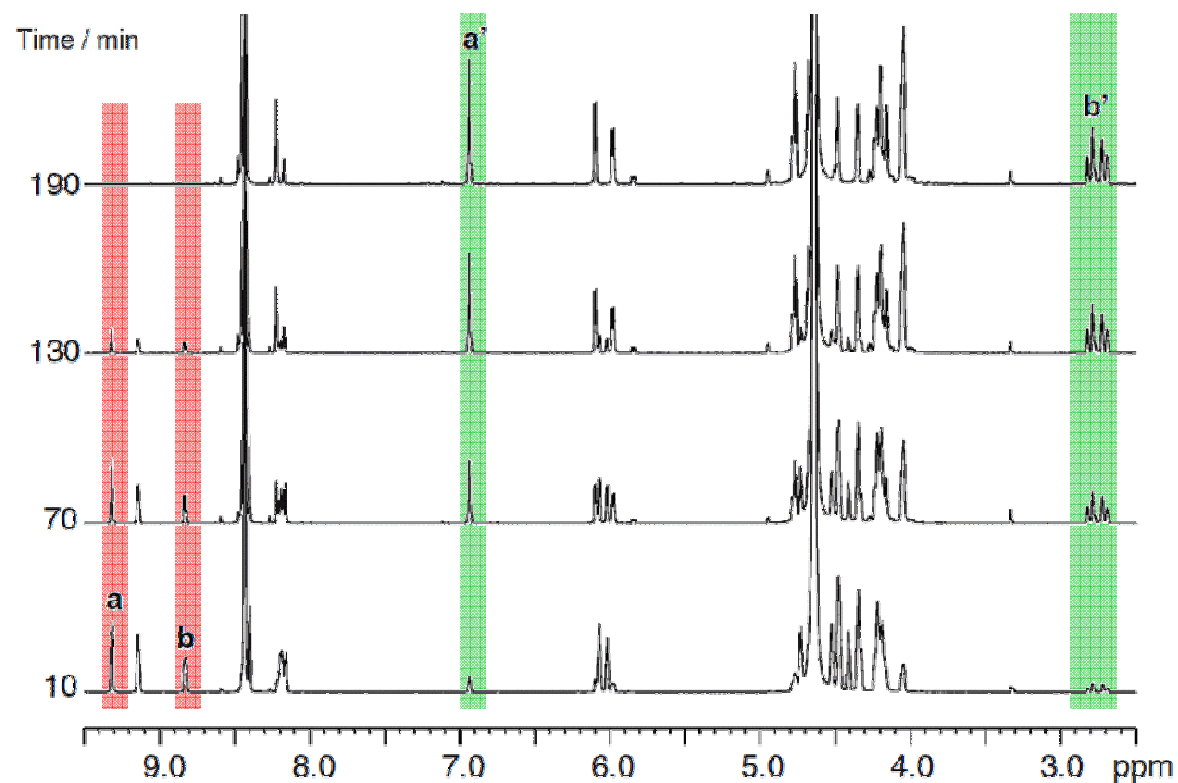
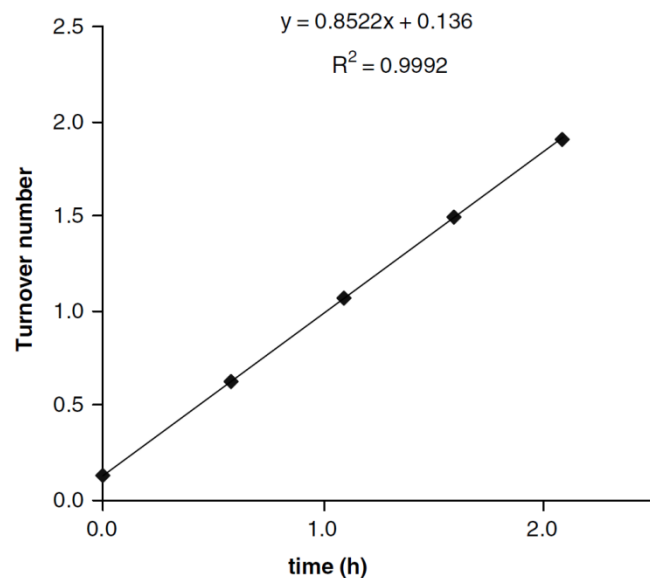
P. Stepnicka *et al.* *Eur. J. Inorg. Chem.* **2007**, 4736-4742

2-2. Reduction by HCO₂H (2)-1

2. Redox



[Ru] complex reduces NAD⁺

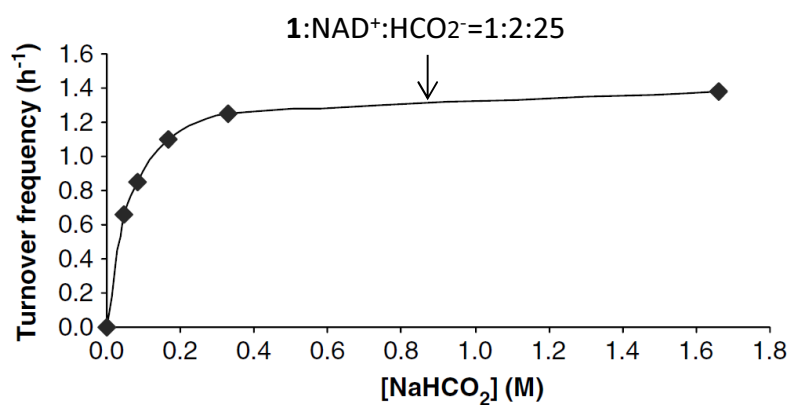
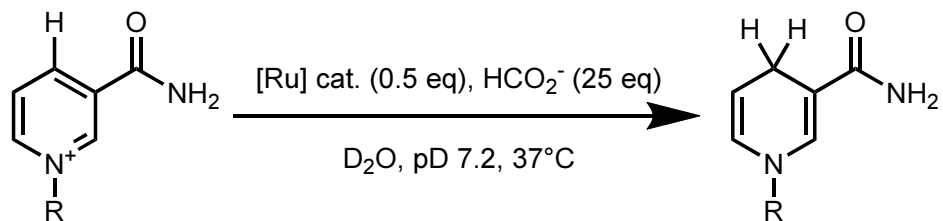


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2-2. Reduction by HCO₂H (2)-2

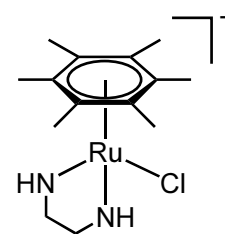
2. Redox

Ligand effects

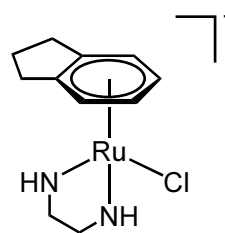


A549 (lung cancer cells) were tolerant to 2.5 mM HCO₂⁻ (highest concentration in this test, data not shown)

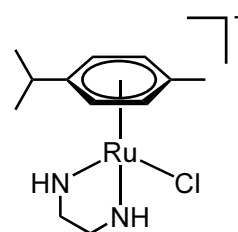
Turnover frequency (h⁻¹)



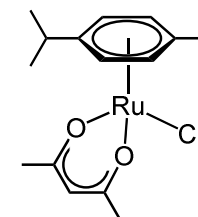
0.85



0.16



0.18

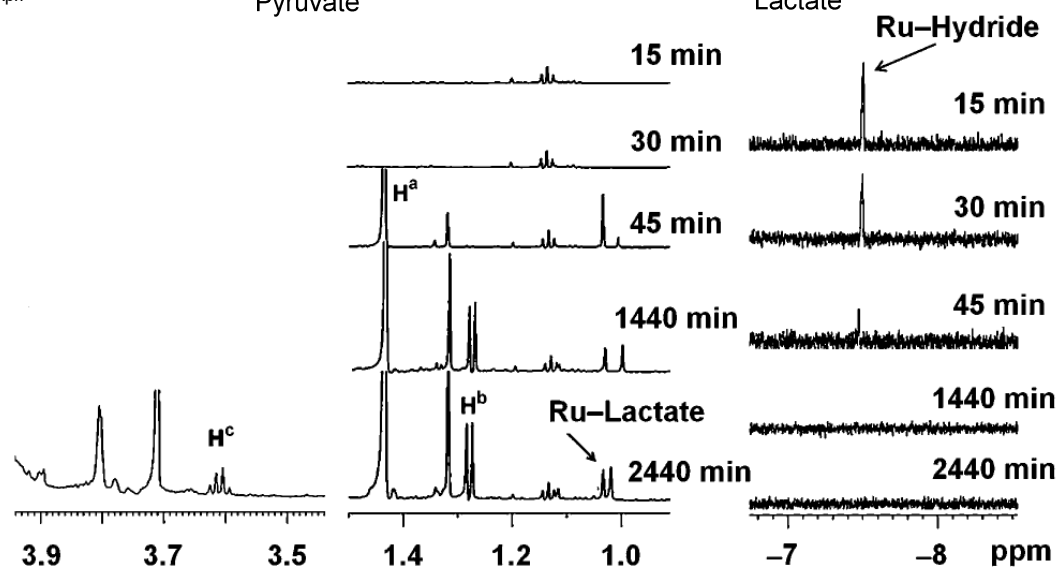
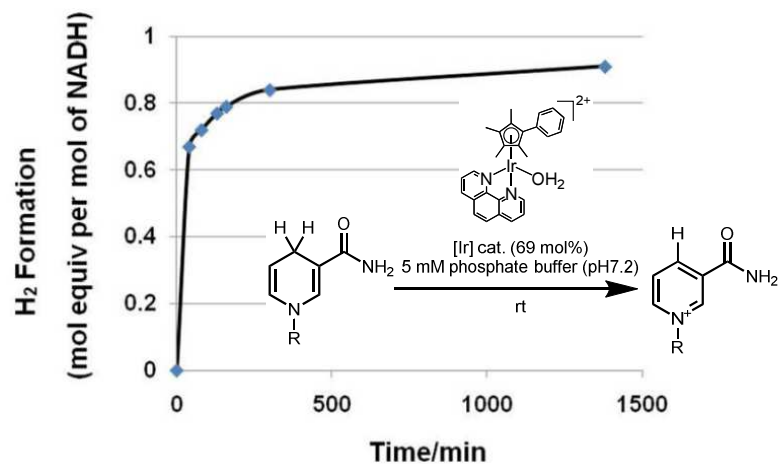
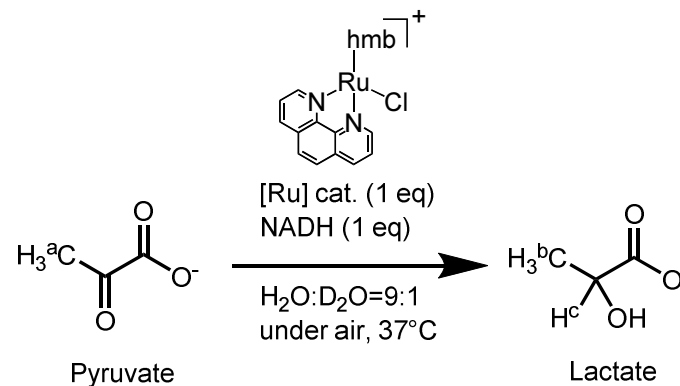
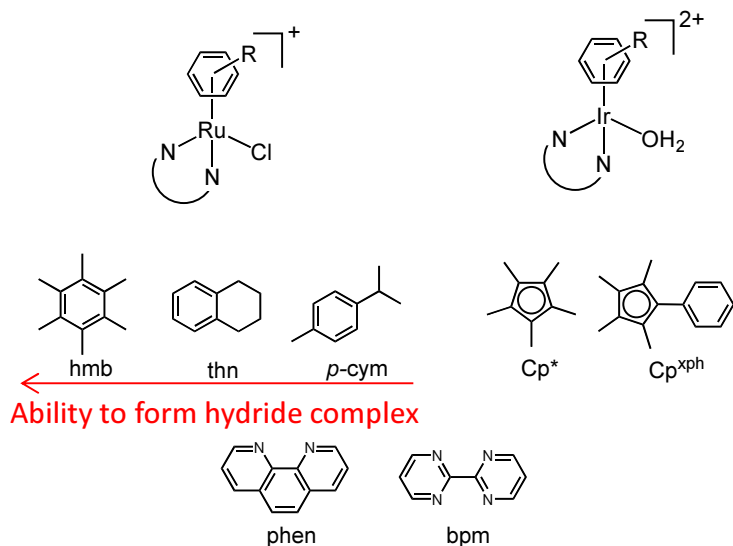


0.05

2-3. Reduction by NADH (1)-1

2. Redox

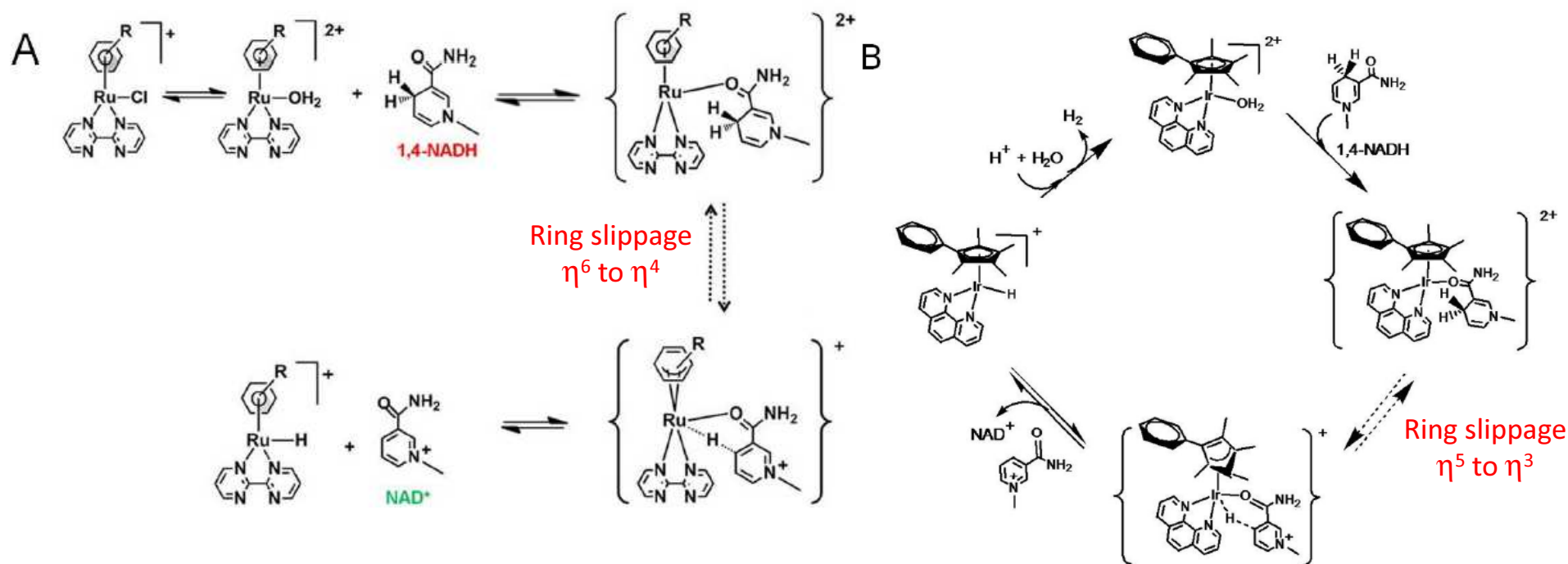
[Ru-H] (also [Ir-H]) reduces pyruvate



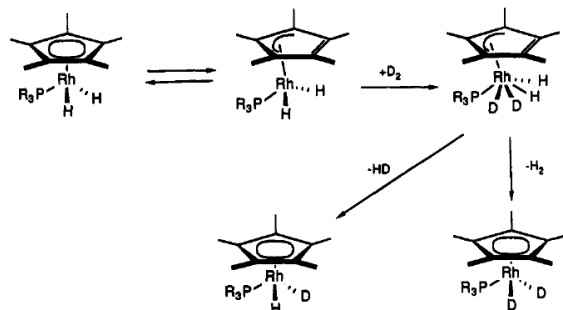
2-3. Reduction by NADH (1)-2

2. Redox

Proposed mechanism of hydride migration



Ring Slip Reaction of Cp*Rh complex



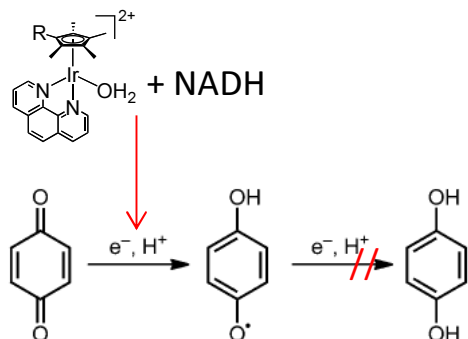
A. D. Selmezy et al. *Organometallics* **1991**, *10*, 1577-1586.

Figure S8. Proposed mechanism for transfer of hydride from 1,4-NADH to (A) Ru^{II} complexes **1**, **2**, and **4**, and (B) Ir^{III} complex (**7**).

2-3. Reduction by NADH (2)-1

2. Redox

[Ir-H] reduces quinones to...?



Scheme 1. One- and two-electron reduction of a quinone (e.g. benzoquinone) leading to semiquinone and hydroquinone, respectively.

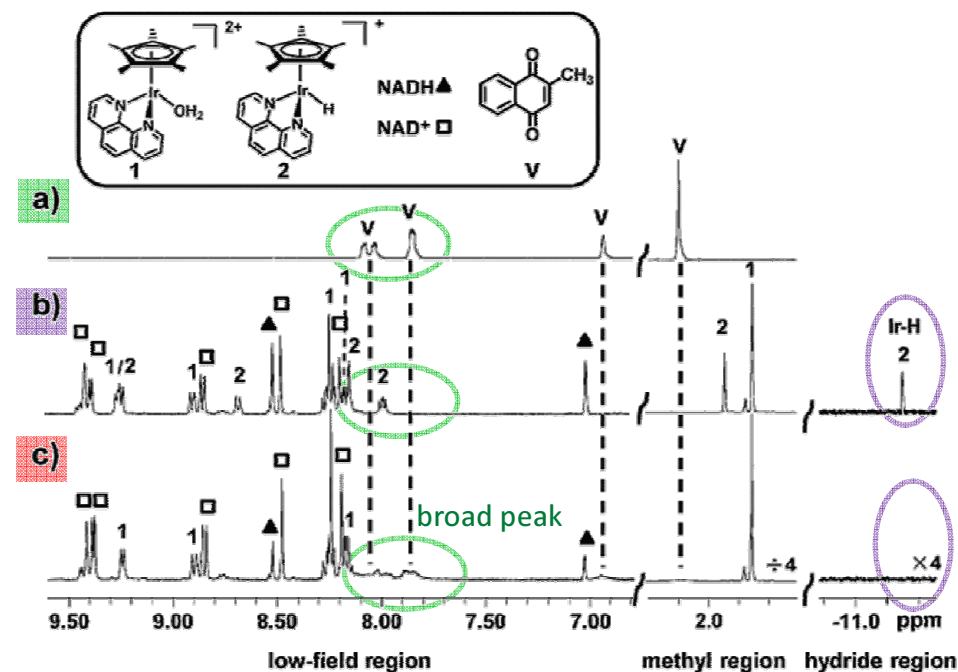
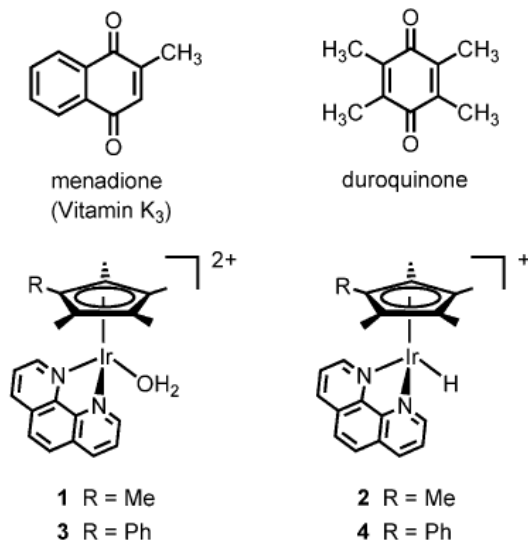


Figure 1. ¹H NMR spectra showing the formation of complex 2 ($[(\eta^5\text{-C}_5\text{Me}_5)\text{Ir}(\text{phen})(\text{H})]^+$) and its reaction with menadione in $[\text{D}_4]\text{MeOD}/\text{H}_2\text{O}$ (1:9) at 298 K. a) Menadione alone. b) Formation of hydrido complex 2 from reaction of $[(\eta^5\text{-C}_5\text{Me}_5)\text{Ir}(\text{phen})(\text{H}_2\text{O})]^{2+}$ (1; 1 mM) with NADH (2 mM). c) Disappearance of signals of hydride 2 and broadening of signals of menadione 10 min after addition of menadione (1 mM) to mixture in (b).

2-3. Reduction by NADH (2)-2

2. Redox

[Ir-H] reduces quinones to semiquinone radical

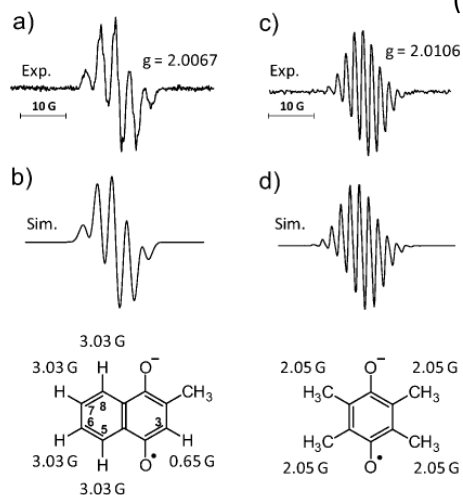
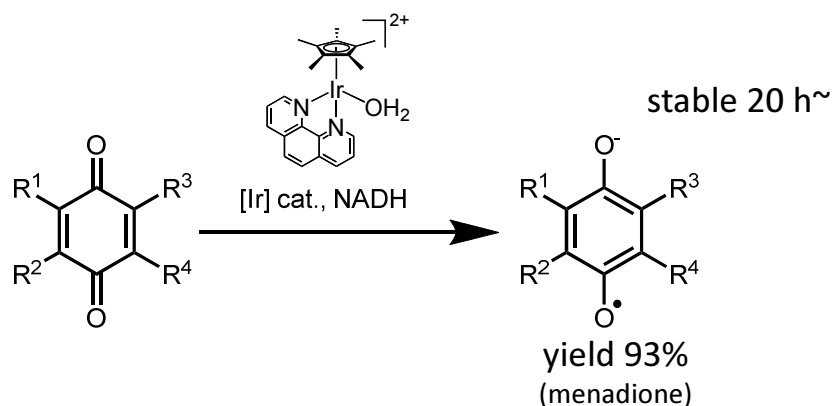


Figure 2. X-band EPR spectra at around 290 K. a) Menadione radical anion ($M^{\bullet-}$) generated by the reduction of menadione (1 mM) by NADH (0.5 mM) catalyzed by complex 1 or 3 (160 μ M) in phosphate buffer (pH 7.2, 9 h). b) Simulated $M^{\bullet-}$ EPR spectrum with hyperfine coupling constants. c) Durosemiquinone radical anion ($D^{\bullet-}$) generated by reduction of duroquinone (2 mM) by NADH (1 mM) catalyzed by complex 1 or 3 (330 μ M) in phosphate buffer (pH 7.2, 10 h). d) Simulated $D^{\bullet-}$ EPR spectrum with hyperfine coupling constants.

