

Total synthesis of (\pm)-Cephanolides

**2021.06.19 Literature Seminar
M1 Kyohei Oga**

contents

1. Introduction

1-1. Total synthesis of Cephalolides (Zhao group)

1-2. Asymmetric total synthesis of Cephalolide (Gao group)

2. Total synthesis of Cephalolides A–D (main paper)



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Communication

Total Synthesis of the *Cephalotaxus* Norditerpenoids (\pm)-Cephalolides A–D

Maximilian Haider,[§] Goh Sennari,[§] Alina Eggert, and Richmond Sarpong*

contents

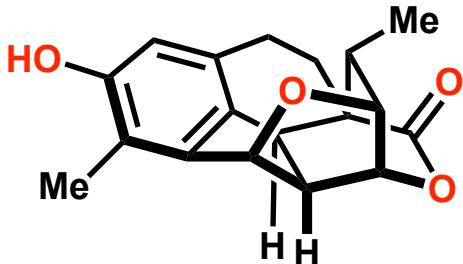
1. Introduction

1-1. Total synthesis of Cephalolides (Zhao group)

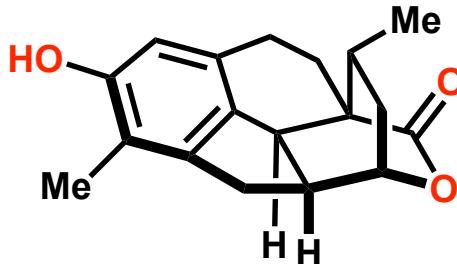
1-2. Asymmetric total synthesis of Cephalolide (Gao group)

2. Total synthesis of Cephalolides A–D (main paper)

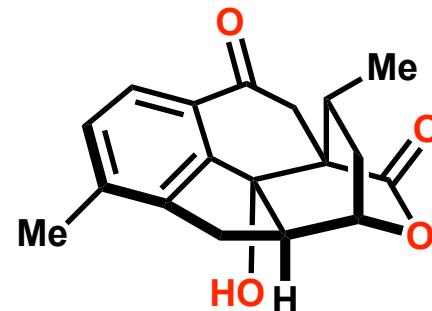
Cephanolide C18 dinorditerpenoids



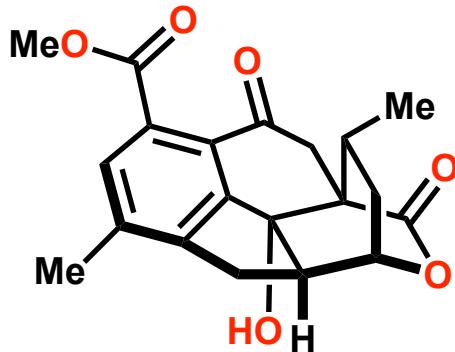
cephanolide A



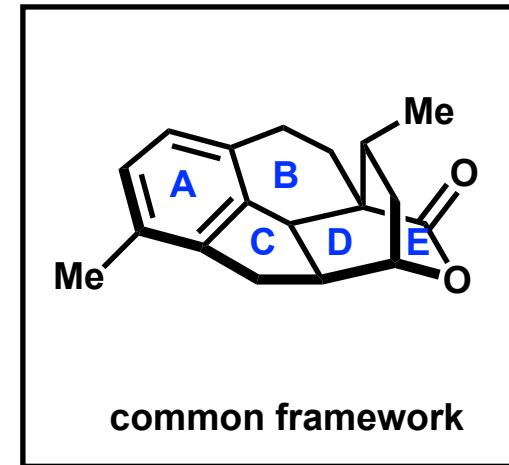
cephanolide B



cephanolide C



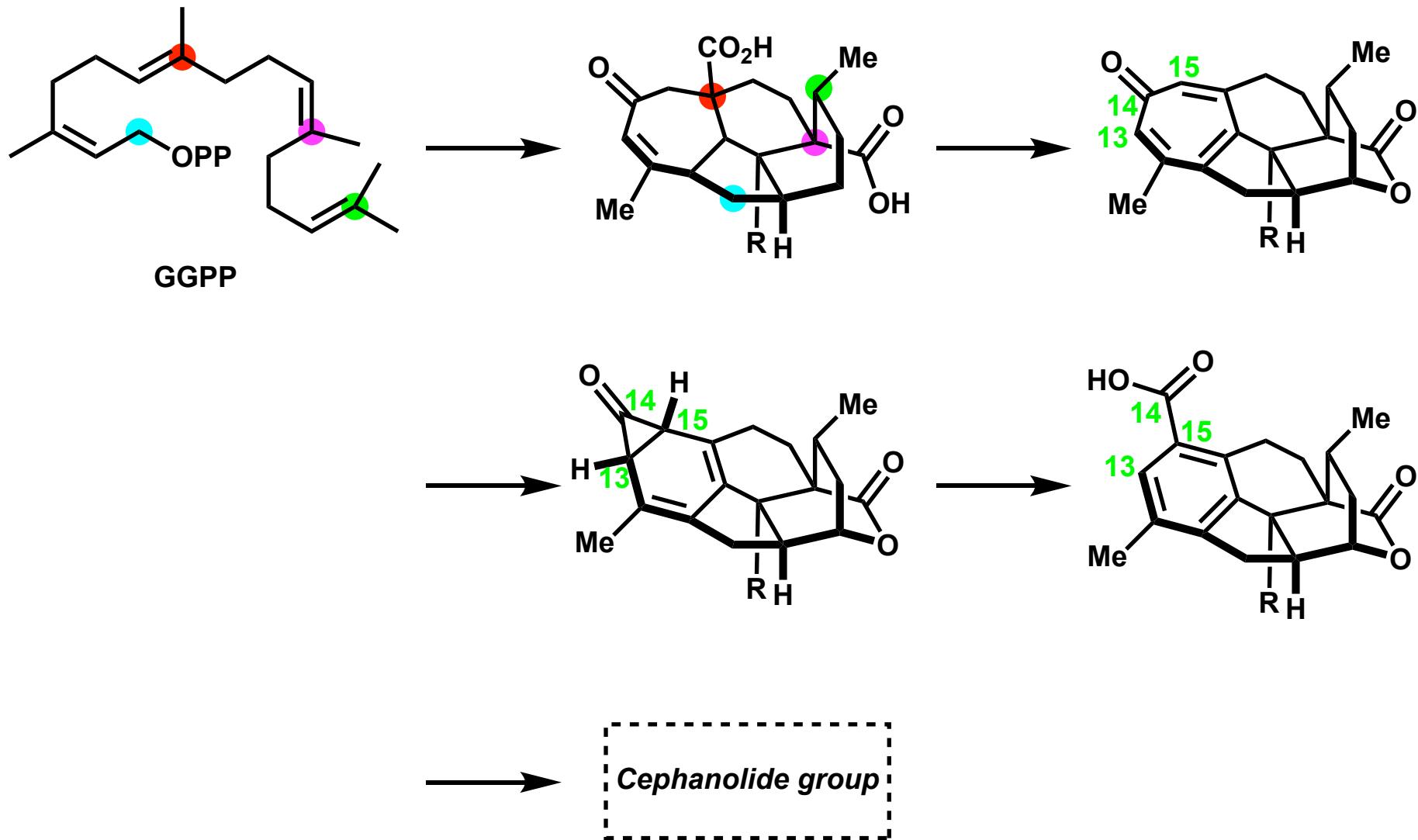
cephanolide D



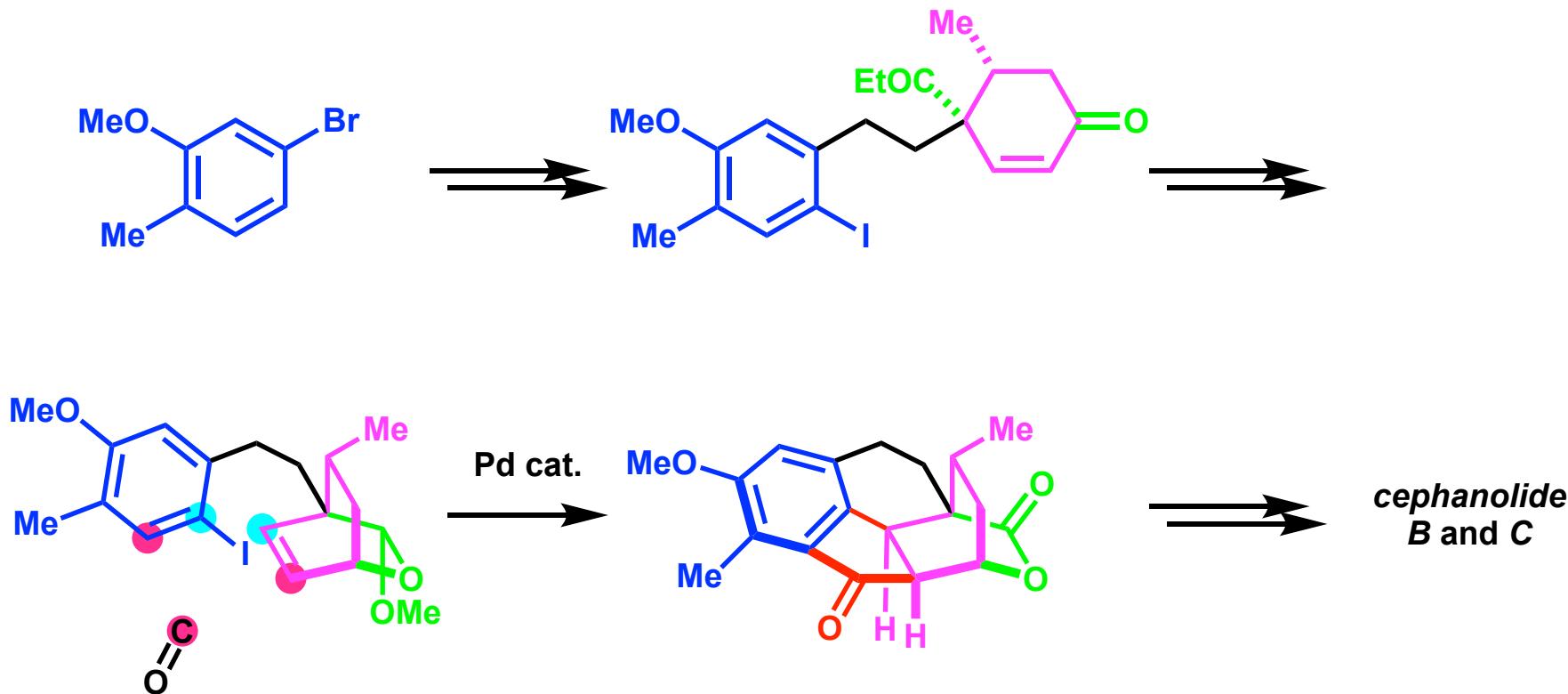
common framework

- isolation : the seeds of *cephalotaxus harringtonia*
- biological activity : plant growth inhibition, antineoplastic, antiviral, antitumor
- structure features : five ring system (A,B,C,D,E)
- total synthesis : two papers

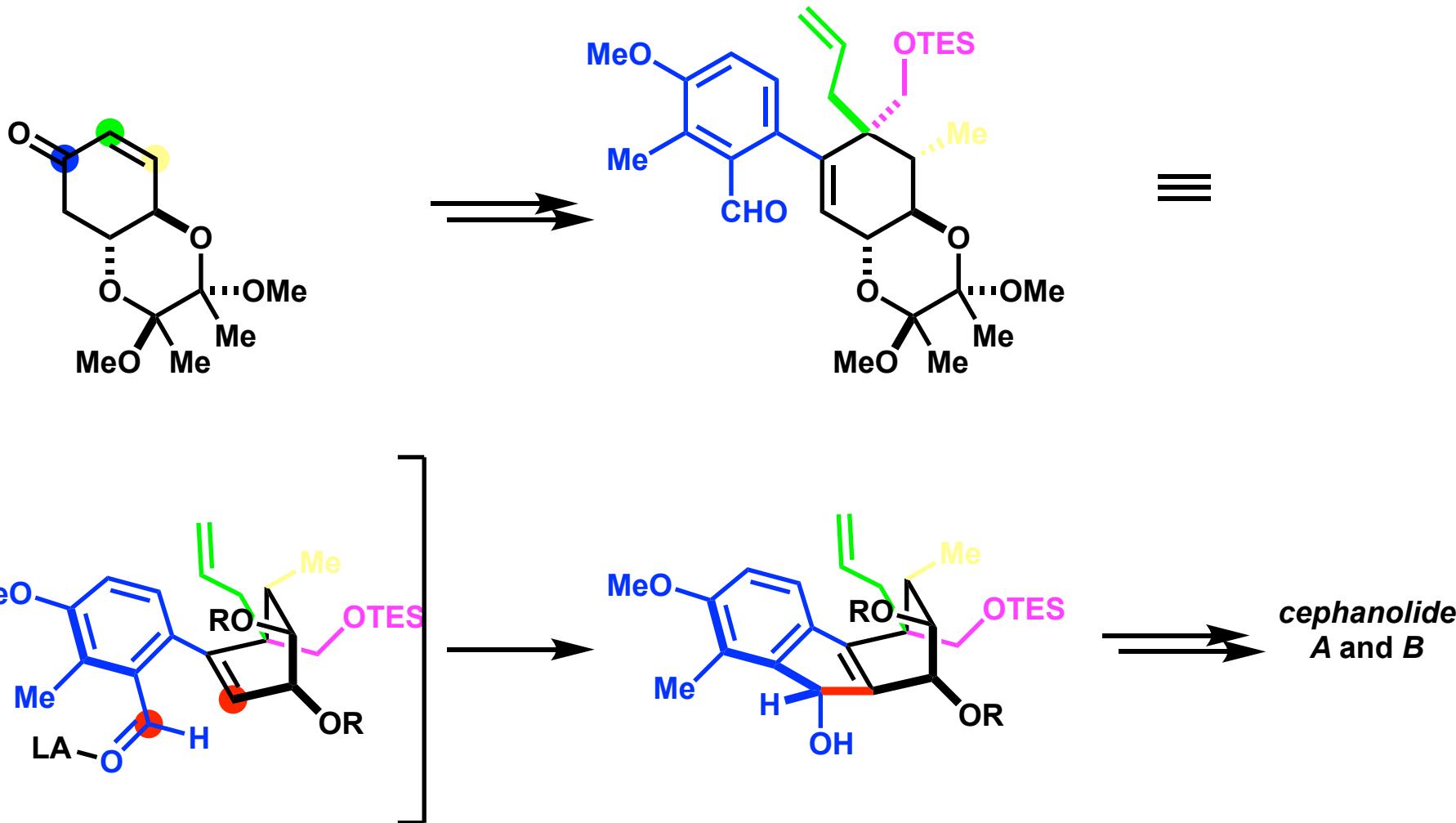
Biosynthesis of Cephanolides



Zhao's total synthesis



Gao's total synthesis



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Communication

Total Synthesis of the *Cephalotaxus* Norditerpenoids (\pm)-Cephalolides A–D

Maximilian Haider,[§] Goh Sennari,[§] Alina Eggert, and Richmond Sarpong*

Richmond Sarpong



Education and academic career:

Richmond Sarpong

- 1995, B.S., Macalester College (Prof. **Rebecca C. Hoyer**)
- 2001, Ph.D., Princeton University (Prof. **Martin F. Semmelhack**)
- 2004, Postdoctoral Fellow, Caltech (Prof. **Brian M. Stoltz**)
- 2010, Assistant Professor, UCB
- 2014, Associate Professor
- Present Full Professor, Executive Associate Dean (Chemistry)

Research topics

1. Total synthesis of natural products



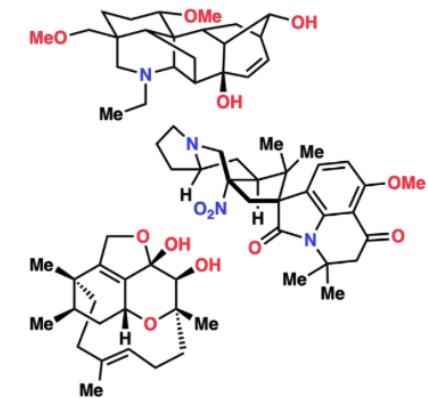
Bioinspired Diversification Approach Toward the Total Synthesis of Lycodine-Type Alkaloids

Hannah M. S. Haley,¹ Stefan E. Payer,¹ Sven M. Papidochia, Simon Clemens, Jonathan Nyenhuis, and Richmond Sarpong*