1 NADH reduces 2 quinone

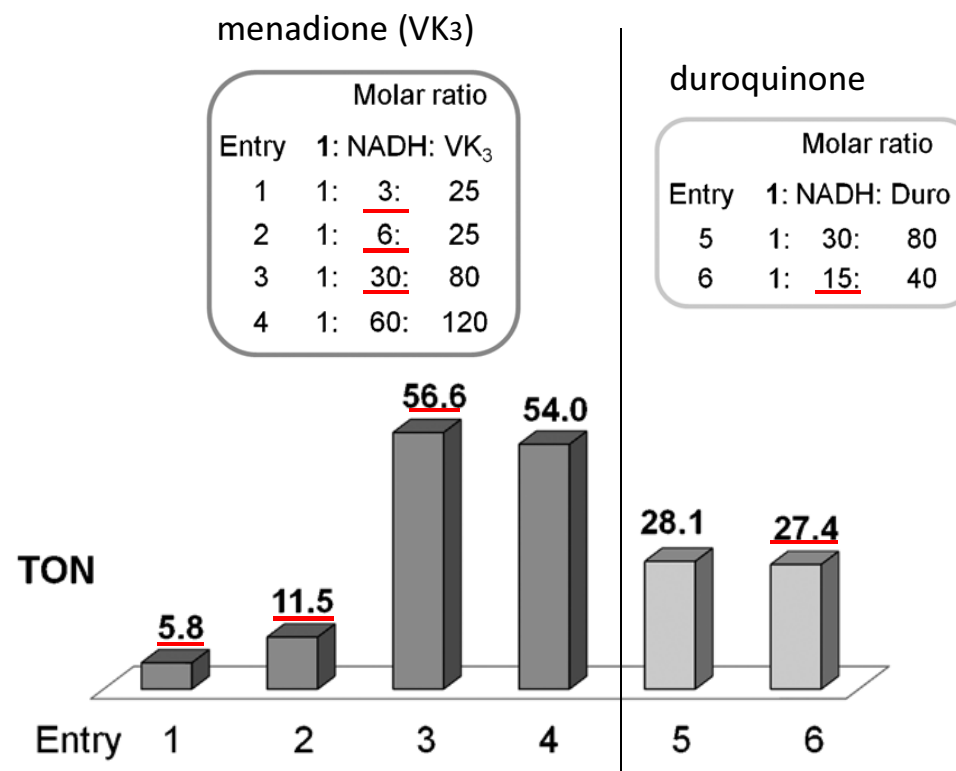
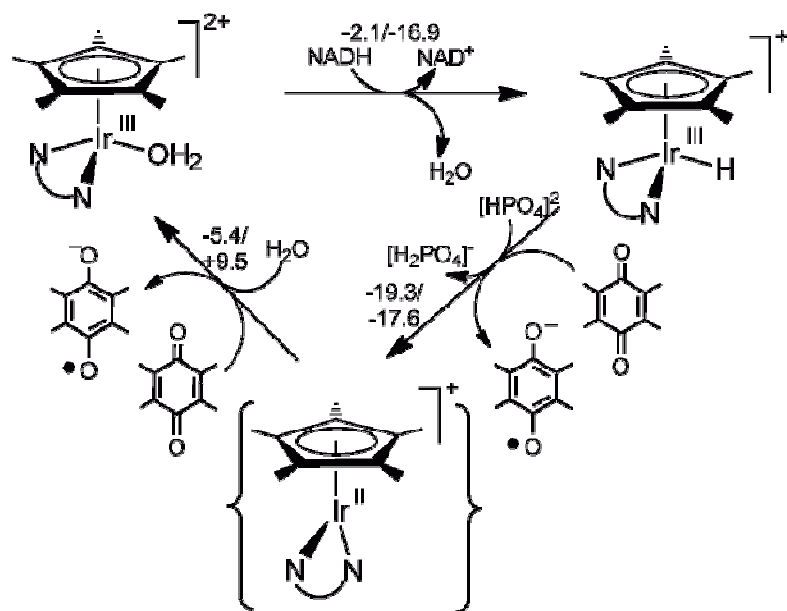


Figure 3. Reduction of menadione (VK₃, entries 1–4) and duroquinone (Duro, entries 5 and 6) by NADH catalyzed by complex 1 (80 μ M).

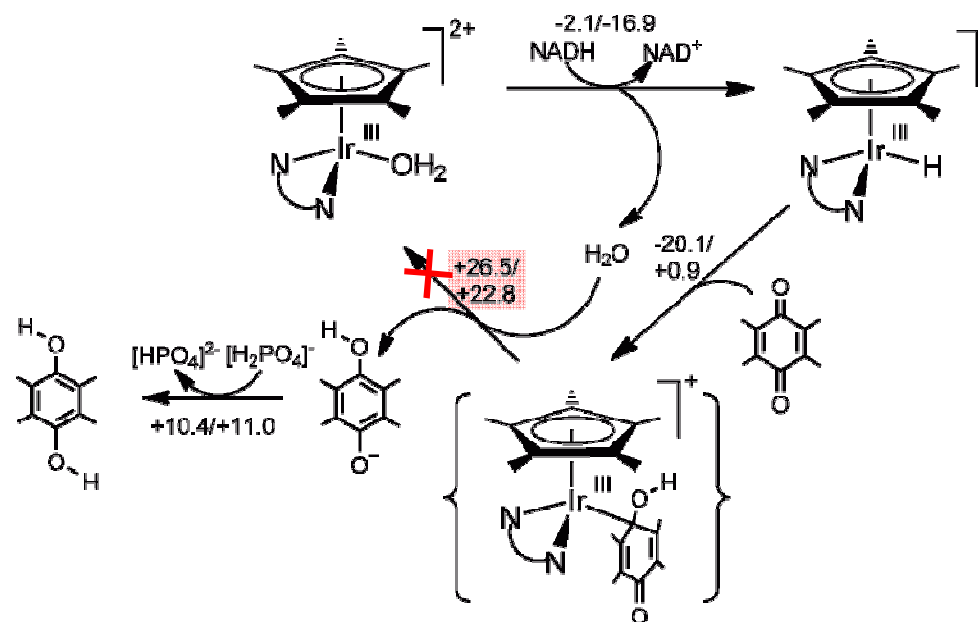
2-3. Reduction by NADH (2)-3

Proposed reaction mechanism

a) One electron transfer $\times 2$ (proposed)



b) Two electron transfer



XX/YY is each potential-energy/free-energy difference ($\text{kcal} \cdot \text{mol}^{-1}$)
by DFT calculation

Contents

1. Introduction

2. Redox catalyst

3. Pd catalyst . . .

Mizorogi-Heck

Sonogashira

Suzuki-Miyaura

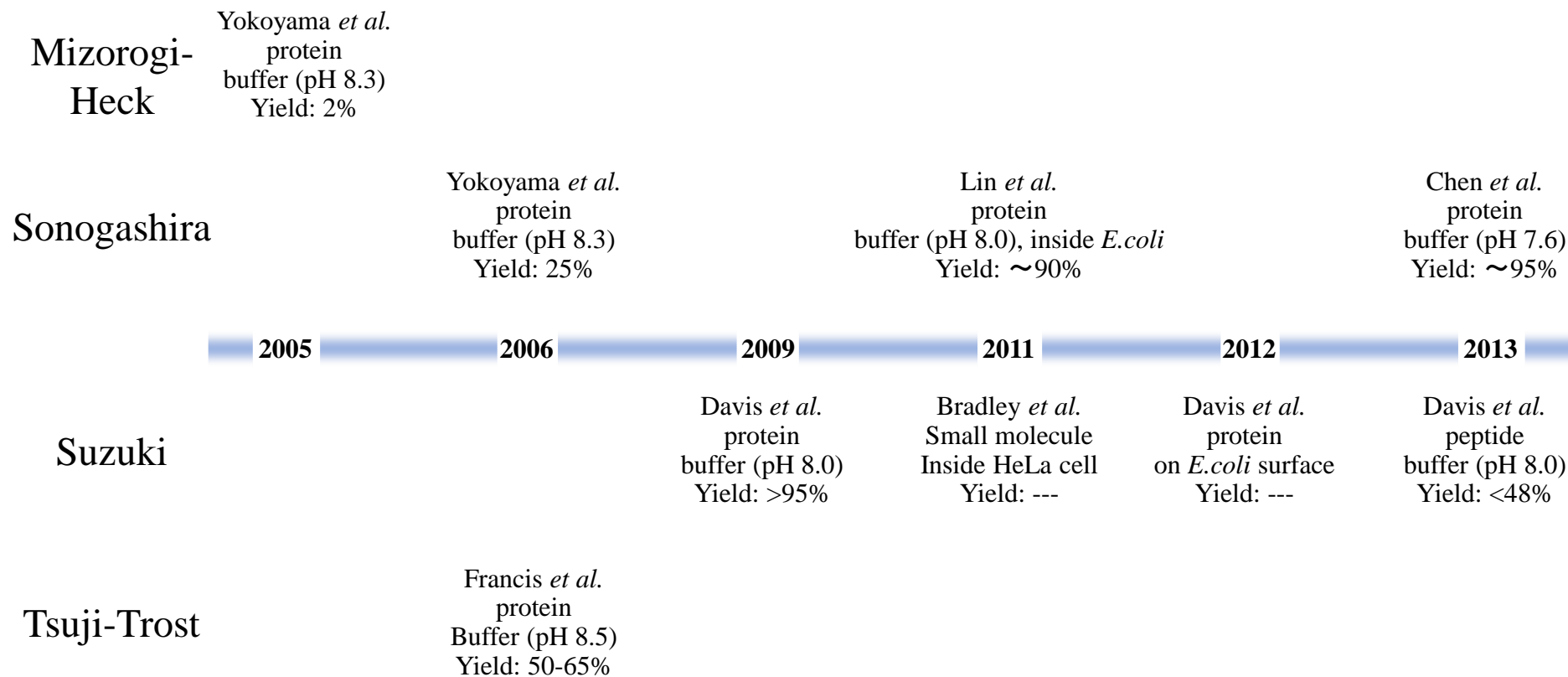
Tsuji-Trost

4. Others

5. Summary

3-0. History

3. Pd



3-1. Mizorogi-Heck reaction

3. Pd

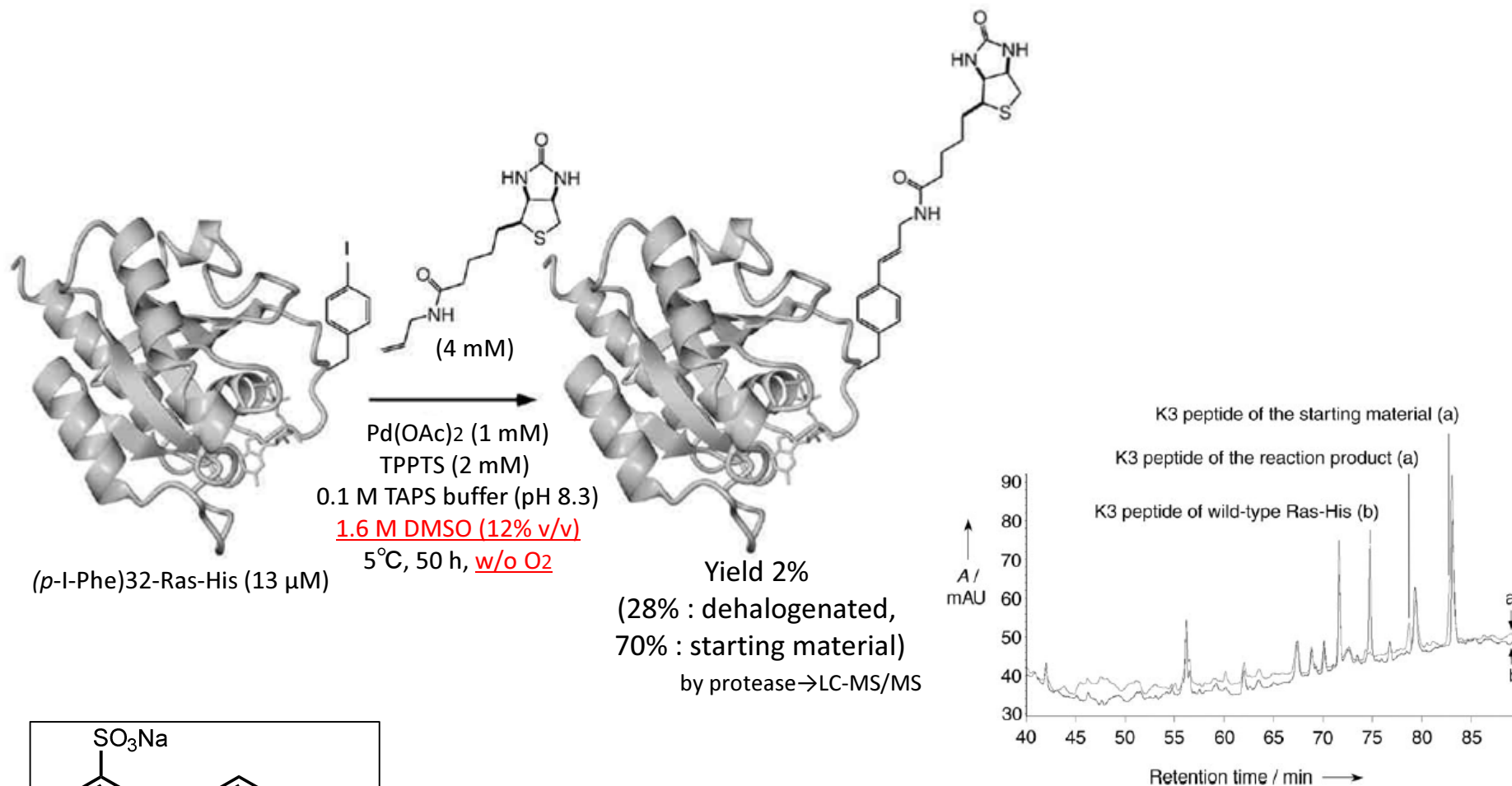
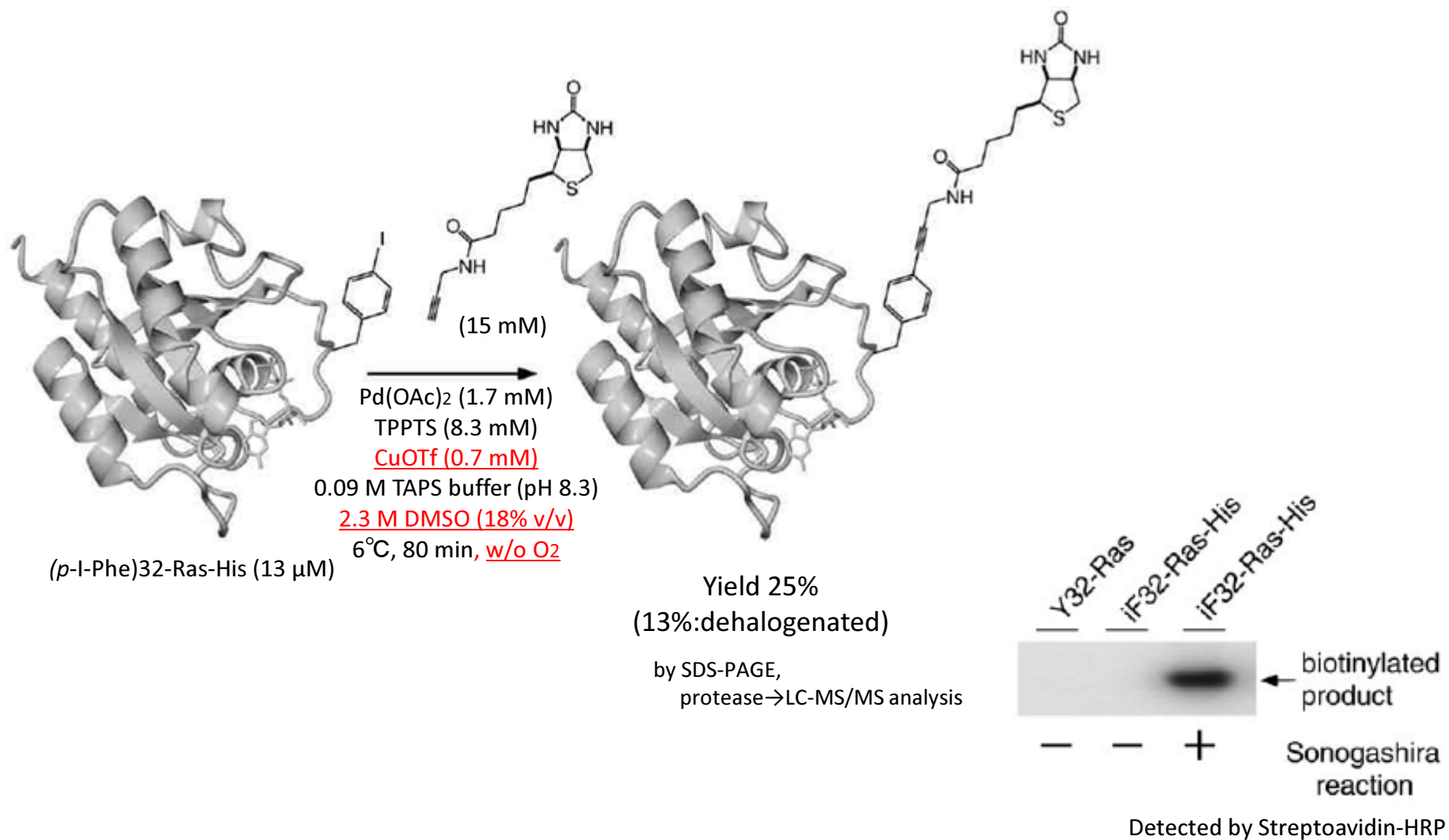


Figure 2. The HPLC chromatogram for Lys-C peptides of a) the crude reaction product and b) those of intact wild-type Ras-His, with ultraviolet (UV) detection at 215 nm. The K3 peptide (17–42) of the biotinylated product (bF32-Ras-His) and that of the dehalogenated product (iF32-Ras-His) were eluted as an overlapped peak at 77.9 min.

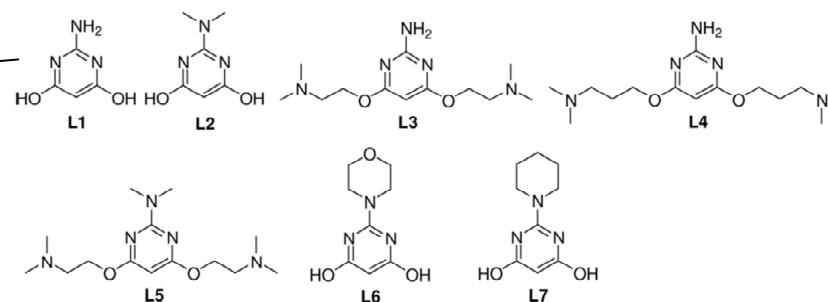
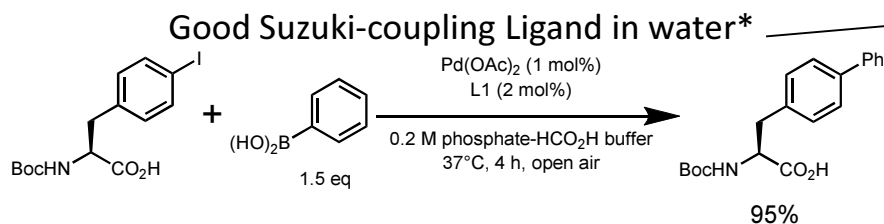
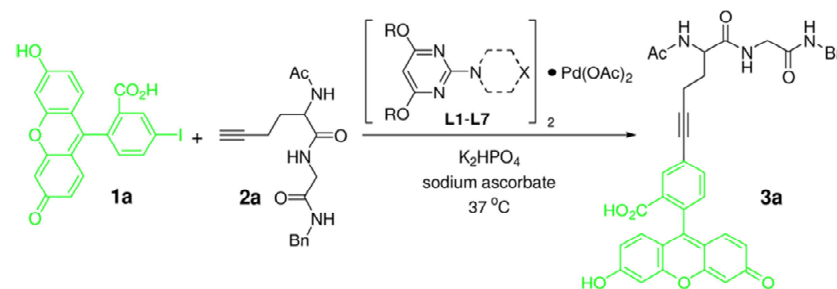
3-2. Sonogashira coupling (1)

3. Pd

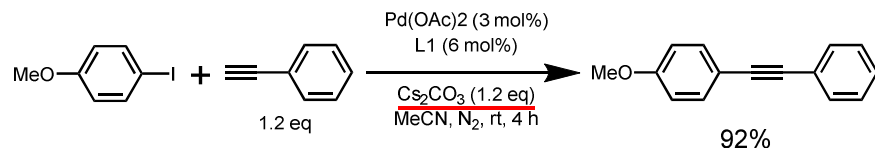


3-2. Sonogashira coupling (2)-1

3. Pd



And Good Cu-free Sonogashira coupling Ligand
in organic solvent**



Entry	1a : 2a	Pd • ligand	Ligand	Time	Yield (%) ^b
1	1.05:1.00	20%	L1	30 min	NR
2	1.05:1.00	20%	L2	30 min	59
3	1.05:1.00	20%	L3	30 min	<5
4	1.05:1.00	20%	L4	30 min	<5
5	1.05:1.00	20%	L5	30 min	29
6	1.05:1.00	20%	L6	30 min	<5
7	1.05:1.00	20%	L7	30 min	13
8	1.00:2.40	30%	L2	40 min	91

*Described more detail in “3-3. Suzuki coupling (1)”

B. G. Davis *et al.* *J. Am. Chem. Soc.* **2009**, *131*, 16346-16347

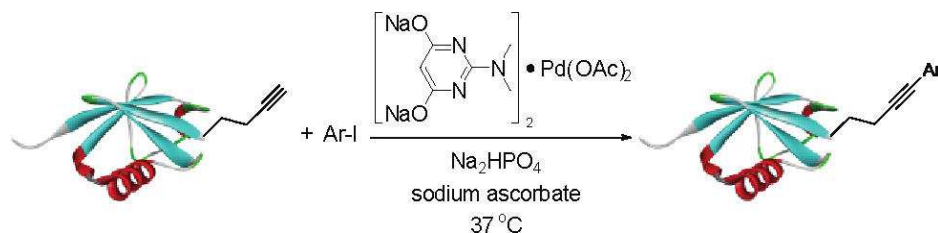
Y. -X. Xie *et al.* *Eur. J. Org. Chem.* **2005, 4256-4259

^bHPLC yield²⁹

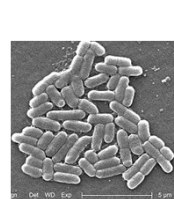
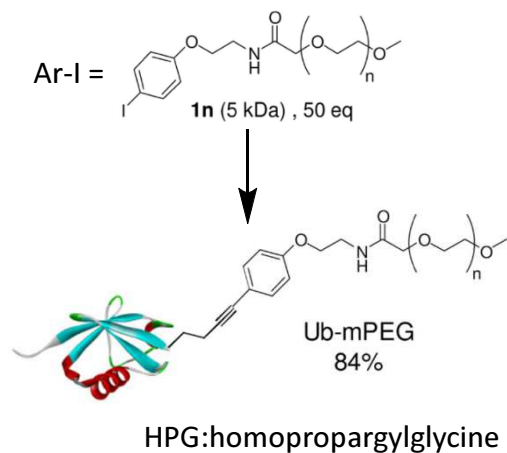
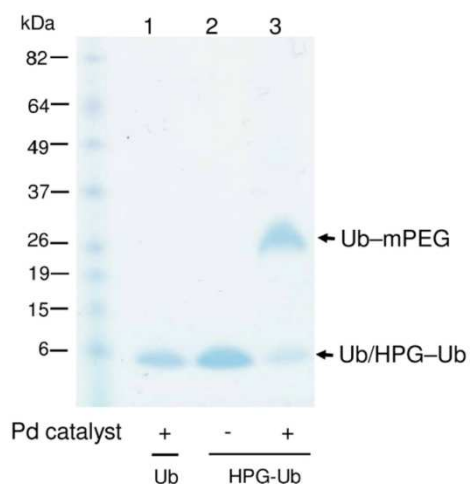
Q. Lin *et al.* *J. Am. Chem. Soc.* **2011**, *133*, 15316-15319

3-2. Sonogashira coupling (2)-2

3. Pd



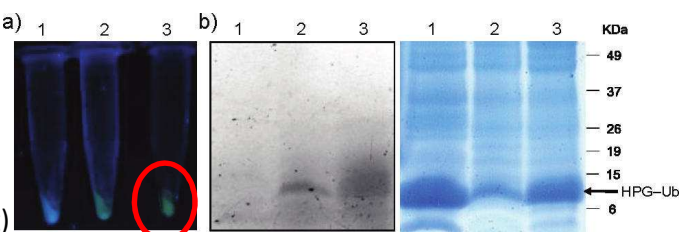
*Mixture of Ar-I, Pd complex, and sodium ascorbate was stirred at 37°C for 60 min to activate



1a (100 μM)
 Pd(OAc) $_2$ ·L $_2$ (1 mM)
 5 mM sodium ascorbate
 2 mM phosphate buffer (pH 8)
 DMSO (0.1 mM), 37°C, 4 h
 → Centrifugation (Figure a) → fluorescence (Figure b, left) + Coomassie blue staining (Figure b, right)

HPG-Ub overexpressing *E. coli*

- 1 : control (w/o Pd complex)
 2 : add **1a** and Pd
 3 : add preactivated **1a** and Pd
 (stirred at 37°C, 1 h before adding)



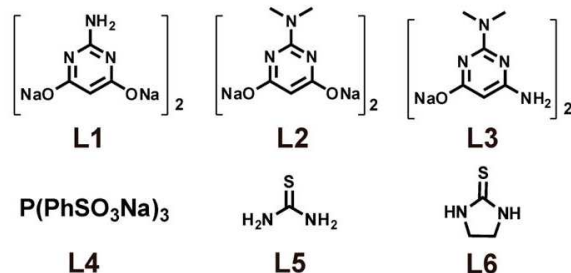
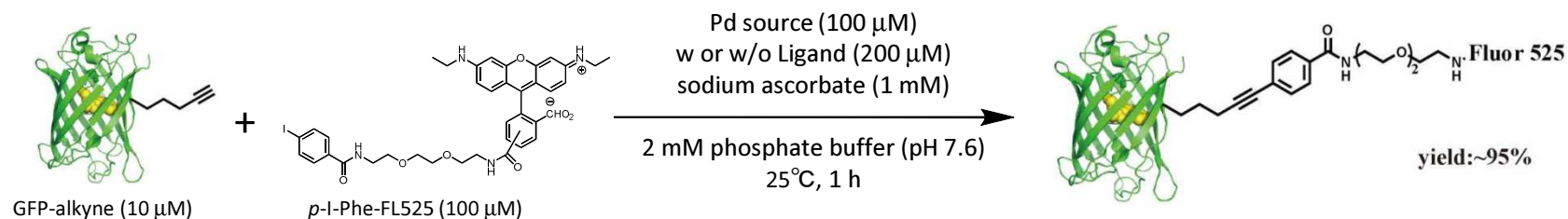
Entry	Ar-I	Time	Yield (%) ^b
1	1a	30 min	93
2	1b	30 min	83
3	1c	30 min	55
4	1d	30 min	86
5	1e	240 min	73 ^c
6	1f	30 min	83
7	1g	180 min	73
8	1h	30 min	82
9	1i	30 min	85
10	1j	30 min	78
11	1k	30 min	79
12	1l	180 min	66
13	1m	180 min	60

^bLC-MS yield

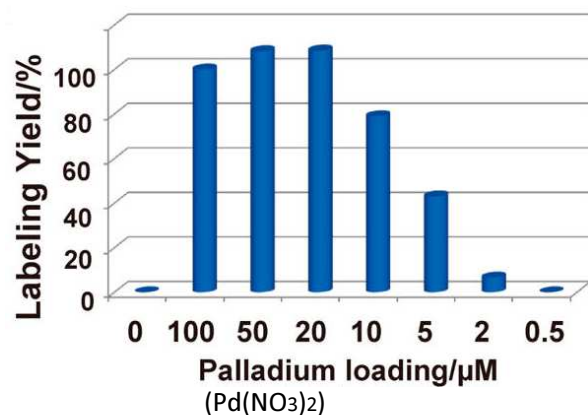
3-2. Sonogashira coupling (3)-1

3. Pd

Ligand-free, Pd(NO₃)₂ shows best conversion



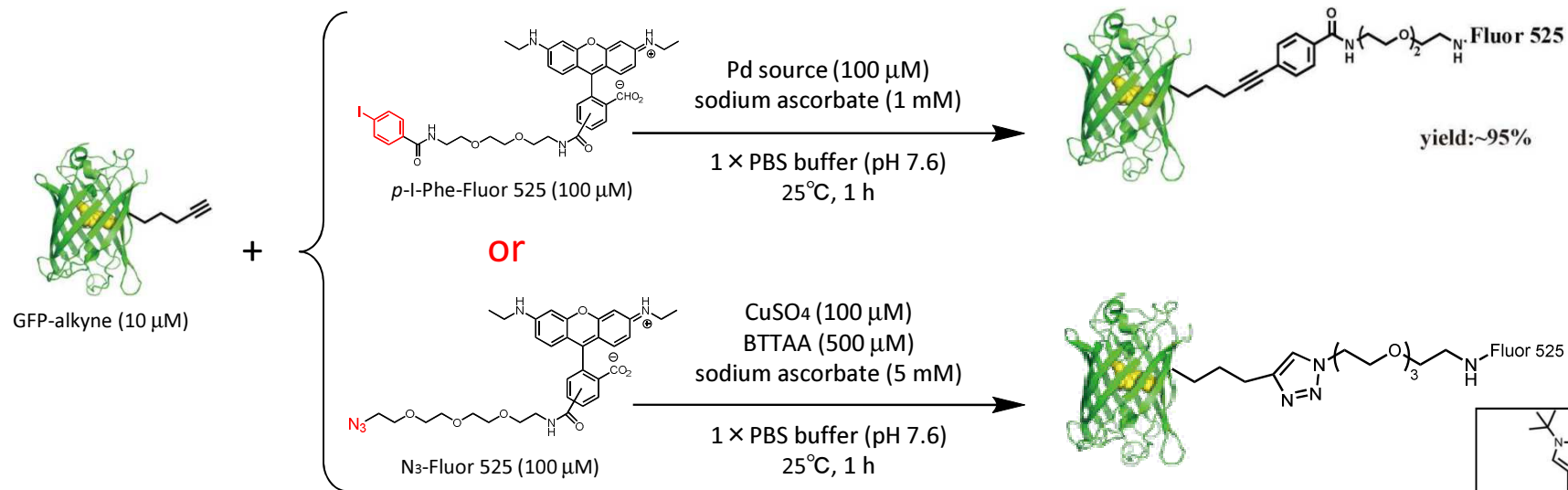
Entry	Ligand	Pd source	Yield(%) ^b
1	L1 ^a	Pd(OAc) ₂	< 5
2	L2 ^a	Pd(OAc) ₂	53
3	L3 ^a	Pd(OAc) ₂	< 5
4	—	Pd(OAc) ₂	11
5	—	Na₂PdCl₄	90
6	—	PdCl ₂	25
7	—	Pd(NO₃)₂	95
8	L4	Na ₂ PdCl ₄	82
9	L4	Pd(OAc) ₂	64
10	—	Pd ₂ (dba) ₃	57
11	L5	Pd(OAc) ₂	< 5
12	L6	Pd(OAc) ₂	< 5



^bcalculated from sample fluorescence on SDS-PAGE

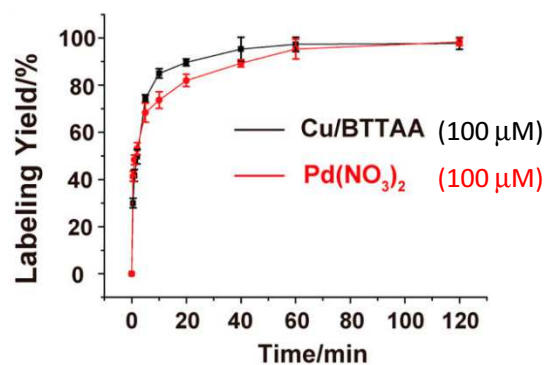
3-2. Sonogashira coupling (3)-2

3. Pd



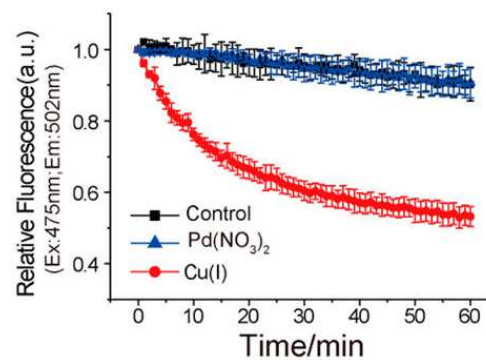
Pd vs Cu

Reaction dynamics



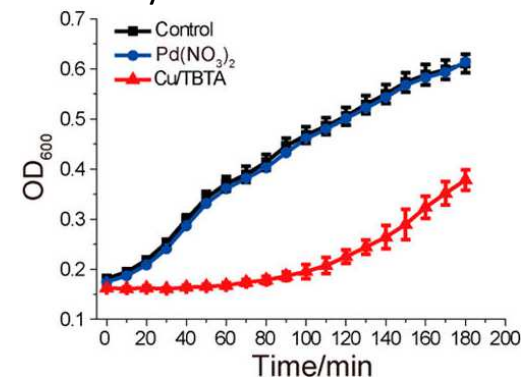
*Same reaction condition as above

Protein deactivation



*GFP (10 μ M) + Pd(NO₃)₂ (100 μ M) or Cu(I) (100 μ M)

Acute toxicity

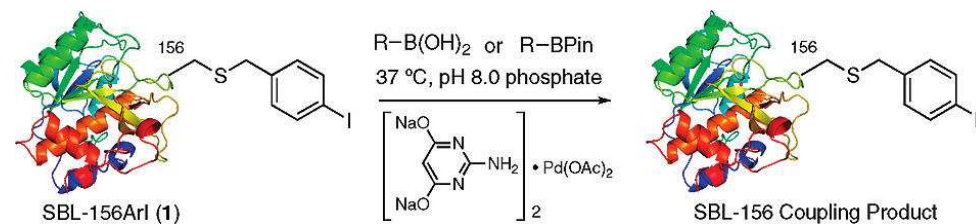
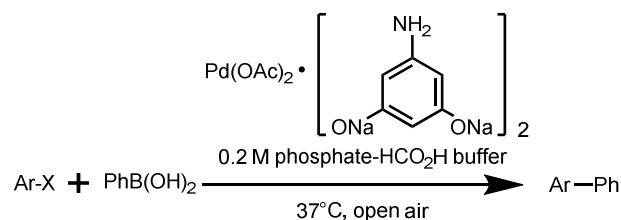


*Pd(NO₃)₂ (200 μ M) or Cu(I) (200 μ M) incubate 25°C, 1 h

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3-3. Suzuki coupling (1)

3. Pd



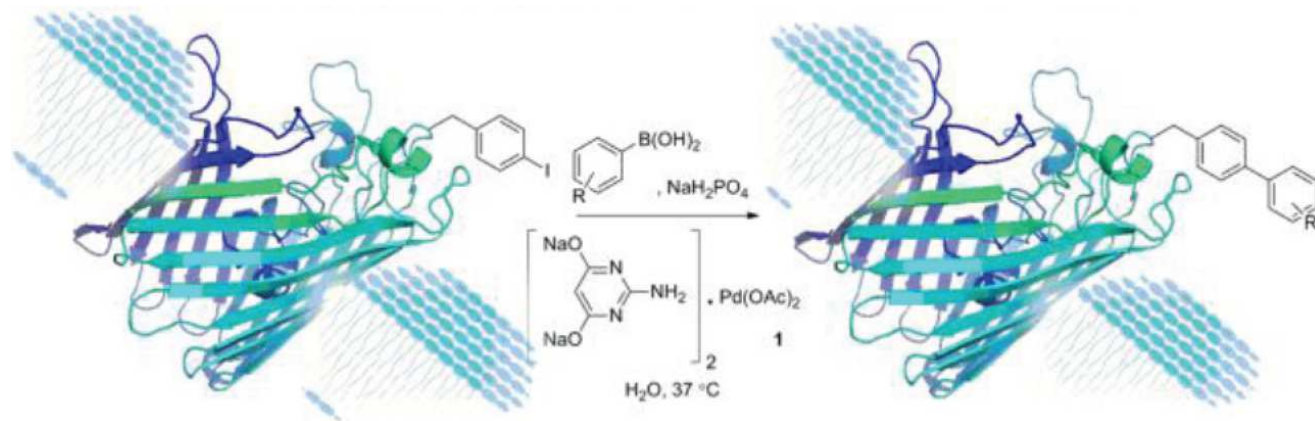
Entry	ArX	Pd Loading	Conditions	Coupling Product	Yield (%)
1		1%	37 °C, 4 h		95
2		1%	37 °C, 4 h		98
3		1%	37 °C, 4 h		0
4		1%	37 °C, 4 h		94
5		2%	37 °C, 4 h		95
6		1%–10%	37 °C, 6 h		0 ?
7		4%	37 °C, 6 h		92

Entry	R-B(OH) ₂ / R-BPin	Time	Conversion
1		30 min	> 95%
2		30 min	> 95%
3		30 min	> 95%
4		30 min	> 95%
5		30 min	> 95%
6		30 min	> 95%
7		30 min	> 95%
8		30 min	> 95%
9		60 min	> 95%

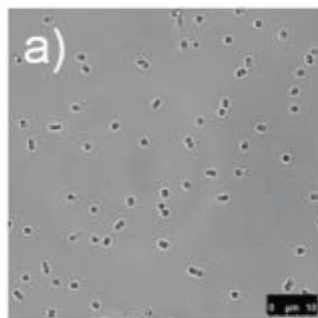
3-3. Suzuki coupling (2)

3. Pd

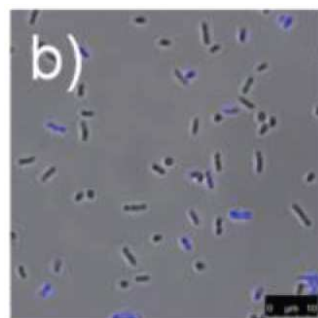
Cell-surface labeling (*E. coli*)



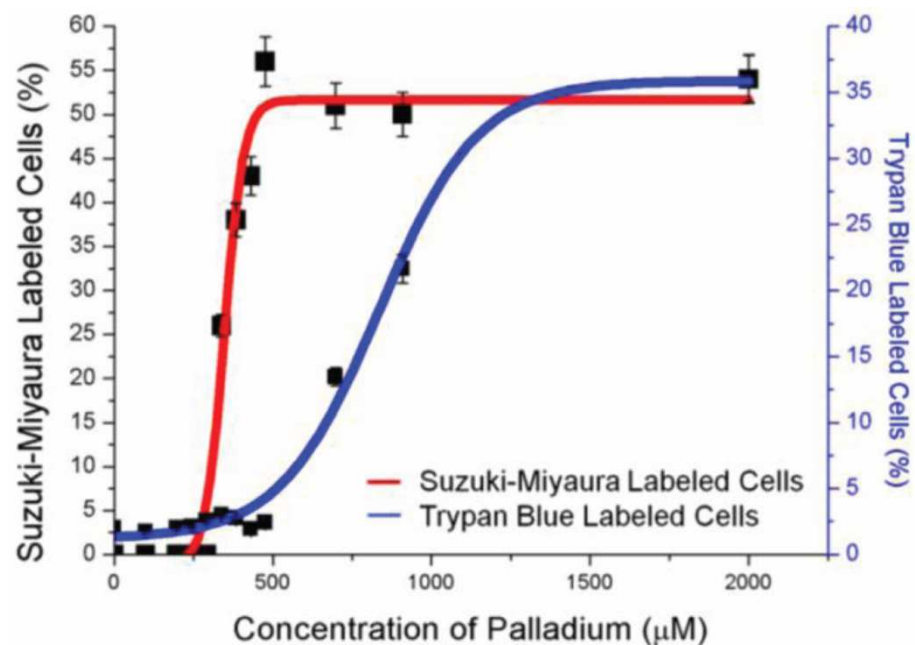
control



p-I-Phe mutant



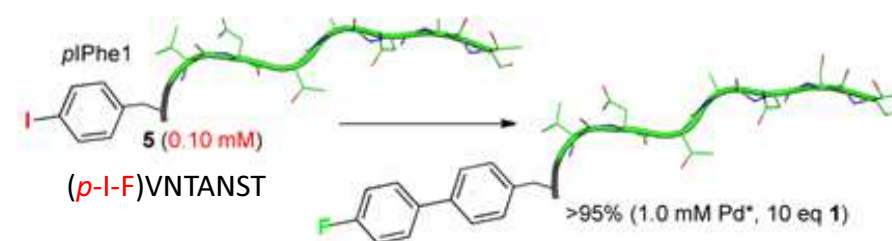
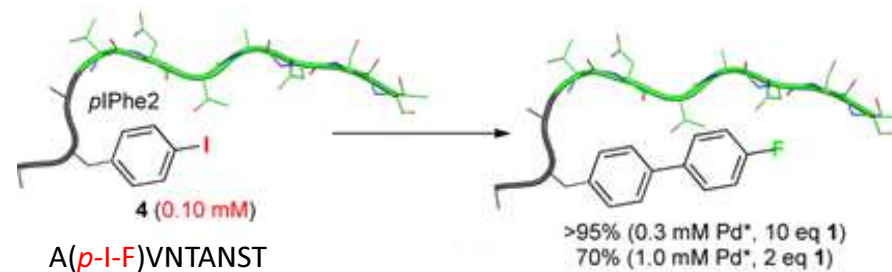
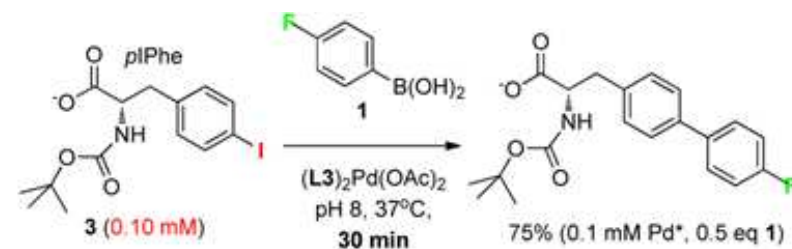
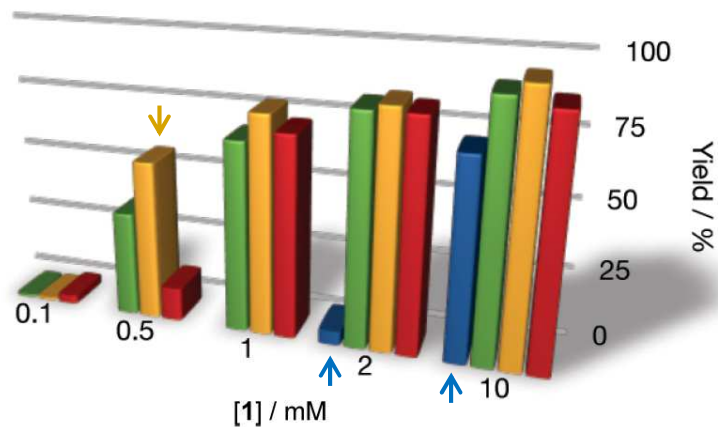
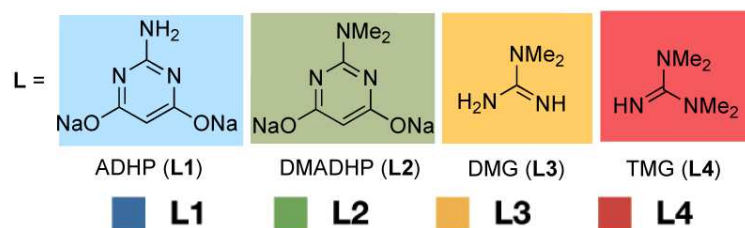
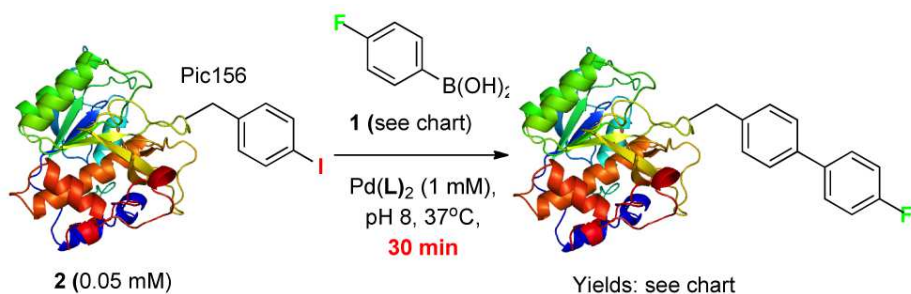
Fluorescence observed