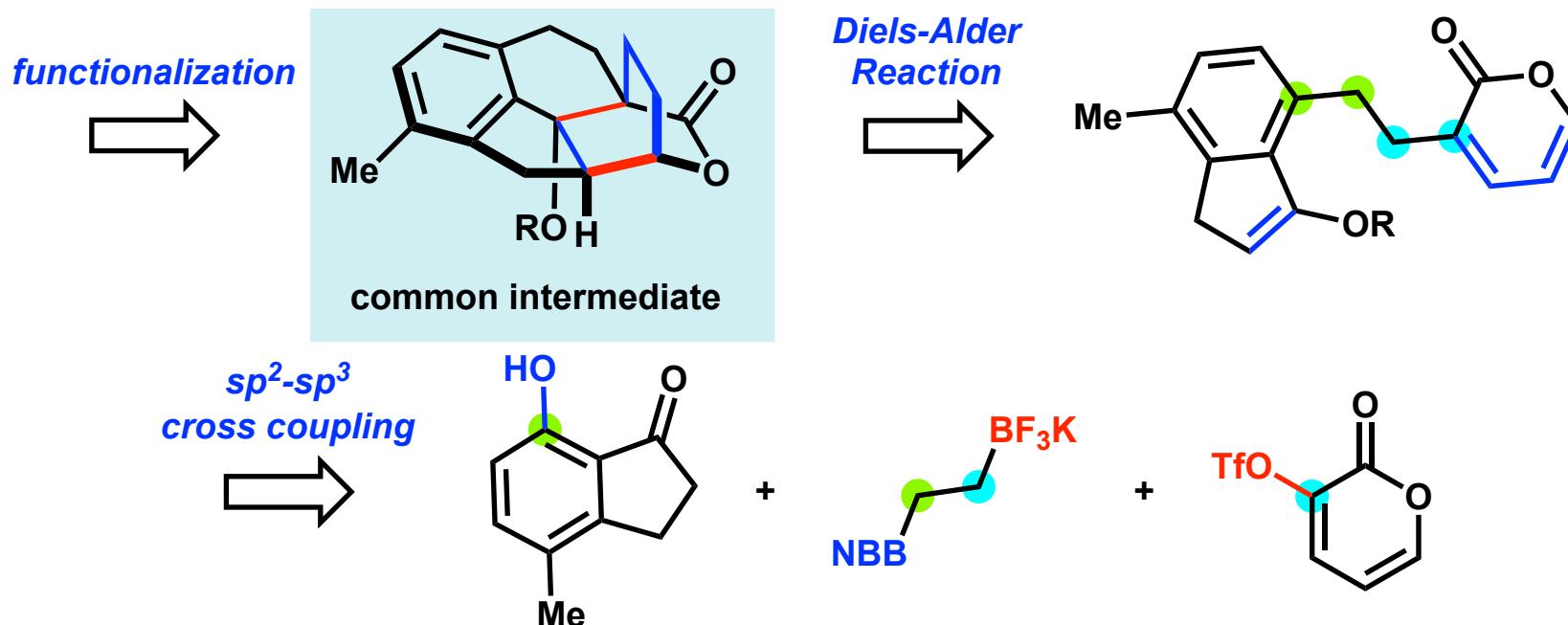
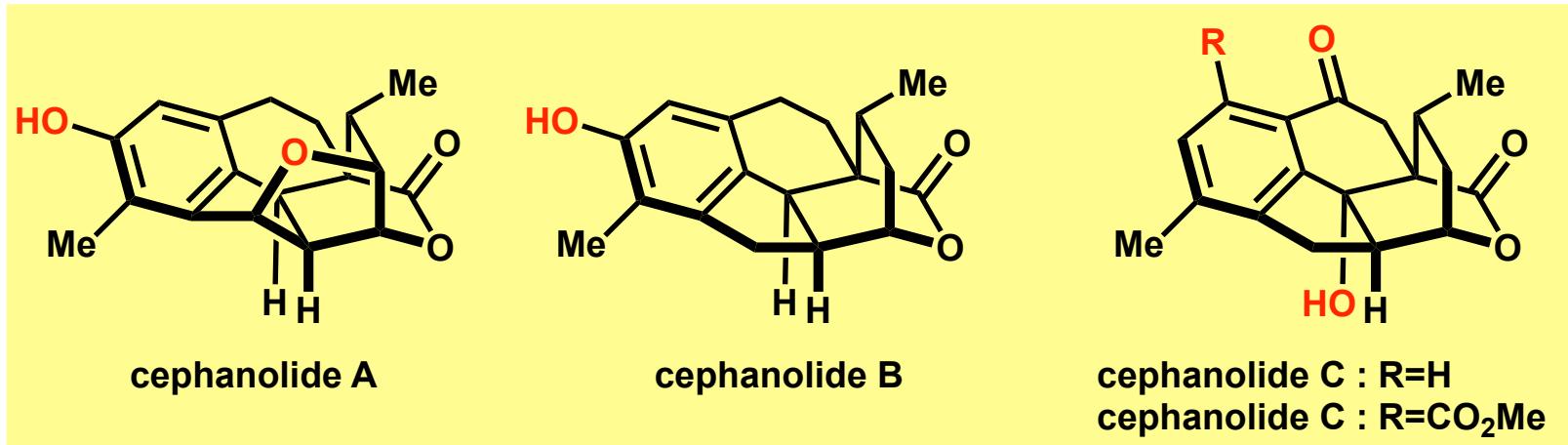
C–C Bond Cleavage Approach to Complex Terpenoids: Development of a Unified Total Synthesis of the Phomactins

Paul R. Leger, Yusuke Kuroda, Stanley Chang, Justin Jurczyk, and Richmond Sarpong*

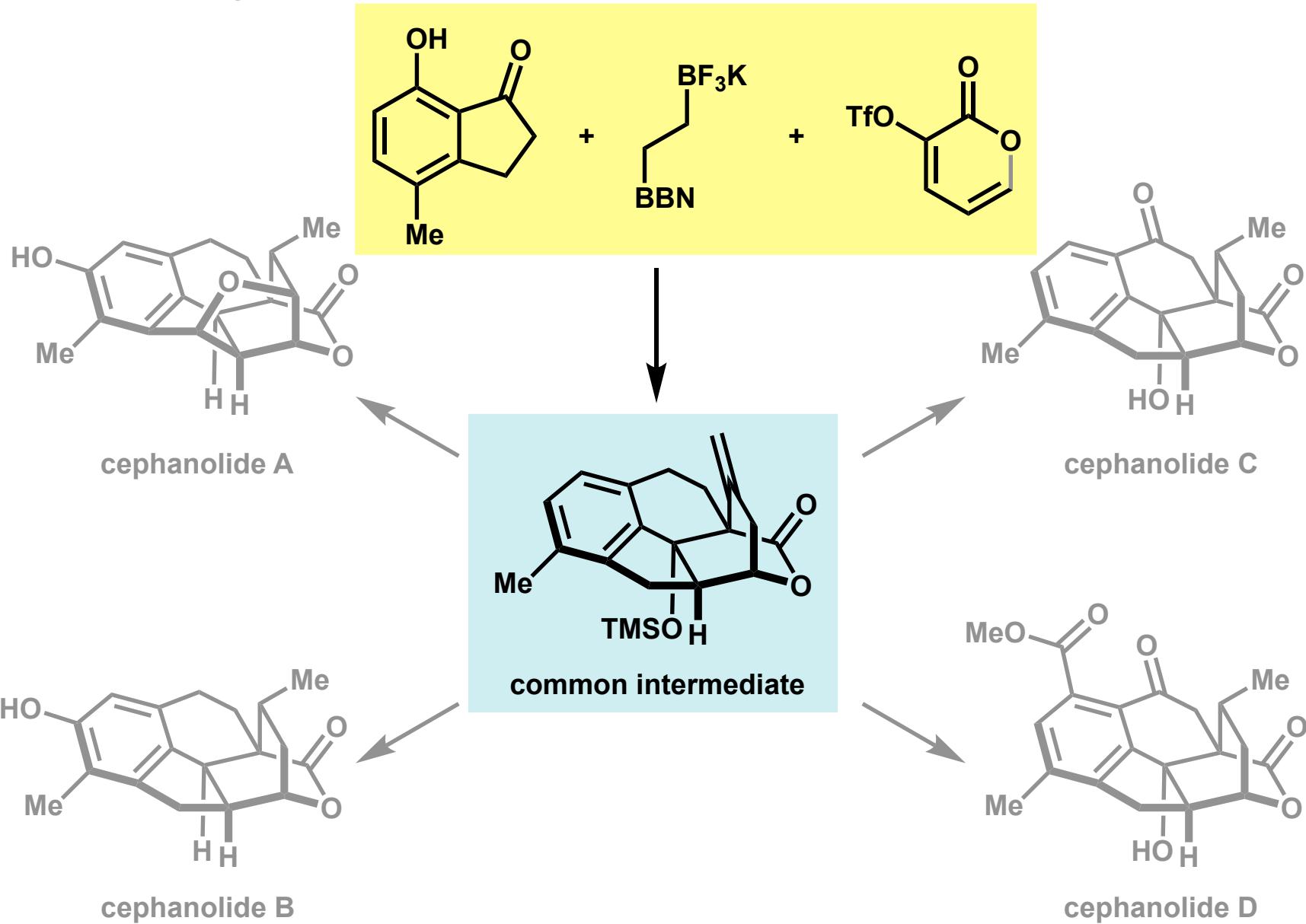


2. Development of synthesis method (C–H activation and C–C cleavage)

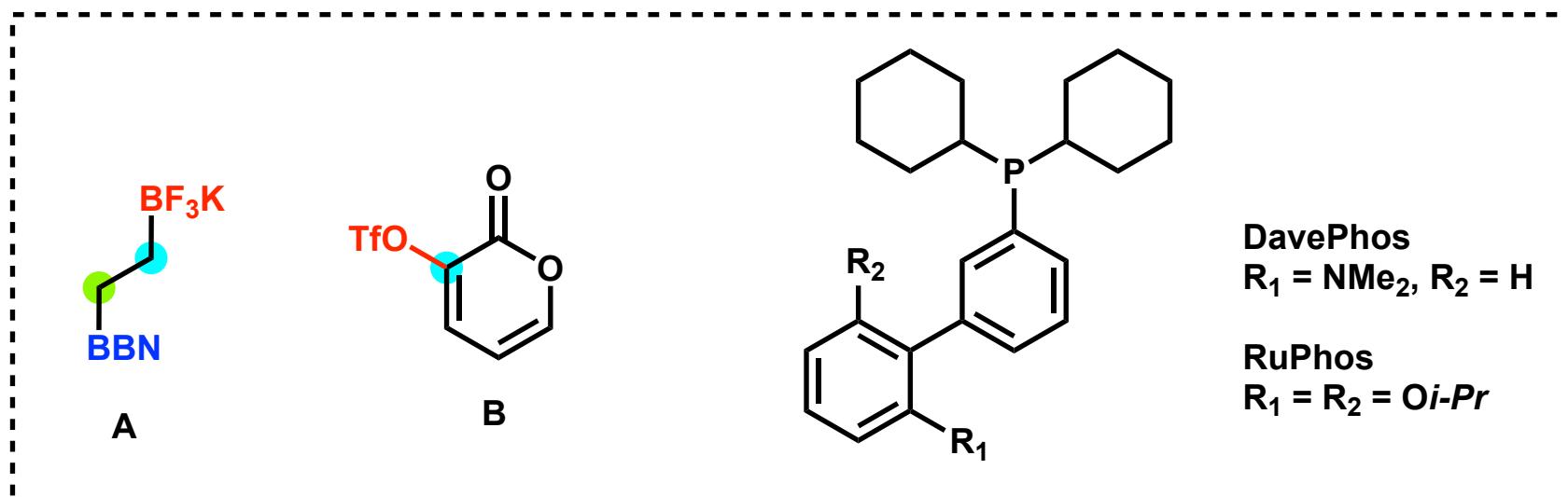
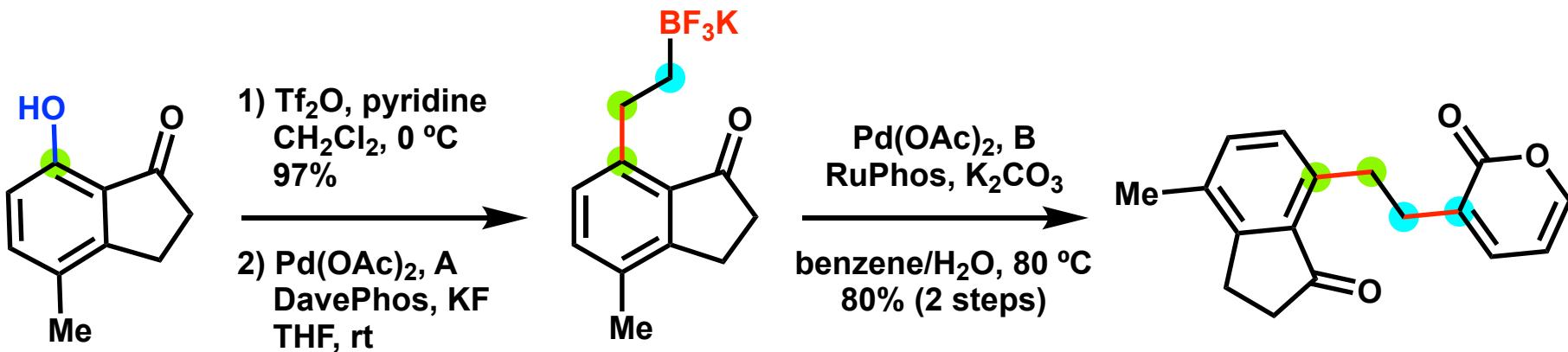
Sarpong's strategies and retrosynthesis