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3-3. Suzuki coupling (3)-1

3. Pd

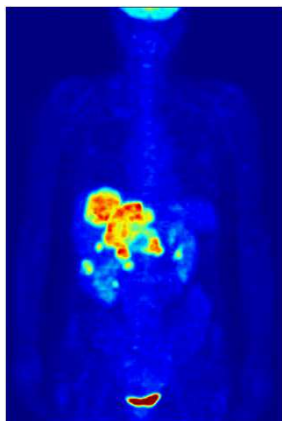
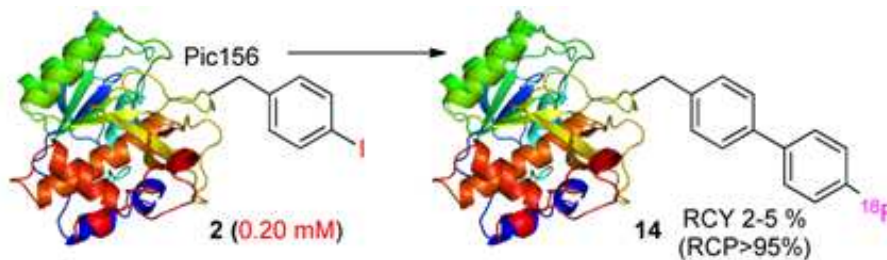
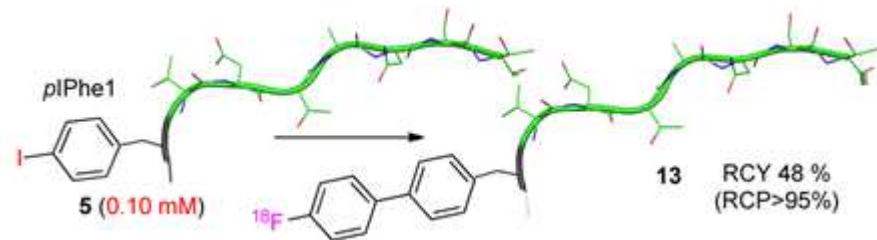
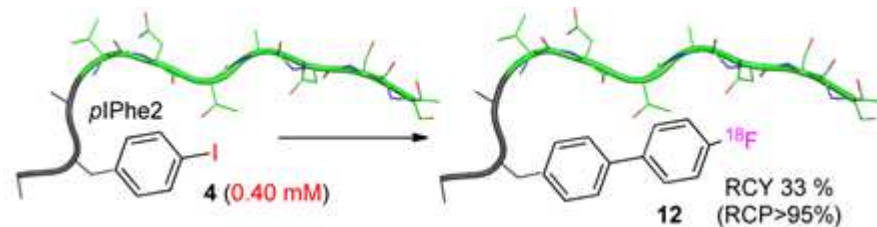
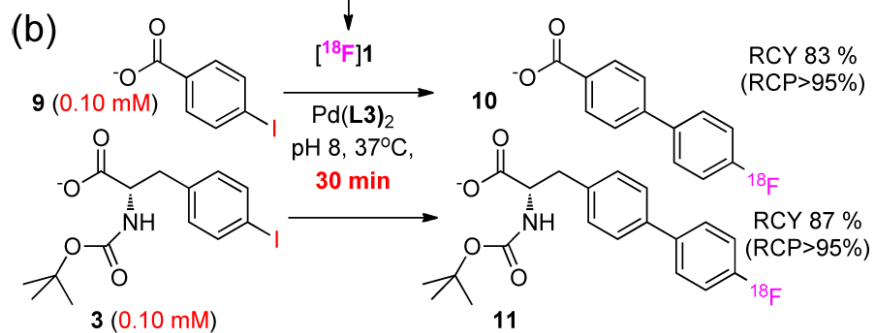
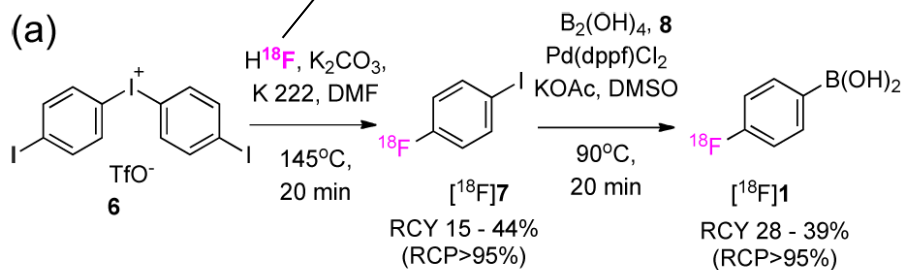


VNTANST sequence :
Comprehensive carcinoma homing peptide

3-3. Suzuki coupling (3)-2

3. Pd

^{18}F ...PET tracer, $t_{1/2}=109$ min
 <1 h total reaction time is desired for PET

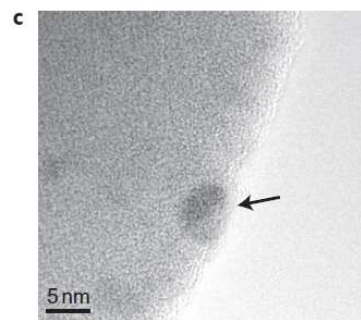
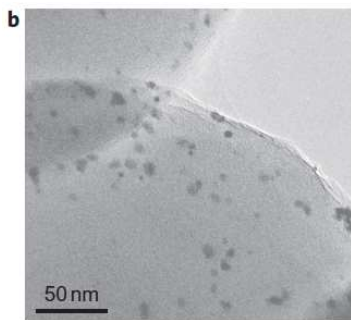
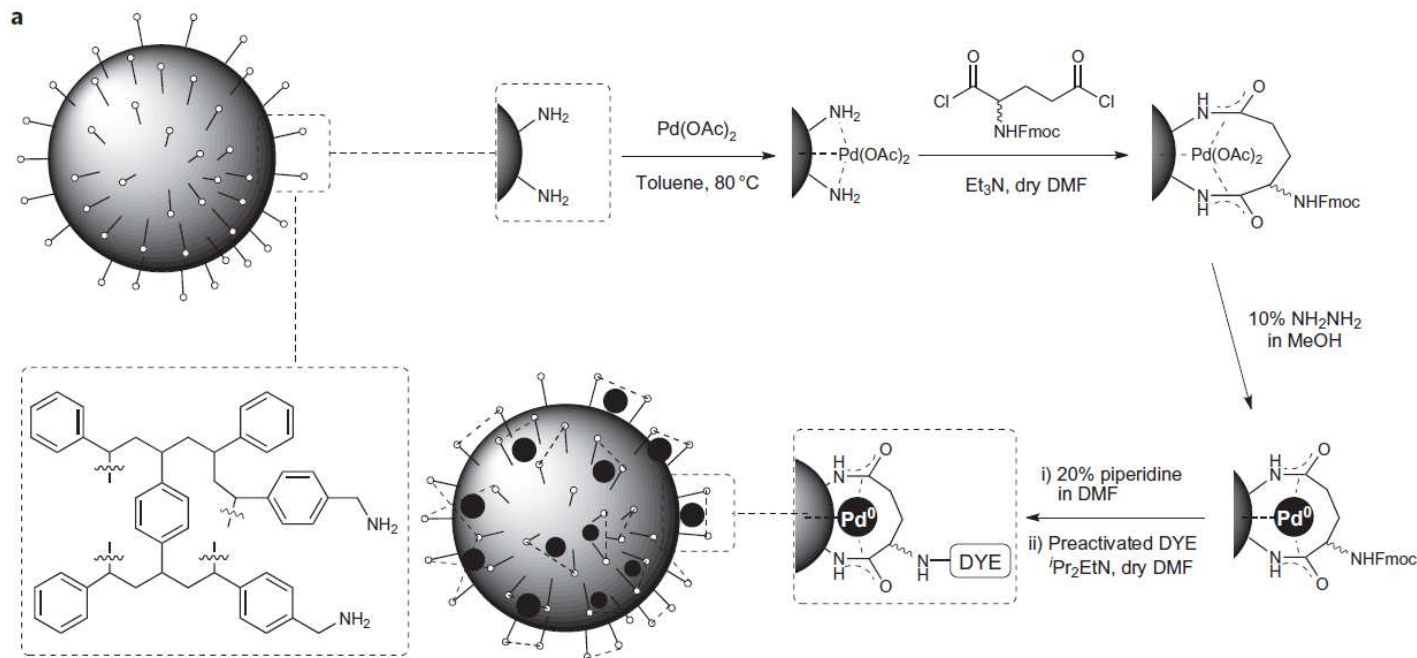


^{18}F -DG whole body PET imaging
 Source:wikipedia, ポジトロン断層法

3-3. Suzuki coupling (4)-1

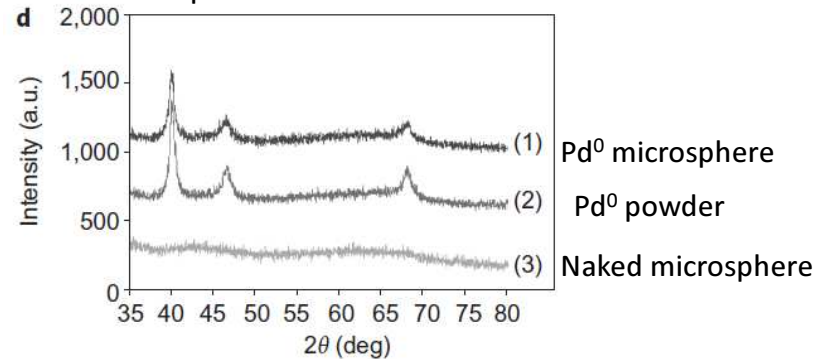
3. Pd

Pd⁰ microsphere synthesis



TEM of Pd⁰ nanoparticle

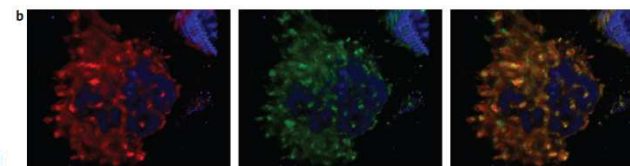
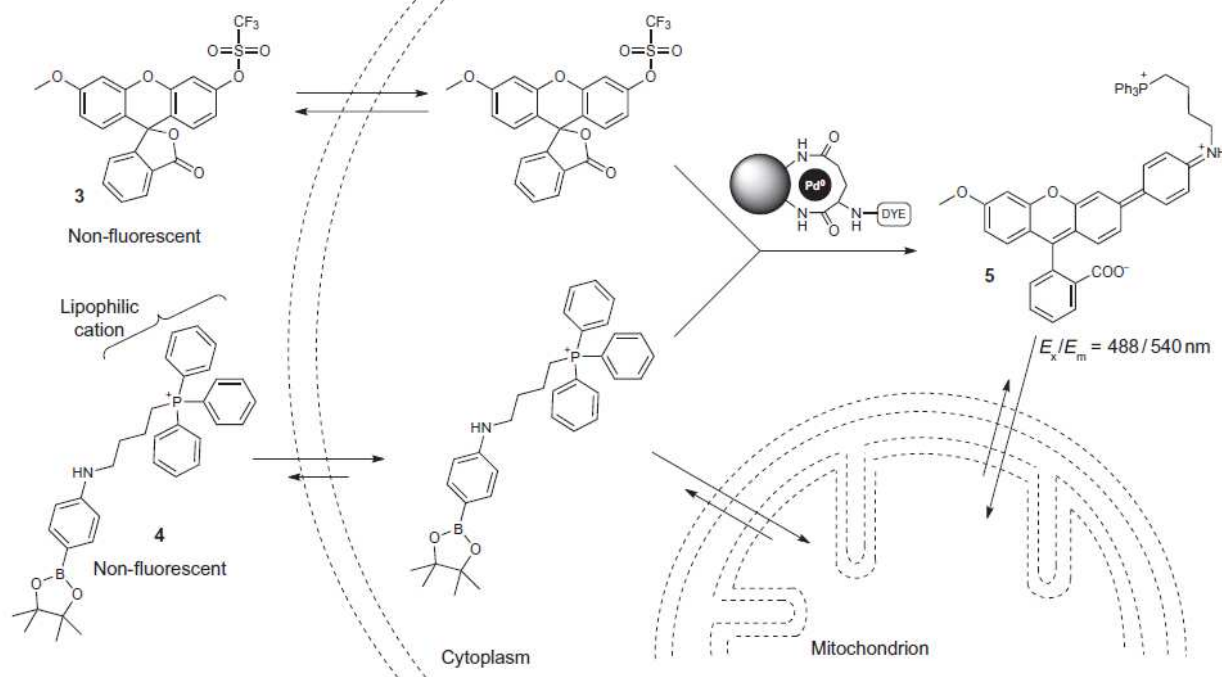
Powder XRD patterns



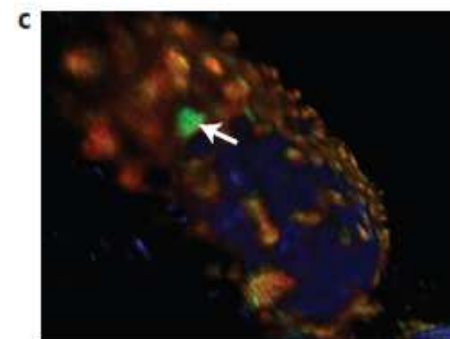
3-3. Suzuki coupling (4)-2

3. Pd

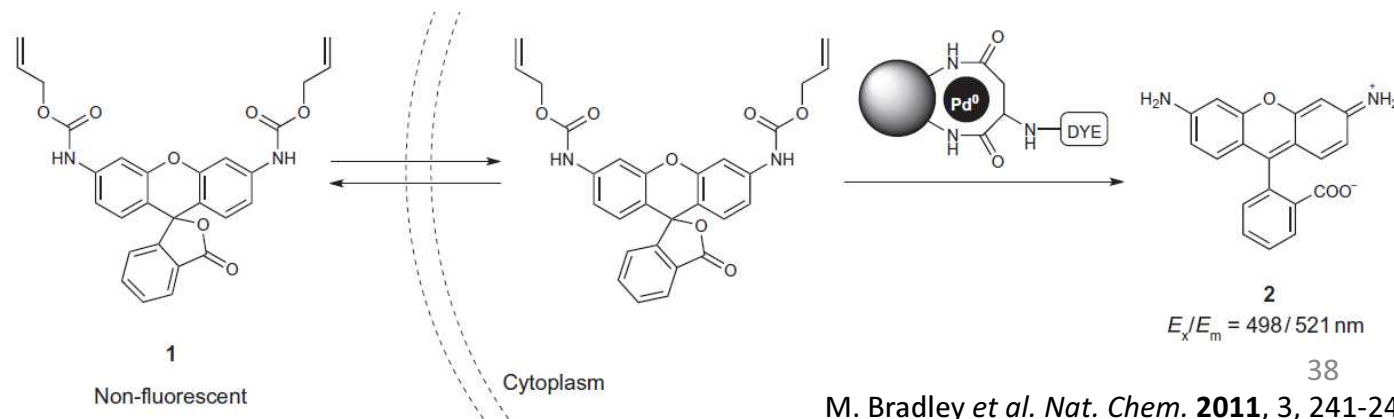
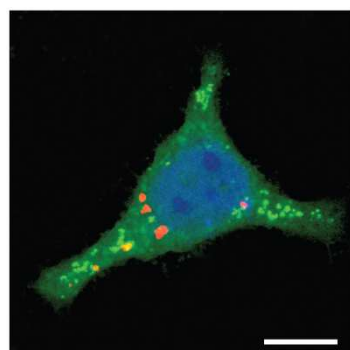
a Pd microspheres are taken into cells (24 h, 75%~)



Mitochondria(red)
Synthesized compound 5(green)
merged



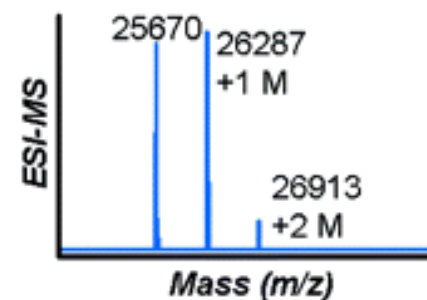
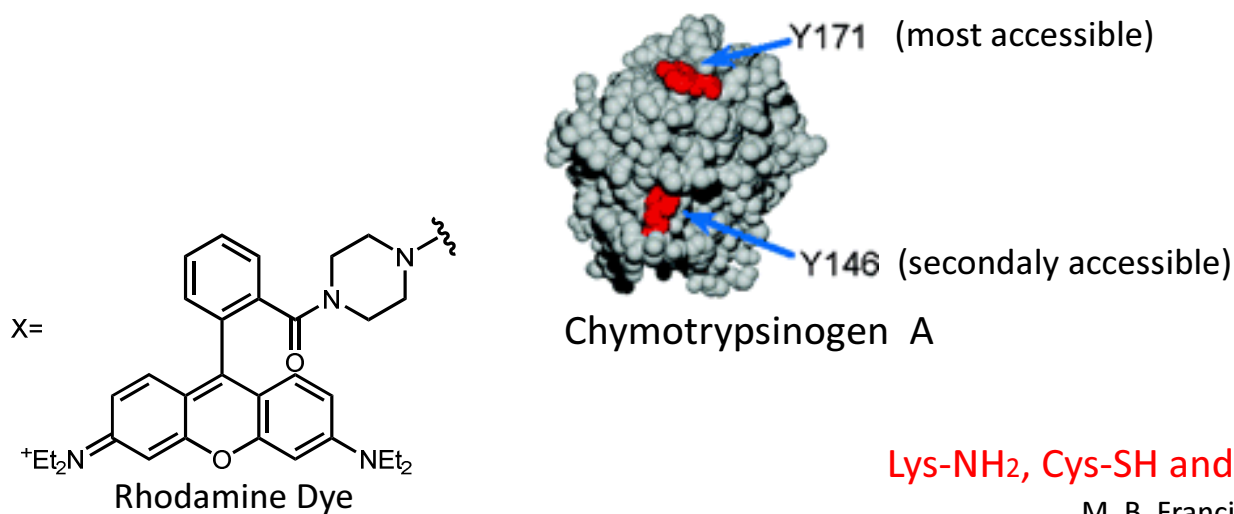
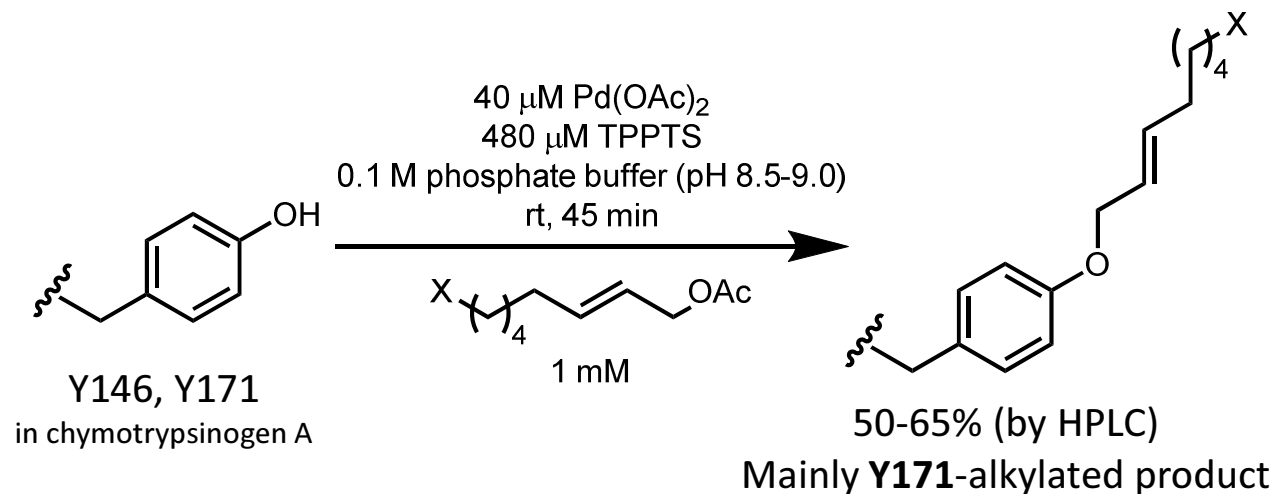
Allow: Pd microsphere
(not localized within mitochondria)



3-4. Tsuji-Trost Reaction

3. Pd

Also see Mr Suzuki's Literature Seminar (2013)



Lys-NH₂, Cys-SH and other residue were intact

39

Contents

1. Introduction

2. Redox catalyst

3. Pd catalyst

4. Others . . .

Photo Cross Linking

OH \cdot generator

Metathesis

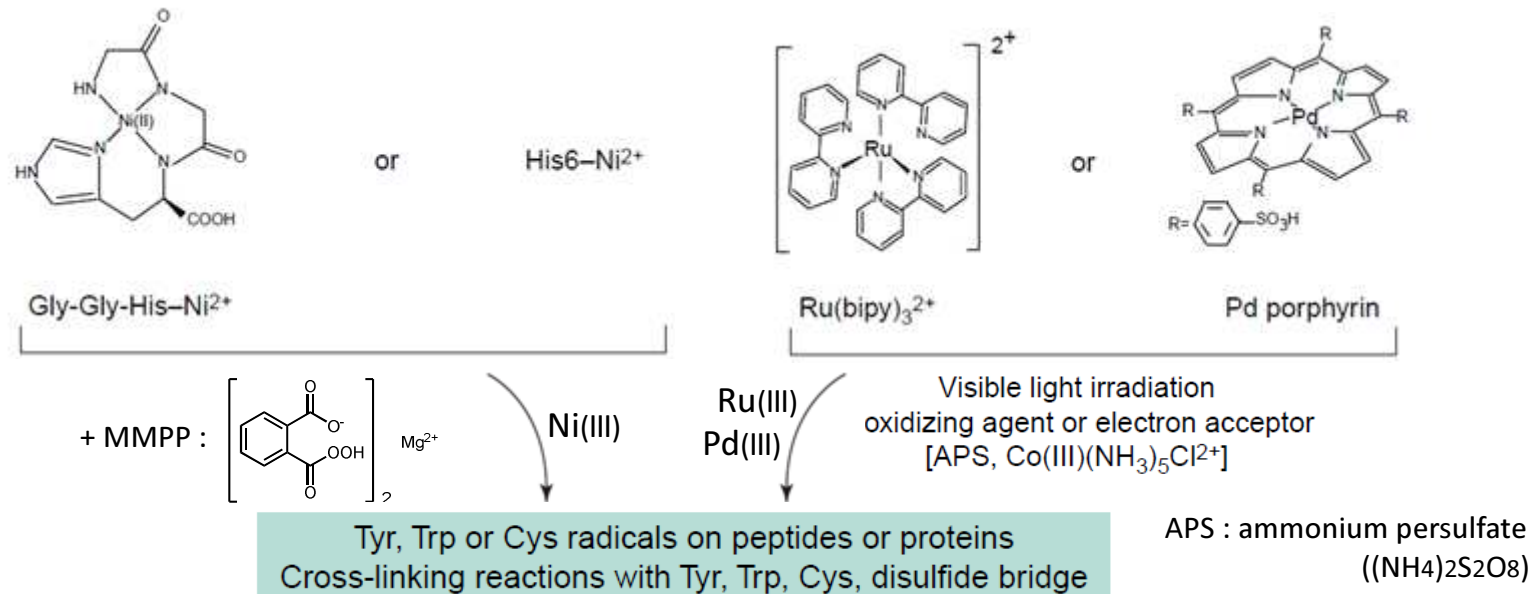
Rh-carbenoid

5. Summary

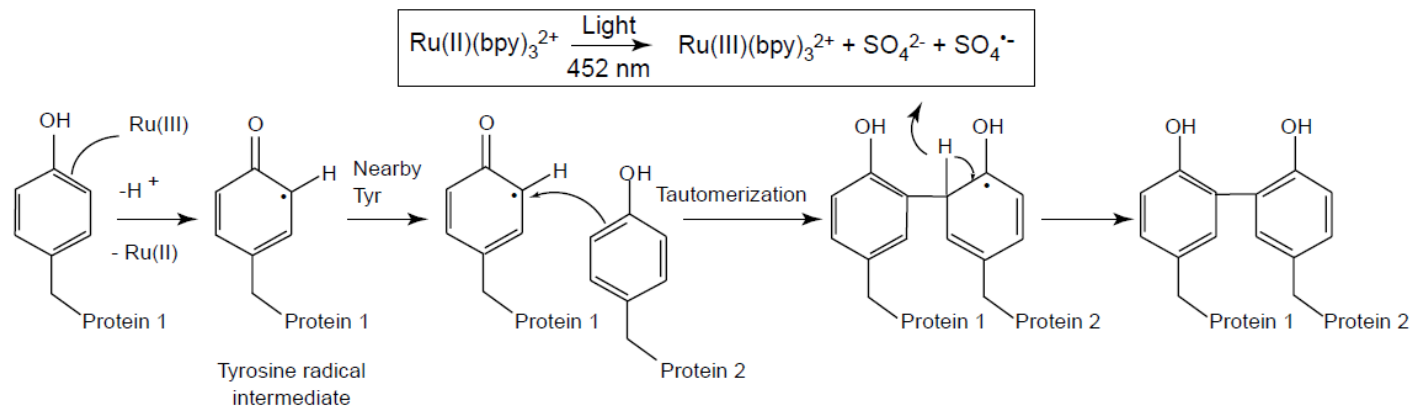
4-1. Photo Cross Linking-1

4. Others

(i) Metal complex-mediated reaction

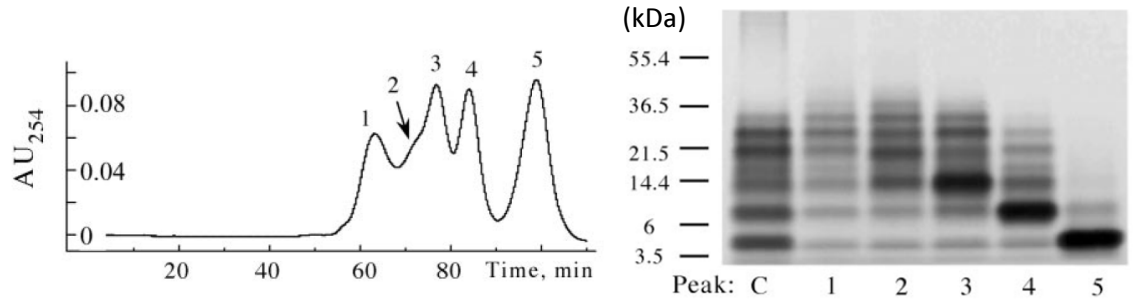
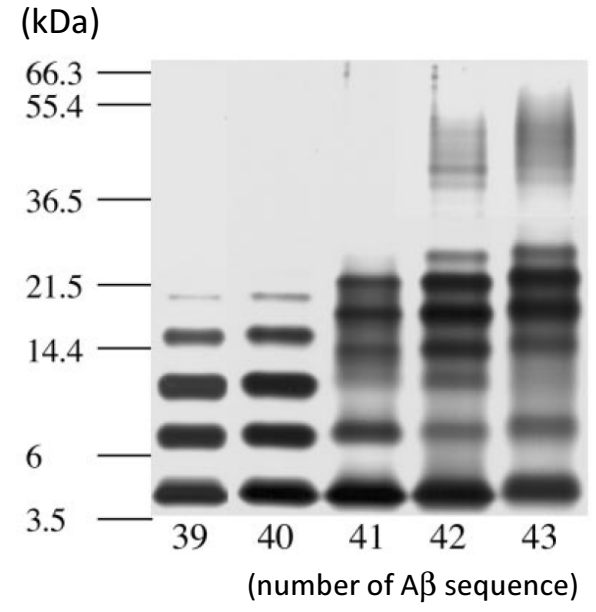
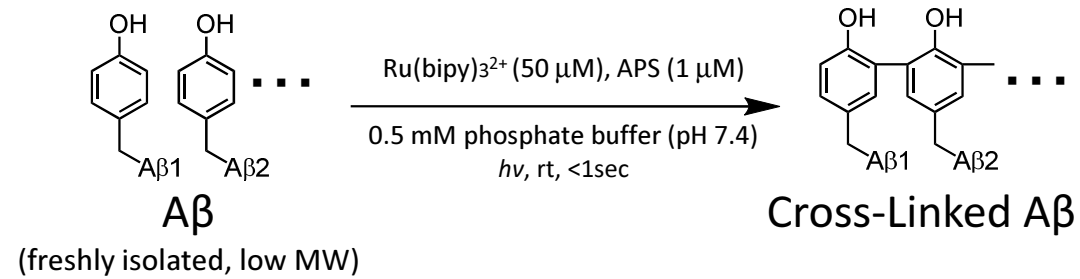
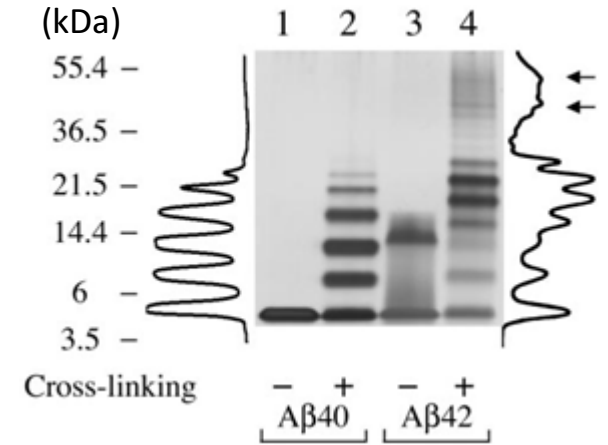
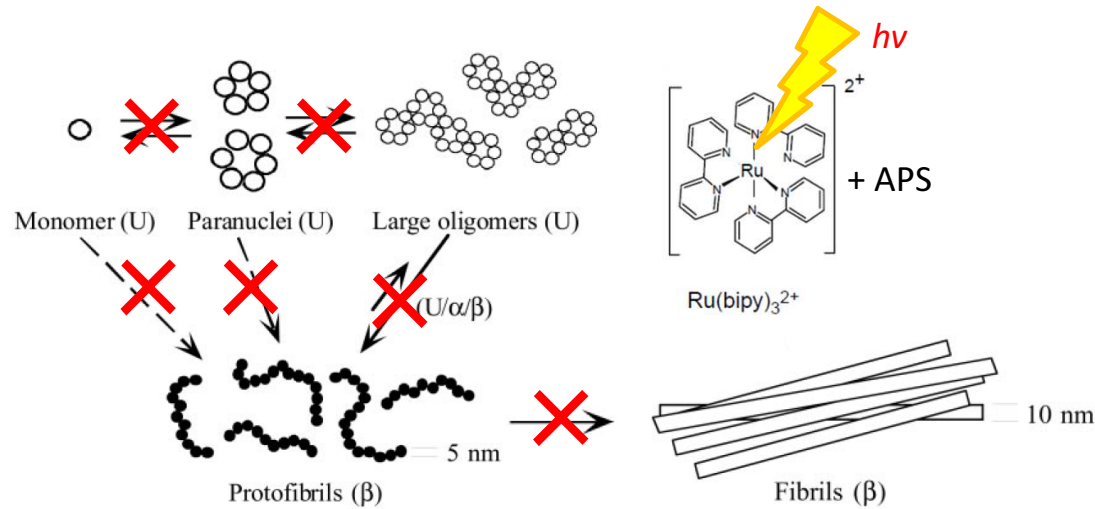


(ii) Radical generation on Tyr residues at ortho positions



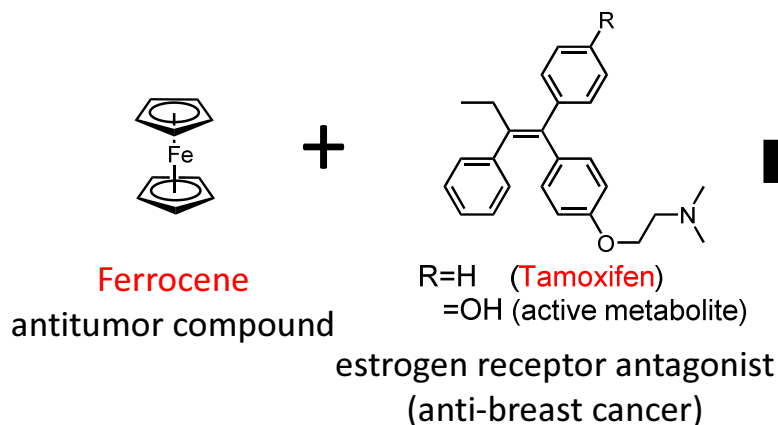
4-1. Photo Cross Linking-2

4. Others



4-2. OH[•] generator

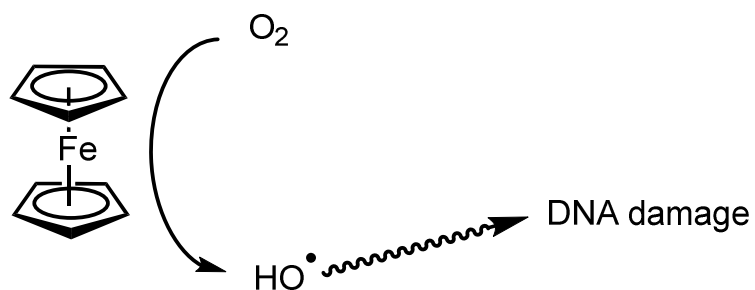
4. Others



compound	binding affinity (%)	IC50 (μM)
oestradiol	100	-
(Z) TAM-OH	107	6.4
(Z)-Fc TAM-OH	40	3.4
(E)-Fc TAM-OH	12	4.9

G. Jaouen *et al. Chem. Commun.* **1996**, 955-956

More active derivatives are reported



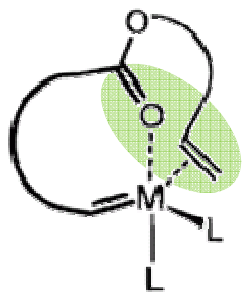
Ferrocene derivatives generate ROS and DNA damage

G. Cavigliolo *et al. Inorg. Chim. Acta.* **2000**, 42-48

in this tissue, relation of OH[•] and DNA damage is not associated
 But there are many reports to relate them.

c.f.) D. MacNamee *et al. J. Biol. Chem.* **1974**, **249**, 2447-2452 ⁴³

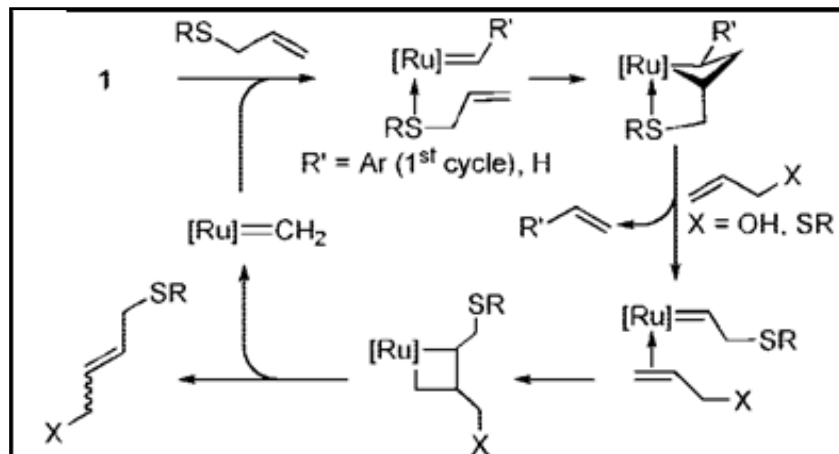
4-3. Olefin metathesis (1)-1



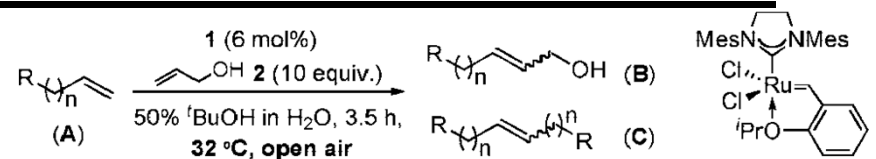
Ester chelates to [Ru] center
(M=Ru)

A. Fürstner et al. *J. Am. Chem. Soc.* **1997**, *119*, 9130-9136

Chalcogen assisted Cross-Metathesis



4. Others

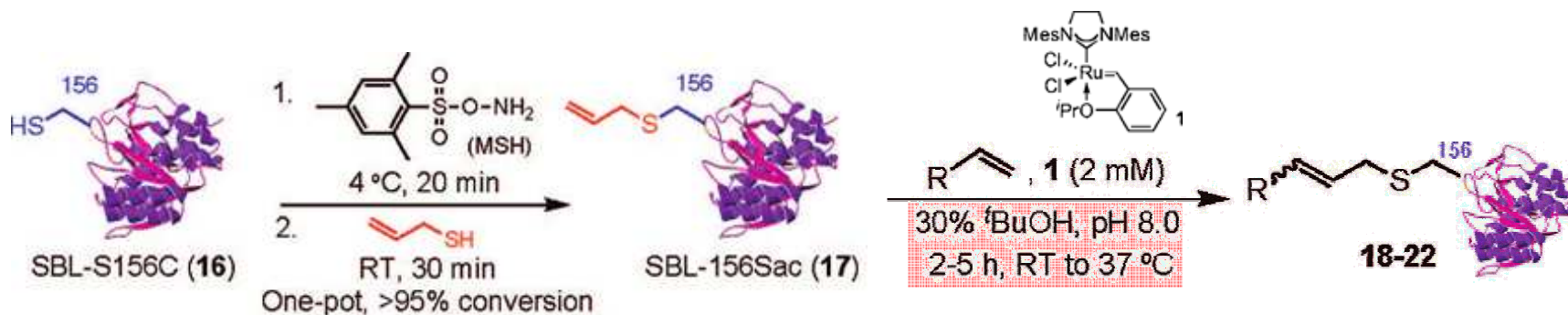


Entry	Alkene A	Cross-Metathesis B (%)^c	Self-Metathesis C (%)^c
1 ^{a,b}		3	0
2 ^a		4	56 (74 brsm)
3 ^a		5	67 (99 brsm)
4 ^a		6	68 ^d
5 ^{a,b}		7	0
6 ^a		8	0
7		9	0
8		10	52 ^d
9		11	19
10		12	8
11		13	11
12		14	31
13		15	28 ^d

B. G. Davis et al. *J. Am. Chem. Soc.* **2008**, *130*, 9642-9643

4-3. Olefin metathesis (1)-2

4. Others



Entry	Alkene (mM)	Additives (mM)	Temp.	Prod.	Conversion (%) ^a
1	2 (100)	None	RT	-	0
2	2 (100)	MgCl ₂ (100)	RT	18	>90
3	2 (100)	NaCl (100)	RT	-	0
4 ^b	23 (30)	MgCl ₂ (80)	37 °C	19	50
5 ^b	24 (130)	MgCl ₂ (130)	37 °C	20	60
6 ^c	25 (75)	MgCl ₂ (160)	37 °C	21	55
7 ^c	26 (66)	MgCl ₂ (130)	37 °C	22	60

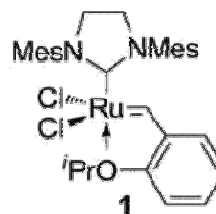
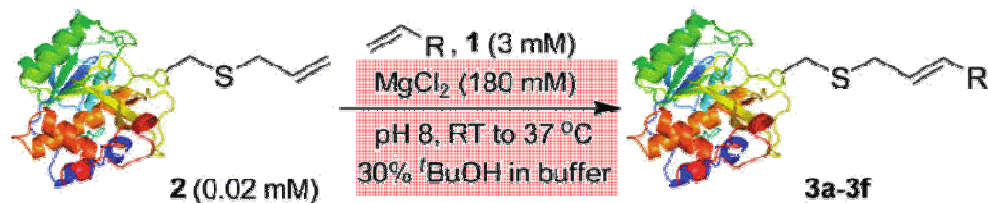
Minor MS signal of **1+17** ...strong chelation?
by LC-MS

MgCl₂ disrupts nonproductive chelation of Ru + protein

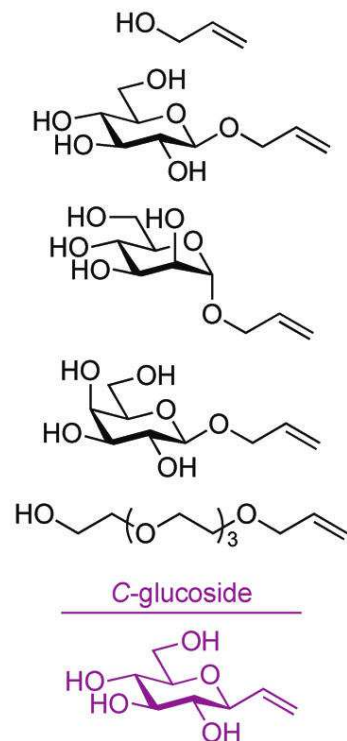
(SBL-S156XXX have no Cys-SH)

4-3. Olefin metathesis (2)-1

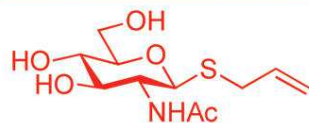
4. Others



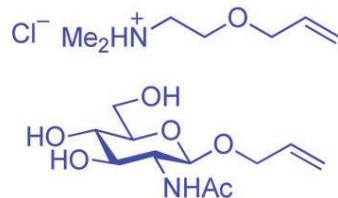
Allyl alcohol, glycosides and OEG



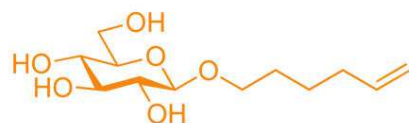
GlcNAc derived allyl sulfide



Allyl ether of ammonium salt and GlcNAc



Hexenyl glucoside



Allyl amine derivatives



Less reactive partner
→cross-metathesis
(entry 1~7)

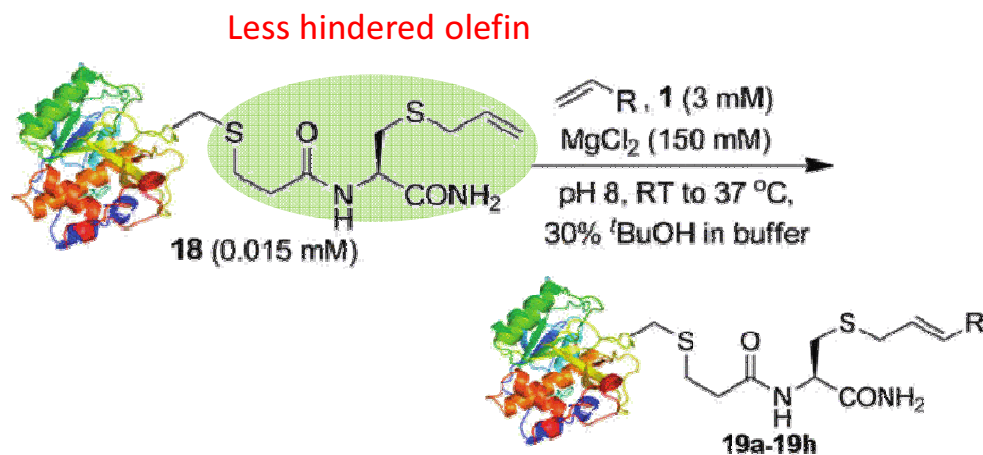
More reactive partner
→self-metathesis
(entry 10)

Much less reactive partner
→no reaction
(entry 8,9,11~14)

Entry	Alkene (mM)	Conditions	Prod.	Conv. (%) ^a
1	$\text{HO-CH}_2\text{-CH=CH}_2$ 4 (180)	RT, 2 h	3a	>95
2	$\text{HO-CH}_2\text{-CH(OH)-CH=CH}_2$ 5 (180)	RT, 2 h then 37 °C, 30 mins	3a	28 >95
3	$\text{HO-CH}_2\text{-CH(OH)-CH}_2\text{-O-CH}_2\text{-CH=CH}_2$ 6 (92)	37 °C, 1 h	3b	30
4	$\text{HO-CH}_2\text{-CH(OH)-CH}_2\text{-O-CH}_2\text{-CH}_2\text{-O-CH}_2\text{-CH=CH}_2$ 7 (89)	37 °C, 1 h	3c	30
5	$\text{HO-CH}_2\text{-CH(OH)-CH}_2\text{-O-CH}_2\text{-CH}_2\text{-O-CH}_2\text{-CH}_2\text{-O-CH}_2\text{-CH=CH}_2$ 8 (92)	37 °C, 1 h	3d	30
6 ^b	$\text{HO-CH}_2\text{-CH}_2\text{-CH}_2\text{-O-CH}_2\text{-CH=CH}_2$ 9 (92)	37 °C, 30 mins	3e	65
7	$\text{HO-CH}_2\text{-CH(OH)-CH}_2\text{-O-CH}_2\text{-CH}_2\text{-O-CH}_2\text{-CH=CH}_2$ 10 (92)	RT, 1 h	3f	>95
8	$\text{Cl}^- \text{Me}_2\text{NH}^+ \text{-CH}_2\text{-CH}_2\text{-O-CH}_2\text{-CH=CH}_2$ 11 (92)	37 °C, 1 h	-	0
9	$\text{HO-CH}_2\text{-CH(OH)-CH}_2\text{-O-CH}_2\text{-CH}_2\text{-O-CH}_2\text{-CH=CH}_2$ 12a (85)	37 °C, 2 h	-	0
10	$\text{HO-CH}_2\text{-CH(OH)-CH}_2\text{-O-CH}_2\text{-CH}_2\text{-O-CH}_2\text{-CH}_2\text{-O-CH}_2\text{-CH=CH}_2$ 13a (85)	37 °C, 1 h	-	0
11	$\text{HO-CH}_2\text{-CH(OH)-CH}_2\text{-O-CH}_2\text{-CH}_2\text{-O-CH}_2\text{-CH}_2\text{-O-CH}_2\text{-CH}_2\text{-O-CH}_2\text{-CH=CH}_2$ 13b (85)	37 °C, 1 h	-	0
12	$\text{HO-CH}_2\text{-CH(OH)-CH}_2\text{-O-CH}_2\text{-CH=CH}_2$ 14 (85)	37 °C, 1 h	-	0
13	$\text{AcHN-CH}_2\text{-CH=CH}_2$ 15 (180)	37 °C, 1 h	-	0
14	$\text{Cl}^- \text{Me}_3\text{N}^+ \text{-CH}_2\text{-CH=CH}_2$ 16 (170)	37 °C, 1 h	-	46 0