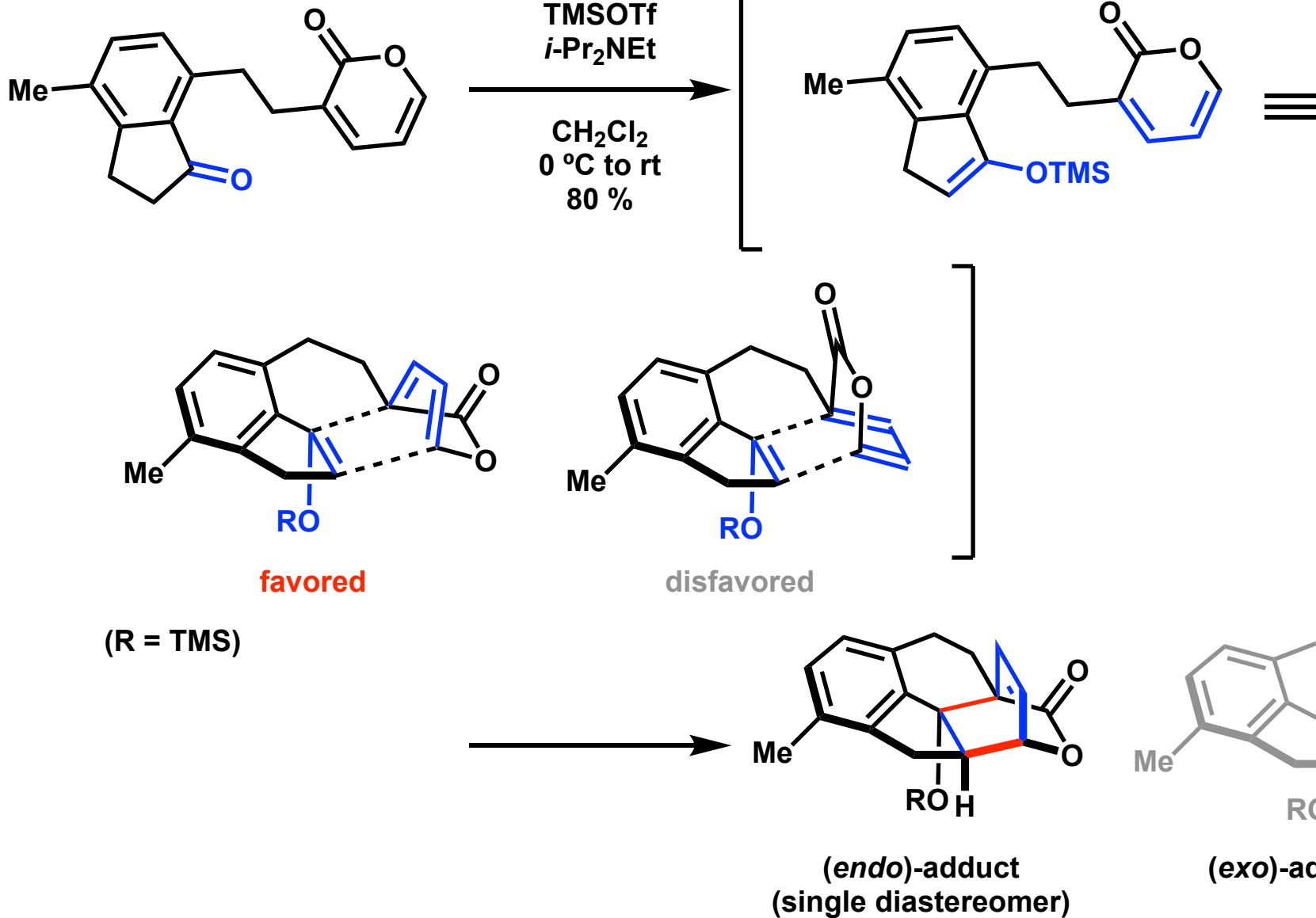
Synthesis of common intermediate



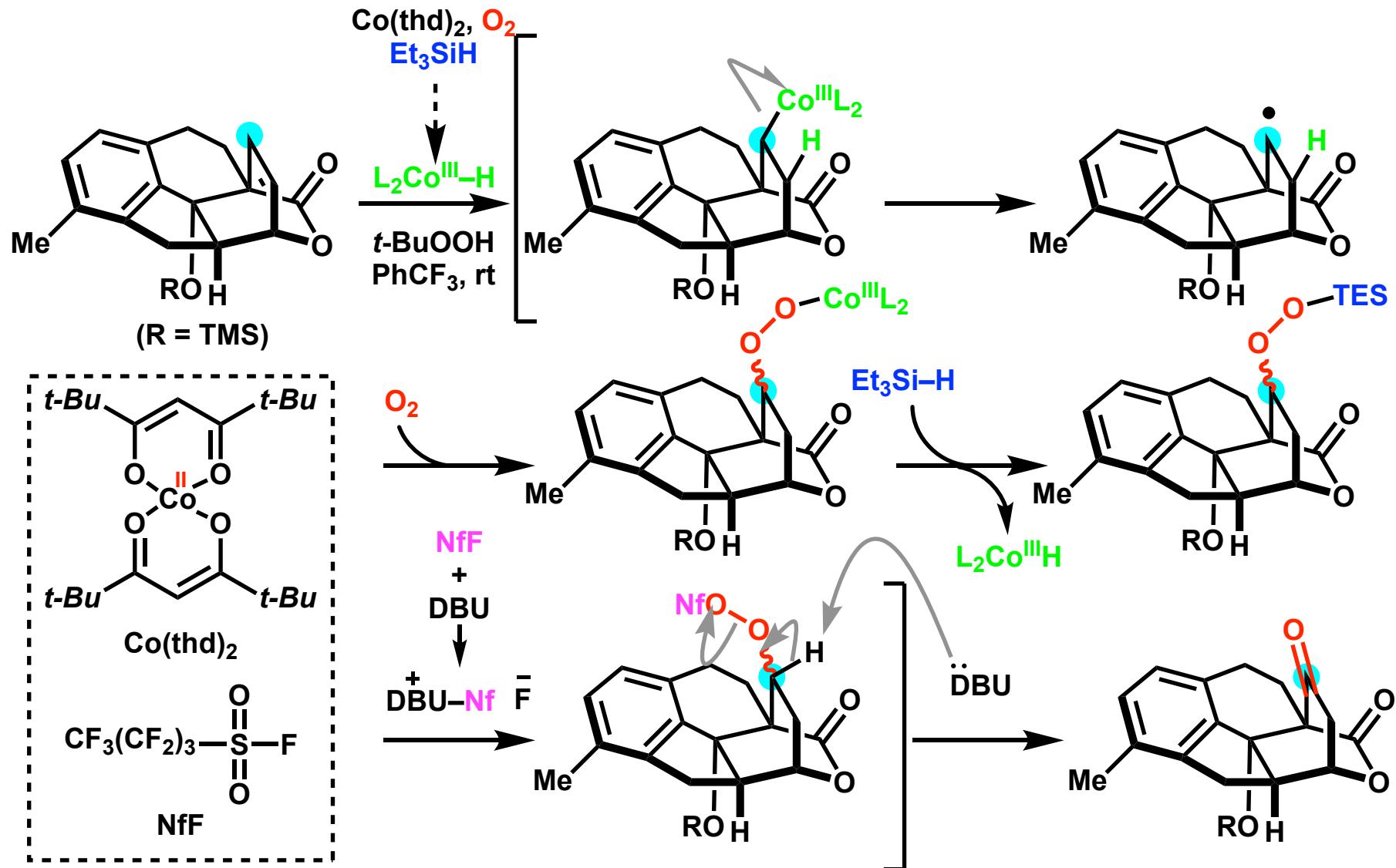
Selective Suzuki cross-coupling



Diels-Alder reaction



Isayama-Mukaiyama hydration

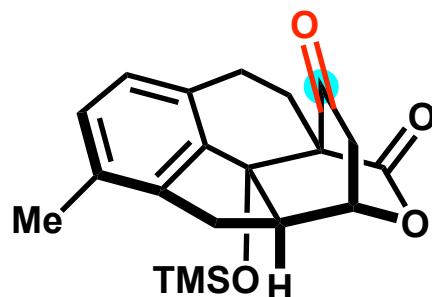


1) Nagatomo, M.; Koshimizu, M.; Masuda, K.; Tabuchi, T.; Urabe, D.; Inoue, M. *J. Am. Chem. Soc.* **2014**, 136, 5916

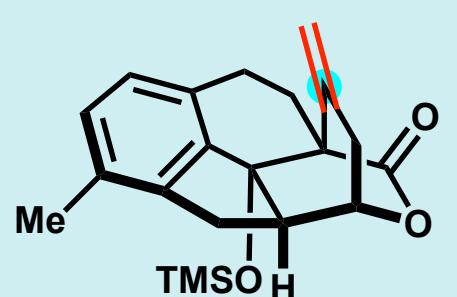
2) Nagatomo, M.; Hagiwara, K.; Masuda, K.; Koshimizu, M.; Kawamata, T.; Matsui, Y.; Urabe, D.; Inoue, M. *Chem. – Eur. J.* **2016**, 22, 222

3) Crossley, S.; Obradors, C.; Martinez, R.; Shenvi, R. *Chem. Rev.* **2016**, 116, 8912

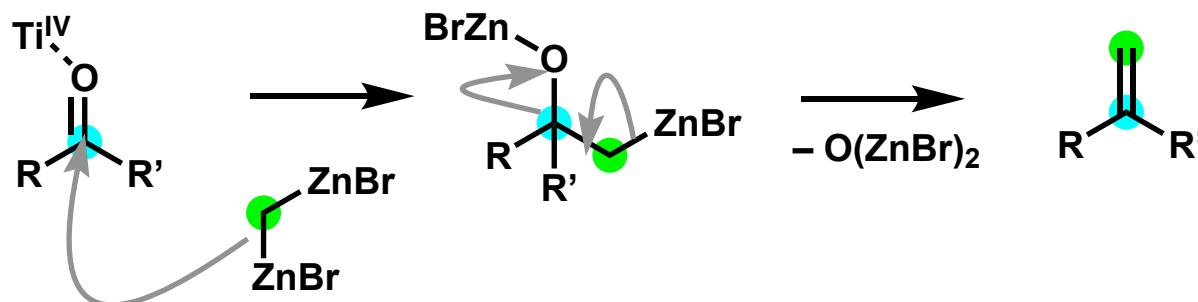
Olefination



$\text{Ti}(\text{O}-\text{i-Pr})_2\text{Cl}_2$
 $\text{CH}_2(\text{ZnBr})_2$
 THF, 15 °C
 53%

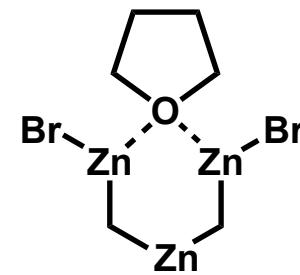
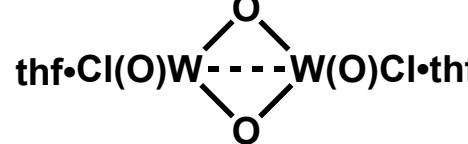
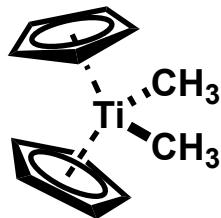
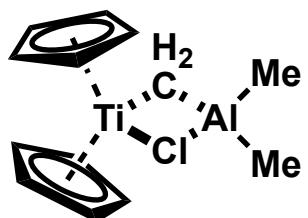


Proposed mechanism



First of all, they attempted *Wittig* olefination but failed resulting from basicity of phosphorus ylides. Therefore, they focus on olefination reagents which are known to be less basic such as Tebbe, Petasis, Johnson-Peterson, Kauffmann, Nystedt and Lombardo olefinations.

Only the highlighted reagents delivered trace amounts of common intermediate.



Tebbe reagent

Petasis reagent

Kauffmann reagent

Lomberdo's reagent

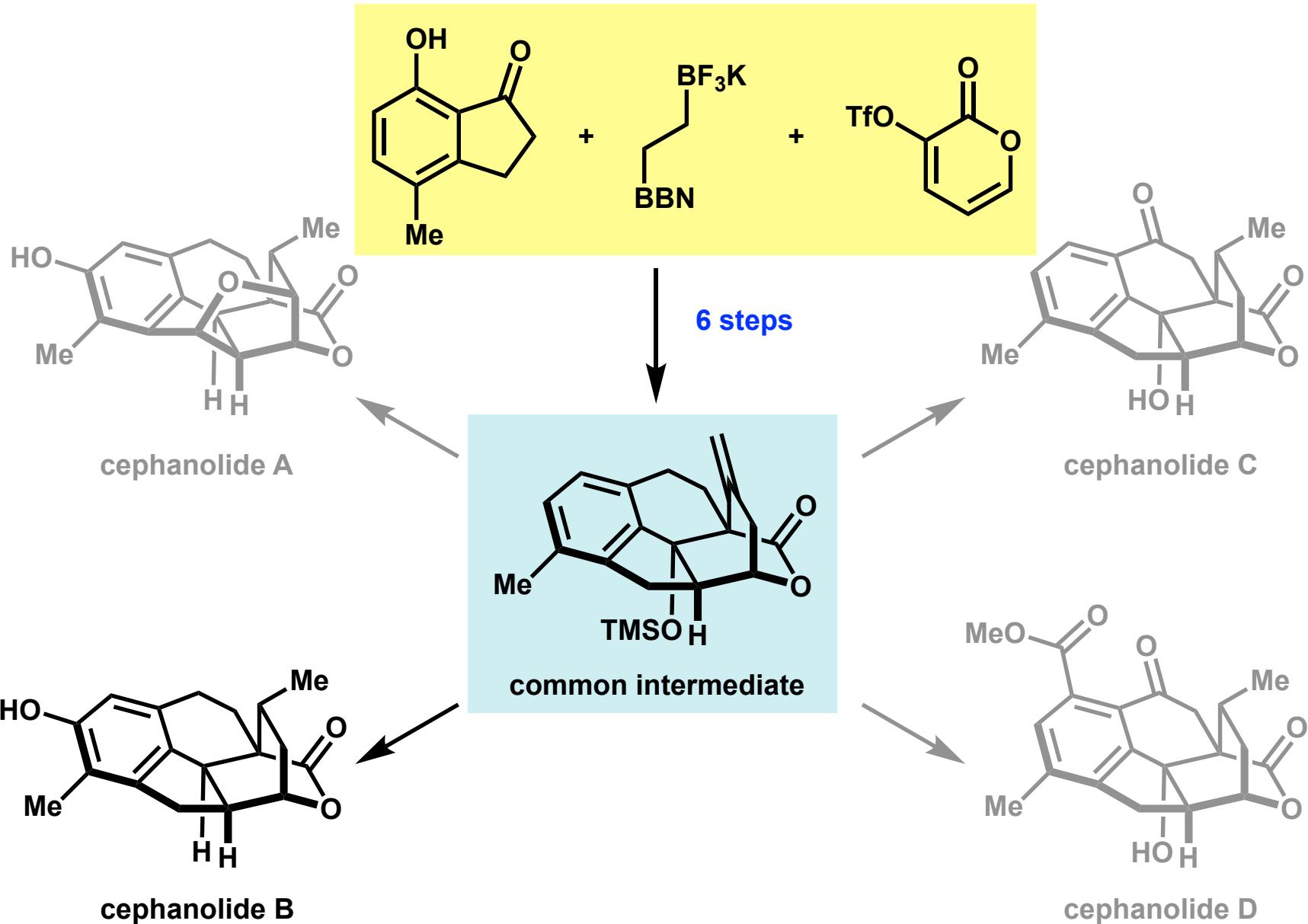
Nysted reagent

1) Takai, K.; Kakiuchi, T.; Kataoka, Y.; Utimoto, K. *J. Org. Chem.* **1994**, 59, 2668 2) Matsubara, S.; Sugihara, M.; Utimoto, K. *Synlett.* **1998**, 1998, 313

3) Barnych, B.; Vatele, J. *Synlett.* **2011**, 13, 1912 4) Chand, H. *Synlett.* **2009**, 15, 2545

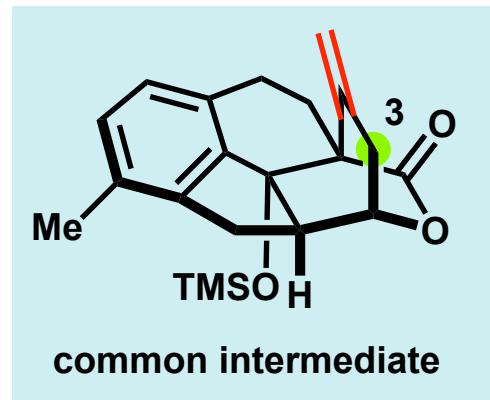
5) Okazoe, T.; Hibino, J.; Takai, K.; Nozaki, H. *Tetrahedron Lett.* **1985**, 26, 45

Total synthesis of Cephalolide B

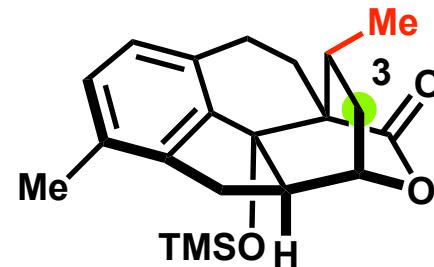


Heterogeneous Hydrogenation (1)

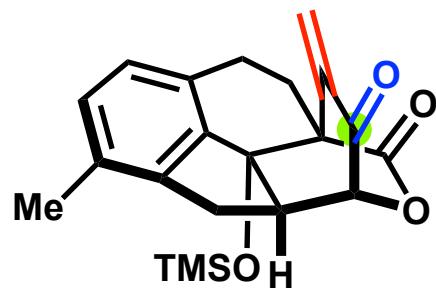
Total synthesis of Cephalolide B,C,D



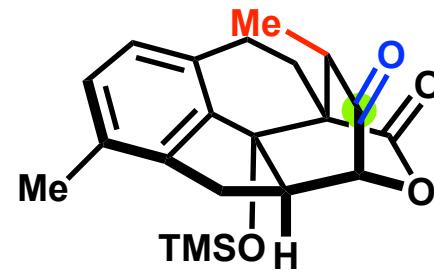
Pd/C, H₂
MeOH, rt
99% (d.r. = >20:1)



Total synthesis of Cephalolide A



Pd/C, H₂
MeOH, rt



1) Woodward, R.; Bader, F.; Bickel, H.; Frey, A. *Tetrahedron* **1958**, *2*, 1–2) Pearlman, B. *J. Am. Chem. Soc.* **1979**, *101*, 6404

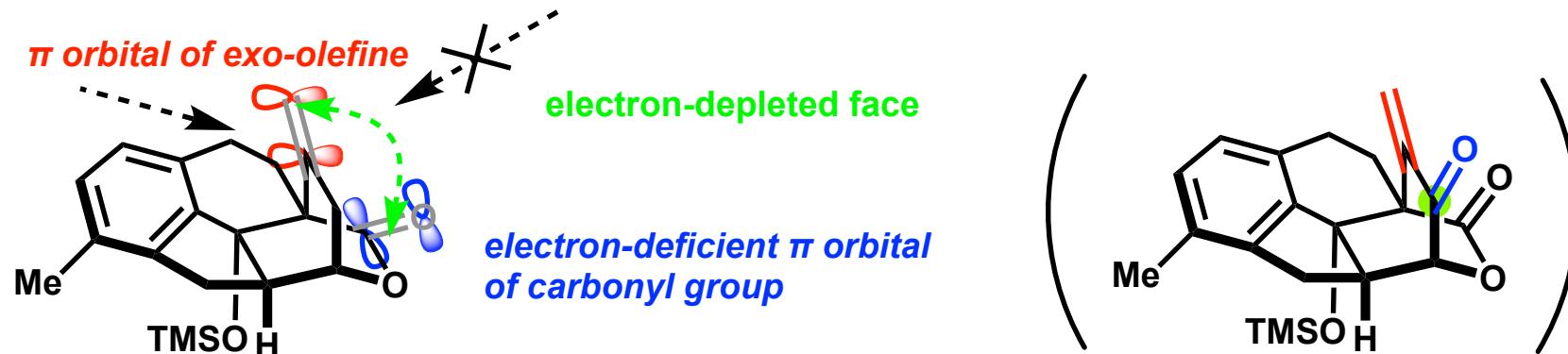
3) Hamlin, A.; Lapointe, D.; Owens, K.; Sarpong, R. *J. Org. Chem.* **2014**, *79*, 6783

Heterogeneous Hydrogenation (2)

Proposal 1

This face is not reactive because of electron-depleted by orbital interaction between exo-olefine π orbital and carbonyl π^* one.

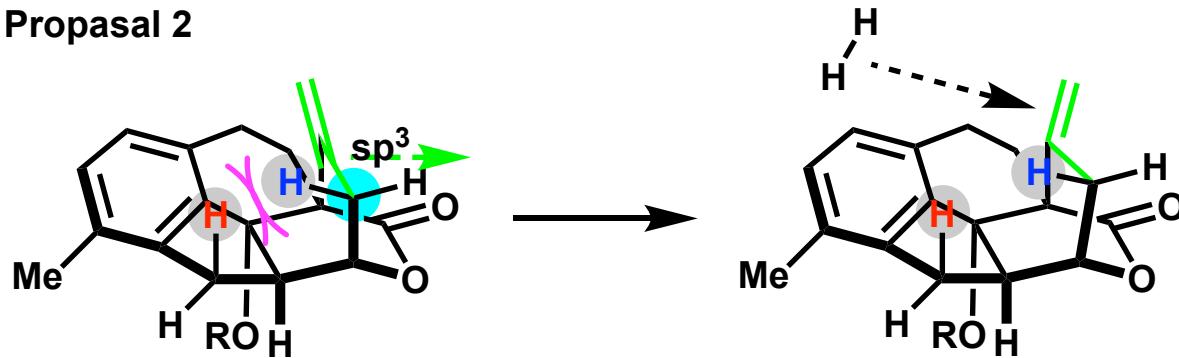
Therefore, hydrolation occurred in opposite side.