4-3. Olefin metathesis (2)-2

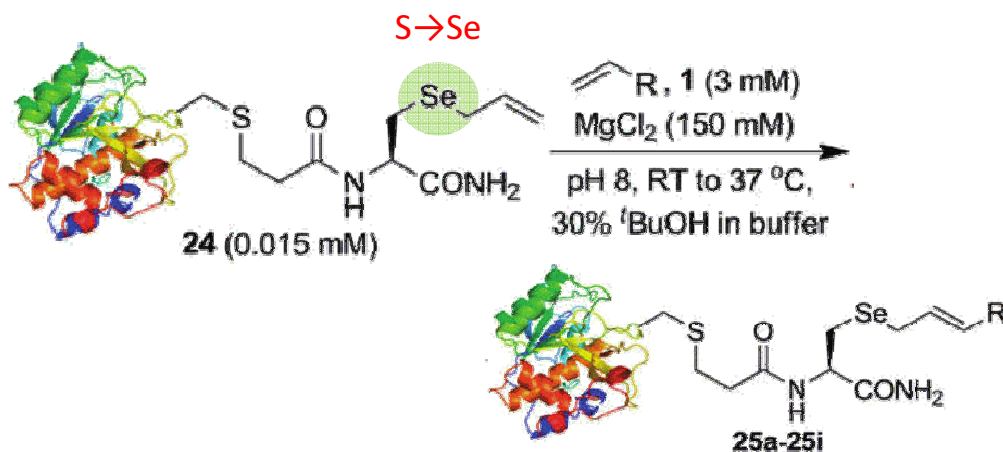
4. Others



Entry	Alkene (mM)	Conditions	Prod.	Conv. (%) ^a	Conv. (hindered)
1	4 (76)	RT, 30 mins	19a	>95	>95 (RT, 2 h)
2	6 (77)	37 °C, 1 h	19b	>95	30
3	7 (72)	37 °C, 1 h	19c	>95	30
4	8 (77)	37 °C, 1 h	19d	>95	30
5	9 (77)	RT, 2 h	19e	>95	65
6	10 (80)	RT, 30 mins	19f	>95	>95
7	11 (77)	37 °C, 30 mins	19g	29	0
8	12a (72)	37 °C, 2 h	19h	53	0
9	13a (77)	37 °C, 1 h	-	0	0
(Only self-metathesis)					
10	14 (72)	37 °C, 1 h	-	0	0
11	15 (77)	37 °C, 30 mins	-	0	0
12	16 (74)	37 °C, 1 h	-	0	0

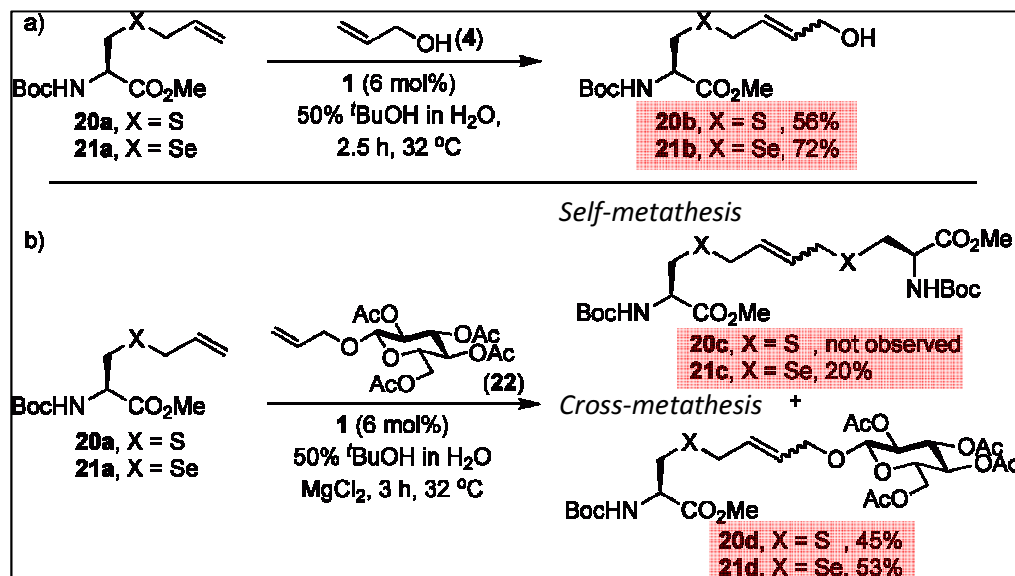
4-3. Olefin metathesis (2)-3

4. Others



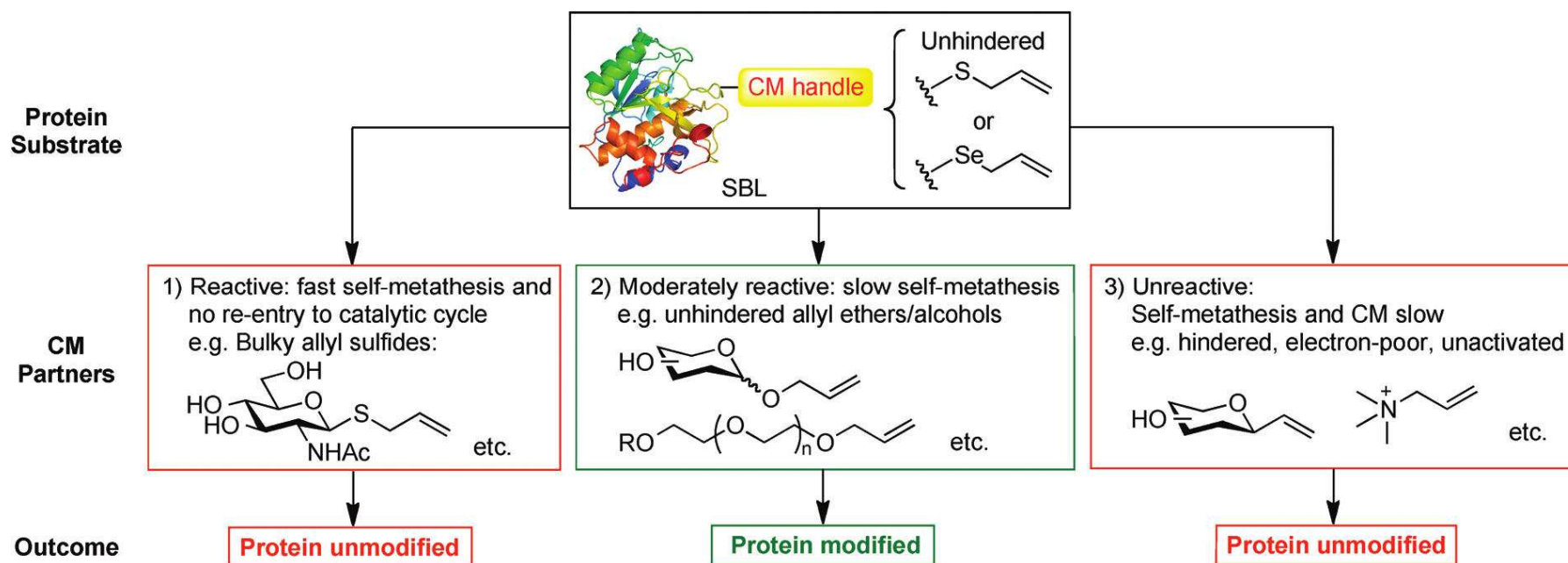
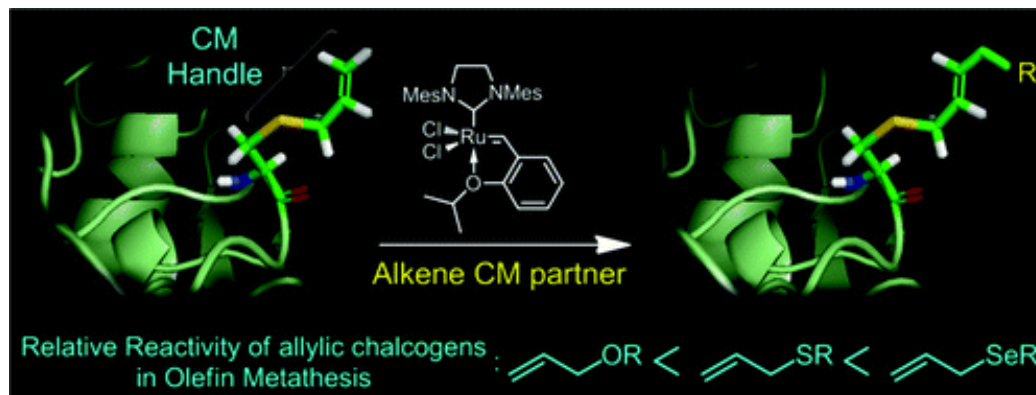
Entry	Alkene (mM)	Conditions	Prod.	Conv. (%) ^a	Conv. (S)
1	4 (76)	RT, 15 mins	25a	>95	>95 (RT, 30 min)
2	6 (77)	37 °C, 1 h	25b	>95	>95
3	7 (74)	37 °C, 1 h	25c	>95	>95
4	8 (77)	37 °C, 1 h	25d	>95	>95
5	9 (77)	RT, 1 h	25e	>95	>95
6	10 (85)	RT, 30 mins	25f	>95	>95
7	11 (76)	37 °C, 30 mins	25g	>95	29
8	12a (73)	37 °C, 1 h	25h	>95	53
9	13a (77)	RT, 1 h	-	0	0
(Only self-metathesis)					
10	14 (82)	37 °C, 1 h	-	0	0
11	15 (77)	37 °C, 30 mins	25i	90	0
12	16 (74)	37 °C, 1 h	-	0	0

S vs Se



4-3. Olefin metathesis (2)-4

4. Others



4-3. Olefin metathesis (3)

4. Others

Water soluble Metathesis catalysts

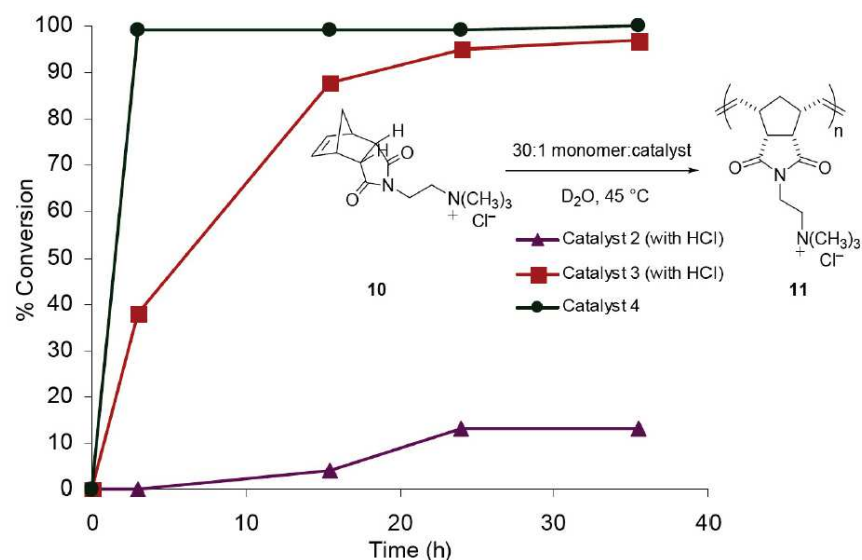
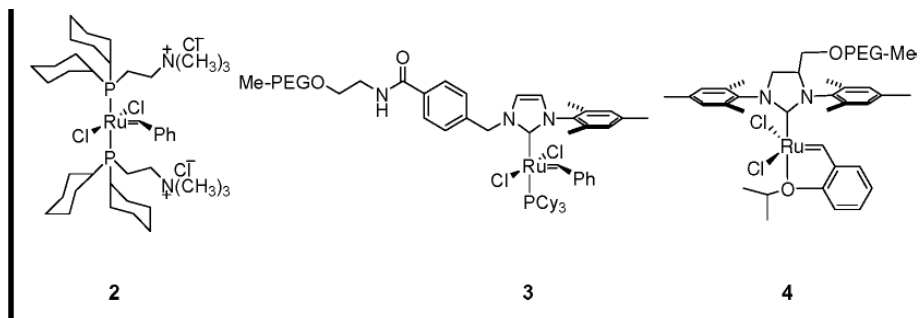


Figure 1. A comparison of the ability of water-soluble catalysts to polymerize *endo*-monomer **10** (data for catalyst **2** and **3** are obtained from ref 4).

PEG : Poly Ethylene Glycol, MW \approx 2000

Table 1. Ring-Closing Metathesis Reactions in Aqueous Media^a

Entry	Substrate	Time	Product	Conversion
1		12 h		>95%
2		24 h		>95%
3		36 h		67% (+28%)
4		24 h		42%
5		24 h		<5%

^a Reactions were carried out at room temperature with 5 mol % of catalyst **4** and an initial substrate concentration of 0.2 M in D₂O or H₂O. Conversions were determined by ¹H NMR spectroscopy.

Table 2. Cross-Metathesis Reactions in Aqueous Media^a

Substrate	Time	Product	Conversion
	12 h		>95% ^b
	12 h		94% ^c

^a Reactions were carried out at 45 °C with 5 mol % of catalyst **4** and an initial substrate concentration of 0.2 M in D₂O or H₂O. Conversions were determined by ¹H NMR spectroscopy. ^b E/Z ~ 15:1. ^c 6% of **24** remains due to thermodynamic equilibrium.

4-4. Rh carbenoid

4. Others

Also see Mr Suzuki's Literature Seminar (2013)

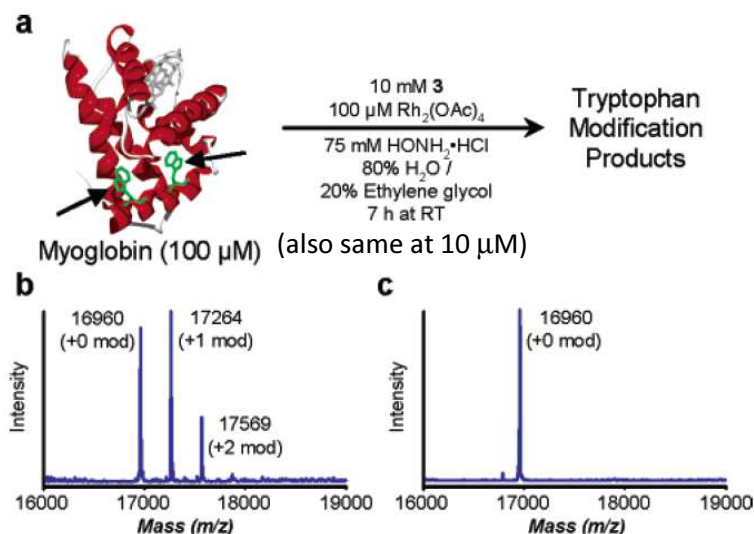
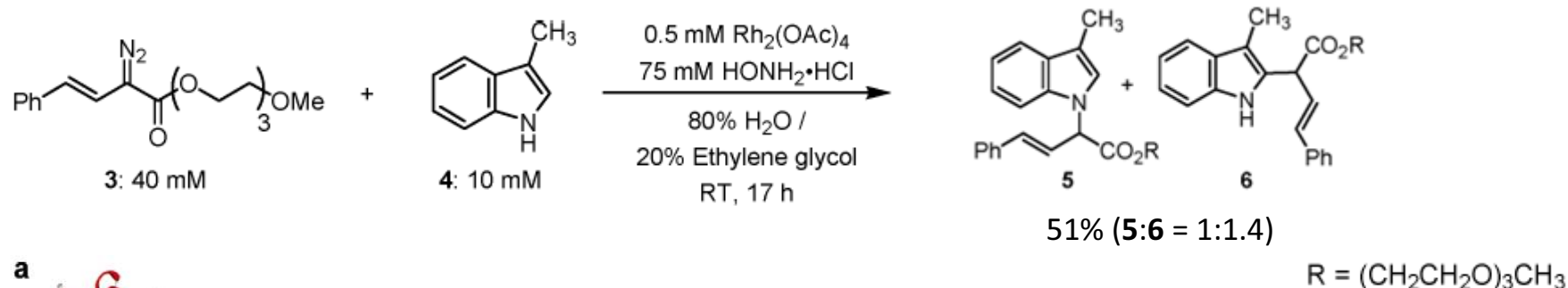


Figure 1. Modification of myoglobin with metalcarbenes. (a) A 100 μM solution of horse heart myoglobin was exposed to **3** and $\text{Rh}_2(\text{OAc})_4$ for 7 h. The two tryptophan residues are shown in green. (b) Following removal of the small molecules via gel filtration, the sample was analyzed by ESI-MS. Both singly and doubly modified protein products were identified in the mass reconstruction. (c) In the absence of $\text{Rh}_2(\text{OAc})_4$, no products were obtained under otherwise identical conditions. (d) After digestion with trypsin, MS/MS analysis of the doubly modified peptide fragment confirmed modification of only the tryptophan residues. All assigned species agree to within 0.1% of the expected mass values.

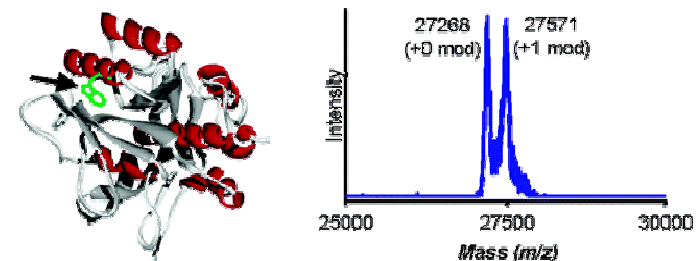


Figure 2. Modification of subtilisin Carlsberg with rhodium carbenoids. Conditions: 100 μM protein, 10 mM **3**, 100 μM $\text{Rh}_2(\text{OAc})_4$ and 75 mM $\text{HONH}_2 \cdot \text{HCl}$ (pH 1.5) in 80% water/20% ethylene glycol, rt, 7 h. Following removal of the small molecules via gel filtration, the sample was analyzed using MALDI-TOF MS. Only the singly modified protein was observed, as would be expected by the single tryptophan residue (shown in green). In the absence of $\text{Rh}_2(\text{OAc})_4$, no reaction occurred under otherwise identical conditions.

These have no Cys ••• Selective?

Contents

1. Introduction

2. Redox catalyst

3. Pd catalyst

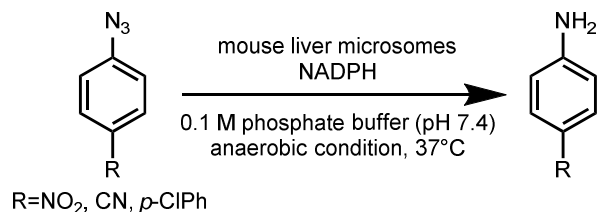
4. Others

5. Summary

Appendix

2-1. R-SH oxidation (4)

Appendix



Effect of microsomal incubation conditions on the rate of formation of *p*-nitroaniline from *p*-nitrophenyl azide.

Incubation conditions	Rate (percentage of control) ^a
Complete system ^b	100
Omit glucose 6-phosphate dehydrogenase ^c	4 ± 1
Heat-denatured microsomes	0 ± 1
Plus NADPH (2 mM)	95 ± 3
Plus NADH (2 mM)	28 ± 1
Plus CO ^d	84 ± 3
Aerobic ^e	0 ± 1

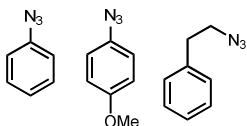
^a Rate determined as nmol of *p*-nitroaniline/30 min per mg of protein.

^b Complete NADPH-generating system under N₂.

^c Glucose 6-phosphate dehydrogenase omitted from the NADPH-generating system.

^d Vials gassed with CO in the same manner as with N₂.

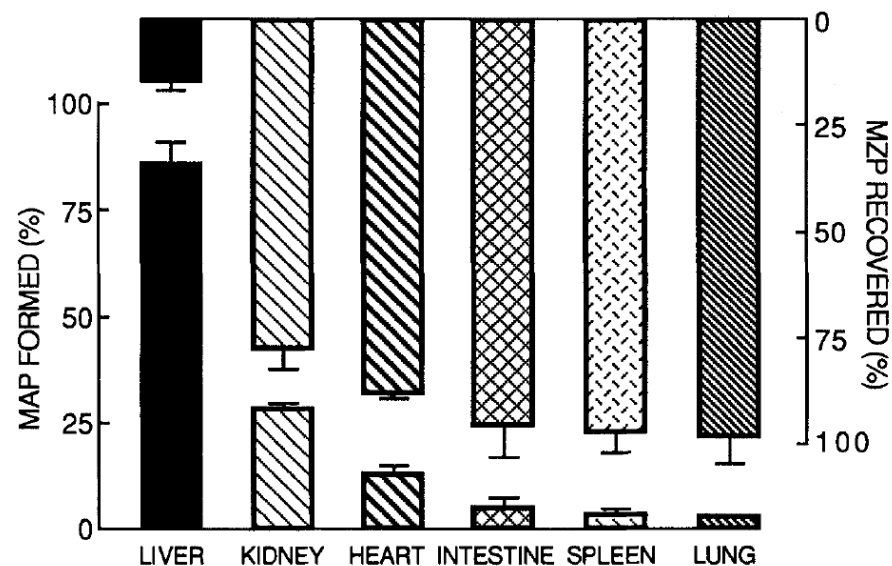
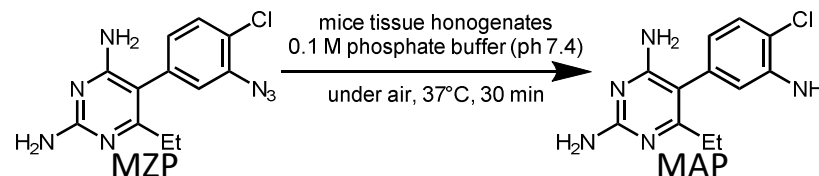
^e Vials sealed with an air atmosphere.



: no reduction occurred

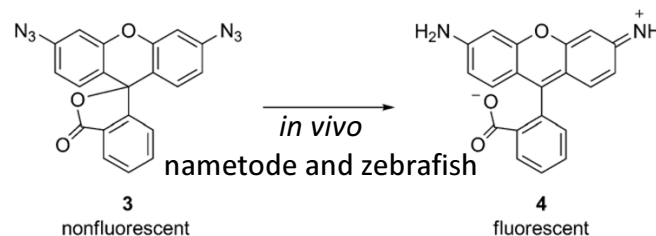
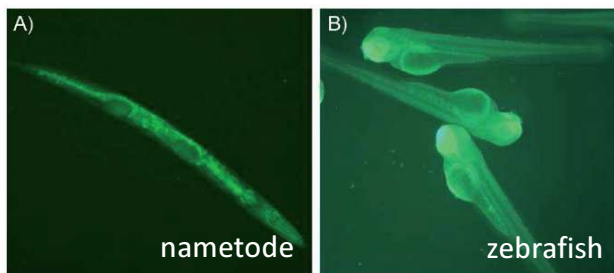
R. J. Griffin *et al. Xenobiotica* **1991**, *21*, 935-943

Ar-N₃ is reduced in Liver, microsome



J. A. Slack *et al. Xenobiotica* **1988**, *18*, 1157-1164

Fluorescence observed *in vivo*, due to reductive metabolism of Ar-N₃

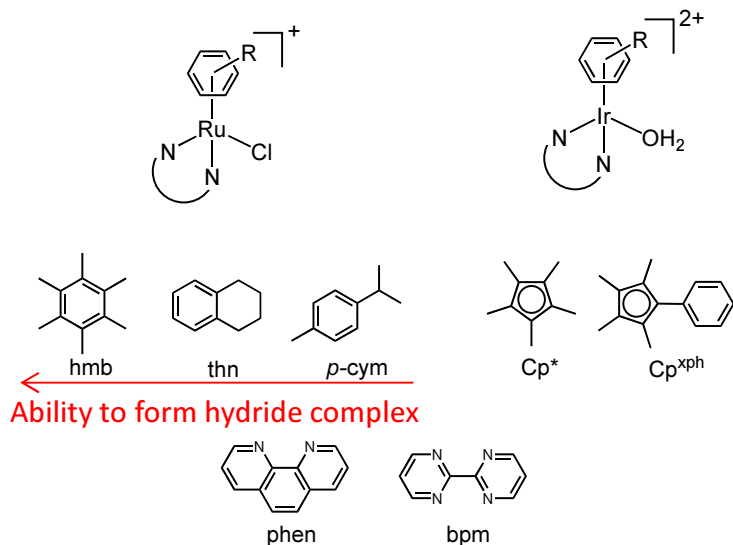


55

E. Meggers *et al. ChemBioChem* **2012**, *13*, 1116-1120

2-3. Reduction by NADH (1)

Appendix



Ability to form [Ru-H]

Table S1. ¹H NMR chemical shifts (ppm) of Ru-H species detected at various times during the reactions of 1,4-NADH with [(η⁶-arene)Ru(N,N')Cl]⁺ complexes **1**, **2**, **4**, and **5** in 90% H₂O/10% D₂O at 310 K.

Complex	Time (min)	Ru-H Chemical shift (ppm)
(1) [(η ⁶ - <i>p</i> -cym)Ru(bpm)Cl] ⁺	31	-6.13
(2) [(η ⁶ -hmb)Ru(bpm)Cl] ⁺	15	-7.44
(4) [(η ⁶ -thn)Ru(bpm)Cl] ⁺	22	-6.61
(5) [(η ⁶ - <i>p</i> -cym)Ru(phen)Cl] ⁺	34	-6.21

[Ru-H] (also [Ir-H]) complex reduce pyrvate

Time course

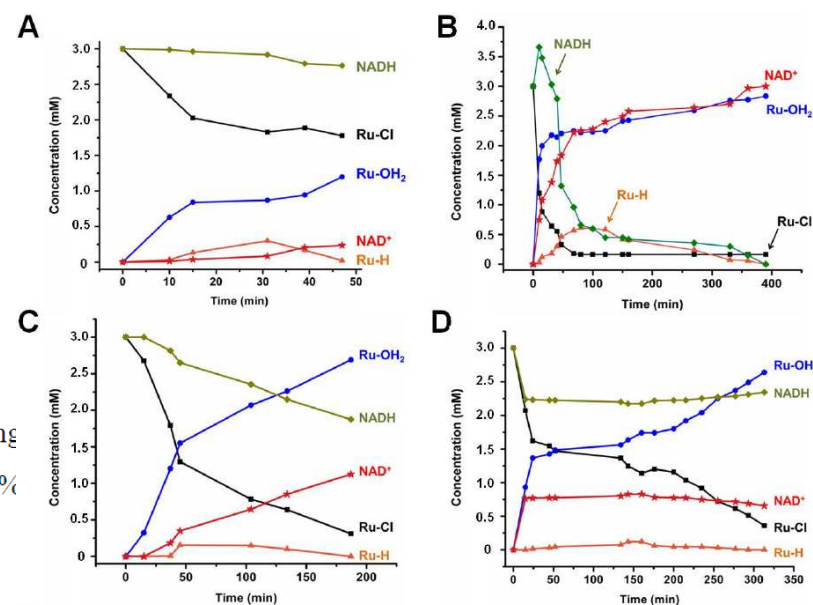


Figure S2. Time dependence of the concentrations of various species observed during the reaction of NADH with ruthenium arene complexes (as determined by integration of ¹H NMR peaks). The solutions contained equimolar amounts of 3 mM 1,4-NADH and (A) [(η⁶-*p*-cym)Ru(bpm)Cl]⁺ (**1**); (B) [(η⁶-hmb)Ru(bpm)Cl]⁺ (**2**); (C) [(η⁶-thn)Ru(bpm)Cl]⁺ (**4**); (D) [(η⁶-*p*-cym)Ru(phen)Cl]⁺ (**5**), in 90% H₂O/10% D₂O at 310 K.

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2-3. Reduction by NADH (1)

Appendix

[Ir-H] complex formed by NADH

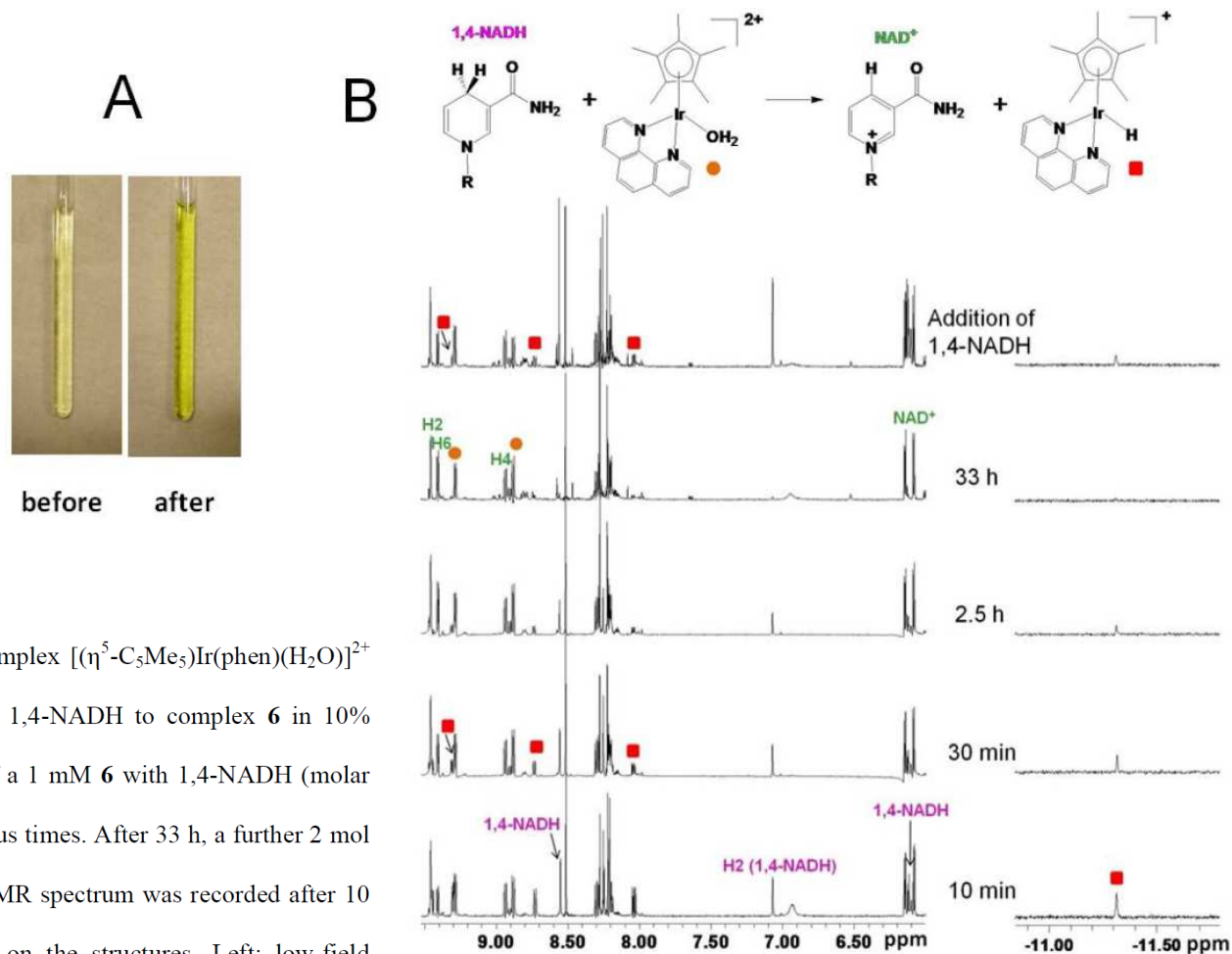


Figure S3. Conversion of 1,4-NADH to NAD⁺ by complex $[(\eta^5\text{-C}_5\text{Me}_5)\text{Ir}(\text{phen})(\text{H}_2\text{O})]^{2+}$ (**6**). (A) Colour change immediately after addition of 1,4-NADH to complex **6** in 10% MeOD-*d*₄/90% H₂O at 298 K. (B) ¹H NMR spectra of a 1 mM **6** with 1,4-NADH (molar ratio 1:2) in 10% MeOD-*d*₄/90% H₂O at 298 K at various times. After 33 h, a further 2 mol equiv of 1,4-NADH was added and the resulting ¹H NMR spectrum was recorded after 10 min (top spectrum). Peak assignments are indicated on the structures. Left: low-field region; right: Ir-H hydride region (×2).

2-3. Reduction by NADH (1)

[Ir-H] complex reduces pyruvate

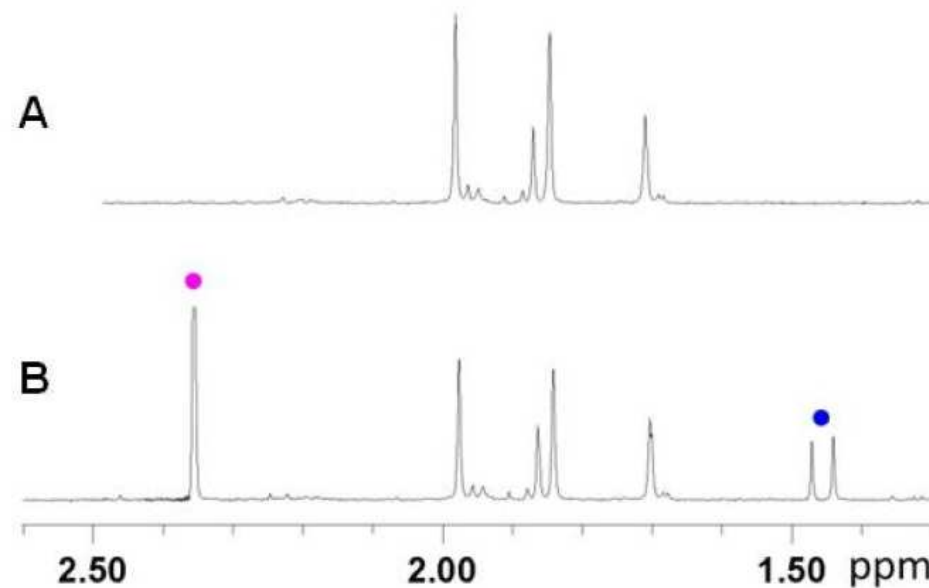
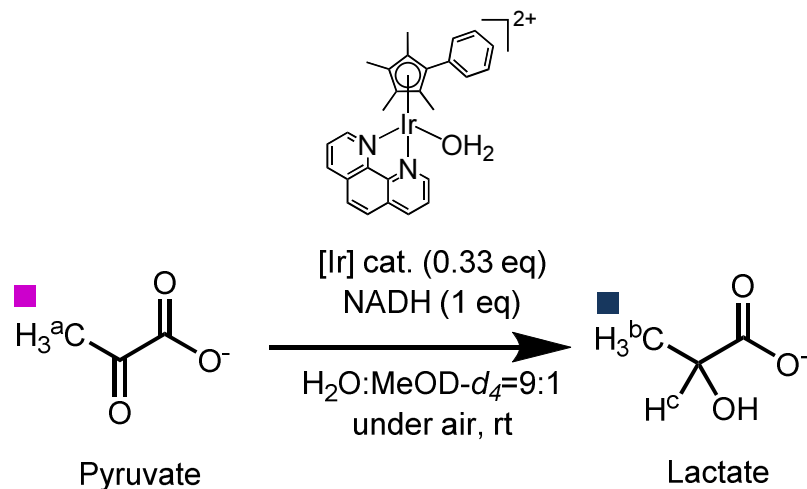


Figure S5. ^1H NMR spectra showing conversion of pyruvate to lactate in the presence of $[(\eta^5\text{-C}_5\text{Me}_4\text{C}_6\text{H}_5)\text{Ir}(\text{phen})(\text{H}_2\text{O})]^{2+}$ (**7**). (A) ^1H NMR spectrum recorded 10 min after addition of 3 mol equiv of 1,4-NADH to a 1 mM solution of **7** in 10% $\text{MeOD-}d_4/90\%$ H_2O (v/v) at 298 K. (B) ^1H NMR spectrum recorded 10 min after addition of 3 mol equiv of sodium pyruvate to the solution in (A). Peak assignments are indicated on the structures.

2-3. Reduction by NADH (2)

Appendix

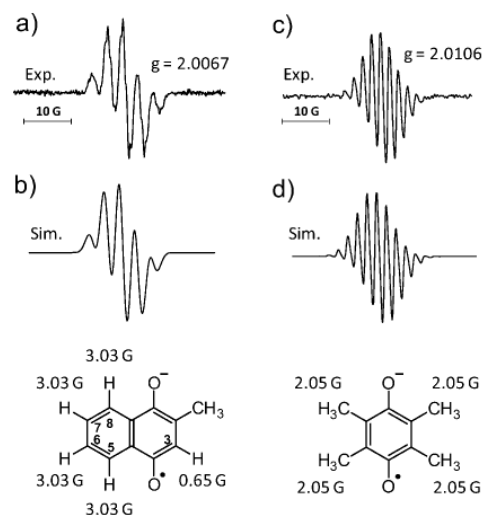
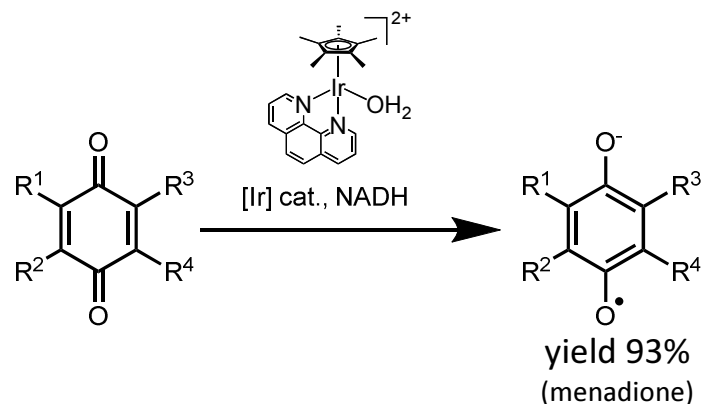
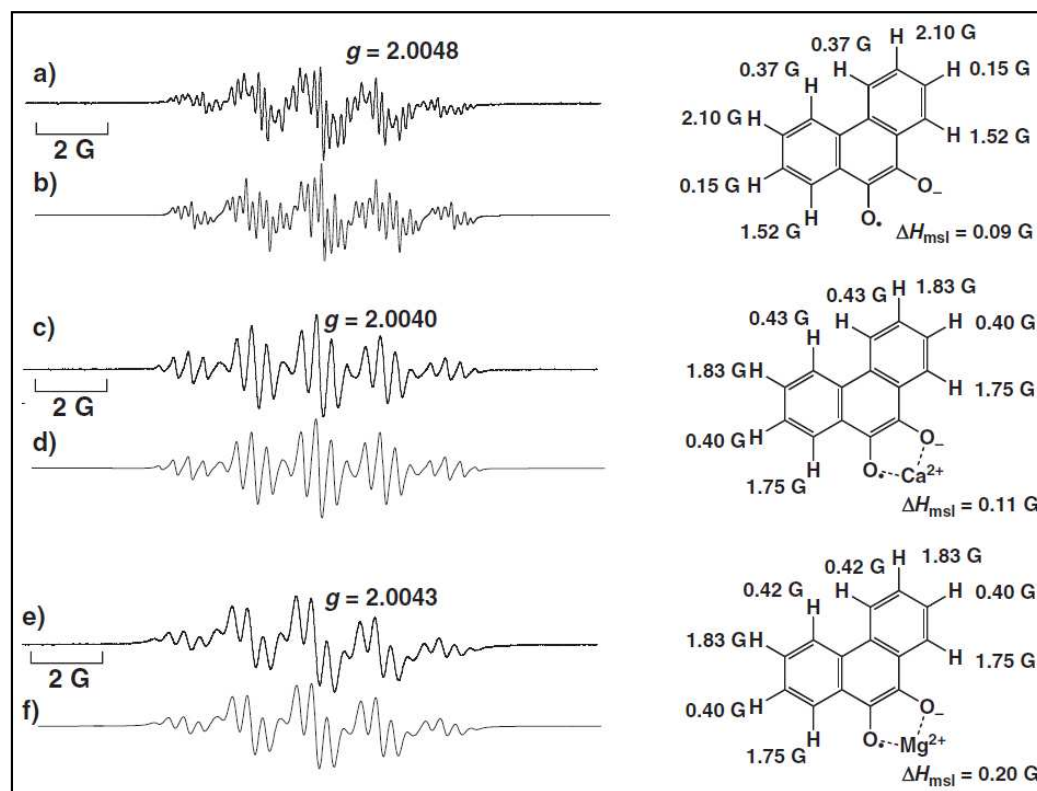


Figure 2. X-band EPR spectra at around 290 K. a) Menadione radical anion ($M^{\bullet-}$) generated by the reduction of menadione (1 mM) by NADH (0.5 mM) catalyzed by complex 1 or 3 (160 μ M) in phosphate buffer (pH 7.2, 9 h). b) Simulated $M^{\bullet-}$ EPR spectrum with hyperfine coupling constants. c) Durosemiquinone radical anion ($D^{\bullet-}$) generated by reduction of duroquinone (2 mM) by NADH (1 mM) catalyzed by complex 1 or 3 (330 μ M) in phosphate buffer (pH 7.2, 10 h). d) Simulated $D^{\bullet-}$ EPR spectrum with hyperfine coupling constants.

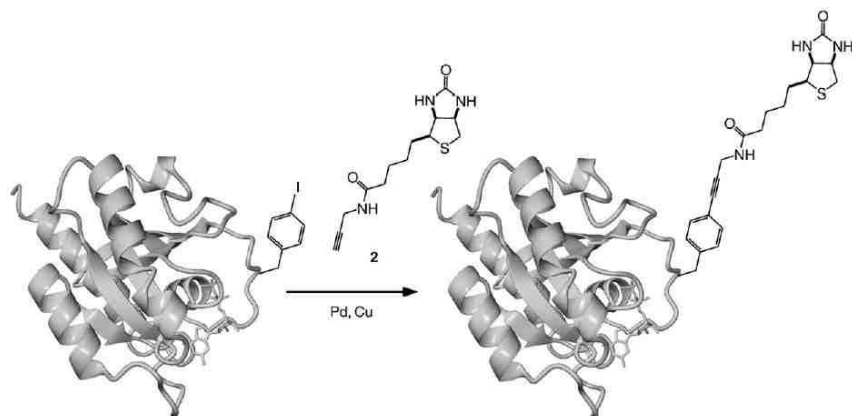
Few semiquinone radical anions ($M^{\bullet-}$) forms [Ir] adducts



S. Fukuzumi *et al.* *ChemPhysChem* 2006, 7, 942-954

3-2. Sonogashira coupling (1)

Appendix



Yield 25%
(13%:dehalogenated)

Condition

in TAPS buffer (0.09 M, pH 8.3)
2.3 M DMSO (18% v/v)
iF32-Ras-His (13 μ M)
propagylated biotin (15 mM)
Pd(OAc)₂ (1.7 mM)
TPPTS (8.3 mM)
CuOTf (0.7 mM)
6°C, 80min, w/o O₂

Detection by SDS-PAGE

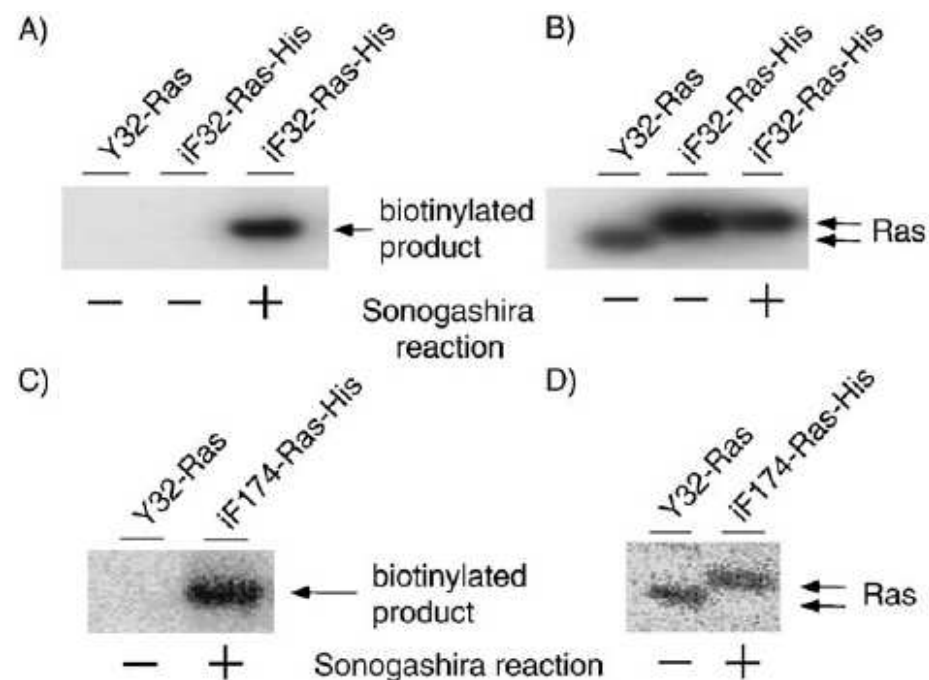


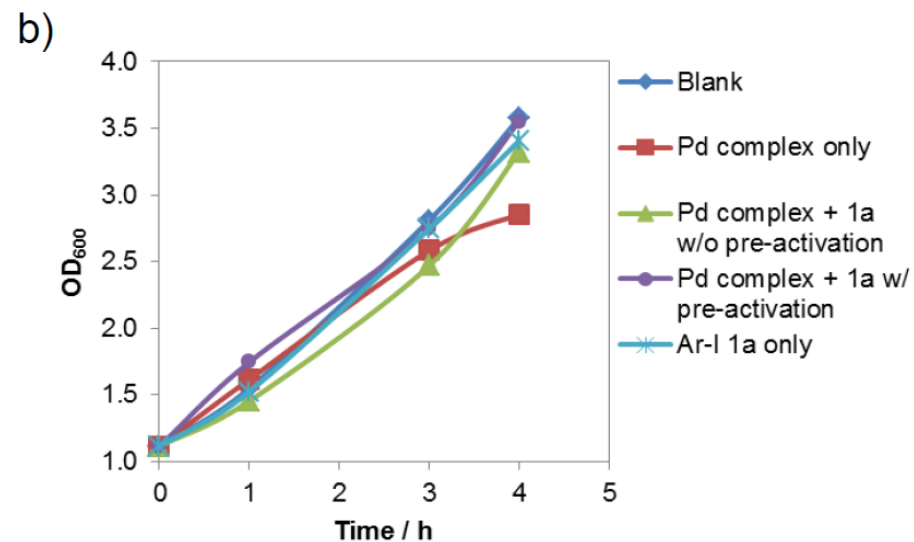
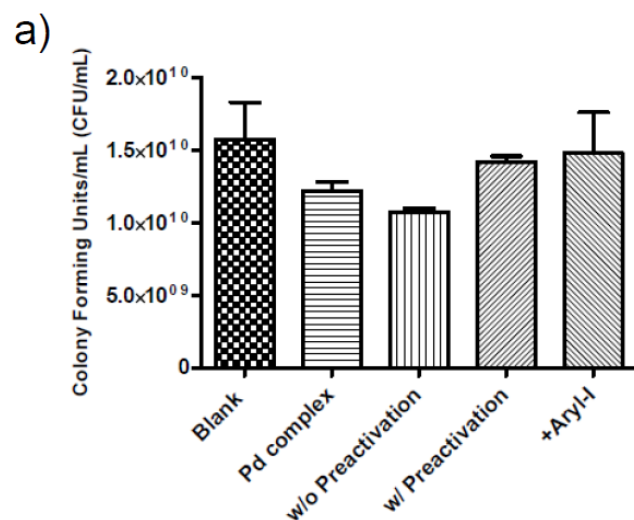
Figure 3. Detection of the biotinylated Ras protein with Streptavidin-HRP (left) or an anti-Ras antibody (right). The A), B) iF32-Ras-His, and C), D) iF174-Ras-His proteins were subjected to the Sonogashira reaction and then were directly analyzed without further purification. The wild-type Ras protein without hexahistidine residues (Y32-Ras) was used as a control.