These effects were proposed by Woodward but was not still clear.

HOMO energy of enone decreased than exo-olefine, so these effects was weak and hydrolation occured from unhindered face.

Proposal 2

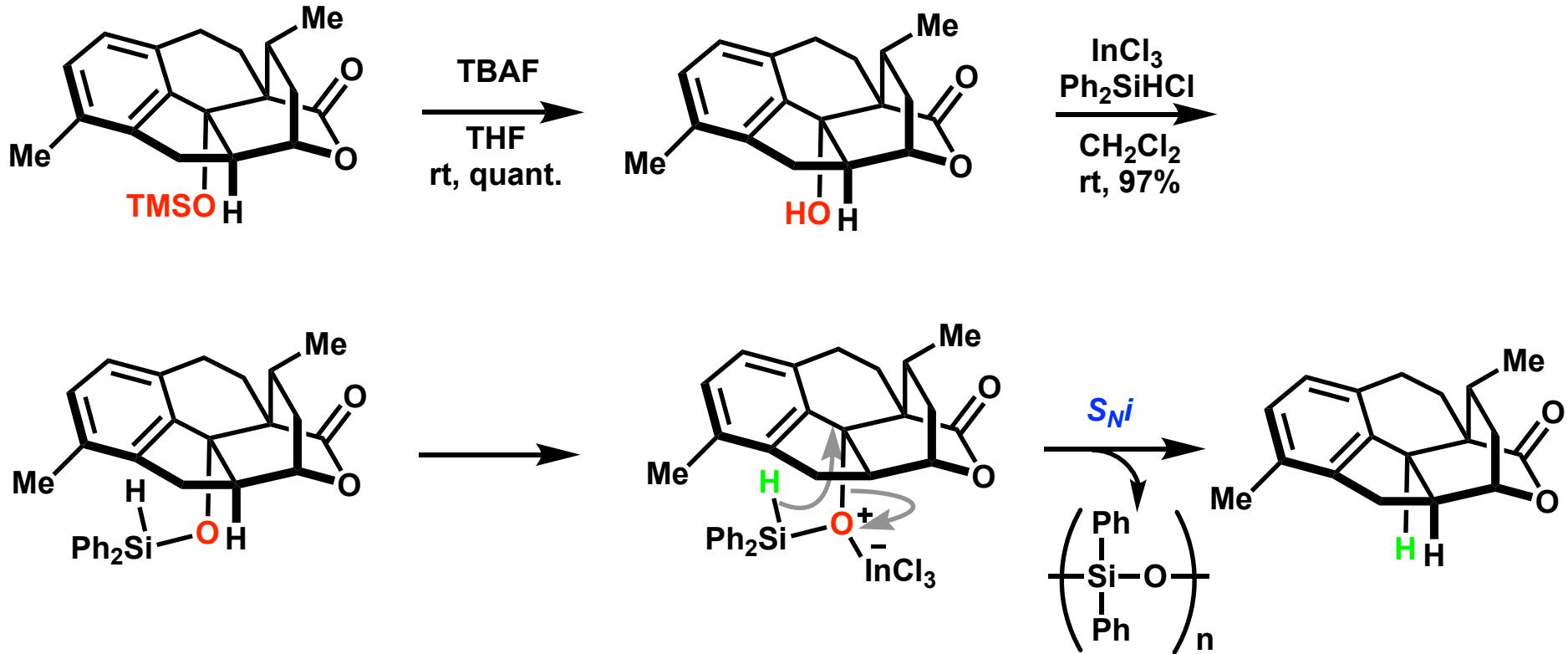


A little conformation change occurred because of avoiding from 1,3-allylic strain

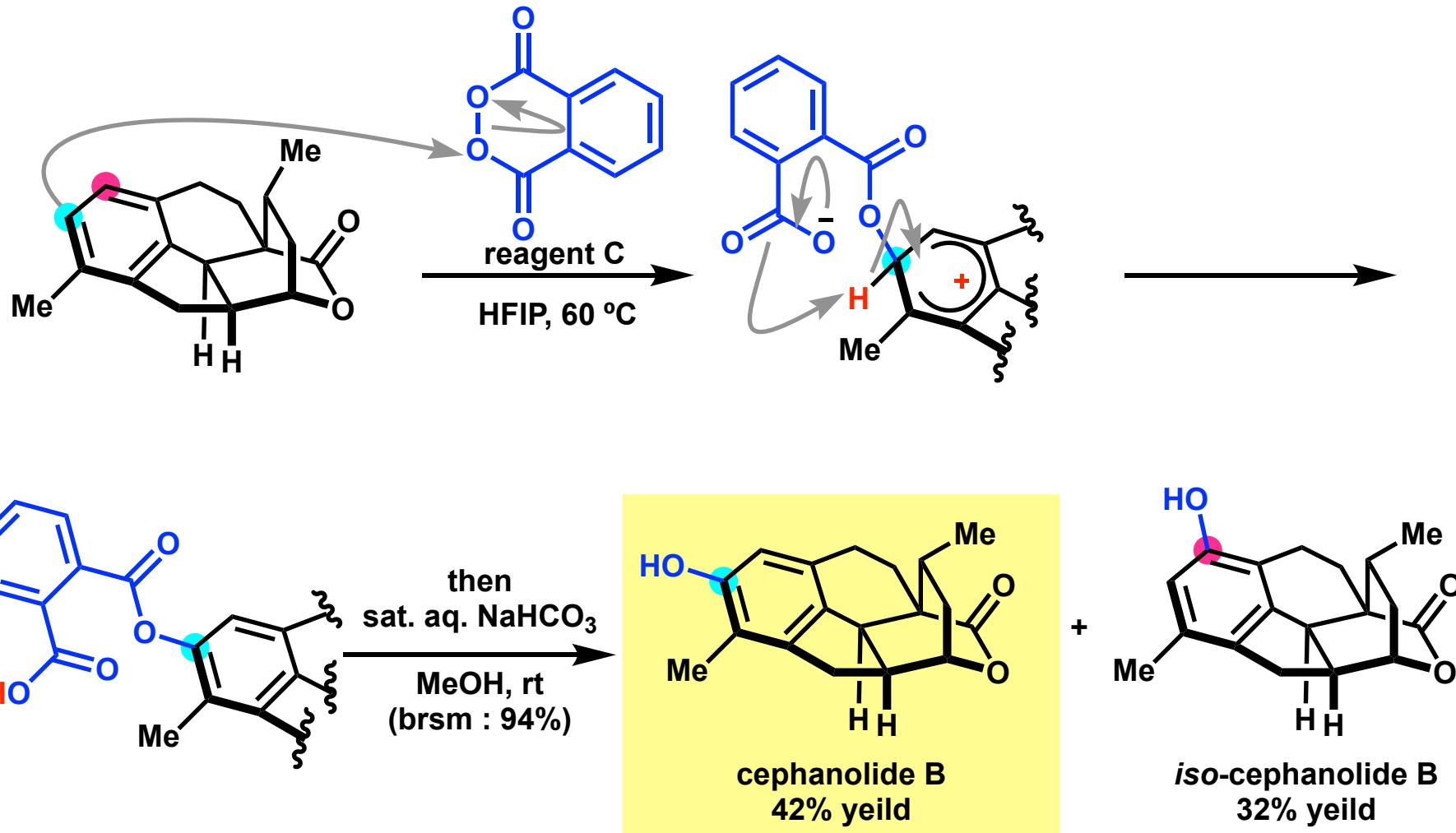
1) Woodward, R.; Bader, F.; Bickel, H.; Frey, A. *Tetrahedron* **1958**, 2, 1–2) Pearlman, B. *J. Am. Chem. Soc.* **1979**, 101, 6404

3) Hamlin, A.; Lapointe, D.; Owens, K.; Sarpong, R. *J. Org. Chem.* **2014**, 79, 6783

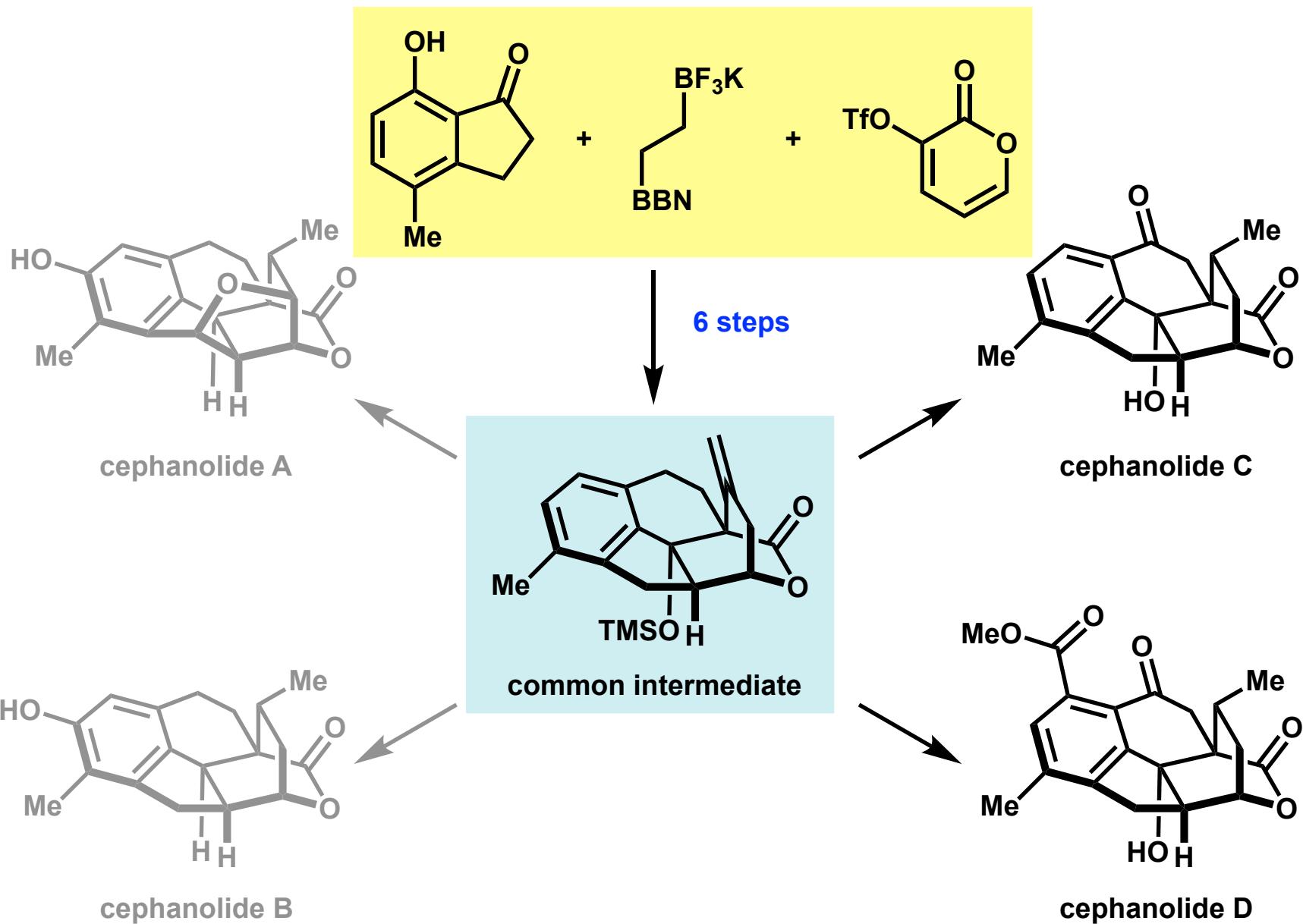
Total synthesis of cephanolide B and *iso*-cephanolide B (1)



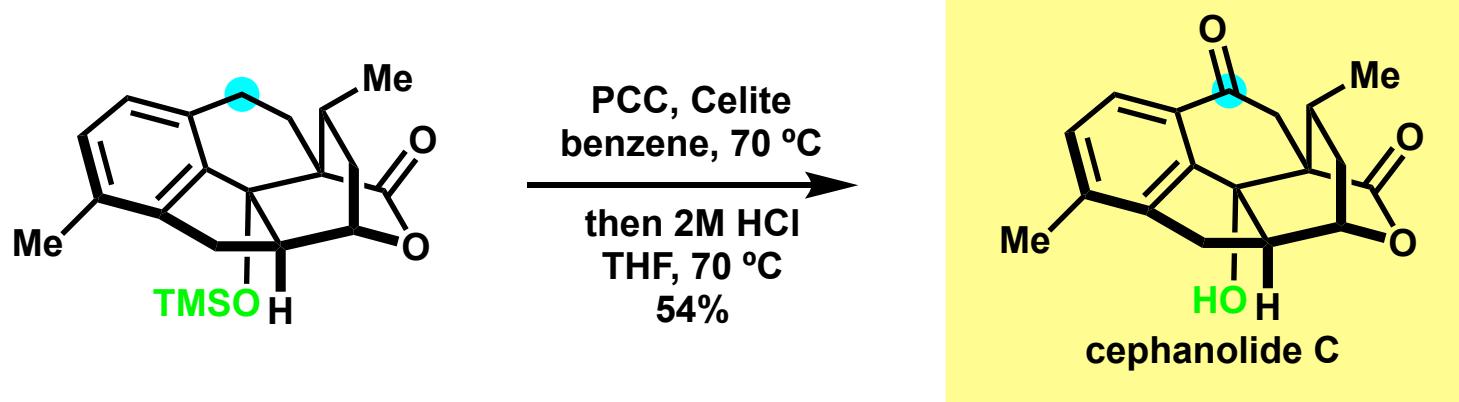
Total synthesis of cephanolide B and *iso*-cephanolide B (2)