3-2. Sonogashira coupling (2)

Appendix

E. coli cytotoxicity of Pd complex

Each treatment
→incubate 4 h
→dilute 10^6
→overnight incubation 37°C
→measurement



Blank : DMSO 1 %

Pd complex : DMSO 1 %, Pd complex 1 mM

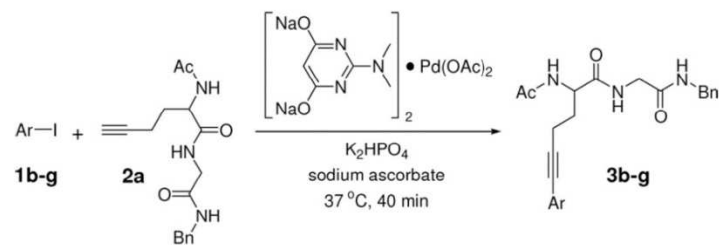
w/o Preactivation : DMSO 1 %, Pd complex 1 mM, **1a** 1 mM, sodium ascorbate 8 mM

w Preactivation : DMSO 1 %, preactivated [Pd complex 1 mM, **1a** 1 mM, sodium ascorbate 8 mM] (stirred 37°C , 1 h before addition)

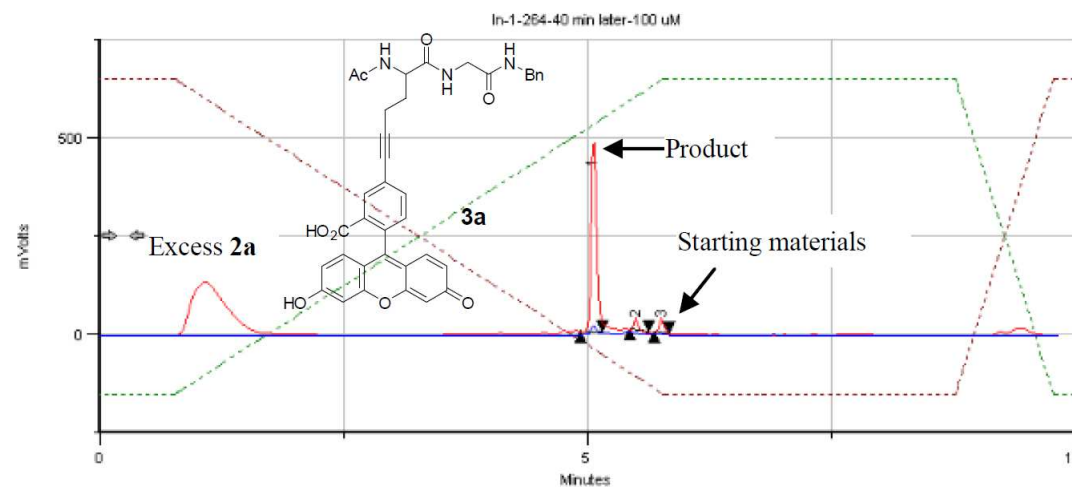
Ar-I : DMSO 1%, **1a** 1 mM

3-2. Sonogashira coupling (2)

Appendix



Entry	Ar-I	Pd • L2	Conversion (%) ^b
1	1b	30%	84
2	1c	30%	95
3	1d	30%	77
4	1e	30%	88
5	1f	30%	70
6	1g	30%	72



The conversions were calculated based on the **disappearance of aryl iodides**, whose amounts were quantified by their absorption at 254 nm (red line).

Reagents (except alkene) are mixed then stirred $37\text{ }^\circ\text{C}$ 1 h, separate each half and added **twice** in all reaction.

Many doubtful points...?

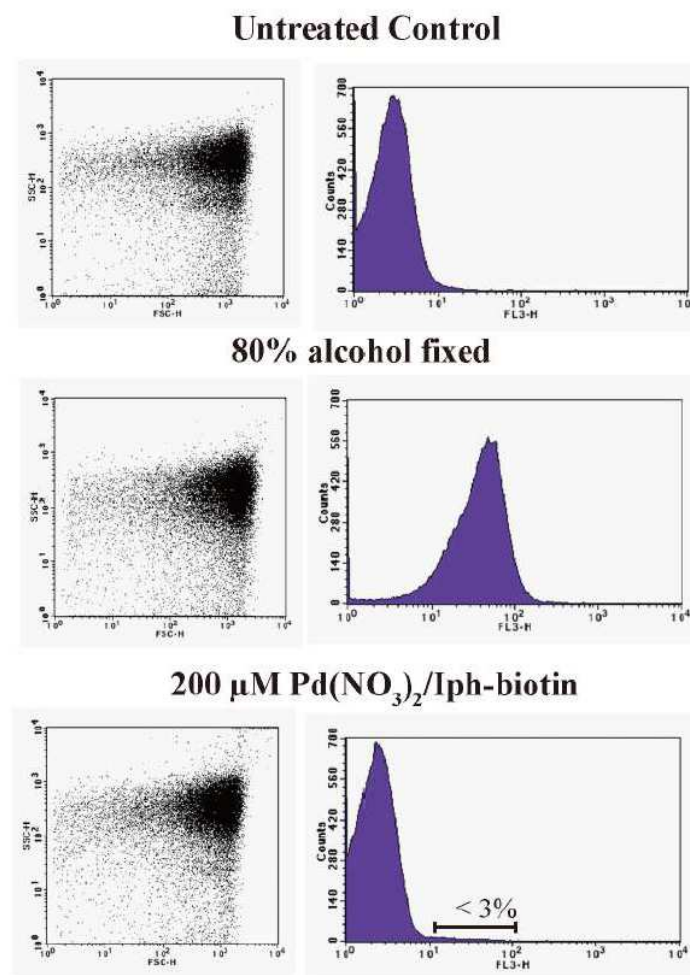
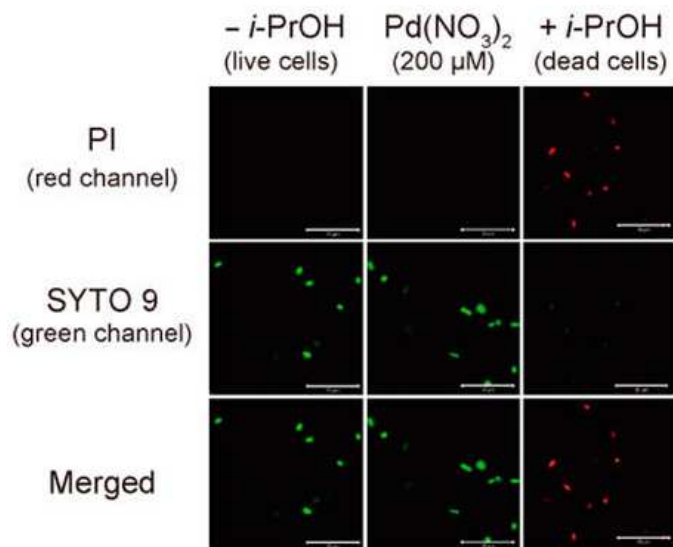
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3-2. Sonogashira coupling (3)

Appendix

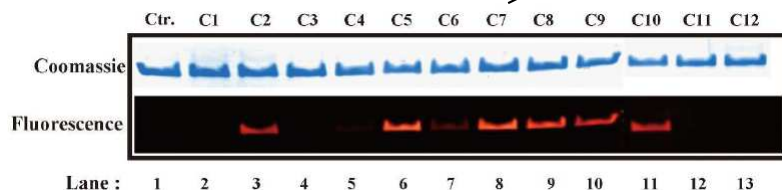
$\text{Pd}(\text{NO}_3)_2$ has no acute toxicity to *E. coli*

$\text{Pd}(\text{NO}_3)_2$ gives few membrane damage



cf) SDS-PAGE fluorescence yield

Yield < 95% by LC-MS



$$RF = \frac{(r.f.i. / r.c.i.)_x}{(r.f.i. / r.c.i.)_{max}}$$

value of C7 as $(r.f.i./r.c.i.)_{max}$
yield = RF × 95 (%)

r.f.i. : relative fluorescence intensity
r.c.i. : relative coomassie stain intensity

Shigella shows almost same results

3-2. Sonogashira coupling (3)

Appendix

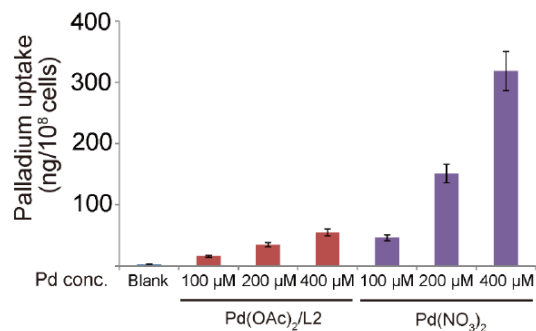
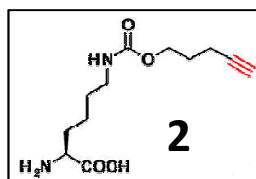
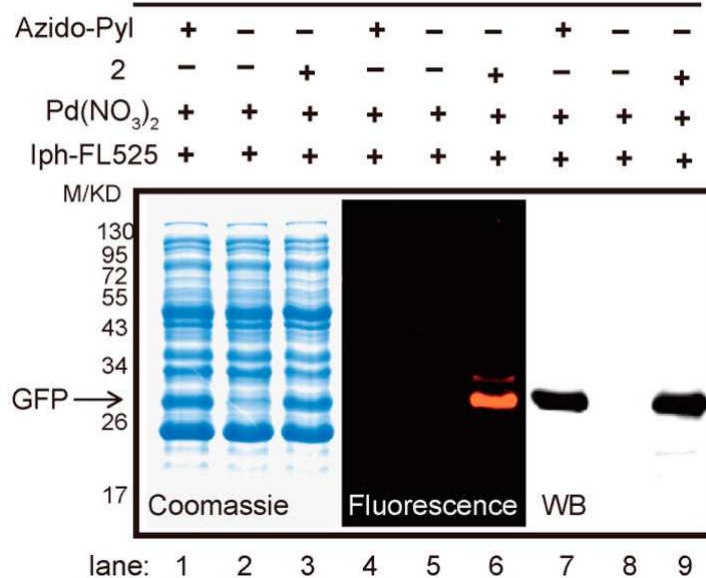


Figure S13. The uptake of palladium measured by ICP-MS analysis. *E. coli* bacterial cells with and without the treatment of 200 μM Pd(OAc)₂/L2 or Pd(NO₃)₂ for 1 h were analyzed according to a protocol described in **Supplementary Method 11**.

	Blank	Pd(OAc) ₂ /L2			Pd(NO ₃) ₂		
Extracellular Pd conc.	N.A.	100 μM	200 μM	400 μM	100 μM	200 μM	400 μM
Pd (ng/10 ⁸ cells) (±10%)	3	16	35	55	46	151	318

GFP-N149TAG (*E. coli*)

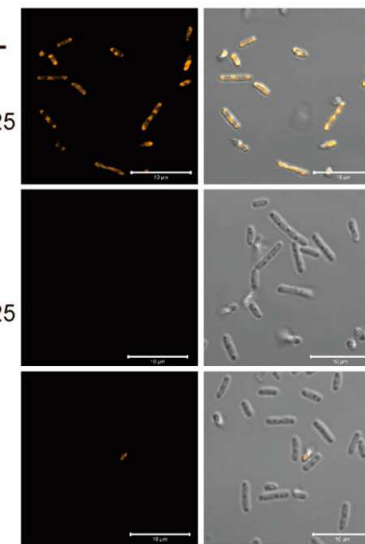


OspF-K102TAG (*Shigella*)

2(1 mM)
+Pd(NO₃)₂/Iph-FL525

Azido-Pyl(1 mM)
+Pd(NO₃)₂/Iph-FL525

2(1 mM)
+Iph-FL525



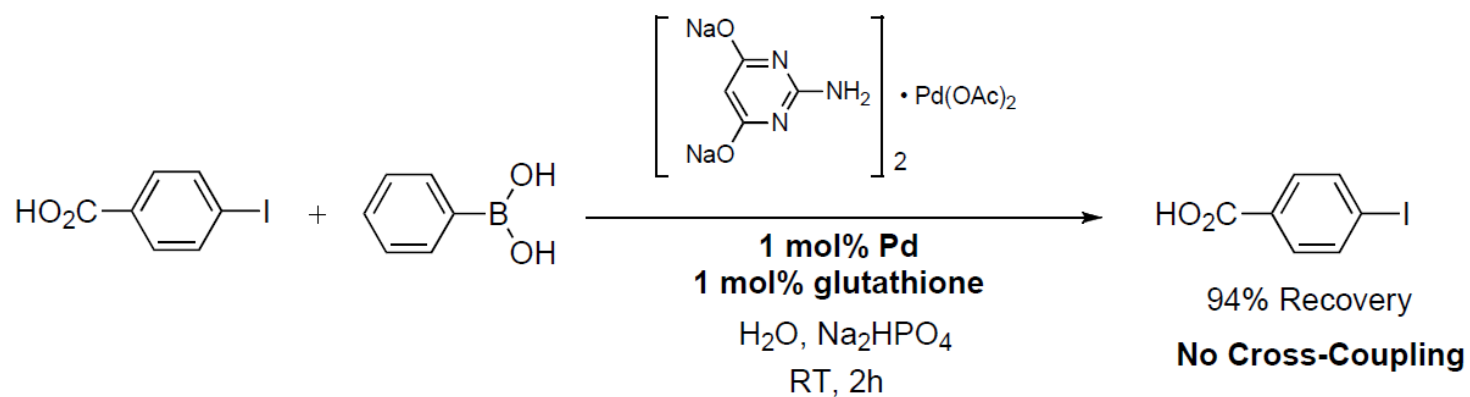
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3-3. Suzuki coupling (1)

Appendix

Free Cys-SH inhibits coupling

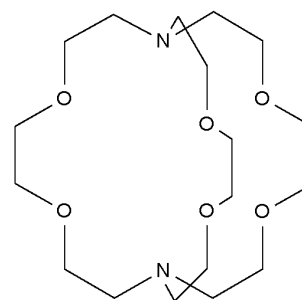


3-3. Suzuki coupling (3)

Toxicity of Guanidine salt

LD₅₀: 500 mg/kg (domestic rabbit)

月刊ファインケミカル 2008年6月号 72-75



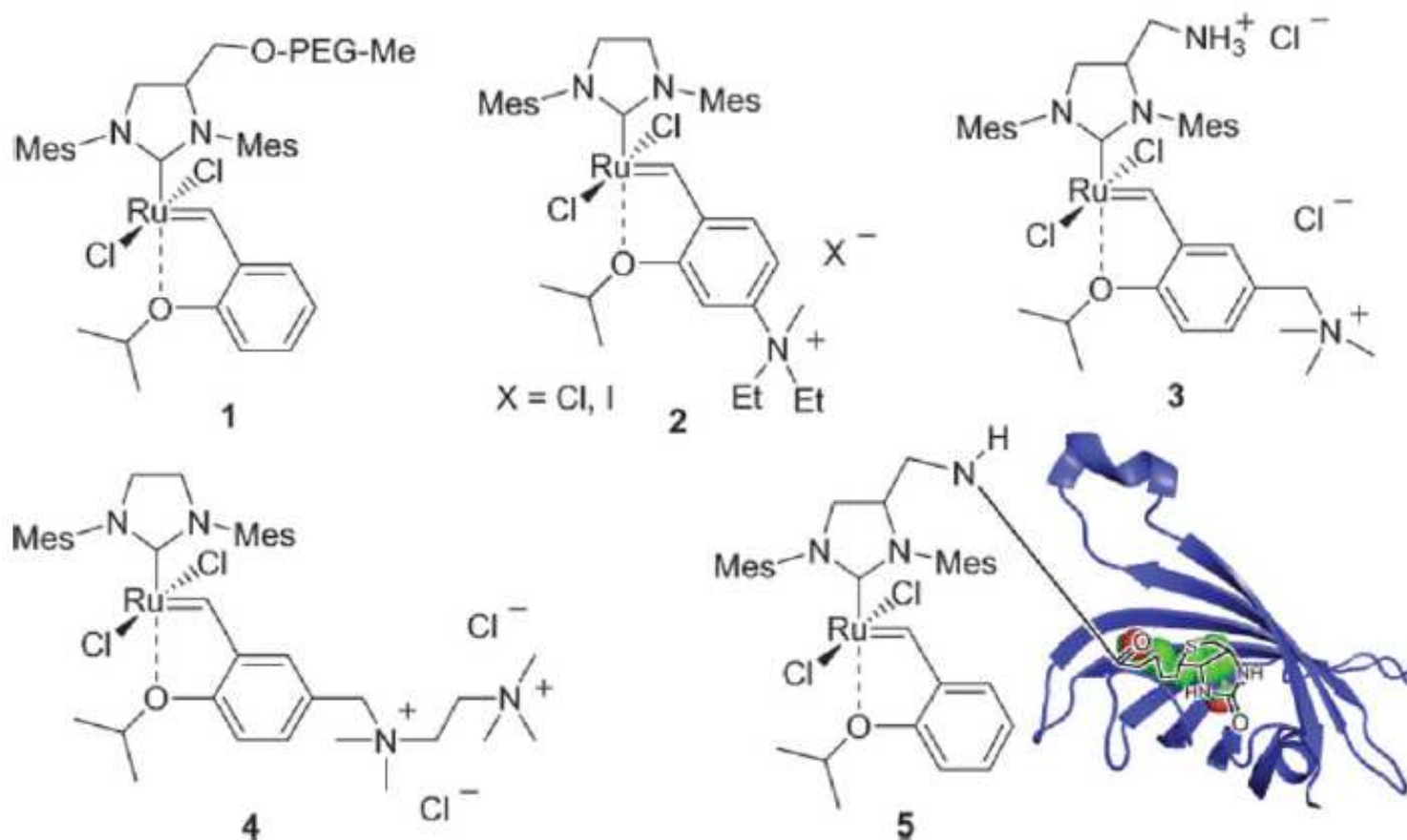
Kryptofix 2.2.2

K⁺ trap

4-2. Olefin metathesis (3)

Appendix

Other water soluble Olefin metathesis catalysts



4-4. Rh carbenoid

4. Others

S-h insertion of Rh Carbenoid

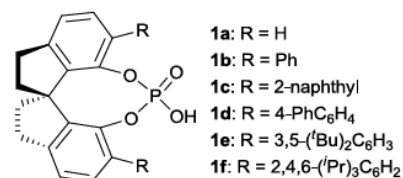
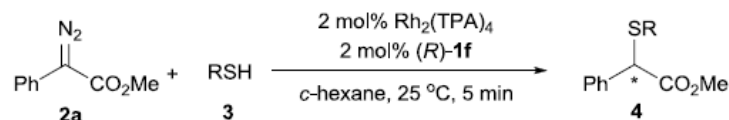


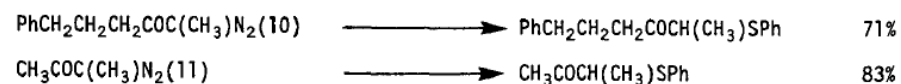
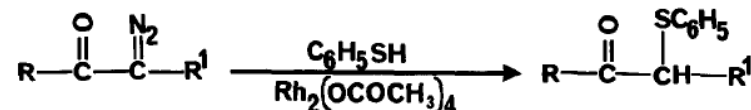
Table 3 S-H insertions of methyl α -phenyl- α -diazo-esters with various mercaptans^a



Entry	R	Product	Yield (%)	ee (%)
1	Bn (3a)	4aa	92	94 (<i>S</i>)
2	4-MeO-Bn (3b)	4ab	97	93
3	4-Cl-Bn (3c)	4ac	94	94
4	<i>n</i> -dodecyl (3d)	4ad	86	93
5	<i>n</i> -octyl (3e)	4ae	87	93
6	<i>n</i> -Pr (3f)	4af	86	93
7	<i>i</i> -Bu (3g)	4ag	89	87
8	<i>i</i> -Pr (3h)	4ah	89	78
9 ^b	trityl (3i)	4ai	88	94
10	EtO ₂ CCH ₂ (3j)	4aj	83	98
11	2-furylmethyl (3k)	4ak	85	96
12 ^b	4-MeOC ₆ H ₄ (3l)	4al	91	77

^a The reaction conditions and analysis methods were the same as those described in Table 1, entry 12. ^b Using (*S*)-1f.

Q-. L. Zhou *et al.* *Chem. Sci.* just accepted



P. Ratananukul *et al.* *Tetrahedron Lett.* **1982**, 23, 2509-2512

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