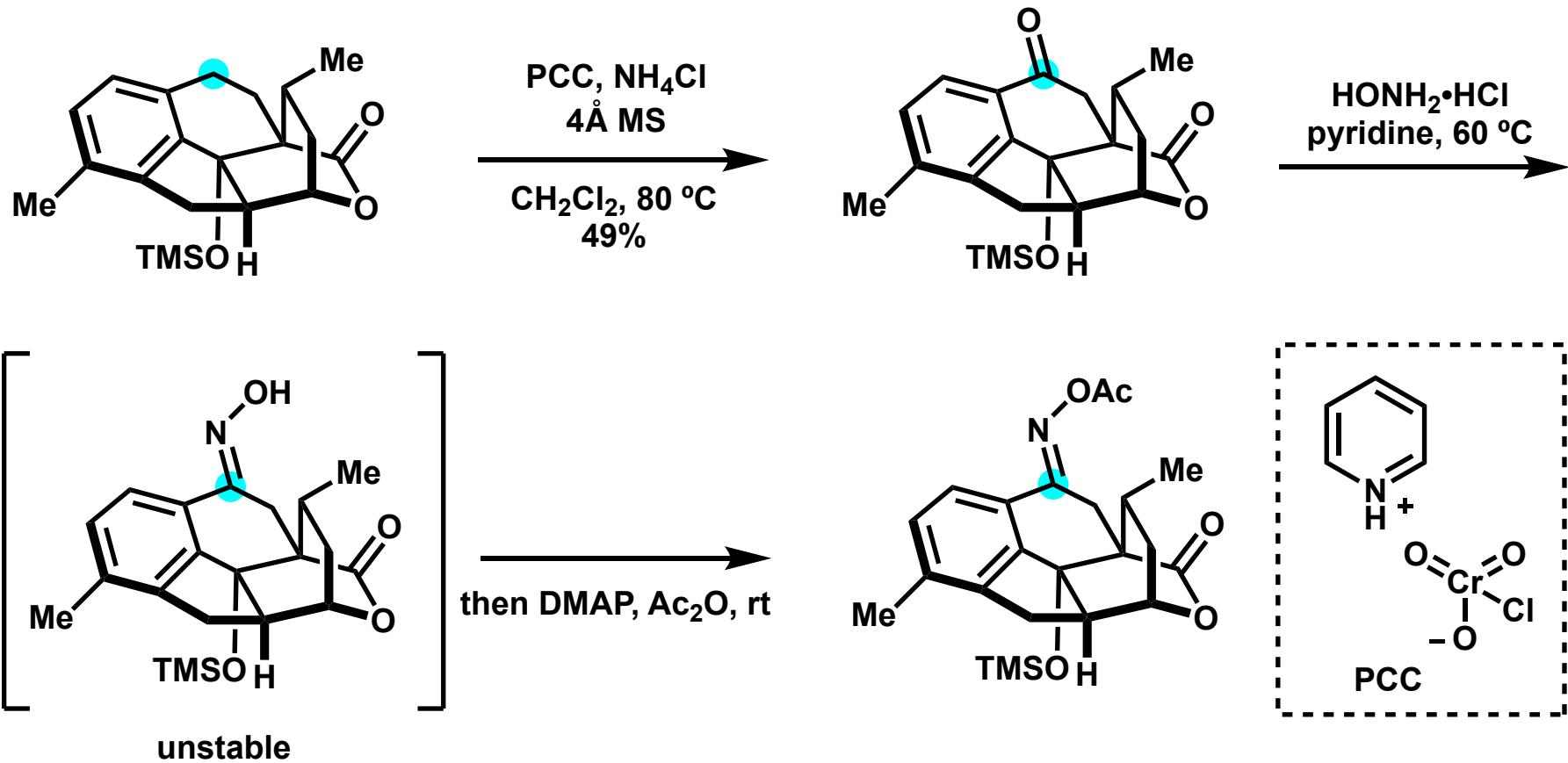
Total synthesis of cephanolide C and D



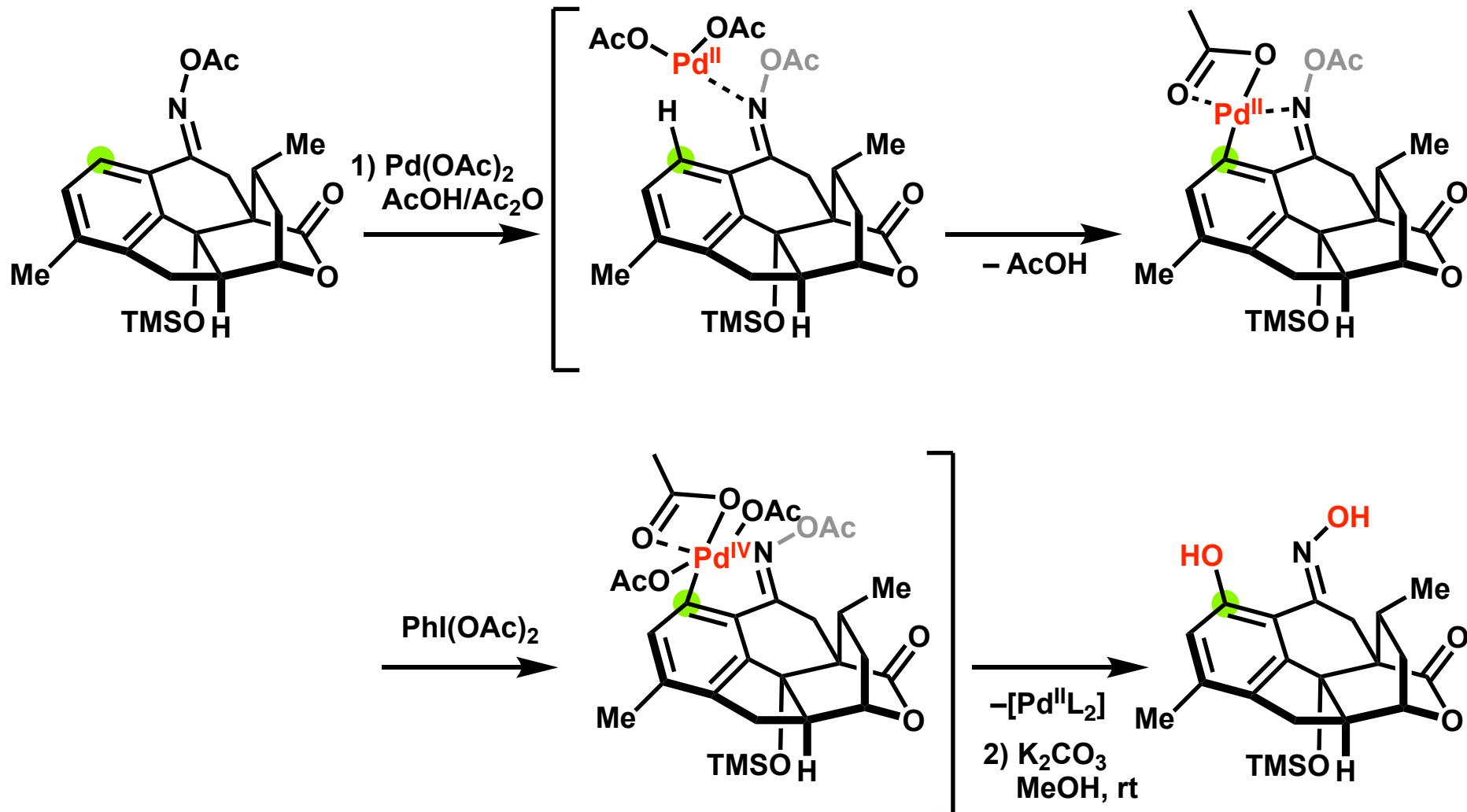
Total synthesis of cephanolide C



Total synthesis of cephanolide D (1)

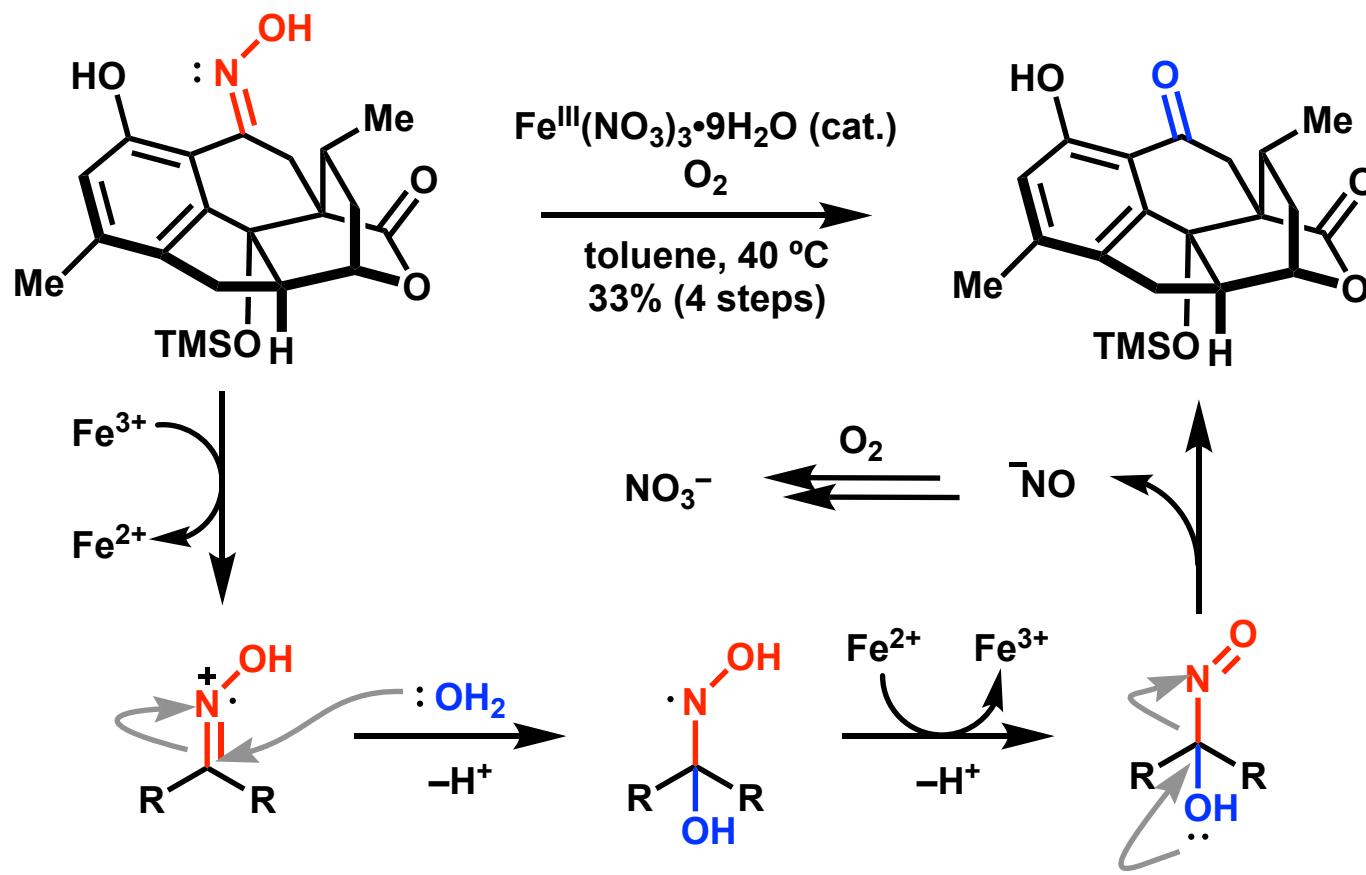


Total synthesis of cephanolide D (2)

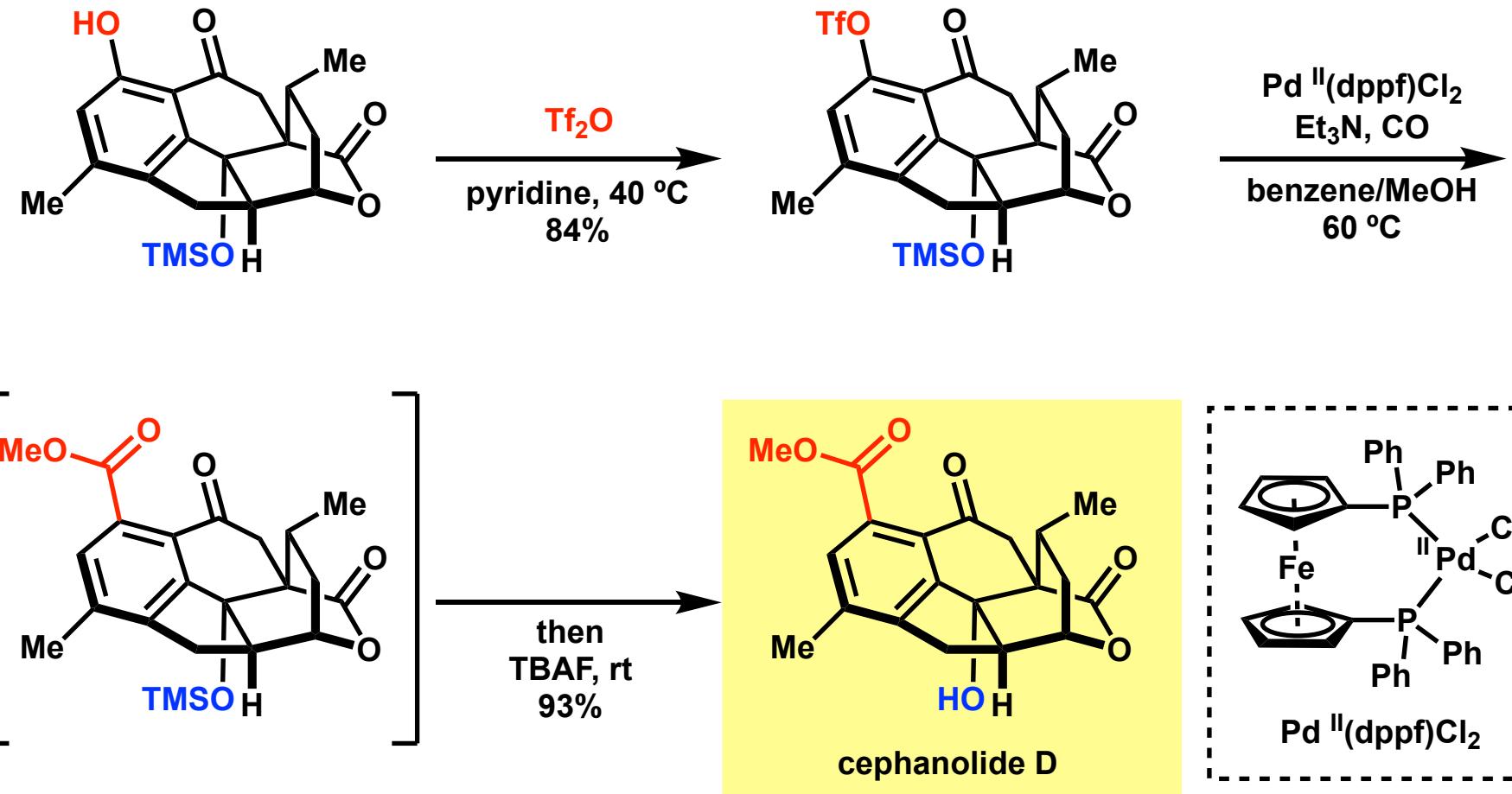


Total synthesis of cephalolide D (3)

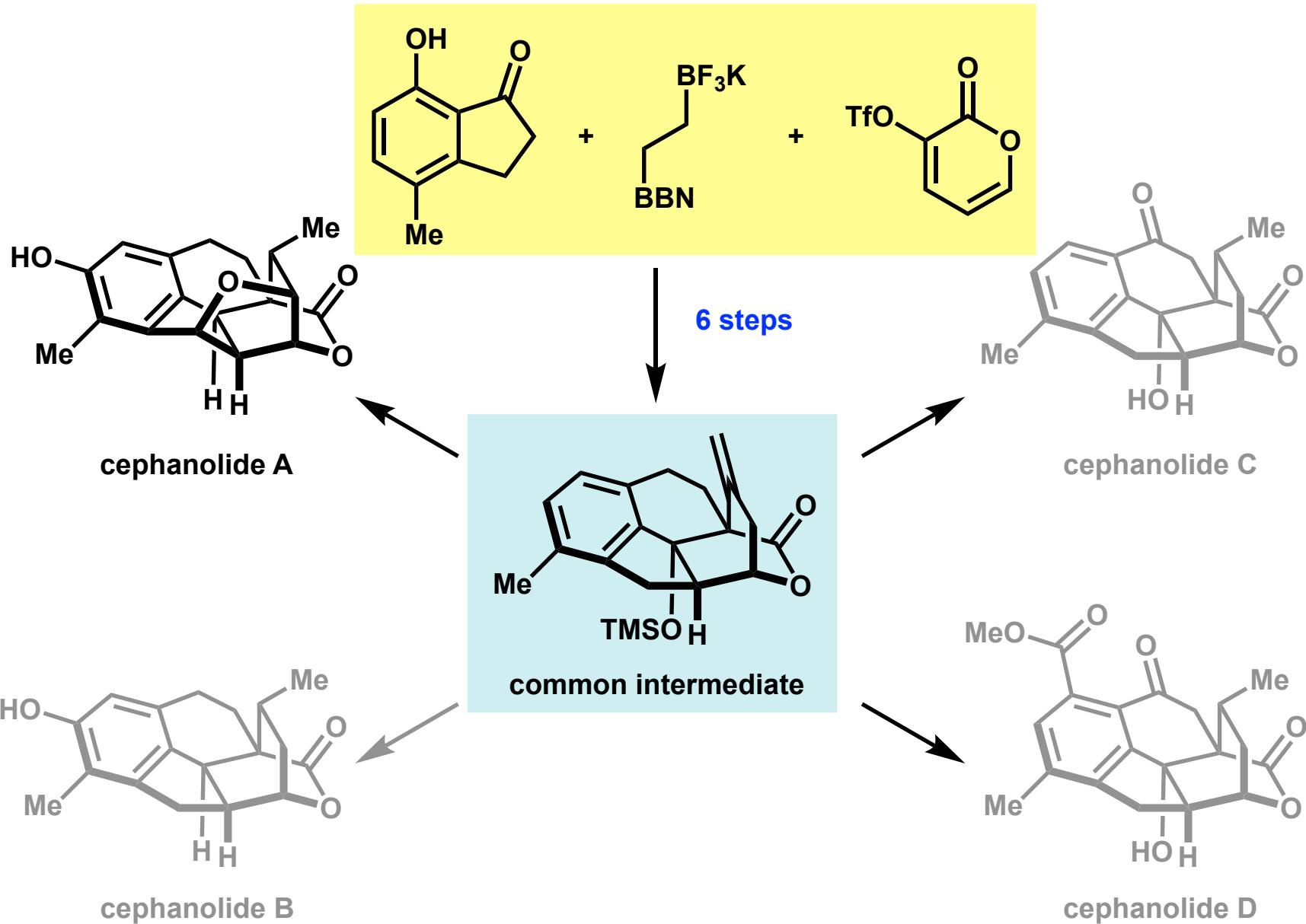
Proposed mechanism



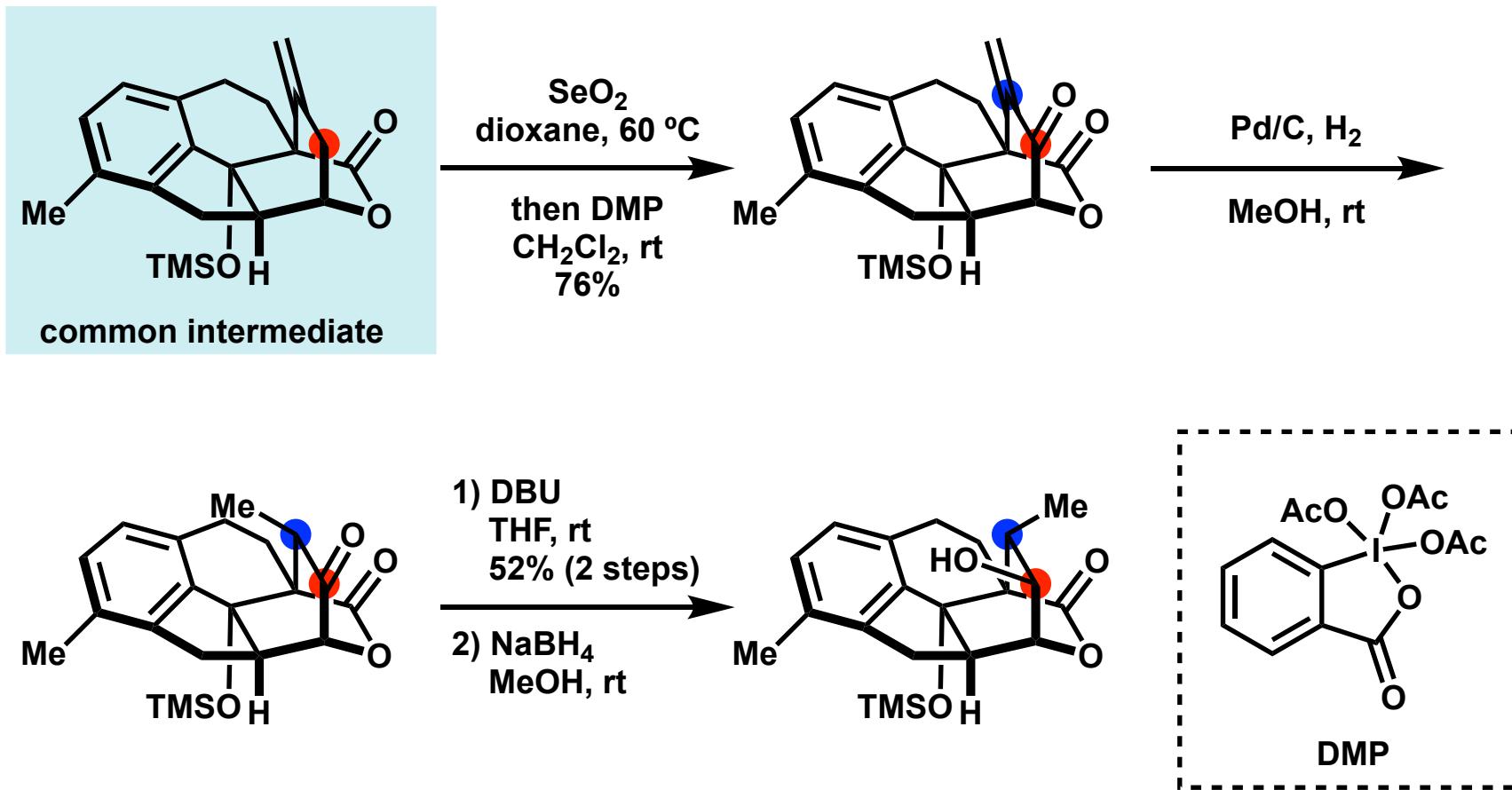
Total synthesis of cephanolide D (4)



Total synthesis of cephanolide A



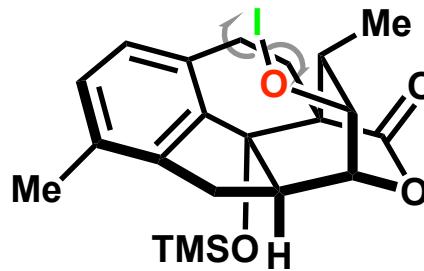
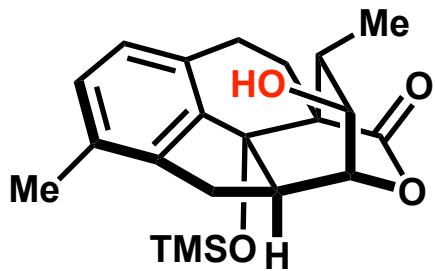
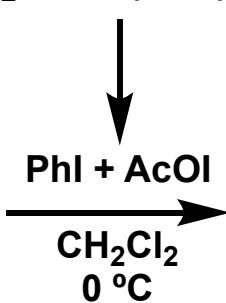
Total synthesis of cephalolide A (1)



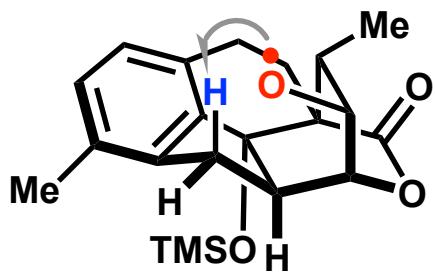
Total synthesis of cephalolide A (2)

Suarez oxidation

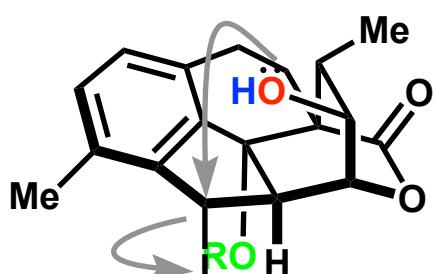
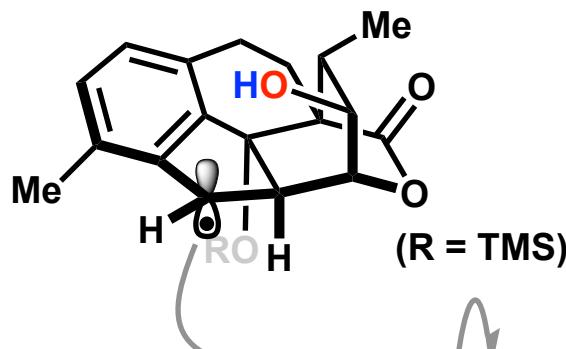
I₂ + PhI (OAc)₂



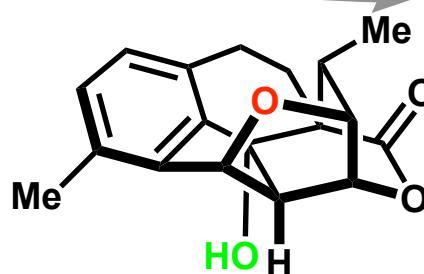
hν



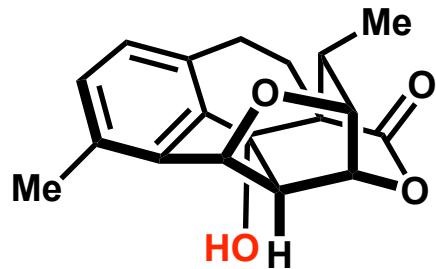
[1.5]-HAT



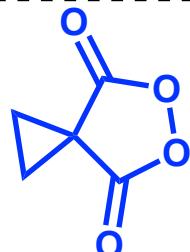
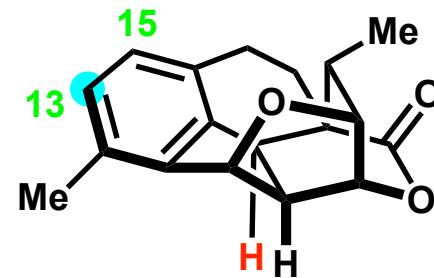
then
TBAF, rt
82% (2 steps)



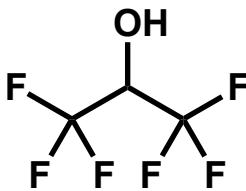
Total synthesis of cephanolide A (3)



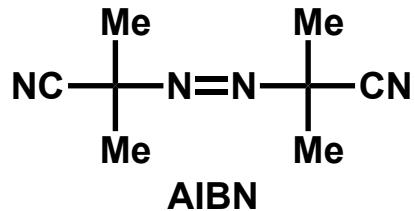
1) NaH, CS₂, MeI
THF, 0 °C
→
2) AIBN, *n*-Bu₃SnH
toluene, 80 °C
82% (2 steps)



reagent D

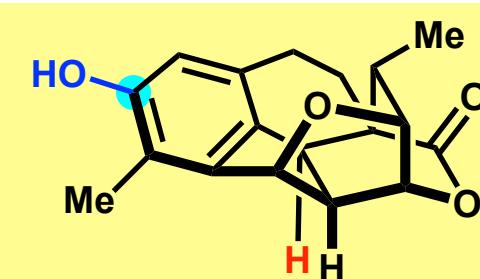


HFIP



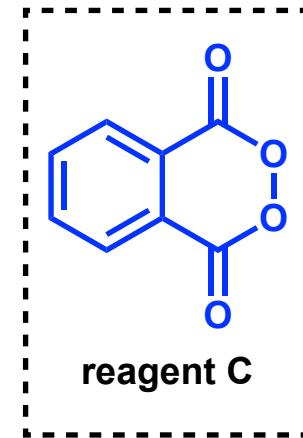
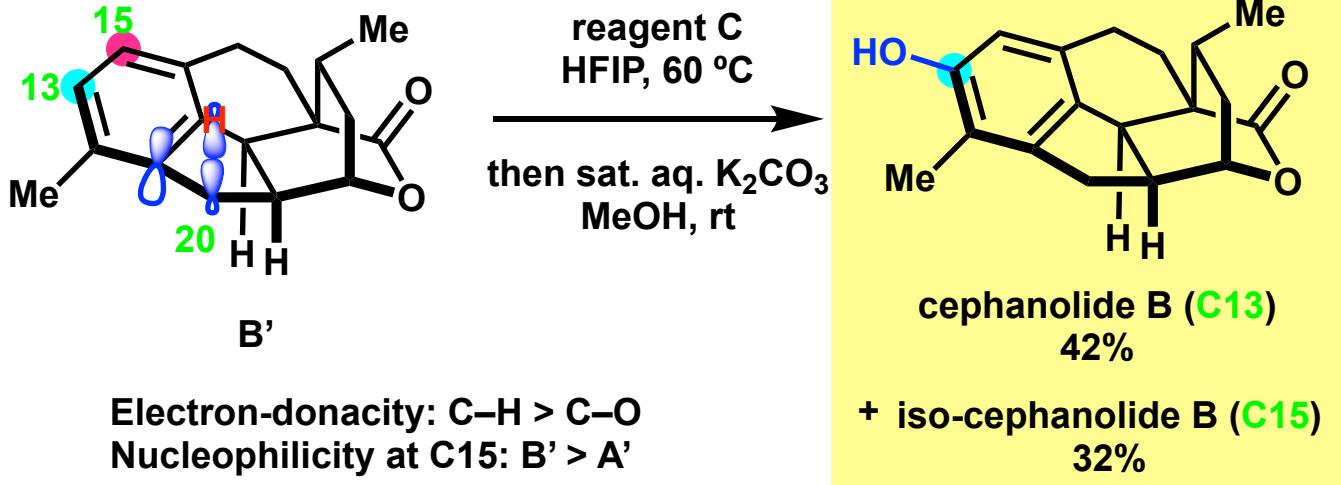
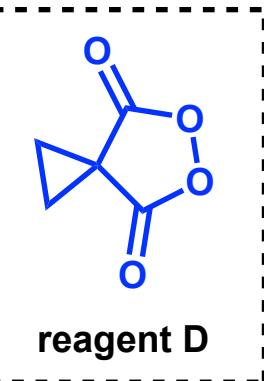
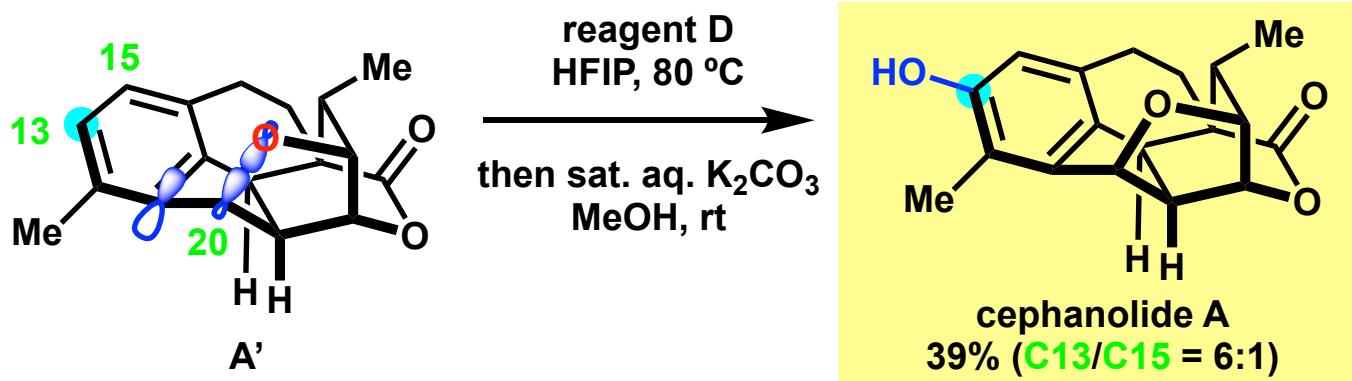
AIBN

reagent D
HFIP, 80 °C
→
then sat. aq. K₂CO₃
MeOH, rt
39%
(C₁₃/C₁₅ = 6:1)



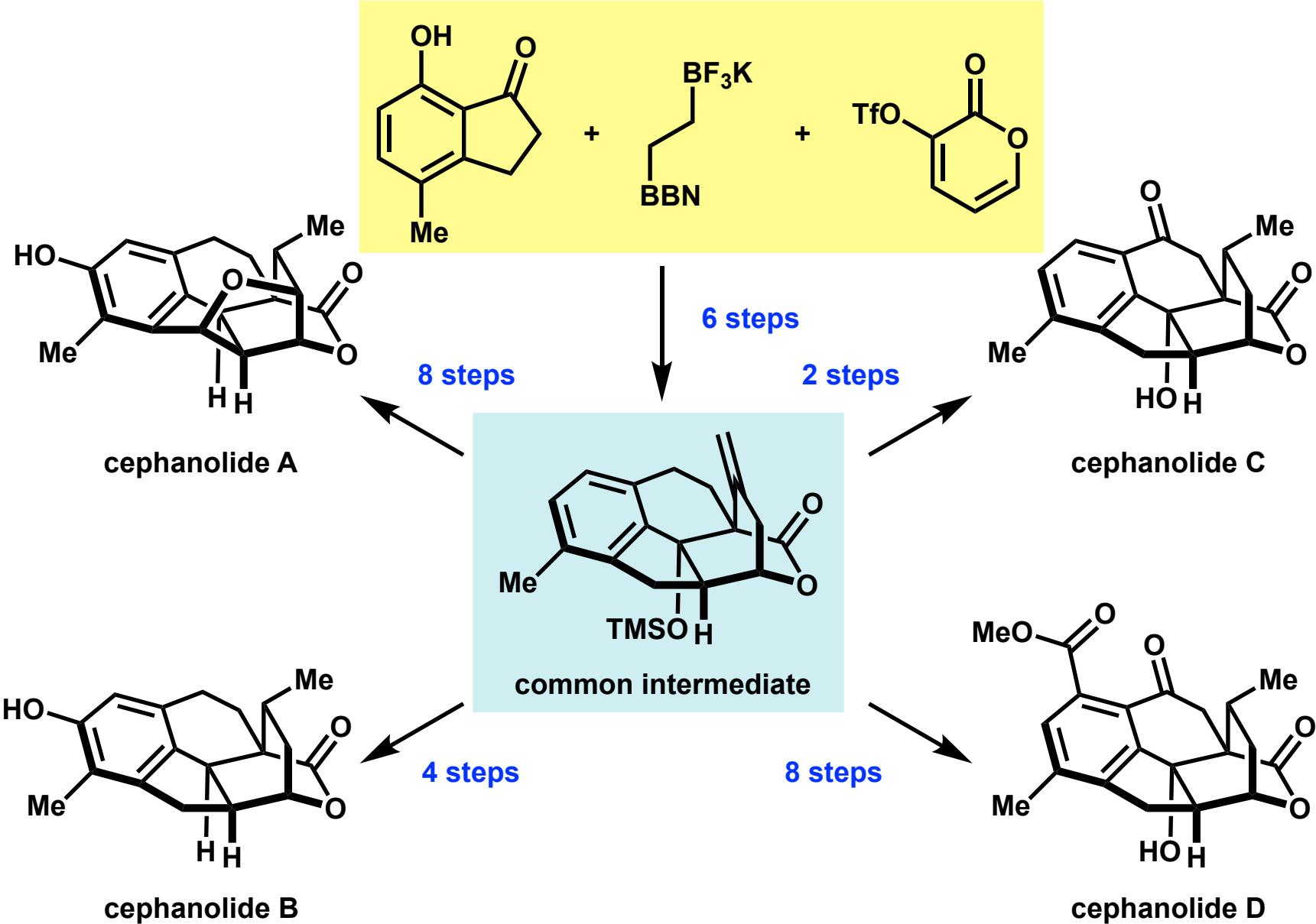
cephanolide A

Regioselectivity of oxidation



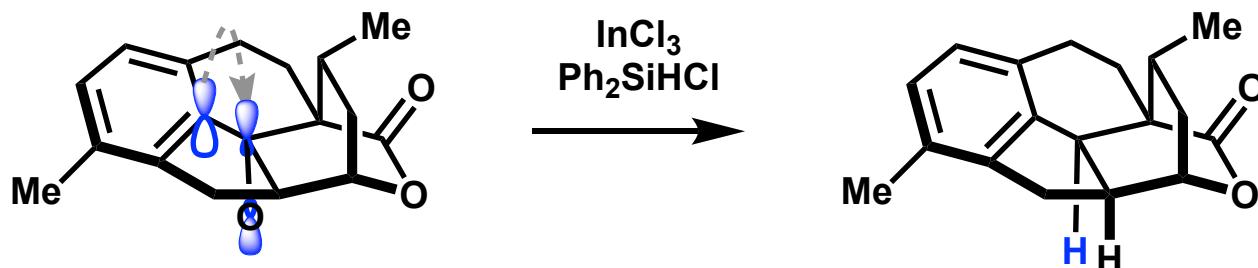
Electron-donacity: C–H > C–O
Nucleophilicity at C15: B' > A'

Summary

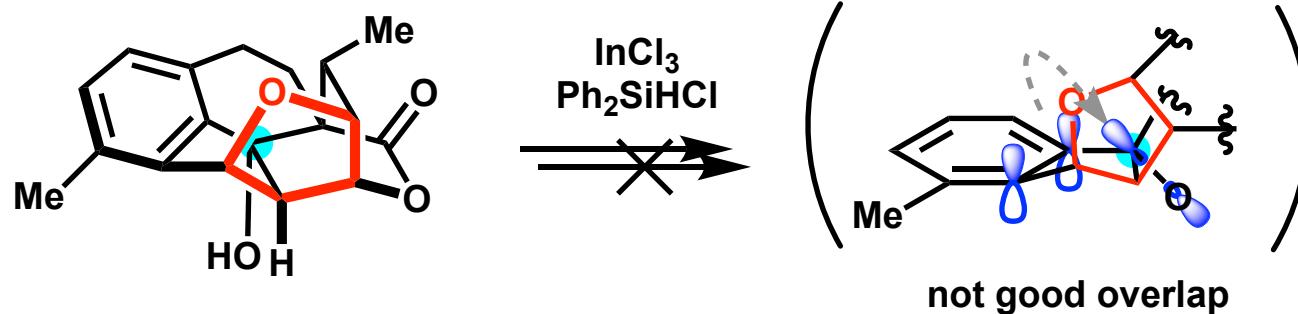


Appx. Conformation change

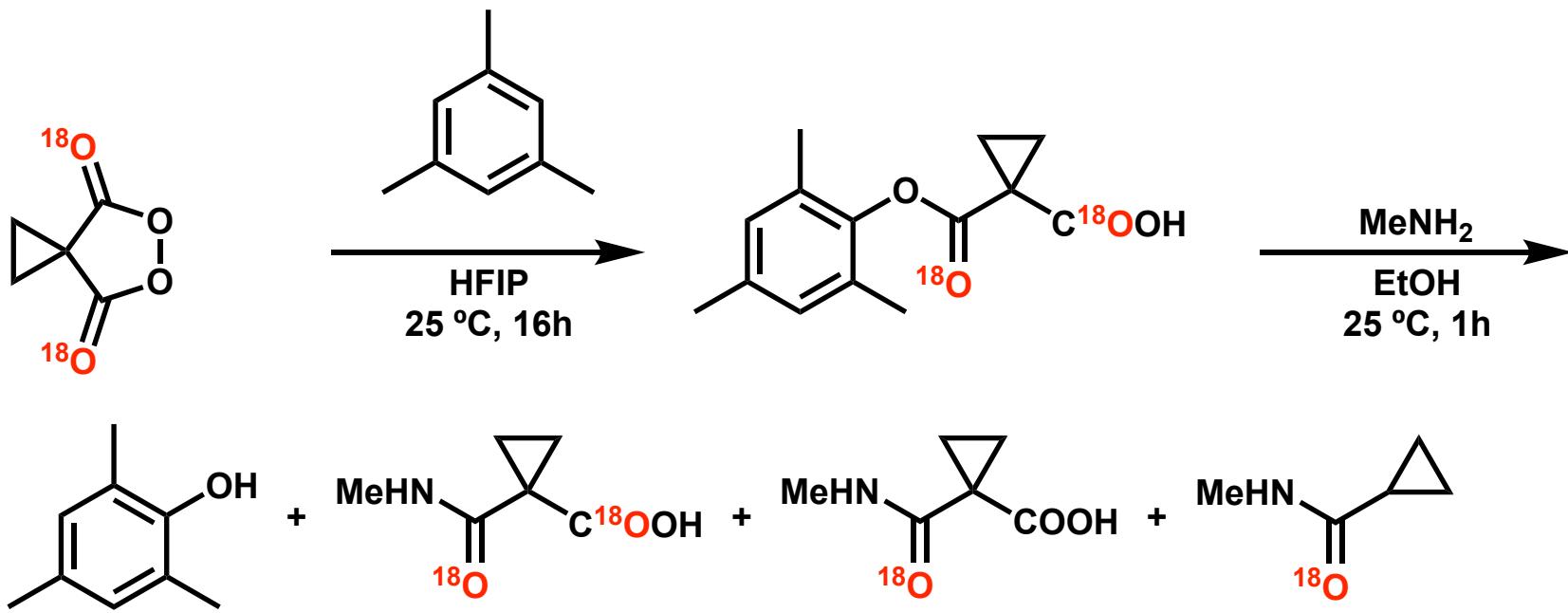
my proposal



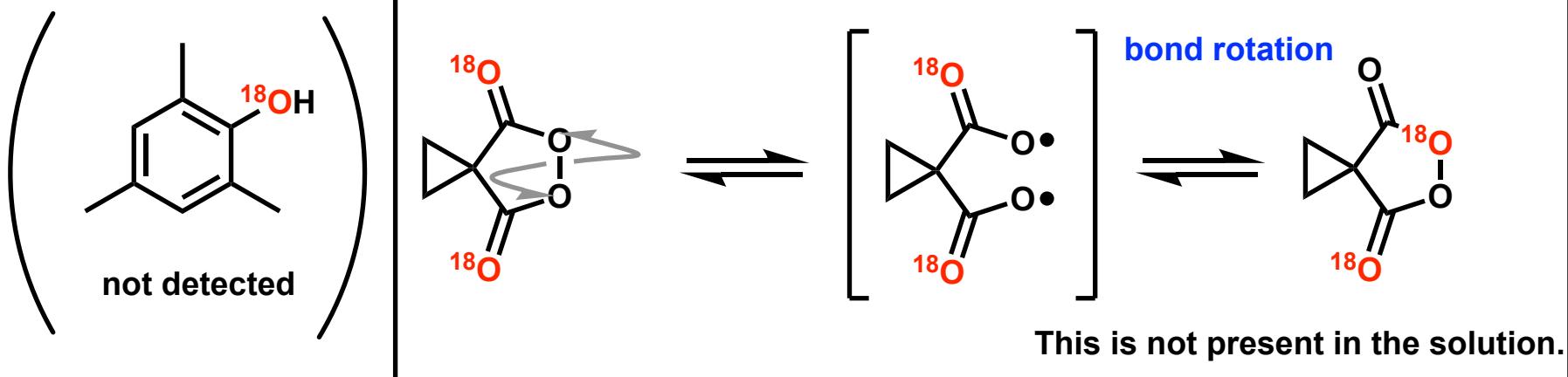
This conformation has good overlap between C-O σ^* orbital and π orbital of aromatic ring, so deoxydation of Sn1-like is occurred.



Appx. Radical pathway

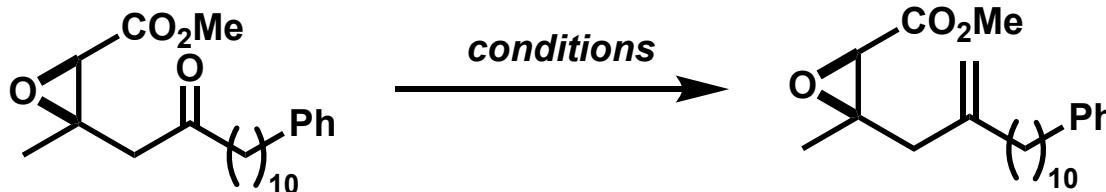


potential of bond rotation of diradical



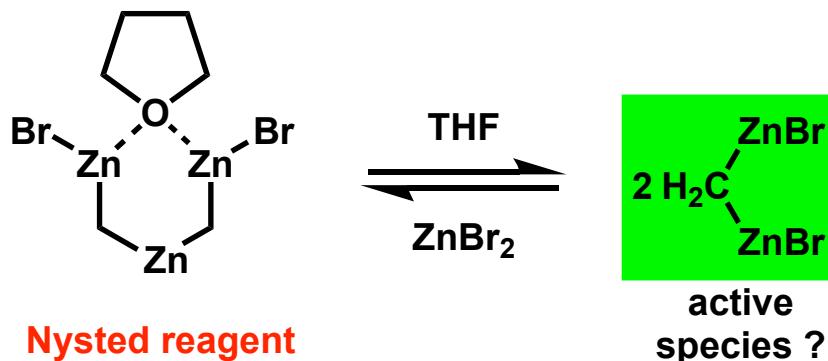
Appx. Zn + Ti (1)

1. Lewis Acid-type (no Zn(0))



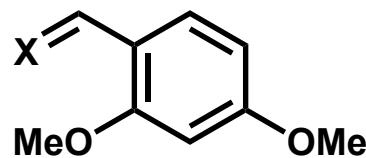
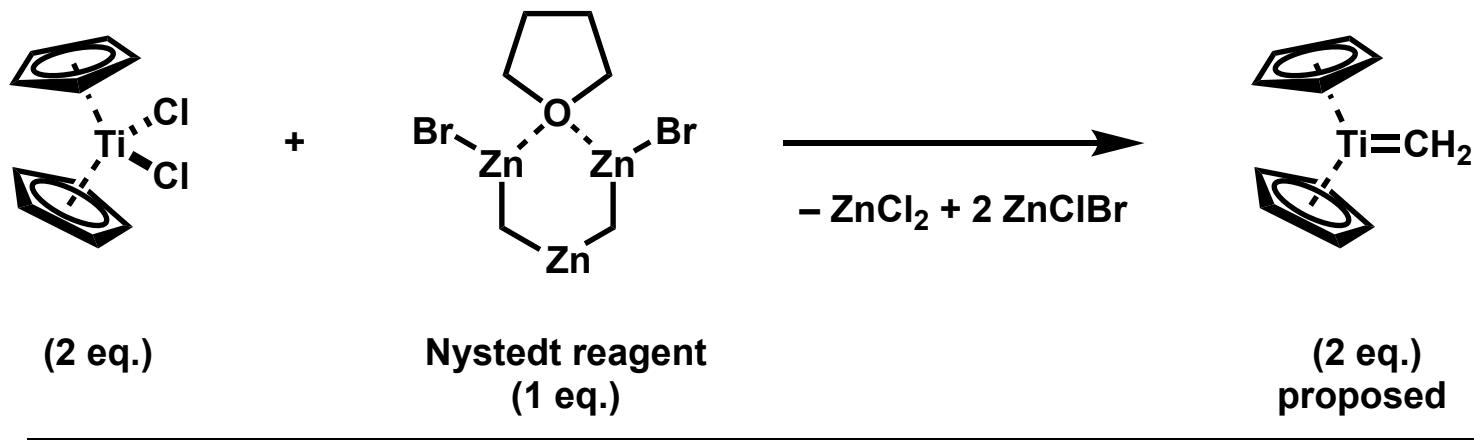
entry	reagents	temp. (°C)	yield
1	Zn, CH ₂ Br ₂ , TiCl ₄	0 to rt	decomp.
2	TiCl ₄ , Nysted reagent	0 to 15	35
3	Ti(O <i>i</i> -Pr) ₂ Cl ₂ , Nysted reagent	0 to 15	70

Lewis acidity: TiCl₄ > Ti(O*i*-Pr)₂Cl₂ > Ti(O*i*-Pr)₄

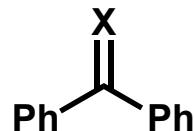


Appx. Zn + Ti (2)

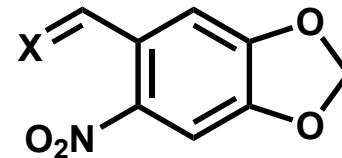
2. Tebbe-type



63%
rt, 2h



59%
-78 °C, 20min

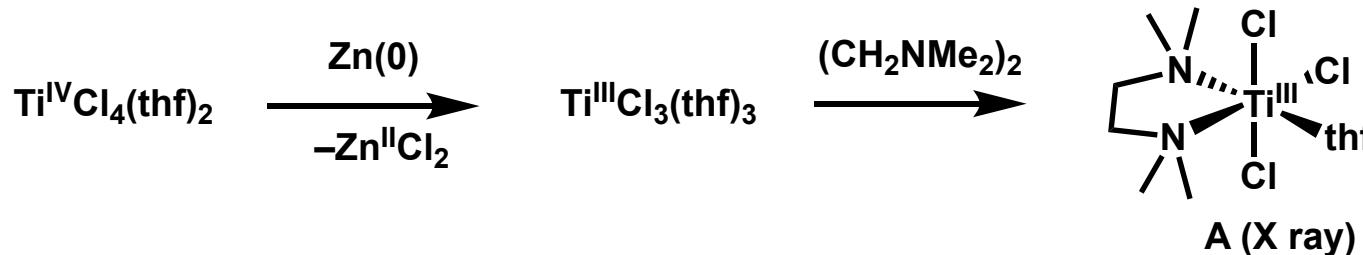


76%
-78 °C, 20min

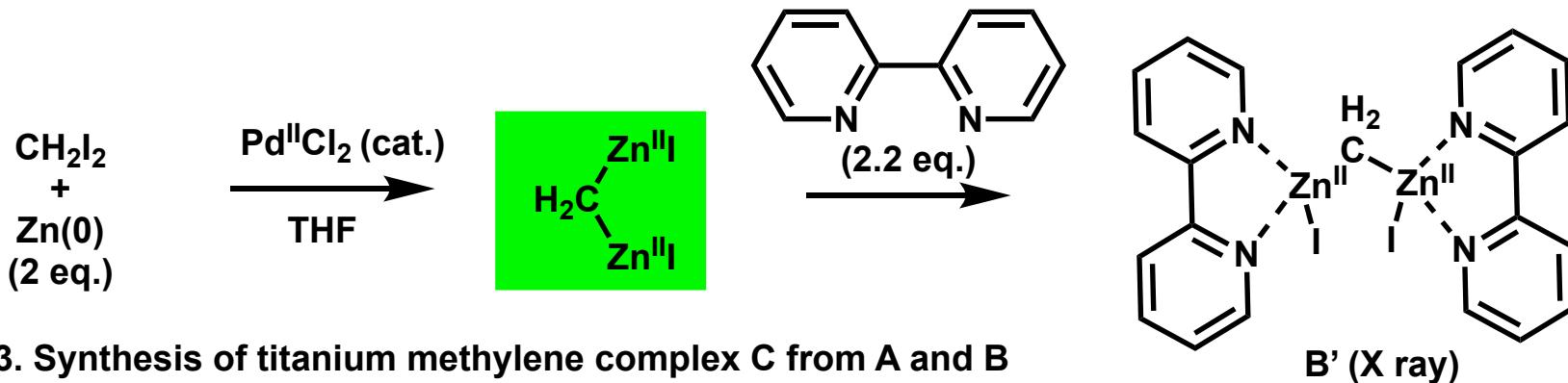
Appx. Zn + Ti (3)

3. Lombardo-type

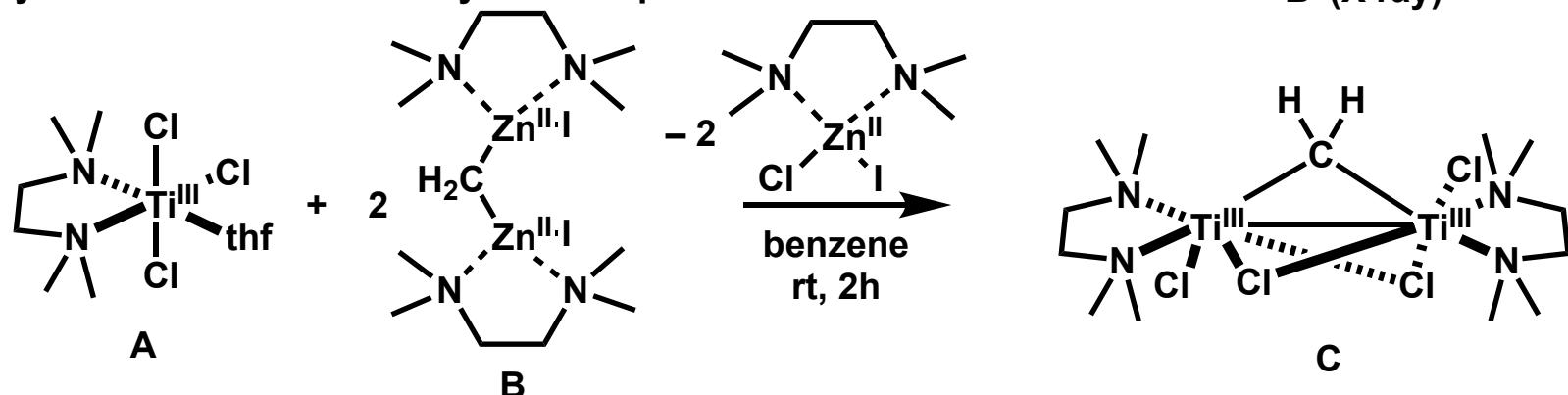
3-1. Reduction of Ti(IV) to Ti(III) occurred by Zn(0)



3-2. Di(idozinic)methane complexes



3-3. Synthesis of titanium methylene complex C from A and B



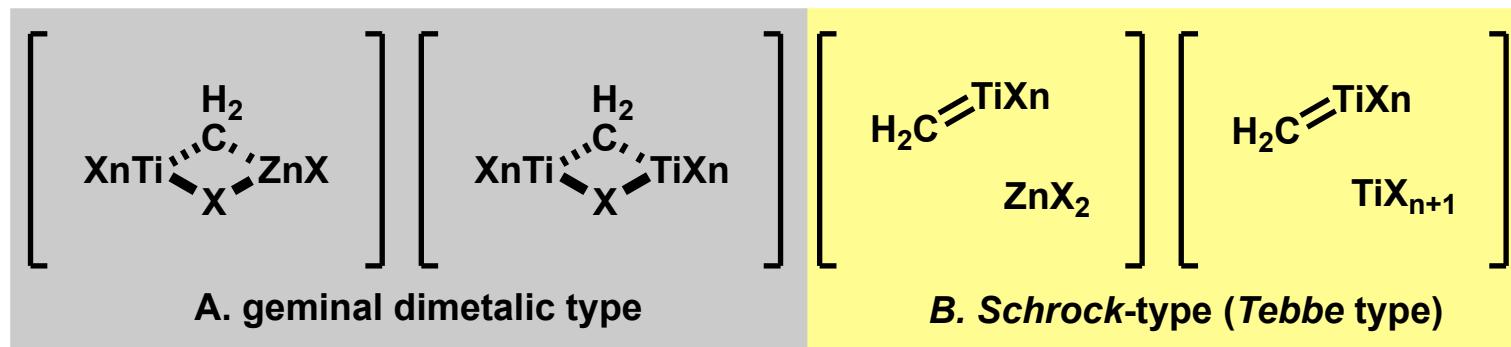
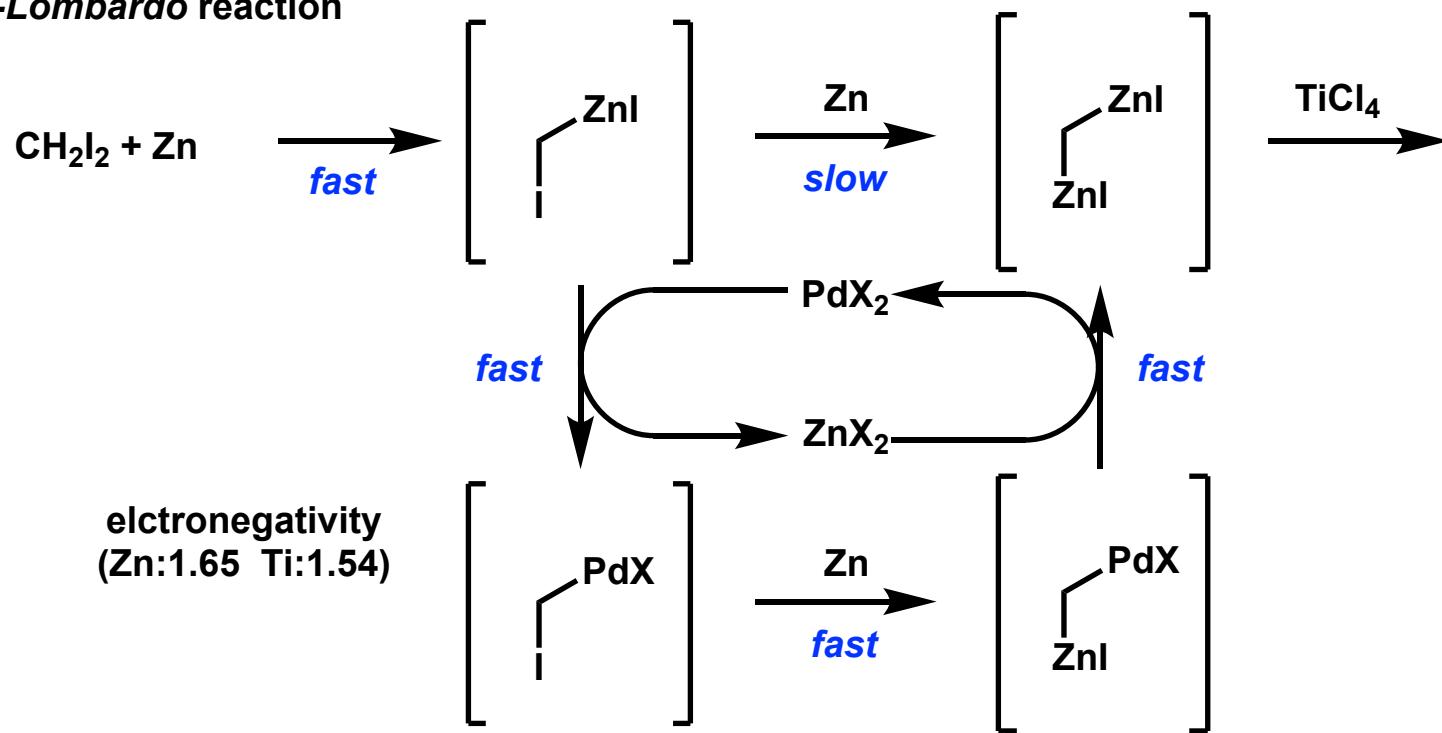
1) Kurogi, T.; Kuroki, K.; Moritani, S.; Takai, K. *Chem. Sci.* **2021**, *12*, 3509

2) Oshiki, T.; Kiriyama, T.; Tsuchida, K.; Takai, K. *Chem. Lett.* **2000**, *29*, 334

3) Nichida, Y.; Hosokawa, N.; Murai, M.; Takai, K. *J. Am. Chem. Soc.* **2015**, *137*, 114

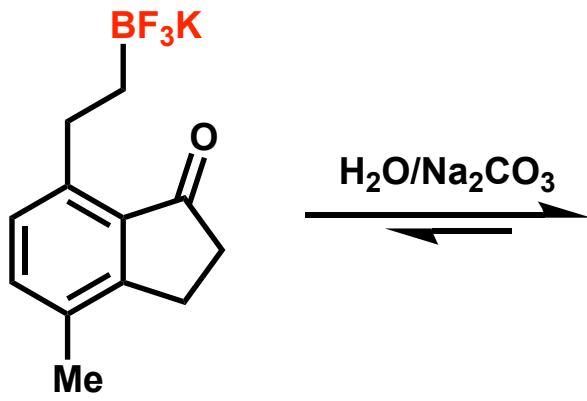
Appx. Zn + Ti (4)

Takai-Lombardo reaction

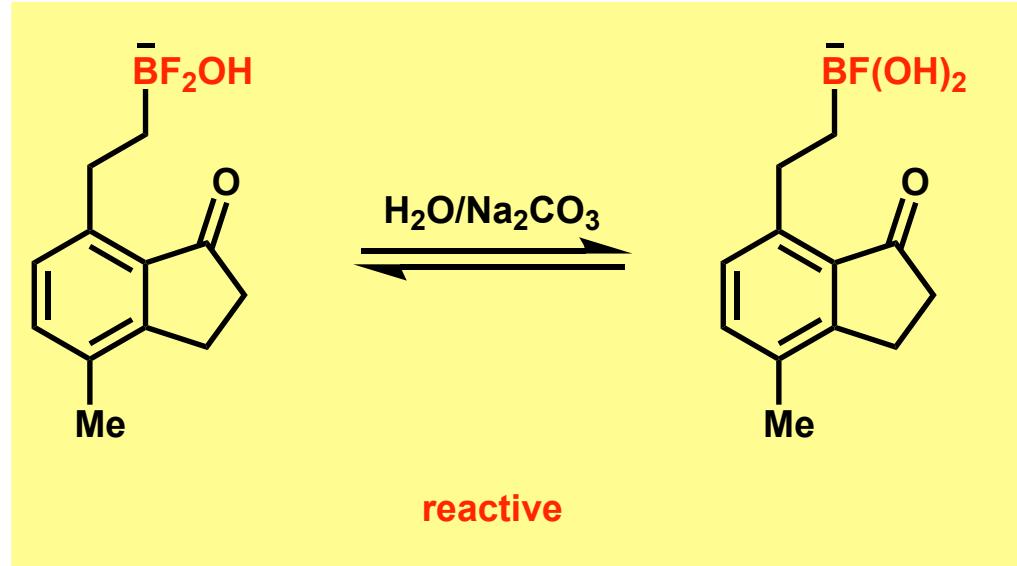


Appx. Hydrous condition

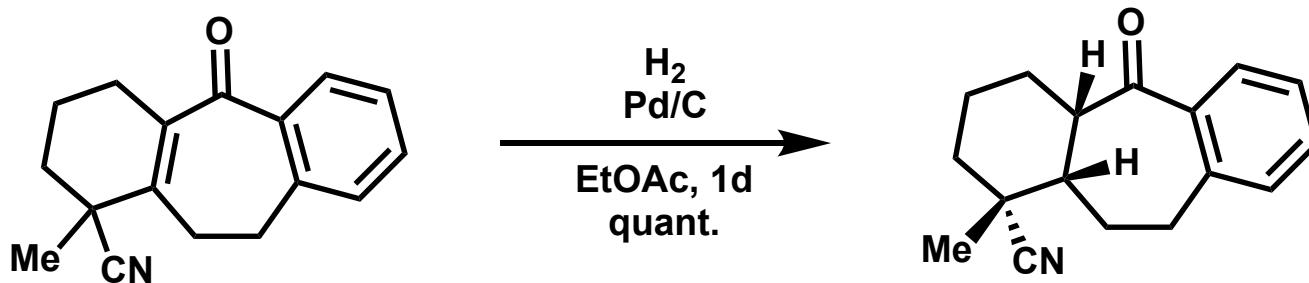
Unhydrous conditions



Hydrous conditions

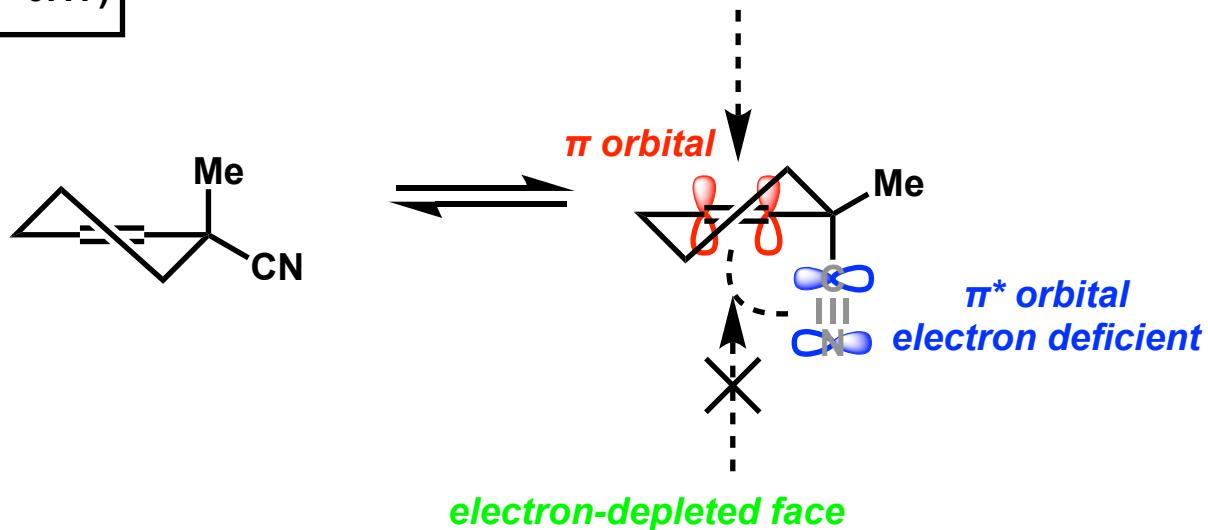


Appx. Stereoelectronic orbital interaction



Me group ($A = 1.7$)
Cyano group ($A = 0.17$)

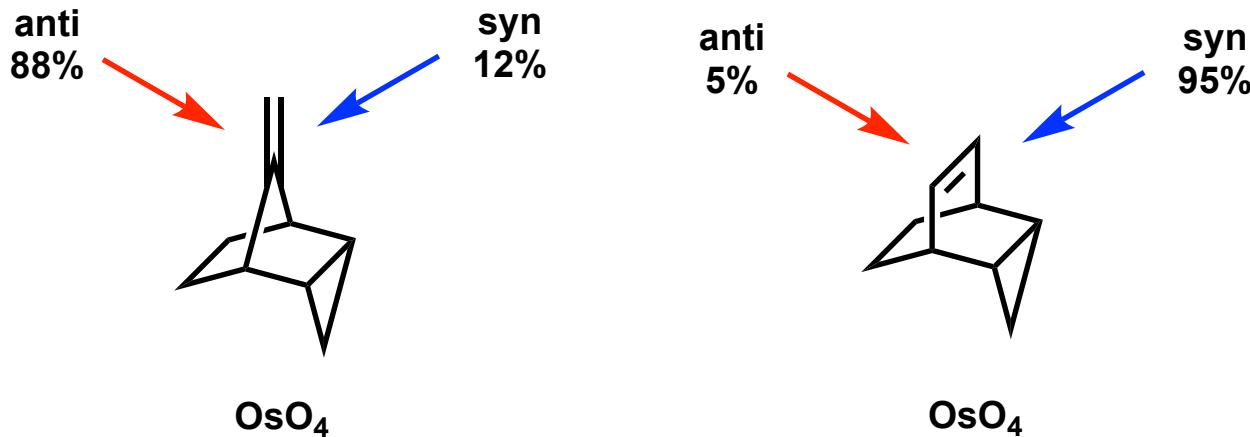
The reaction was occurred in electron-rich face.



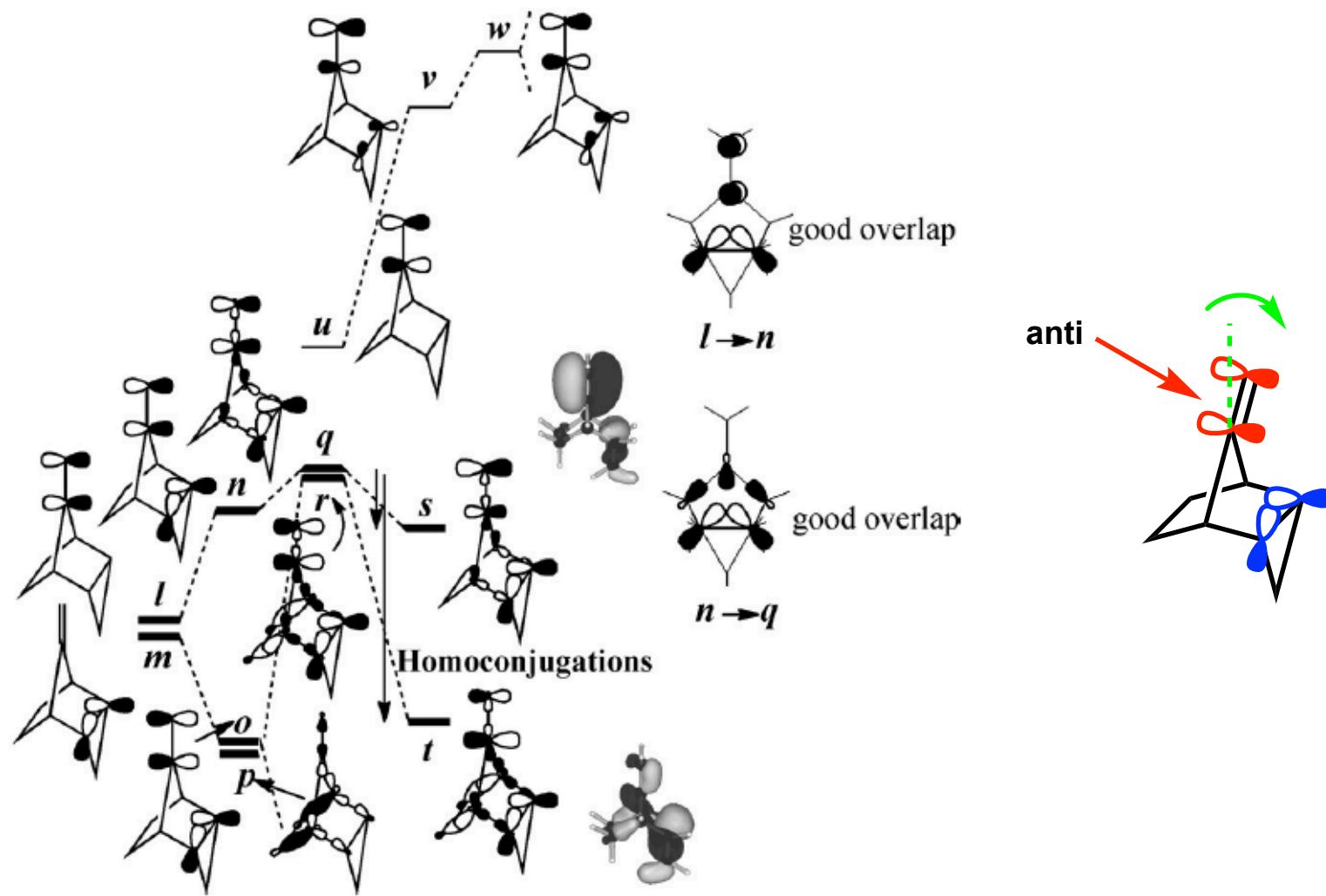
Appx. π -facial selectivity (1)

π -facial selectivity {
1. Steric factor
2. Orbital phase factor}

ex) dihydroxylation of tricyclic olefins with OsO_4



Appx. π -facial selectivity (2)



Appx. π -facial selectivity (3)

